Stereochemistry of Pyrrolizidine Alkaloid Biosynthesis: Incorporation of Chiral [2-²H]Putrescines into Retrorsine

Ellen K. Kunec and David J. Robins*

Department of Chemistry, University of Glasgow, Glasgow G12 800, U.K.

Feeding experiments with (2R)- and (2S)- $[2-^2H]$ putrescine on *Senecio isatideus* plants have shown that hydroxylation at C-7 of retronecine proceeds with retention of configuration, and formation of the 1,2-double bond involves removal of the *pro-S* hydrogen and retention of the *pro-R* hydrogen at C-2 of retronecine.

Retrorsine (4) is one of many pyrrolizidine alkaloids that contain retronecine (3) as the base portion.^{1,2} The C₈ skeleton of retronecine is derived biosynthetically from L-ornithine or L-arginine³ via putrescine (1).^{4–7} A later intermediate in the biosynthetic pathway is homospermidine (2), formed from two molecules of putrescine (Scheme 1).^{6,8} We have shown

that ²H n.m.r. spectroscopy can be used to establish complete labelling patterns in retrorsine derived biosynthetically from $[1,4-^2H_4]$ - and $[2,3-^2H_4]$ -putrescine in *S. isatideus* plants.⁹ With the latter precursor, ²H is retained at C-2, C-6 α , C-6 β . and C-7 α of retronecine (**3**). The stereochemical course of a number of enzymic processes involving removal of hydrogen



Scheme 2. *Reagents:* i, HCl, HNO₃, urea; ii, SOCl₂, MeOH; iii, di-isobutylaluminium hydride; iv, LiAlD₄; v, chromic acid; vi, HBr; vii, NaN₃; viii, LiAlH₄.

from the 1- and 4-positions of putrescine has been established using chiral [1-2H]putrescine.^{10,11} We have now prepared samples of chiral [2-2H]putrescine and used them to obtain ²H-labelled retrorsine. The stereochemical course of further enzymic processes in retronecine biosynthesis has been established.

(2R)-[2-2H]Succinic acid (7) was prepared from (2S)aspartic acid (5) as outlined by Arigoni and Eliel (Scheme 2).¹² An o.r.d. curve of this material (7) was in accord with literature values,12,13 and mass spectral data on the deuteriated succinic anhydride indicated ca. 98% ²H₁ species. The immediate precursor of the deuteriated succinic acid (7), (2R)-[2-2H]butane-1,4-diol (6) was converted into (2R)-[2-²H]putrescine (8) dihydrochloride (Scheme 2) $[^{2}H{^{1}H}$ n.m.r. (H₂O) δ 1.76 p.p.m. (s)]. Analogous treatment of (2R)aspartic acid yielded samples of (2S)-[2-2H]succinic acid and (2S)-[2-²H]putrescine dihydrochloride. The latter sample contained an impurity of ca. 15% [1-2H]putrescine dihydrochloride [from ${}^{2}H{}^{1}H{}^{1}$ n.m.r. (H₂O) δ 3.08 p.p.m. (s)]. This impurity may have arisen by lithium aluminium deuteride reduction of a small amount of incompletely reduced diester, left after the reduction with di-isobutylaluminium hydride (Scheme 2). Each precursor (200 mg) together with [1,4-¹⁴C]putrescine dihydrochloride (5 μ Ci) was pulse fed into the xylems of four S. isatideus plants over a period of two weeks. After a further two weeks, retrorsine (4) was isolated and recrystallised to constant specific radioactivity.

The ${}^{2}H{}^{1}H{}$ n.m.r. spectrum⁹ of retrorsine derived from (2*R*)-[2-²H]putrescine (8) was obtained in chloroform at 60 °C



Figure 1. 30.72 MHz ²H{¹H} N.m.r. spectra of retrorsine in chloroform at 60 °C: (a) sample of (9) derived from (2R)-[2-²H]-putrescine (8); (b) sample of (11) derived from (2S)-[2-²H]putrescine (10). Natural abundance ²H in CHCl₃ was the internal reference at δ 7.15.

[Figure 1(a)]. This showed two major signals at δ 2.15 and 6.20 p.p.m. corresponding to retrorsine (9) labelled with ²H at C-6 α and C-2, respectively (Scheme 3). The enrichment factors[†] for the two labelled sites are nearly equal at *ca*. 0.18% ²H. The specific incorporation of ²H per C₄ unit is therefore 0.18% [0.18/2 × 2/98 × 100%, where 98/2 atom % ²H is the average enrichment of ²H at each labelled site of putrescine (8)]. This value is about half the ¹⁴C specific incorporation of 0.30% per C₄ unit. Removal of ²H from the carbon atoms destined to become C-1 and C-7 in retronecine accounts for this loss of about half of the ²H from (2*R*)-[2-²H]putrescine (8) on its conversion into retrorsine (9).‡

Using (2S)-[2-²H]putrescine (10) as a precursor, a sample of retrorsine was obtained which showed two major signals in the ²H{¹H} n.m.r. spectrum, obtained in chloroform at 60 °C, at δ 2.38 and 5.00 p.p.m. [Figure 1(b)]. This corresponds to retrorsine (11) labelled with ²H at C-6 β and C-7 α , respectively (Scheme 4). Enrichment factors† of 0.34% ²H for C-6 β and 0.20% for C-7 α were estimated, leading to a specific incorporation of ²H of 0.65% (0.54/2 × 2/83 × 100%) for the C₄ unit in which most of the ²H appears. This value is similar to the ¹⁴C specific incorporation of 0.73% per C₄ unit.‡ About half of the

[†] The enrichment factor for a labelled site in retrorsine is calculated from [(integral of labelled site in retrorsine) –(natural abundance integral/concentration of retrorsine)]/(natural abundance integral of ²H in chloroform at δ 7.25 p.p.m./concentration of chloroform) × 0.0156%.

 $[\]ddagger$ Differences between the ²H and ¹⁴C specific incorporations, after taking into account the loss of ²H from some carbon atoms in retronecine, may be due to the operation of ²H isotope effects in the biosynthetic pathway.



Scheme 3§



Scheme 4§

²H is lost from (2*S*)-[2-²H]putrescine (10) as it is transformed into retrorsine (11). Removal of ²H from the carbon atoms which become C-1 and C-2 of retronecine leads to the appearance of most of the ²H in one C₄ unit derived from putrescine.

The hydroxylation at C-7 of retronecine thus occurs with retention of configuration, in agreement with the stereospecificity generally observed for direct hydroxylation at sp³ carbon atoms.¹⁴ Formation of the 1,2-double bond in retronecine

§ No molecules of putrescine can contain more than one ²H atom. The structures in Schemes 3 and 4 are therefore representations of all the ²H-labelled species present. takes place with retention of the *pro-R* hydrogen and loss of the *pro-S* hydrogen at the carbon atom which becomes C-2 of retronecine.

We are grateful to Dr. D. S. Rycroft (Glasgow) for obtaining the n.m.r. spectra and to Dr. A. F. Drake (King's College, London) for providing the o.r.d. data. We thank the S.E.R.C. for a Research Assistantship (to E. K. K.).

Received, 1st July 1985; Com. 925

References

- 1 D. J. Robins, Fortschr. Chem. Org. Naturst., 1982, 41, 115.
- 2 D. J. Robins in 'The Alkaloids,' ed. M. F. Grundon, Specialist Periodical Reports, The Royal Society of Chemistry, London, 1978—1983, vols. 8—13; D. J. Robins, *Nat. Prod. Rep.*, 1984, 1, 235; 1985, 2, in the press.
- 3 D. J. Robins and J. R. Sweeney, Phytochemistry, 1983, 22, 457.
- 4 D. J. Robins and J. R. Sweeney, J. Chem. Soc., Perkin Trans. 1, 1981, 3083 and references cited therein.
- 5 H. A. Khan and D. J. Robins, J. Chem. Soc., Chem. Commun., 1981, 146.
- 6 H. A. Khan and D. J. Robins, J. Chem. Soc., Chem. Commun., 1981, 554.
- 7 G. Grue-Sorensen and I. D. Spenser, J. Am. Chem. Soc., 1981, 103, 3208; Can. J. Chem., 1982, 60, 643.
- 8 J. Rana and D. J. Robins, J. Chem. Res. (S), 1983, 146.
- 9 J. Rana and D. J. Robins, J. Chem. Soc., Chem. Commun., 1983, 1222.
- 10 J. Rana and D. J. Robins, J. Chem. Soc., Chem. Commun., 1984, 517.
- 11 G. Grue-Sorensen and I. D. Spenser, J. Am. Chem. Soc., 1983, 105, 7401.
- 12 D. Arigoni and E. L. Eliel, Top. Stereochem., 1969, 4, 200.
- 13 J. S. Chickos, M. Bausch, and R. Alul, J. Org. Chem., 1981, 46, 3559; D. Portsmouth, A. C. Stoolmiller, and R. H. Abeles, J. Biol. Chem., 1967, 242, 275.
- 14 R. Bentley, 'Molecular Asymmetry in Biology,' Academic Press, New York, 1970.