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Biosynthesis of Asukamycin. Formation of the 2-Amino-3-hydroxycyclopent-2-enone Moiety

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Feeding experiments with labelled precursors suggest that the 2-amino-3-hydroxycyclopent-2-enone moiety of the anticoccidial antibiotic, asukamycin, is formed by a novel intramolecular cyclization of 5-aminolevulinic acid.

A number of antibiotics have been discovered in recent years which contain as a unique structural feature a 2-amino-3hydroxycyclopent-2-enone (1) moiety. These include in addition to asukamycin (2),^{1,2} the subject of this report, the related compounds manumycin³ and U-56,407,⁴ as well as reductiomycin,⁵ moenomycin A,⁶ senacarcin A,⁷ virustomycin A,⁸ and bafilomycin.⁹ We report results which suggest a mode of biosynthesis of this novel structural unit.

Experiments were carried out using cultures of *Strepto-myces nodosus* subsp. *asukaensis* grown in 100 ml of medium (2% glucose, 2% soybean meal, 0.3% NaCl, pH adjusted to 7.0) in 500 ml indented Erlenmeyer flasks at 28 °C on a rotary shaker (300 r.p.m.). Precursors were fed 24 h after inoculation and 48 h later asukamycin (2) was extracted with ethyl acetate and purified by preparative layer chromatography (silica gel; benzene-acetone, 3:1). The purified (2) was quantitated by

h.p.l.c. (μ -Bondapak C₁₈; MeOH-H₂O, 3:1) and subjected to liquid scintillation counting (Beckman LS7500) and/or ¹³C n.m.r. spectroscopy (Bruker WM-300).

Inspection of the structure of (1) suggests that C-1 and the attached nitrogen may be derived from glycine. In agreement with this notion, $[2^{-14}C]$ glycine gave specific incorporations into (2) ranging from 2 to 21%, depending on the amount of material fed (Table 1, expts. 1—4). Incorporation of glycine specifically into the (1) moiety was demonstrated by feeding $[2^{-13}C, {}^{15}N]$ glycine (500 mg; 90 atom % ${}^{13}C$, 99 atom % ${}^{15}N$; 10 flasks). The signal for C-2" in the ${}^{13}C$ n.m.r. spectrum of the resulting (2) (δ 115.39) appears as a 3-line pattern (J_{C-N} 17.6 \pm 0.6 Hz) (Figure 1), indicating predominantly intact incorporation of the C-2–N assembly of glycine into (1).

It occurred to us that (1) contains the entire carbonnitrogen skeleton of 5-aminolevulinic acid (3). If (3) were the

Table 1. Incorporation of ¹⁴C-labelled precursors into asukamycin by cultures of Streptomyces nodosus subsp. asukaensis.

Expt.	Precursor	Spec. act. (d.p.m./µmol) of precursor	Amount/µmol of precursor fed	Amount/µmol of (2) isolated	Spec. act. (d.p.m./µmol) of isolated (2)	Specific incorporation (%)
1	[2-14C]Glycine	1.98×10^{4}	105	4.76	504	2.54
2	2-14ClGlycine	1.00×10^{4}	207	3.51	553	5.53
3	2-14C Glycine	5.2×10^{3}	403	2.88	473	9.1
4	2-14C Glycine	2.6×10^{3}	807	2.49	553	21.3
5	[1,4-14C]Succinic acid	5.5×10^{4}	49	4.16	192	0.35
6	[1,4-14C]Succinic acid	2.56×10^{4}	106	4.56	81	0.32
7	[1,4-14C]Succinic acid	1.30×10^{4}	209	3.24	59	0.45
8	[1,4-14C]Succinic acid	$6.5 imes 10^{3}$	414	4.10	40	0.61
9	[5-14C]-(3)-HCl	2.03×10^{4}	200ª	5.57ª	49	0.24
10	[5-14C]-(3)·HCl	9.7×10^{3}	418ª	4.87ª	110	1.14
11	5-14C1-(3)•HC1	$6.8 imes 10^{3}$	599ª	2.12ª	96	1.41
12	[5-14C]-(3)·HCl	$4.0 imes 10^3$	1010ª	1.63ª	87	2.18

^a Two 100 ml cultures.







Figure 1. Signal for C-2" in the ${}^{13}C$ n.m.r. spectrum of asukamycin derived from [2- ${}^{13}C$, ${}^{15}N$]glycine.

precursor of (1), C-1 of glycine should not be incorporated and carbon atoms 1, 3, 4, and 5 of (1) should arise from succinic acid. Although [1,4-14C]succinic acid gave only low specific incorporations into (2) (Table 1, expts. 5-8), perhaps reflecting the Krebs cycle activity in this organism, other evidence supports the above idea. [1-13C]Acetate labels C-1 and C-3 in the (1) moiety of (2) but not C-2, and the ${}^{13}Cn.m.r.$ spectrum of (2) derived from [U-13C3]glycerol shows strong one-bond ¹³C coupling satellites on the signals for C-1 and C-3 of the (1) moiety but only statistical coupling for C-2 (data not shown). The absence of more than statistical coupling in the signal for C-2" rules out the alternative possibility that C-1 and C-2 of glycine are incorporated intact, since the glycerol would label glycine contiguously in both carbons via serine. The non-utilization of C-1 of glycine was confirmed directly by a comparison of the incorporations of [1-14C]glycine (6.57 × 10⁶ d.p.m., 56 µCi/µmol) and [2-14C]glycine (7.35 × 106 d.p.m., 57 µCi/µmole), each fed to two flasks. The two experiments gave 4 628 and 45 666 total d.p.m. in asukamycin, corresponding to incorporations of 0.070% and 0.62%, respectively, i.e., C-2 of glycine was utilized about 9 times better than C-1.

To evaluate further the role of (3), 5-amino- $[5^{-14}C]$ levulinic acid was fed under a variety of conditions. Although (3) seemed to inhibit antibiotic synthesis and resulted in the

formation of copius amounts of dark pigments, moderately good specific incorporation values were consistently obtained with $[5^{-14}C]$ -(3) (e.g., Table 1, expts. 9–12). That these values were lower than the ones for glycine is not too surprising since apparently the bulk of the added (3) was diverted into other products. In addition, it may reflect low permeability of the cells for (3).

In view of the above results we suggest that the (1) moiety of (2) and other antibiotics is formed by a novel transformation of 5-aminolevulinic acid involving an intramolecular cyclization. A plausible mechanism for this conversion, involving a pyridoxal phosphate (PLP) enzyme, is shown in Scheme 1.

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