

Synthesis and Structures of $\text{Re}_6(\text{CO})_{18}(\mu_4\text{-PMe})_3$ and $\text{Re}_5(\text{CO})_{14}(\mu_4\text{-PMe})(\mu\text{-PMe}_2)(\mu_3\text{-P}[\text{Re}(\text{CO})_5])$; Two New Rhenium Cluster Geometries

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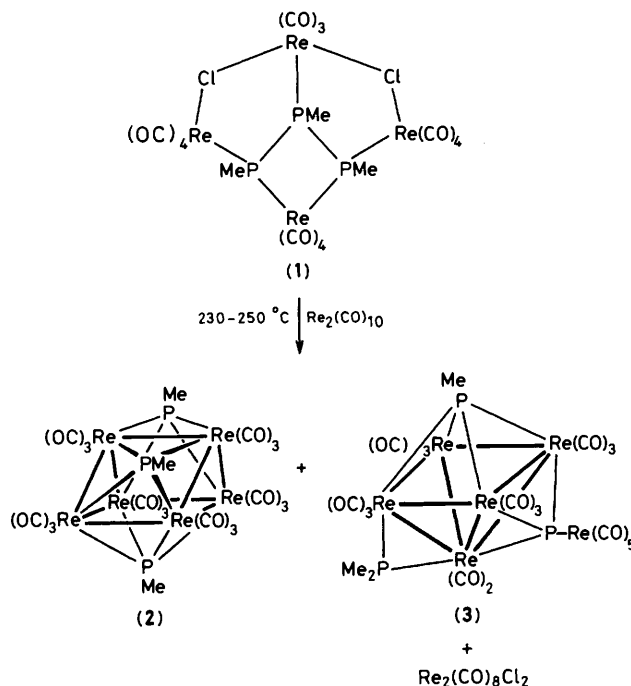
Reaction of $\text{Re}_4\text{Cl}_2(\text{CO})_{15}\{\text{MePP}(\text{Me})\text{PMe}\}$ with $\text{Re}_2(\text{CO})_{10}$ at 230–250 °C leads to formation of the trigonal prismatic $\text{Re}_6(\text{CO})_{18}(\mu_4\text{-PMe})_3$ and square pyramidal $\text{Re}_5(\text{CO})_{14}(\mu_4\text{-PMe})(\mu\text{-PMe}_2)\{\mu_3\text{-P}[\text{Re}(\text{CO})_5]\}$, new rhenium cluster geometries which have been characterized by X-ray crystal analysis.

In rhenium carbonyl chemistry, polynuclear species have until recently been limited to the numerous tri- and tetra-nuclear hydridocarbonyls and hydridocarbonyl anions. Only in the last couple of years have a few large rhenium clusters based on the interstitial carbide octahedron Re_6C been synthesised by pyrolysis of $(\text{Et}_4\text{N})\{\text{ReH}_2(\text{CO})_4\}$ as the anions $[\text{Re}_6(\mu_3\text{-H})_2\text{C}(\text{CO})_{18}]^{2-}$,¹ mono-capped $[\text{Re}_7\text{C}(\text{CO})_{21}]^{3-}$,^{2a} and bi-capped $[\text{Re}_8\text{C}(\text{CO})_{24}]^{2-}$.^{2b} We are currently attempting to develop a chemistry associated with rhenium–phosphorus clusters.

Pyrolysis of the novel cyclometallophosphine complex $\text{Re}_4\text{Cl}_2(\text{CO})_{15}\{\text{MePP}(\text{Me})\text{PMe}\}$ (**1**)³ with $\text{Re}_2(\text{CO})_{10}$ under argon at 230–250 °C leads to the formation of two hexanuclear clusters of stoichiometry $\text{Re}_6(\text{CO})_{18}(\text{P}_3\text{Me}_3)$ (**2**) and $\text{Re}_6(\text{CO})_{19}(\text{P}_3\text{Me}_3)$ (**3**) (Scheme 1). Total yields are essentially stoichiometric with little decomposition, the formation of (**2**) being favoured at 230 °C, (**3**) at higher temperature.

Both compounds are dark red with relatively simple i.r. spectra: (**2**) $\nu(\text{CO})$, tetrahydrofuran (THF): 2035s, 2028s, 2010wbr, 1980m, and 1953wbr cm^{-1} ; (**3**) $\nu(\text{CO})$, toluene: 2057sh, 2052s, 2033sh, 2024s, 2008s, 1995w, 1978m, 1966m, and 1954m cm^{-1} .

The structures of (**2**) and (**3**) have been determined by single-crystal X-ray techniques† and both proved to be new cluster geometries for rhenium. The structure of (**2**) (Figure 1) is that of a symmetrical, tri-capped trigonal prism $\text{Re}_6(\text{CO})_{18}(\mu_4\text{-PMe})_3$, the three methylphosphinidene groups probably resulting from P–P bond cleavage of the $\text{MeP}(\text{Me})\text{-PMe}$ moiety in (**1**). The trigonal prism angles are fixed at 60 and 90°, the molecule possessing crystallographically imposed $\bar{6}$ symmetry. In keeping with the known trigonal prismatic clusters of group 8 {for example, $[\text{Pt}_6(\text{CO})_{12}]^{2-}$ (ref. 4a) and $[\text{Ni}_6(\text{CO})_{12}]^{2-}$ (ref. 4b)}, the Re–Re bonds of the triangular faces [3.092(1) Å] are somewhat shorter than the



Scheme 1

other three edges [3.156(1) Å] but the difference is much less. The deceptively simple i.r. spectrum of this compound is a reflection of its intrinsic symmetry.

The structure of (**3**) (Figure 2) is that of a pentanuclear square pyramid with the base capped by a methylphosphinidene moiety, one triangular face capped by a $-\text{P}[\text{Re}(\text{CO})_5]$ phosphinidene-like moiety and one edge bridged by a dimethylphosphido group to give a stoichiometry $\text{Re}_5(\text{CO})_{14}(\mu_4\text{-PMe})(\mu\text{-PMe}_2)\{\mu_3\text{-P}[\text{Re}(\text{CO})_5]\}$. The variety of phosphorus ligands in this structure leads one to suggest that prior to P–P bond cleavage in (**1**), $\text{Re}_2(\text{CO})_{10}$ inserts into the central Re–P bond with expulsion of $\text{Re}_2(\text{CO})_8\text{Cl}_2$ and that in the ensuing scrambling of functionalities, methyl group migration is more favoured at higher temperatures. However, any hope of fully comprehending a mechanism for this reaction is probably unlikely in view of the drastic conditions employed. The average apical Re–Re bond length in (**3**) of 2.929 Å is markedly shorter than the average basal distances (3.044 Å) in common with other square pyramidal structures such as $\text{Ru}_5(\text{CO})_{15}(\mu_4\text{-PPh})$.⁵

Although the average Re–Re bond lengths in (**3**) (2.986 Å) are considerably shorter than those of (**2**) (3.113 Å), they are not too dissimilar to the 3.041(1) Å of $\text{Re}_2(\text{CO})_{10}$ ⁶ [Δ for (**2**) +0.072; (**3**) –0.055 Å] and are comparable to the few known large rhenium clusters $[\text{Re}_6(\mu_3\text{-H})_2\text{C}(\text{CO})_{18}]^{2-}$ (av. 3.020 Å),¹ $[\text{Re}_7\text{C}(\text{CO})_{21}]^{3-}$ (av. 2.992 Å),² and $[\text{Re}_8\text{C}(\text{CO})_{24}]^{2-}$ (av. 2.985 Å).³ In respect to the rhenium–phosphinidene phosphorus bond lengths, the average Re–P of 2.453 Å in (**2**) is

† Crystal Data: (**2**), dark-red capped hexagonal prisms (from THF), $\text{C}_{22}\text{H}_9\text{O}_{19}\text{P}_3\text{Re}_6$, $M = 1759.46$; hexagonal, $a = 11.311(2)$, $c = 15.506(2)$ Å, $U = 1718.1(5)$ Å³; space group $P6_3/m$; $Z = 2$, $D_c = 3.401$ g cm^{-3} ; $F(000) = 1548$; $\lambda = 0.71069$ Å; $\mu(\text{Mo-K}\alpha) = 224.57$ cm^{-1} . From 1378 unique diffractometer (Syntex $P2_1$) measured data ($2\theta \leq 55^\circ$) obtained using a crystal of dimensions 0.04 dia. \times 0.05 mm, the structure was solved by Patterson and Fourier techniques using 783 absorption corrected reflections with $I \geq 3\sigma(I)$. The structure was refined to an R of 0.039 ($R_w = 0.046$).

Crystal Data: (**3**) dark-red prisms (from toluene), $\text{C}_{22}\text{H}_9\text{O}_{19}\text{P}_3\text{Re}_6 \cdot \text{C}_7\text{H}_8$, $M = 1879.61$; monoclinic, $a = 18.651(2)$, $b = 11.028(2)$, $c = 20.846(2)$ Å, $\beta = 105.89(1)^\circ$, $U = 4123.8(9)$ Å³; space group $P2_1/n$; $Z = 4$, $D_c = 3.027$ g cm^{-3} ; $F(000) = 3352$; $\lambda = 0.71069$ Å; $\mu(\text{Mo-K}\alpha) = 187.26$ cm^{-1} . From 5767 diffractometer measured data ($2\theta \leq 46^\circ$) obtained using a 0.1 mm³ crystal, the structure was solved by direct methods (MULTAN 80) and Fourier techniques using 3404 absorption corrected reflections with $I \geq 3\sigma(I)$. The structure was refined to an R of 0.046 ($R_w = 0.054$).

The atomic co-ordinates for these structures are available from the Director of the Cambridge Crystallographic Data Centre, University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW. Any request should be accompanied by the full literature citation for this communication.

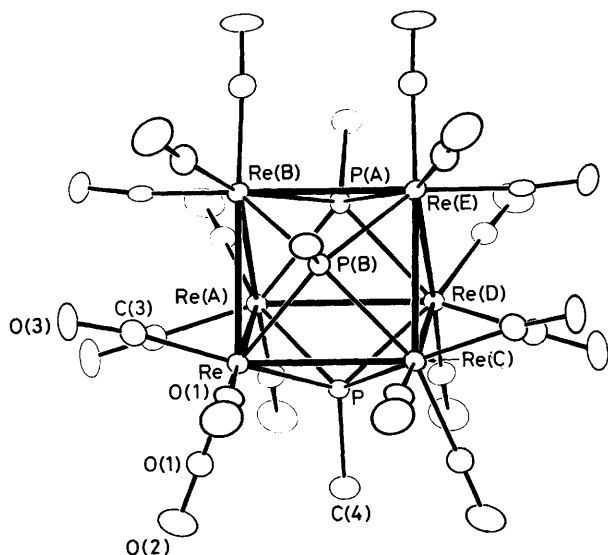


Figure 1. ORTEP plot of $\text{Re}_6(\text{CO})_{18}(\mu_4\text{-PMe})_3$. Re–Re(A) = Re–Re(B) 3.092(1); Re–Re(C) 3.156(1); Re–P 2.432(6); Re–P(B) 2.474(6) Å; (A) = $-y, x-y, z$; (B) = $-x+y, -x, z$; (C) = $x, y, 1/2-z$; (D) = $-y, x-y, 1/2-z$; (E) = $-x+y, -x, 1/2-z$.

identical to the average Re–P [Re–P(1), Re(1)–P(2), Re(4)–P(2), Re(6)–P(2)] 2.453 Å in (3) but with the Re(5)–P(2) bond length of 2.281(7) Å being 0.17 Å shorter, this may be viewed as a possible rhenium–phosphorus double bond.

Both these $2n$ structures conform to electron counting rules. Although (3) is readily recognisable as the mono-capped octahedron expected for a $14e-7$ vertex polyhedron, the tri-capped trigonal prismatic geometry of (2) is somewhat misleading since an $18e-9$ vertex polyhedron should be either a mono-capped square antiprism or dodecahedron. However, Figure 1 does indeed demonstrate the presence of both predicted forms with Re, Re(A), Re(C), and Re(D) half a square antiprism, Re(B), Re(E), P(A), and P(B) half a dodecahedron, and P the capping vertex.

Compounds (2) and (3) represent the first examples of high nuclearity rhenium–phosphorus clusters and are also possibly the first examples of homonuclear phosphinidene clusters outside the group 8 triad.

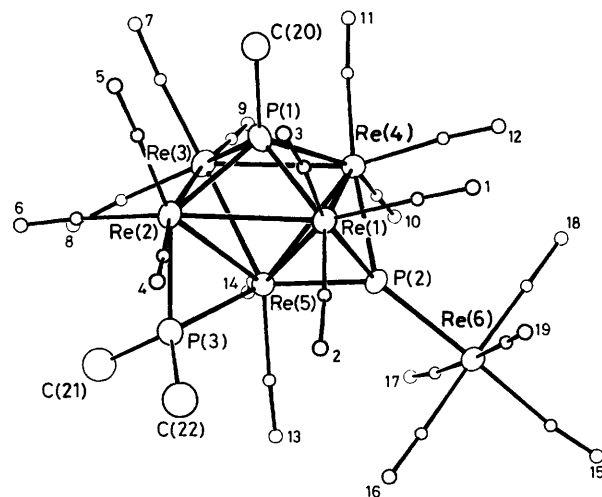


Figure 2. ORTEP plot of $\text{Re}_5(\text{CO})_{14}(\mu_4\text{-PMe})(\mu\text{-PMe}_2)\{\mu_3\text{-P}[\text{Re}(\text{CO})_5]\}$. Re(1)–Re(2) 3.101(1); Re(1)–Re(4) 2.983(1); Re(1)–Re(5) 2.973(1); Re(2)–Re(3) 3.067(2); Re(2)–Re(5) 2.925(1); Re(3)–Re(4) 3.024(1); Re(3)–Re(5) 2.856(2); Re(4)–Re(5) 2.960(1); Re(1)–P(1) 2.437(7); Re(1)–P(2) 2.488(7); Re(2)–P(1) 2.436(7); Re(2)–P(3) 2.425(8); Re(3)–P(1) 2.450(7); Re(4)–P(1) 2.411(7); Re(4)–P(2) 2.465(7); Re(5)–P(2) 2.281(7); Re(5)–P(3) 2.334(8); Re(6)–P(2) 2.486(7) Å.

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