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One-step Synthesis of Tricyclo[5.3.1.0^{3,8}]undecan-4,11-diones by Three Consecutive Michael Reactions. A Formal Synthesis of (\pm) -Seychellene

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Divinylketone and the trimethylsilyl enol ethers of cyclohex-2-en-1-ones undergo Lewis acid assisted triple Michael reactions yielding tricyclo[$5.3.1.0^{3,8}$]undecan-4,11-diones; the application of this reaction enabled a formal synthesis of (\pm)-seychellene (5).

The efficient construction of polycyclic systems is of great importance in synthetic organic chemistry.¹ Previously we reported a one-step synthesis of tricyclo[$3.2.1.0^{2,7}$]octan-6ones, and its application to the total synthesis of (\pm)ishwarane, employing successive Michael–Michael–alkylation reactions.² Our continuing interest in the development of a new methodology of forming multiple C–C bonds leads us to our present report: Lewis acid assisted three-fold Michael reactions of divinylketone with trimethylsilyl enol ethers of cyclohex-2-en-1-ones give tricyclo[$5.3.1.0^{3,8}$]undecan-4,11diones (equation 1), which form the key framework of several polycyclic natural products such as seychellene (**5**) or patchouli alcohol.

In a typical procedure, to a stirred solution of Lewis acid in CH_2Cl_2 was added at -80 °C a solution of divinylketone and trimethylsilyl enol ether in CH_2Cl_2 successively. The reaction temperature was gradually raised and stirring was continued at

 Table 1. Lewis acid assisted three-fold Michael reaction of the divinylketone with the trimethylsilyl enol ethers of various cyclohex-2-en-1-ones.

		Yield (%)	
Entry	Product	TiCl ₄ /Ti(OPr ⁱ) ₄ ^a	Et ₂ AlCl ^a
1	$(2a) (R^1 = H, R^2 = R^3 = Me)$	30	34
2	(2b) $(R^1 = R^2 = Me, R^3 = H)$	20	43ь
3	$(2c) (R^1 = Me, R^2 = R^3 = H)$	5	13
4	(2d) (R ¹ = R ² = H, R ³ = Me)	3	13
5	$(2e) (R^1 = Me, R^2 = H, R^3 =$		
	$-C(Me)=CH_2$ and H)	4	8
6	$(2f)(R^1 = R^2 = R^3 = H)$	trace	trace

 a Two equivalents of Lewis acid were used. b Three equivalents of $Et_{2}AlCl$ were used.



room temperature overnight. After extractive work-up, the product was isolated by preparative h.p.l.c.[†]

Among the Lewis acids investigated, satisfactory results were obtained using a mixture of $TiCl_4$ and $Ti(OPr)_4$ (ratio 0.8:1), or Et_2AlCl which gave somewhat higher yields.[‡] The results are listed in Table 1.

When the reaction was quenched at -80 °C after 1 h in entry 1 using titanium, the single Michael product (3) and the two-fold Michael product (4)§ were isolated in 9 and 39% yields, respectively. The progress of the reaction could be followed (t.l.c.) by the disappearance of the spots due to these compounds and appearance of the spot due to the three-fold Michael product (2a). From these results, the reaction proceeded via the sequential three-fold Michael pathway, not a successive Diels-Alder then Michael reaction pathway.

Since the tricyclic diketone (2b) has previously been transformed into (\pm) -seychellene (5) by Jung and coworkers,³ the present work constitutes a formal synthesis of this naturally occurring tricyclic sesquiterpene (5).⁴ The spectral data (n.m.r., i.r., and mass) of (2b) matched completely those of an authentic specimen. Since the average yield for each of the three C-C bond forming steps is 75%, the yield is quite satisfactory. Bicycloannulation by the three-fold Michael reaction described above is synthetically useful, since construction of a tricylic ring system can be achieved in a single step from readily available precursors, although yields are modest.

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[†] Satisfactory ¹H n.m.r., i.r., high resolution mass spectra and/or elemental analyses were obtained for all new compounds.

[‡] The reaction of divinylketone with the kinetic enolates generated by cleavage of the trimethylsilyl enol ether by MeLi gave only trace amounts of the desired products.

§ The stereochemistry was tentatively assigned as indicated.