Stereo- and Enantio-controlled Synthesis of Chiral Intermediates for the Total Synthesis of Thienamycin and Related β -Lactam Antibiotics from 3-Hydroxybutyrates

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The synthesis of (1'R,3S,4S)- and (1'R,3S,4R)-3-(1'-t-butyldimethylsilyloxyethyl)-4-hydroxymethyl-2-azetidinone from ethyl (S)-3-hydroxybutyrate is reported.

One major difficulty in the synthesis of thienamycin (1) and related β -lactam antibiotics (2) and (3) is the control of the relative and absolute stereochemistry of the three contiguous chiral centres.¹

Since we and Hart first demonstrated² that 3-hydroxybuty-rates³ can be used as chiral building blocks for the synthesis of 3-(1'-hydroxyethyl)-2-azetidinones in a one-flask procedure a

number of other groups have reported similar findings. In this paper we now convey the synthesis, from ethyl (S)-3-hydroxybutyrate (Scheme 1), of optically active *trans*- and *cis*-3-(1'-t-butyldimethylsilyloxymethyl)-4-hydroxymethyl-2-azetidinones, intermediates which can be converted into thienamycin and olivanic acids by a known procedure. We used ethyl (S)-3-hydroxybutyrate with 86% optical purity,

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(3)

Scheme 1. Reagents: i, tetrahydrofuran (THF), 2 LICA (lithium isopropylcyclohexylamine), $-20-10\,^{\circ}\text{C}$; ii, THF, PPh₃, diethyl azodicarboxylate (DEAD), HCO₂H, 0 °C—room temp., 1.5 h; MeOH, HCl, room temp., 15 h; iii, dimethylformamide (DMF), t-butyldimethylsilyl chloride, imidazole, room temp., 15 h; iv, MeCN, ammonium cerium(IV) nitrate, $-20\,^{\circ}\text{C}$, 20 min; v, CH₂Cl₂, O₃, NaBH₄, $-78-0\,^{\circ}\text{C}$.

Scheme 2. Reagents: i, DMF, t-butyldimethylsilyl chloride, imidazole, room temp., 15 h; ii, CH₂Cl₂, O₃, NaBH₄, -78 to -20 °C; iii, acetone, Jones reagent, -5 °C, 2 h; iv, MeCN, Pb(OAc)₄, Cu(OAc)₂, 70 °C, 0.5 h.

obtained from reduction of ethyl acetoacetate using baker's yeast.3 The dianion imine condensation-cyclisation reaction of ethyl (S)-3-hydroxybutyrate (4) and N-anisylcinnamylidenimine (5) gave a 1:1 mixture of trans- and $cis-\beta$ -lactam (6) in 77—100% yield.6 Mitsunobu inversion^{2a,7} of (6) followed by acid hydrolysis of the intermediate formyl esters cleanly produced the inverted (R)-hydroxyethyl-2-azetidinones (7) in 84% yield. We also isolated a small amount (3%) of the elimination product (8). After protection of (7) as its t-butyldimethylsilyl ether we were able to separate the trans and cis isomers (9) by flash chromatography. We then subjected them separately to oxidative dearylation⁸ with ammonium cerium(iv) nitrate, to obtain the N-unprotected β-lactam trans (10a) {97% yield, m.p. 85—86 °C (light petroleum– CH_2Cl_2), [α]_D²⁵ +40.7° (c 0.8555 CHCl₃)} and cis (10b) {86% yield, m.p. 80 °C (light petroleum– CH_2Cl_2), $[\alpha]_{D}^{25}$ -45.9° (c 1.06 CHCl₃)). Ozonolysis of (10a) and (10b), followed by reductive work-up⁹ produced the 4-hydroxymethyl-2-azetidinones *trans* (11a){(yield 97%, m.p. 89—90 °C (light petroleum), $[\alpha]_D^{25}$ –14.1° (c 0.595 CHCl₃)} and cis (11b)10 (yield 81%, m.p. 94 °C (light petroleum), $[\alpha]_{D^{25}}$ -45.6° (c 0.8 CHCl₃)}.†

The overall yield from ethyl (S)-3-hydroxybutyrate for *trans* (11a) was 30% and 23% for *cis* (11b). ¹¹ These yields are comparable to other reported methods for the synthesis of chiral thienamycin precursors. ^{8b,11} The optical rotation of (11a) was found to be identical to the literature value for (1'R,3S,4S)-3-(1'-t-butyldimethylsilyloxyethyl)-4-hydroxymethyl-2-azetidinone. ^{11b,c} After we had unequivocally established the absolute stereochemistry of the *trans* products, we decided to determine the configuration of the *cis*-azetidinones through *cis*-*trans* isomerization at position 4 (Scheme 2). Reider and Grabowski had previously shown, ^{11a} that oxidative decarboxylation of 3-(1'-hydroxyethyl)-4-(carboxylic acid)-2-azetidinone results in exclusive introduction of the acetoxy group at C-4 opposite the hydroxyethyl group at C-3.

The cis-trans mixture (6) (racemic) was quantitatively converted into the t-butyldimethylsilyl ether (12). Ozonolysis of (12) followed by reductive work-up with NaBH₄ and then oxidation with Jones reagent gave the carboxylic acid (13).

[†] All intermediates showed spectroscopic data in agreement with their structures.

After oxidative decarboxylation with lead tetra-acetate¹² we obtained *trans*- β -lactam (14) as single isomer in an overall yield of 40% (Scheme 2).

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