A Novel Quinquedentate N₅ Macrocycle containing a Pendant Amino Group

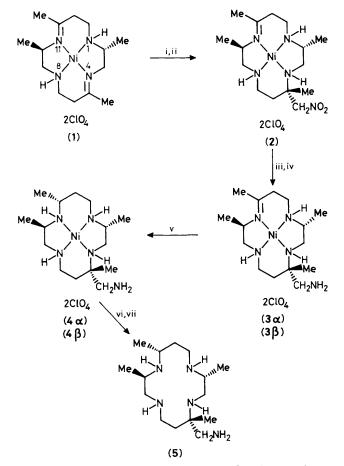
Bohdan Korybut-Daszkiewicz

Institute of Organic Chemistry, Polish Academy of Sciences, Warsaw 01-224, Poland

The addition of the monoanion of nitromethane to the C=N group of $[Ni(C-rac.-N-meso-2,5,10,12-tetra-methyl-1,4,8,11-tetra-azacyclotetradeca-4,11-diene)][CIO_4]_2 (1) followed by reduction gives the nickel(II) complex (4), containing the pendant –CH_2NH_2 group; the free ligand (5) may be obtained by treating the nickel complex with dry HCl gas in acetonitrile solution.$

The preparation of pendant arm macrocycles has attracted considerable attention in recent years.¹ House² has reported the preparation and crystal structure of a quinquedentate N₄O ligand prepared by dioxygen oxidation of *trans-(RSSR)*-[CoCl₂(tet a)]ClO₄.[†] In this reaction oxidation of one of the *gem-*dimethyl substituents to a co-ordinated –CH₂OH group occurs. The present communication describes the preparation of a *C*-functionalised cyclam derivative with a –CH₂NH₂ side chain.

The ligand synthesis is summarised in Scheme 1. Addition of Et_3N to a nitromethane solution of the nickel(II) complex $[Ni(2,5,10,12-Me_4[14]4,11-dieneN_4)][ClO_4]_2^{\dagger}$ (1)³ leads to



Scheme 1. Reagents: i, MeNO₂, Et₃N; ii, HClO₄; iii, Zn, HCl; iv, NaOH; v, NaBH₄; vi, HCl, MeCN; vii, NaOH.

addition of $-CH_2NO_2$ to the less hindered -C=N group of (1) to give the sparingly soluble paramagnetic monoperchlorate salt of (2). The paramagnetic product was then protonated with perchloric acid in methanol-nitromethane (9:1) solution giving the yellow diamagnetic diperchlorate (2).

Reduction of the nitro group with zinc powder and hydrochloric acid gave two paramagnetic N-isomeric 5-aminomethyl derivatives $(3\alpha) \cdot H_2O$ and $(3\beta) \cdot 2H_2O$. Further reduction with sodium borohydride in methanol solution reduced the remaining C=N double bond, yielding the paramagnetic, N-isomeric complexes $(4\alpha) \cdot H_2O$ and $(4\beta) \cdot 2H_2O$. Demetallation of (4α) and (4β) in acetonitrile saturated with dry hydrogen chloride gave the same free ligand (5) (m.p. 120 °C).

All 5-aminomethyl derivatives (3) and (4)[‡] were also isolated in the form of their protonated diamagnetic triperchlorates, which were precipitated with benzene from nitromethane solutions acidified with perchloric acid.

Compounds (2)—(5) as well as the triperchlorate salts gave satisfactory analytical data and were characterised by the i.r., u.v., and visible spectra. The ${}^{13}C$ and ${}^{1}Hn.m.r.$ spectra (Table 1) of the diamagnetic triperchlorates are also consistent with the structures (2)—(5).

We have found that the addition of nitromethane to the C=N bonds in other readily accessible nickel(II) complexes of

Table 1. ¹ H N.m.r. data of compounds (1)—(5). ^a				
		1		
Compound	Solvent	=CCMe	>CHMe	NCH ₂ CMe
$(1) [CIO_4]_2$	CD_3NO_2	2.36s (3),	1.38d (3),	ļ
		2.23s (3)	1.22d (3)	
$(2) [ClO_4]_2$	CD_3NO_2	2.28s (3)	1.41d (3),	1.45s (3)°
			1.21d (3)	
(3α) [ClO ₄] ₃		2.37s(3)	1.47d (3),	1.50s (3)
			1.20d (3)	
$(3\beta) [ClO_4]_3$		2.36s (3)	1.43d (3),	1.53s (3)
			1.27d (3)	• /
(4 α) [ClO ₄] ₃	CD_3NO_2		$2.00d^{b}(3),$	
	+CD ₃ OD		1.22d (3),	1.50s (3)
	+HClO₄		1.17d (3)	. ,
$(\mathbf{4\beta}) [ClO_4]_3$			1.62d (3),	
			$1.17d^{b}(3),$	1.40s (3)
			1.17d (3)	()
(5)	CDCl ₃		1.05d(3),	1.07s (3)
	5		1.04d (6)	

^a 60 MHz, δ relative to SiMe₄; relative intensities in parentheses. ^b Singlet when NaBD₄ was used for reduction of the double bond. ^c NCH₂CMe at 5.39d (1) and 6.11d (1).

 $[\]dagger$ (2,5,10,12-Me₄[14]4,11-dieneN₄) = C-rac-N-meso-2,5,10,12-tetramethyl-1,4,8,11-tetra-azacyclotetradeca-4,11-diene; tet a = C-meso-5,5,7,12,12,14-hexamethyl-1,4,8,11-tetra-azacyclotetradecane.

[‡] The relative C- and N-configurations in complexes (3) and (4) established on the basis of ¹H n.m.r. spectra^{3.4} is as follows: (3 α), (1RS,2RS,4RS,5SR,8RS,10RS); (3 β), (1SR,2RS,4RS,5SR, 8RS,10RS); (4 α), (1RS,2RS,4RS,5SR,8RS,10RS,11RS,12RS); (4 β), (1SR,2RS,4RS,5SR,8RS,10RS,11SR,12RS).

unsaturated cyclam derivatives3,4 proceeds with equal ease to give macrocycles carrying a pendant arm.

This work was supported by the Polish Academy of Sciences.

Received, 25th September 1984; Com. 1364

References

- 1 T. A. Kaden, *Top. Curr. Chem.*, 1984, **121**, 157. 2 D. A. House, M. Harnett, W. T. Robinson, and M. C. Couldwell, J. Chem. Soc., Chem. Commun., 1984, 979.
- R. A. Koliński, B. Korybut-Daszkiewicz, Z. Kubaj, and J. Mroziński, *Inorg. Chim. Acta*, 1982, 57, 269.
- 4 R. A. Koliński and B. Korybut-Daszkiewicz, Inorg. Chim. Acta, 1975, 14, 237.