He₂²⁺, the Experimental Detection of a Remarkable Molecule

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The smallest possible doubly charged molecule, He_2^{2+} , has been detected spectroscopically by charge-stripping mass spectrometry.

Helium, the second most abundant form of matter in the universe, is the chemically least reactive element. It owes this property to the filled K shell and to the uniquely high first ionization energy, -24.586 eV.¹ Nevertheless, several singly charged helium molecules with appreciable binding energies are known, e.g. HeH⁺ $(D_0^{\circ} = 178.07 \text{ kJ/mol})^2$ and He₂⁺ $(D_0^{\circ})^2$ = 228.27 kJ/mol).² The smallest possible doubly charged molecule, † He₂²⁺, hitherto only characterized through quantum mechanical calculations,^{2,3,4} is a remarkable species of potential astrophysical interest and importance in fusion processes. It is indicated, at definitive (James-Coolidge) theoretical levels, to have the shortest internuclear separation in any molecule, 0.704 Å. 3 For comparison, the experimental distances in H₂ and in HeH⁺, which also have two-electron bonds, are 0.741 and 0.774 Å, respectively.² The threeelectron bond in He₂+ is somewhat longer, 1.081 Å.² The electrostatic repulsion in He_2^{2+} is enormous. According to

Coulomb's law, the energy released when two point charges 0.704 Å apart separate to infinity is 1975 kJ/mol. Indeed, the calculated energy of He_2^{2+} lies far above that of two separated He⁺ ions. Despite this huge driving force for dissociation.

$$He_2^{2+} \rightarrow He^{++} + He^{++}; D_0^{\circ} = -835.9 \text{ kJ/mol}^3$$
 (1)

He₂²⁺ is predicted theoretically to have a bound (metastable) ground state $X^{1}\Sigma_{g}^{+}$. The dissociation (equation 1) is inhibited by an energy barrier calculated to be 144.8 kJ/mol.³ How does this barrier arise? For short increases in length from the equilibrium value, the loss in binding energy (which depends roughly on $1/r^{2}$) is greater than the gain due to the reduction in electrostatic repulsion (which depends on 1/r). Many metastable doubly charged molecules comprised of only a few atoms are now known,^{5,6} and owe their existence to the same principle. The energy barrier for the dissociation of He₂²⁺ should be high enough to permit its experimental observation, but we find no evidence for its generation in the mass spectra of helium or in the collision of α -particles with helium. We now report the first spectroscopic detection of this species.

[†] HeH²⁺ is not bound, see ref. 2.

This has been achieved, albeit with considerable difficulty, by means of the charge-stripping technique which is being used extensively to generate multiply charged species.⁷

The precursor He_2^+ ions were formed by chemical ionization at relatively high pressures (up to ca. 0.5 Torr) in the triple sector mass spectrometer described previously.⁸ These ions, detected in the third field-free region are accelerated to an energy of 7 keV, mass selected by the magnetic sector, and then allowed to interact with the neutral target gas in a collision cell (equation 2). The voltages applied to the electrostatic sector are scanned so as to transmit ions whose energy-to-charge ratio is V/2 or slightly less. However, considerable difficulty is encountered in attempts to observe ${}^{4}\text{He}_{2}{}^{2+}$, owing to serious interference from the peak arising from the fragmentation reaction (equation 3). The noise associated with the relatively intense broad peak for 4He+* obscured detection of the ${}^{4}\text{He}_{2}{}^{2+}$ signal expected in the same V/2 region of the translational energy spectrum. Hence recourse was made to the examination of a 1:1 mixture of ³He and 4He. However, this experiment was foiled by N+ products formed via reaction (4), in the field-free region before the magnet, from N_2^+ generated from a low level of residual N_2 in the ion source. These fragment ions have an apparent mass-to-charge ratio of 7 (= $14^{2}/28$) and an energy-to-charge ratio of V/2.

$$He_{2^{+}} + N_{2} \rightarrow He_{2^{2+}} + e^{-} + N_{2}$$
 (2)

$${}^{4}\text{He}_{2}{}^{+}{}^{\cdot} + N_{2} \rightarrow {}^{4}\text{He}{}^{+}{}^{\cdot} + {}^{4}\text{He} + N_{2}$$
 (3)

$$N_2^+ \to N^+ + N^{\cdot} \tag{4}$$

This second source of interference was overcome by employing a second electrostatic analyser. The first two sectors mass select ${}^{3}\text{He}{}^{4}\text{He}{}^{+}$ and are adjusted to transmit only ions of energy-to-charge ratio V; ions produced from reaction (4) are thus excluded. After charge-stripping in the third field-free region of the spectrometer, a spectrum due to ${}^{3}\text{He}{}^{4}\text{He}{}^{2}$ was clearly evident, albeit at a very low signal intensity (ca. 6×10^{-19} A). The minimum energy defect (taken to be the onset energy) was measured to be $37 \pm 2 \text{ eV}$.

We assign this peak to the bound-bound transitions from the $X^2\Sigma_u^+$ state of He₂⁺ to the $X^1\Sigma_g^+$ state of He₂²⁺. There was also some evidence for possible transitions to the $1^1\Sigma_u^+$ or $2^1\Sigma_g^+$ states^{3.4} of He₂²⁺ as well.⁹ The observation of He₂²⁺ suggests the possibility that other multiply charged homo- or hetero-nuclear clusters comprised of two or more noble gas atoms might exist. We are investigating this possibility both computationally and experimentally.

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