

Glycinoeclepin A, a Natural Hatching Stimulus for the Soybean Cyst Nematode

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A large-scale of isolation of glycinoeclepin A, a natural hatching stimulus for the soybean cyst nematode, and its structure elucidation by spectroscopic and X-ray methods are described.

We recently reported¹ the isolation of the title compound, glycinoeclepin A (**1a**), as its bis(*p*-bromophenacyl) ester (**1b**). We now describe the determination of the structure of this hatching stimulus.

Compound (**1b**) was isolated in large quantities, required for the structure elucidation, as follows. Hatch-stimulating concentrates (*ca.* 1 kg) were prepared from dried and powdered roots of kidney beans (*ca.* 1 ton) collected in late July at Memuro (10 hectare), Hokkaido, and were fractionated according to an improved modification of the reported

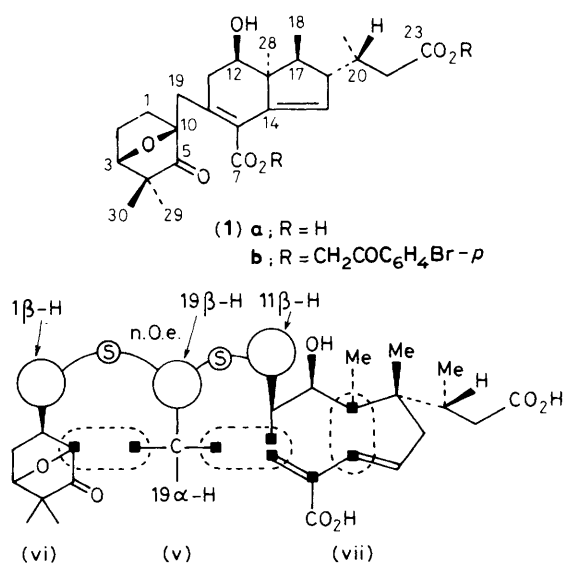


Table 1. Isolation of glycinoeclepin A as its bis(*p*-bromophenacyl) ester.

Dried and powdered kidney bean roots	1058 kg (10 ha)
↓ Extraction with water (<10 °C, 15 kl)	
↓ Evaporation (<30 °C) under reduced pressure	
Concentrates	3801
↓ Continuous extraction with chloroform at pH 2–3	
Chloroform extracts A	1011 g 10 ⁻⁵ g/ml
↓ EtOAc-aq. NaHCO ₃	
Acidic extracts B	636 g 10 ⁻⁵ –10 ⁻⁶ g/ml
↓ Charcoal–Celite column (aq. acetone)	
Fraction C	197.3 g 10 ⁻⁶ g/ml
↓ Sephadex LH-20 column (MeOH)	
Fraction D	88.9 g 10 ⁻⁶ –10 ⁻⁷ g/ml
↓ DEAP-Sephadex LH-20 column (72% EtOH–AcOH–NH ₄ OH)	
Fraction E	18.15 g 10 ⁻⁷ g/ml
↓ Sephadex LH-20 column (CH ₂ Cl ₂ –MeOH, 95:5)	
Fraction F	3.59 g 10 ⁻⁷ –10 ⁻⁸ g/ml
↓ Sephadex LH-20 column (CH ₂ Cl ₂ –MeOH, 97:3)	
Fraction G	951 mg 10 ⁻⁸ g/ml
↓ <i>p</i> -Bromophenacylation (<i>p</i> -BrC ₆ H ₄ COCH ₂ Br, Pr ₂ NEt in MeCN, 20 °C); Silica gel column	
Fraction H as <i>p</i> -BPE ^a	2057 mg 10 ⁻⁸ g/ml
↓ Hitachi Gel 3019 column (MeOH–CH ₂ Cl ₂ , 8:2)	
Fraction I as <i>p</i> -BPE	336 mg 10 ⁻⁹ g/ml
↓ μBondapak C ₁₈ (MeOH–H ₂ O, 8:2)	
Fraction J as <i>p</i> -BPE	79 mg 10 ⁻¹⁰ g/ml
↓ μBondapak NH ₂ (C ₆ H ₁₄ –CH ₂ Cl ₂ –MeCN, 76:14:10)	
Fraction K as <i>p</i> -BPE	8.8 mg 10 ⁻¹¹ g/ml
↓ μBondapak NH ₂ (C ₆ H ₁₄ –CH ₂ Cl ₂ –MeCN, 63:33:4)	
Glycinoeclepin A as <i>p</i> -BPE	1.25 mg 10 ⁻¹¹ –10 ⁻¹² g/ml

^a *p*-BPE = bis(*p*-bromophenacyl) ester.

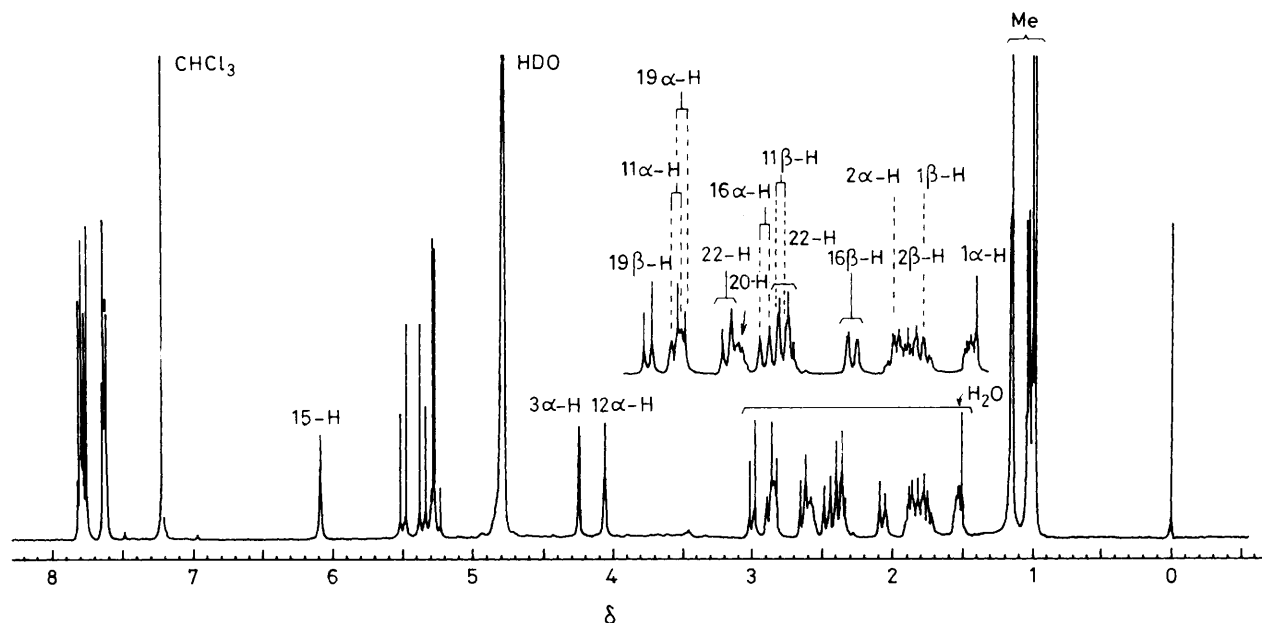


Figure 1. ^1H N.m.r. spectrum of (**1b**); 400 MHz, 500 μg in $\text{CDCl}_3\text{-D}_2\text{O}$.

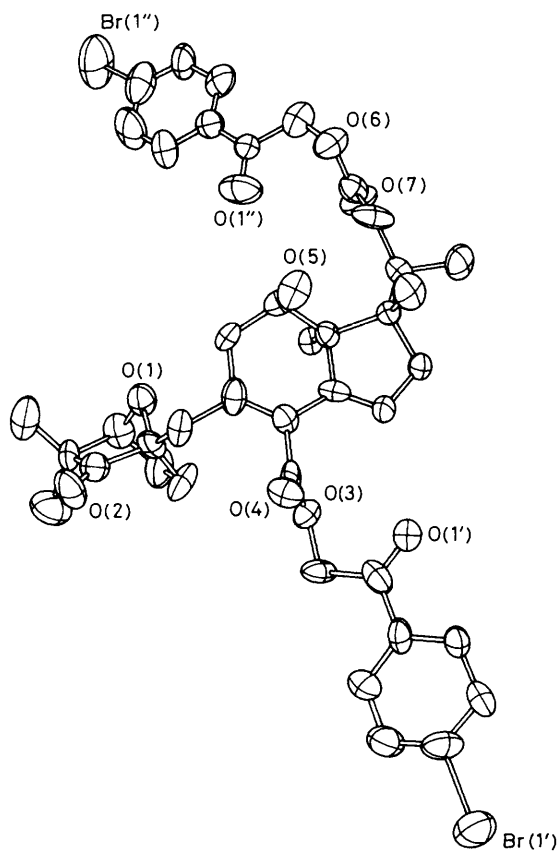


Figure 2. X-Ray structure of (**1b**).

procedure¹ (Table 1). Repeated chromatography, guided by bioassays,¹ yielded (**1b**) (1.25 mg), showing a single h.p.l.c. peak on various columns; following hydrolysis to give (**1a**), hatching and larval emergence were stimulated at a concentration of 10^{-11} – 10^{-12} g/ml in water at 25 °C.

The molecular formula of (**1a**) was established as $\text{C}_{25}\text{H}_{34}\text{O}_7$ on the basis of the field-desorption (m/z 838) and chemical-ionisation (NH_3 ; m/z 856) mass spectra of (**1b**) as well as the ^1H n.m.r. spectra of (**1b**) (Figure 1). The ^{13}C n.m.r. spectra of (**1b**) under completely decoupled and off-resonance conditions, combined with INEPT studies, and the ^1H n.m.r. spectra of (**1b**), coupled with examination of the COSY spectra and extensive decoupling studies, indicated the presence of the following structural units:† $2 \times (\text{CH}_3\text{-}\blacksquare)$, $\text{CH}_3\text{-}\blacksquare\text{-CH}_3$, $>\text{C}=\text{O}$ (δ 215.7), $\text{sp}^3\text{-C-CO}_2\text{H}$ (δ 173.1), $\blacksquare=\blacksquare\text{-CO}_2\text{H}$ (δ 169.5), $\blacksquare\text{-CH(O-}\blacksquare\text{)CH}_2\text{CH}_2\text{-}\blacksquare$ (i), $\blacksquare\text{-CH(CH}_3\text{)CH}_2\text{-}\blacksquare$ (ii), $\blacksquare\text{-CH(OH)CH}_2\text{-}\blacksquare$ (iii), $\blacksquare=\text{CH}(\delta$ 6.02) $\text{CH}_2\text{-}\blacksquare$ (iv), and $\blacksquare\text{-CH}_2\text{-}\blacksquare$ (v). Measurement of the nuclear Overhauser effect (n.O.e.) difference spectra of (**1b**) involving the methyl and other protons, coupled with the i.r. (1758 cm^{-1}) and u.v. [251 nm (ϵ 8300)] spectra of (**1a**), led to the presumption that the unit (i) can be extended to a structural moiety (vi) and units (ii)–(iv) to the unit (vii). In view of the small n.O.e. between $19\beta\text{-H}$ in (v) and $1\beta\text{-H}$ (vi) and $11\beta\text{-H}$ in (vii), the partial structures (vi), (v), and (vii) must be connected straightforwardly, and hence glycoeclepin A most probably has the structure (**1a**).

This structure assignment was confirmed, and the absolute configuration established, by X-ray analysis of a single crystal of (**1b**) obtained by slow evaporation of a solution in hexane and dichloromethane containing a trace of methanol. *Crystal data*: $\text{C}_{41}\text{H}_{44}\text{O}_9 \cdot \frac{1}{2}\text{CH}_2\text{Cl}_2$, monoclinic, space group $P2_1$, $a = 18.596(6)$, $b = 15.010(4)$, $c = 16.112(5)$ Å, $\beta = 114.41(2)^\circ$, $Z = 4$, $D_c = 1.432\text{ g cm}^{-3}$, $\mu(\text{Cu-K}\alpha) = 35.6\text{ cm}^{-1}$. There are two molecules of (**1b**) and one of dichloromethane in the asymmetric unit. The structure was solved by the Monte Carlo direct method,² and the absolute configuration was determined by taking account of the anomalous dispersion of the halogen atoms for Cu-K α radiation. The final R value was 7.8%. The two molecules of (**1b**) have almost the same

† \blacksquare denotes a quaternary carbon atom.

geometry, and the structure of one is shown in Figure 2.‡ The structure of (1a) is characterized by migration of two methyl groups involved in the C and D rings and oxidative cleavage of the B ring with loss of one carbon atom, compared with those

of cycloartanes such as cimigenol,³ and is regarded as a pentanortriterpene.

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‡ The atomic co-ordinates for this work are available on request from the Director of the Cambridge Crystallographic Data Centre, University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW. Any request should be accompanied by the full literature citation for this communication.

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