A General Strategy for the Synthesis of Ipecac and Heteroyohimbine Alkaloids

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The synthesis of the ester (**6b**) and lactone (**7a**) *via* reductive photocyclisation, kinetically controlled alkylation and acylation, and reductive cleavage of γ -lactone illustrates a new general method for the preparation of the ipecac and heteroyohimbine alkaloids.

Whereas the structural and biosynthetic parallel between the ipecac and heteroyohimbine types of alkaloid has been mentioned,1,2 only few studies have been devoted to the synthesis of 2,3-disubstituted benzo- or indolo-quinolizine derivatives as common intermediates for the synthesis of both types of alkaloid. We now describe a general strategy applicable to the total synthesis of both types of alkaloid by the efficient synthesis of compounds (6a-d) and (7a and b) via the furanoquinolizine (3a) as a common precursor. The ester (6b) is a known key intermediate for the total synthesis of emetine^{3,4} and the lactone (7a) is an essential fragment of the C, D, and E rings of the well known key intermediate for the total synthesis of aimalicine and other heteroyohimbine alkaloids. 5—8 The overall strategy is based on three key steps: reductive photocyclisation of enamides, kinetically controlled alkylation or acylation α to a lactam carbonyl group, and reductive cleavage of a γ-lactone ring.

Acylation of the isoquinoline (1) with furan-2-carbonyl chloride in the presence of triethylamine gave the enamide (2) in 98% yield. Reductive photocyclisation⁹ of the enamide (2)

in the presence of sodium borohydride in MeCN-MeOH (9:1) at 5-10 °C proceeded smoothly to give the furanoquinolizine (3a) in 77% yield [δ_H (CDCl₃) 4.84 (d, J 10.5 Hz, 8a-H), 4.64 (br. dd, J 11.5 and 2.5 Hz, 12a-H), and 3.37 (m, 11a-H)]. The c/p-cis fusion of the lactam (3a) was deduced from comparison of the coupling constant (J 10.5 Hz) between 8a- and 11a-H with those (J 10-11 Hz) of analogous cis-compounds¹⁰ which were also prepared by reductive photocyclisation of enamides having a furan ring and firmly characterised by both spectral and chemical means. Lithiation of (3a) with lithium di-isopropylamide in tetrahydrofuran at -78 °C followed by quenching with either ethyl iodide or acetic anhydride led to the formation of the corresponding 8a-ethyl (3b) or 8a-acetyl derivatives (3c) in 86 or 76% yield respectively [(3b) $\delta_{\rm H}$ (CDCl₃) 3.13 (dddd, J 12, 5.5, 3, and 1.5 Hz, 11a-H), 1.94 (q, J 7.5 Hz, CH_2Me), and 0.96 (t, J 7.5 Hz, Me); (3c) $\delta_{\rm H}$ (CDCl₃) 3.64 (dddd, J 12.5, 5.5, 3, and 2 Hz, 11a-H) and 2.49(s, COMe)]. The stereochemistry of both products (3b and c) thus obtained was deduced from comparison of their n.m.r. spectra, particularly signals due to protons

in the c- and p-rings, with the spectrum of the starting lactam (3a). These lactams (3b and c) both contain a versatile benzoquinolizine nucleus bearing a C_2 -unit at both the 2- and 3-positions appropriate for a key precursor for conversion into ipecac alkaloids such as emetine.

The furanoquinolizine (3b) was then converted into the known intermediate,3,4 the keto-ester (6b), as follows. The dihydrofuran (3b) was hydrated with 15% sulphuric acid to give the hemiacetal (4a) as an epimeric mixture which was then oxidised with pyridinium chlorochromate (PCC) to furnish the γ-lactone (5a) in 60% overall yield. Reductive cleavage of the lactone (5a) using methods developed for the deoxygenation of alcohols11 was unsuccessful. However, reduction of (5a) with calcium in liquid ammonia^{12,13} gave the desired carboxylic acid (6a), m.p. 186.5—188 °C (lit., 3 187— 188 °C) in 62% yield which was converted into the corresponding methyl ester (6b), m.p. 54-55 °C (lit., 3 56-57 °C) [v(CHCl₃) 1735 and 1625 cm⁻¹; $\delta_{\rm H}$ (CDCl₃) 4.67 (br. dd, J 12 and 4 Hz, 11b-H), 2.56 (dt, J 12 and 4 Hz, 1-H_{eq}), 2.37 (m, 2-H), and 1.44 (q, J 12 Hz, 1-H_{ax})] with diazomethane. Both the acid (6a) and the ester (6b) were identical (i.r. spectra)

with the authentic samples which are known to be key intermediates^{3,4} for the synthesis of emetine.

Similarly, the 8a-acetyl furan (3c) was converted into the acetyl lactone (5b) via the route involving hydration with 15% sulphuric acid–tetrahydrofuran (THF) (quantitatively) followed by oxidation of the resulting hemiacetal (4b) with either Me₂SO–Ac₂O (76%) or PCC (51%). Cleavage of the lactone ring in the 8a-acetyl lactone (5b) with aluminium amalgam in aqueous ethanolic THF¹⁴ proceeded more smoothly than in the case of the ethyl lactone (5a) to afford the desired carboxylic acid (6c) in 95% yield. The stereochemistry of the acid (6c) was deduced from the spectral data of the corresponding methyl ester (6d) [v(CHCl₃) 1740—1720 and 1630 cm⁻¹; $\delta_{\rm H}$ (CDCl₃) 4.74 (dd, J 12 and 5 Hz, 11b-H) and 3.46 (d, J 11 Hz, 3-H)].

Among conditions investigated for chemoselective reduction of the acetyl and lactam groups, treatment of the acetyl ester (6d) with diborane in THF at -10 °C for 1 h afforded a mixture of two amino lactones (7a) (15%) and (7b) (40%) which were readily separated by p.l.c. and characterised by their n.m.r. spectra [(7a) $\delta_{\rm H}$ (CDCl₃) 4.76 (br. qd, J 7 and 4 Hz, 9-H) and 1.36 (d, J 7 Hz, CHMe); (7b) $\delta_{\rm H}$ (CDCl₃) 4.23 (dq, J 10 and 6 Hz, 9-H), 1.88 (br. qt, J 11 and 5 Hz, 12a-H), 1.68 (br. qd, J 11 and 4 Hz, 8a-H), and 1.42 (d, J 6 Hz, CHMe)]. These two lactones (7a and b) contain the essential tricyclic skeletal structure of the c, d, and e rings of heteroyohimbine alkaloids such as ajmalicine⁵ and corynantheine.⁶

Thus, we have established a new useful synthetic method for preparing a common potential intermediate for ipecac and heteroyohimbine alkaloid synthesis.

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