

Total Synthesis of Optically Active 17 β -t-Butoxy-3-methoxy-7 α - or -7 β -18-dimethyl-1,3,5(10)-estratriene

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In the preparation of 7-methyl-19-nor-steroids, a new method for the synthesis of key intermediates (10), which involves alkylation of the C α ring fragment (4) and subsequent hydrogenation, cyclization, and hydrogenation is described.

The introduction of a 7-methyl group into the steroid nucleus usually results in an enhancement of the basic hormonal activities of the molecules.^{1,2} 7-Methyl steroids have been obtained by partial³ and total synthesis.⁴ In this paper, we report an enantioselective synthesis of the key intermediate (10a) for 7 α -methyl-19-nor-steroids, which can be secured by Birch reduction from (10a) or (9a).

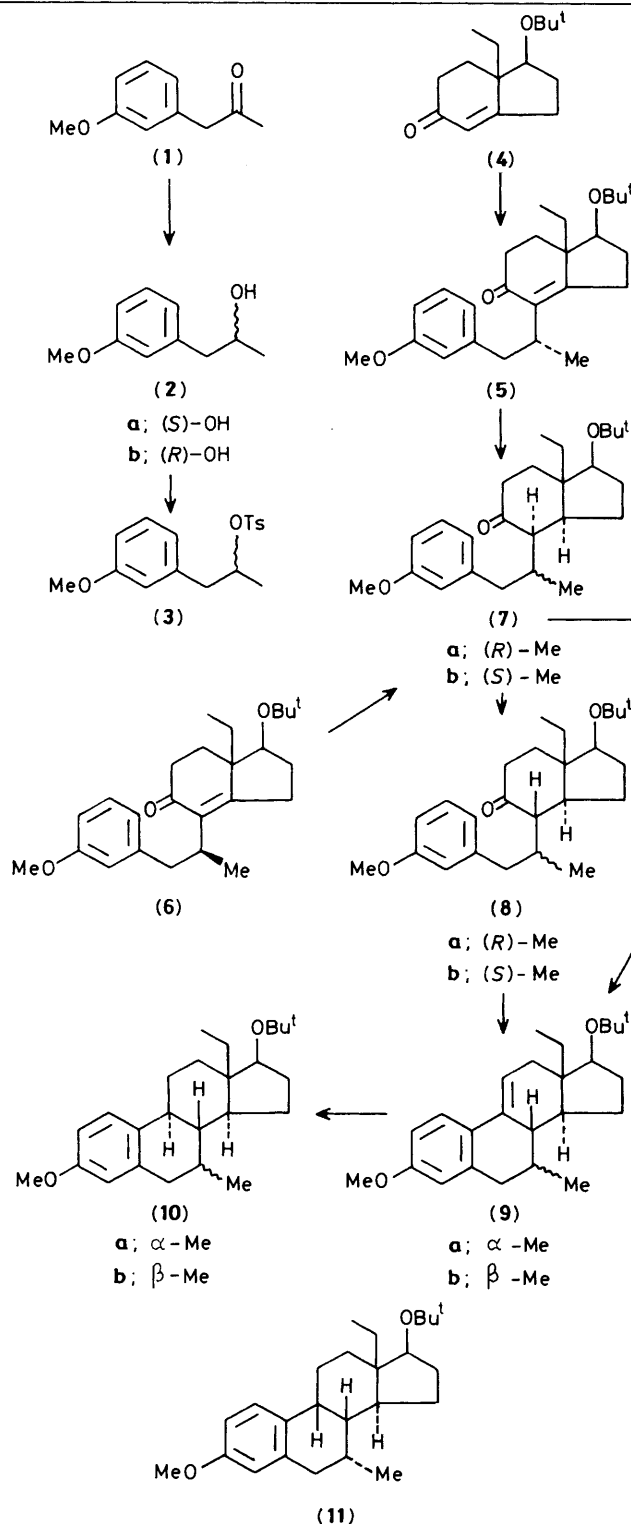
By *Saccharomyces cerevisiae* asymmetric reduction, the compound (1) obtained from eugenol⁵ was transformed to (2a) {yield 96%; [α]_D¹² +31.8° (c 1.53, CHCl₃); i.r.: 3350 cm⁻¹; ¹H n.m.r. (CCl₄): δ 1.1 [d, 3H, *J* 6 Hz, -CH(OH)Me]; *m/z*: 166 (*M*⁺)}. The hydroxy group of (2a) was determined to be in the *S*-configuration by the Horeau method.⁶ The optical purity[†] of (2a) was shown to be 98–99% enantiomeric excess (e.e.) by ¹⁹F n.m.r. analysis of the (*R*)-(+)-methoxy(trifluoromethyl)phenylacetyl (MTPA) ester⁷ of (2a).

The compound (4)⁸ was alkylated with the toluene-*p*-sulphonate of (2a) in the presence of NaH in dimethoxyethane (DME) to afford compound (5) {yield 72%; [α]_D²⁷ -25° (c 1.06, CHCl₃); λ_{\max} 255 nm (log ϵ 4.39); ¹H n.m.r. (CCl₄, 200 MHz): δ 1.16 (d, 3H, *J* 8 Hz, 7 α -Me), 3.44 (t, 1H, *J* 8 Hz, 17- α H); *m/z*: 385 (*M*⁺ + 1)}. Hydrogenation of compound (5) (5% Pd/C, MeOH, 2 atm, room temp., 24 h), gave *trans*-perhydroindane (7a) with 8 α -H {yield 47.4%; [α]_D¹⁰ +8.9° (c 0.99, CHCl₃); λ_{\max} 280 nm (log ϵ 3.36), 273 (3.39)}. Compound (7a) was cyclized to tetracyclic major product (9a) {yield 71.6%; [α]_D⁹ +29.56° (c 0.56, EtOH); λ_{\max} 266 nm; ¹H n.m.r. (CCl₄, 200 MHz): δ 6.15 (m, 1H, *w*_{1/2} 12 Hz, 11-H)} and the minor product (8a) with 8 β -H, [α]_D¹⁸ +20.2° (c 1.54, CHCl₃).

The *trans* junction of ring c and d in compound (7a) was established by c.d. spectroscopy, λ_{\max} 293 nm ($\Delta\epsilon$ +0.78).^{4b} Treatment of (8a) with acid also gave (9a), confirming a β -H on C(8).

Hydrogenation of compound (9a) (10% Pd/C, MeCO₂Et, room temp., 38 h), afforded the 7 α ,18-dimethyl estradiol derivative (10a) {[α]_D²¹ +23.1° (c 0.53, CHCl₃); m.p. 151–154 °C; λ_{\max} 279, 287 nm; c.d. (MeOH): λ_{\max} 279 nm ($\Delta\epsilon$ +0.23); ¹H n.m.r. (CDCl₃, 200 MHz): 1.00 (d, 3H, *J* 8 Hz, 7- α Me); *m/z*: 370 (*M*⁺)}, and the compound (11) with 9- β H, c.d. (MeOH): λ_{\max} 276 nm ($\Delta\epsilon$ -0.22). It is easy to envisage that in the presence of 7- α Me, hydrogen added from the α side would be hindered and eventually result in a mixture of compound (10a) and (11), although the compound (10a) is formed preferentially.

The 7 β -methyl steroid was prepared by a similar reaction sequence as described above. Alkylation of compound (4) with racemic (3), obtained by reduction of compound (1) (NaBH₄) and subsequent toluene-*p*-sulphonation of (2), led to



[†] ¹⁹F N.m.r. studies were performed on a Varian XL-200 MHz instrument, using CF₃CO₂H as internal standard. Chemical shifts of the (*R*)-(+)-MTPA esters of (2a,b) were 6.747, 6.776 p.p.m. respectively.

a mixture of (5) and (6) (g.c. 44:33).[‡] The isolated pure compound (6) was converted into (7b), then (9b), and finally compound (10b) {[α]_D²⁷ -65.5° (c 0.8, CHCl₃); m.p. 53—54 °C; λ_{max} 273, 284 nm; c.d. (MeOH): λ_{max} 274 nm ($\Delta\epsilon$ -0.71), 225 (+0.86); ¹H n.m.r. (CDCl₃, 200 MHz): δ 0.81 (d, 3H, *J* 6 Hz, 7- β Me)}.

The preparation of (10a) and (10b) also provides a case for the comparison of coupling constants between axial and equatorial methyl groups. Our observations were consistent with those made by Danneels and Anteunis.⁹

Received, 3rd April 1985; Com. 457

[‡] G.c. was carried out on a Varian 3700 instrument, H₂ flame detector. Column: FFAP, 25 m \times 0.2 mm programmed from 160 to 240 °C, 15 °C/min, retention time 12.29 min for (2a) and 10.25 min for (2b).

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