

Biosynthesis of Vitamin B₁₂: Order of the Later C-Methylation Steps

Horst C. Uzar and Alan R. Battersby*

University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW, U.K.

Pulse labelling experiments with [*methyl*-¹³C]S-adenosylmethionine (SAM) had previously shown that during the biosynthesis of vitamin B₁₂, the fourth C-methylation site is C-17; new inverted pulse studies show that the remaining four SAM-derived methyl groups are introduced in the order 12, 1, 15, 5.

Biosynthesis of cobyrinic acid (4), the precursor of vitamin B₁₂, from uro'gen-III (1) involves the introduction of eight C-methyl groups by transfer from S-adenosylmethionine (SAM); these C-methylations occur at positions 1, 2, 5, 7, 12, 15, 17, and 20¹ [see structure (1)]. Earlier work based on isolation of intermediates and structure determination had shown¹ that the first three methylations occur in the order C-2, followed by C-7 and then C-20. The dimethylated intermediate was proved by isolation to be the dihydroisobacteriochlorin² (2) and since C-methylation does not affect the oxidation state, it is reasonable to assume that the trimethylated system is also formed at this same dihydro level (3).

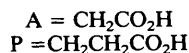
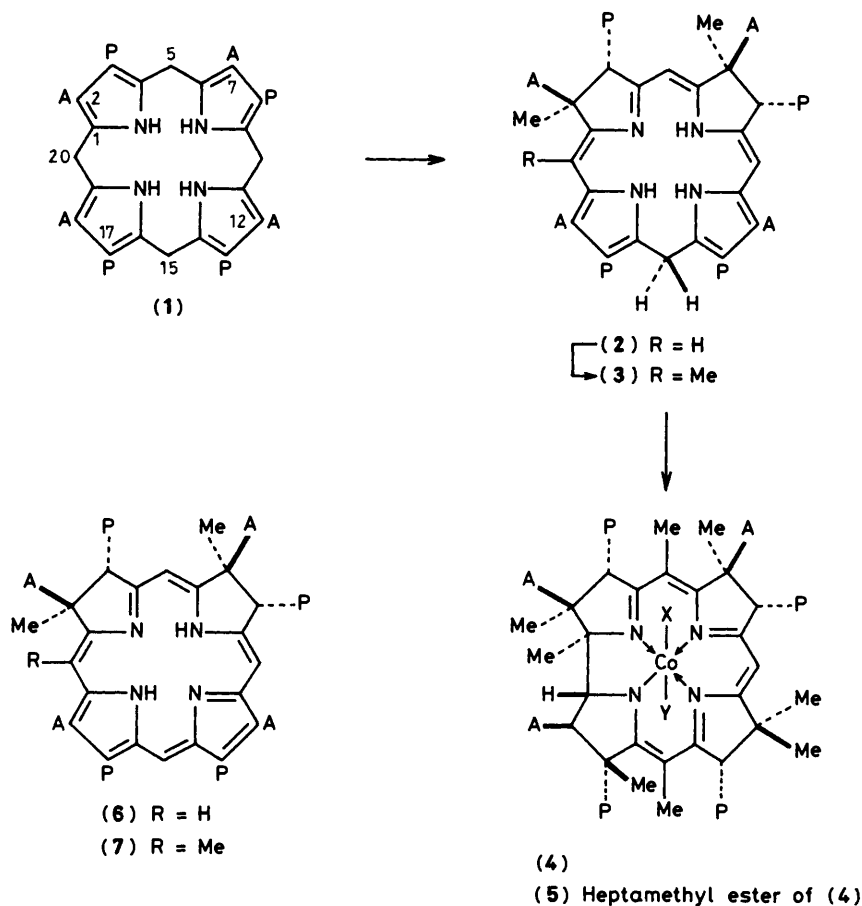
Under normal conditions of work-up (aerobic), the dimethylated and trimethylated intermediates are isolated as the aromatised macrocycles (6) and (7), respectively, but these can evidently be reduced again to the dihydro level by enzyme preparations from B₁₂-producing organisms.¹

So far, no intermediates on the pathway to cobyrinic acid (4) have been isolated carrying 4, 5, or more C-methyl groups. Accordingly, we developed an approach based on pulse labelling which established^{3,4} that the fourth C-methyl group is introduced at C-17 of the trimethyl system (3). It must be borne in mind that enzymic decarboxylation of the C-12 acetate residue in (3), which is necessary for the formation of

Table 1. Relative ¹³C-incorporations^a into C-methyl groups of cobyrinic acid (4).

	Position of methyl group				
	1	5	12	15	17
Cobester (5) from Inverted Pulse (IP)	100	95	106	102	155
Cobester (5) from Normal Pulse ^b (NP)	100	109	93	107	72
Difference ^c (IP - NP)	0	-14	13	-5	83

^a The relative amounts of ¹³C at each methyl group were obtained by comparison of the peak areas (obtained by line-fitting) with those from a standard sample of (5) uniformly labelled at all the SAM-derived methyl groups. The results are expressed as percentages of the ¹³C relative value for the C-1 methyl group. Statistical analysis of many intensity determinations gave the worst error for each peak as <2%, mostly *ca.* 1%. ^b The samples for the Normal Pulse and the standard are those described in ref. 3. For each set of ¹³C-determinations, the 3 spectra (IP, NP, and standard) were recorded consecutively under identical conditions. ^c The more positive this number, the earlier the corresponding methyl group is added in the biosynthetic sequence.

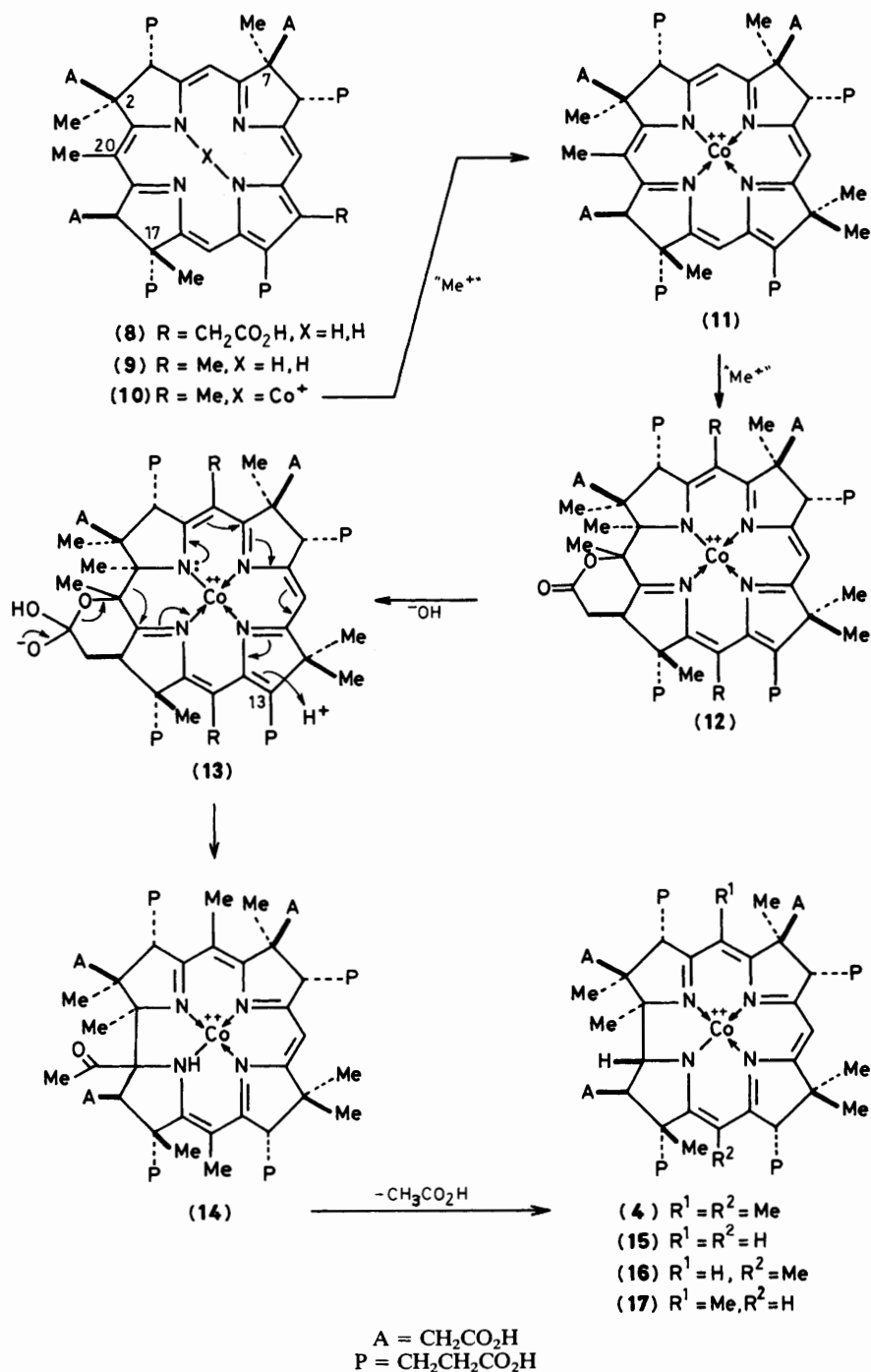


(4), may occur before the C-17 methylation. So the pulse labelling work showed that the tetramethylated intermediate is either the 12-acetate pyrocorphin (8) or the 12-methyl pyrocorphin (9).

The foregoing experiments made use of a cell-free enzyme system from *Clostridium tetanomorphum* which converts the dimethylated system (6) into cobyrinic acid (4). The enzyme preparation containing (6) was incubated with a deficiency of unlabelled SAM and after a period to allow C-methylated intermediates to be produced, a large excess of [*methyl*-¹³C]-SAM was added (this is the Normal Pulse). Further incubation then generated cobyrinic acid (4) in which the C-methyl groups introduced early in the sequence could be detected by their having a lower ¹³C-content (assay by ¹³C n.m.r.³ or by ¹H n.m.r. spectroscopy using a heteronuclear spin echo method⁴).

These studies have now been extended by performing the pulse labelling experiments in the inverted sense, *i.e.* adding the [*methyl*-¹³C]SAM first and then, after an incubation period to generate intermediates, completing the biosynthesis of cobyrinic acid (4) by further incubation with a large excess of unlabelled SAM (Inverted Pulse). This approach requires the amount of endogenous SAM to be small relative to the amount of added [*methyl*-¹³C]SAM and this was established first by dilution analysis. Now the C-methyl groups which are enzymically transferred early in the methylation sequence will

have a higher ¹³C-content and the later ones will have lower levels. The labelled product was isolated as cobester (5) and assayed by ¹³C n.m.r. spectroscopy (see Figure 1). The high ¹³C-content in the C-17 methyl group is clear by inspection, a result in agreement with our earlier finding³ that the fourth methylation site is C-17. Such inspection is only sufficient for large differences in ¹³C-content; for medium sized or smaller differences, it is essential to carry out accurate determinations of signal intensities and to compare them with those from a standard sample of (5) uniformly labelled at the SAM-derived methyl groups. Accordingly, the intensities of the signals from the other C-methyl groups were measured by several methods (*e.g.* integration, cutting and weighing, line fitting). The Lorentzian line fitting programme consistently gave values very close (within 1–2%) to the average from the other methods. Four long-term ¹³C n.m.r. runs were carried out on each labelled sample followed by many determinations of peak area for each signal in every spectrum. The data in Table 1 are from one such set of determinations and the key values are the differences between the values from the Inverse and Normal Pulses since these increase the discrimination of the approach. The more positive this difference value, the earlier the corresponding methyl group is added in the biosynthetic sequence. It should be emphasised that the same methylation sequence emerged from the difference values for each of the 4 separate ¹³C n.m.r. runs. The results extend our earlier



Scheme 1

findings beyond the fourth C-methylation at C-17³ to showing the fifth is at 12, then C-1, 15, and 5.†

This knowledge allows some important features of the biosynthetic pathway beyond the tetramethylated intermediate (8) or (9) to be briefly picked out to help future research (Scheme 1). The stage on the pathway where cobalt is inserted is unknown but recent experiments using *Propionibacterium*

shermanii indicate that this step is not the last one.⁵ Also, Nussbaumer and Arigoni⁶ found that 5,15-bisnorcobyrinic acid (15) and the 5-nor (16) and 15-nor (17) analogues are not incorporated into cobyrinic acid (4) by an active enzyme preparation from *P. shermanii*. These two studies together suggest that cobalt is inserted at some earlier stage. As one possibility, Scheme 1 shows cobalt insertion at the pyrrocorphin stage (9) but it is emphasised that the chemistry shown could also proceed on the metal-free macrocycles which at present must not be excluded.⁷ Methylation of the pyrrocorphin (10) at C-12 yields the corphin⁸ (11) and subsequent methylation at C-1 sets up the system (12) for ring-

† In the unlikely event that errors accumulated in one direction for all 4 runs, then the discrimination between the methylation order at C-1 and C-15 would become doubtful.

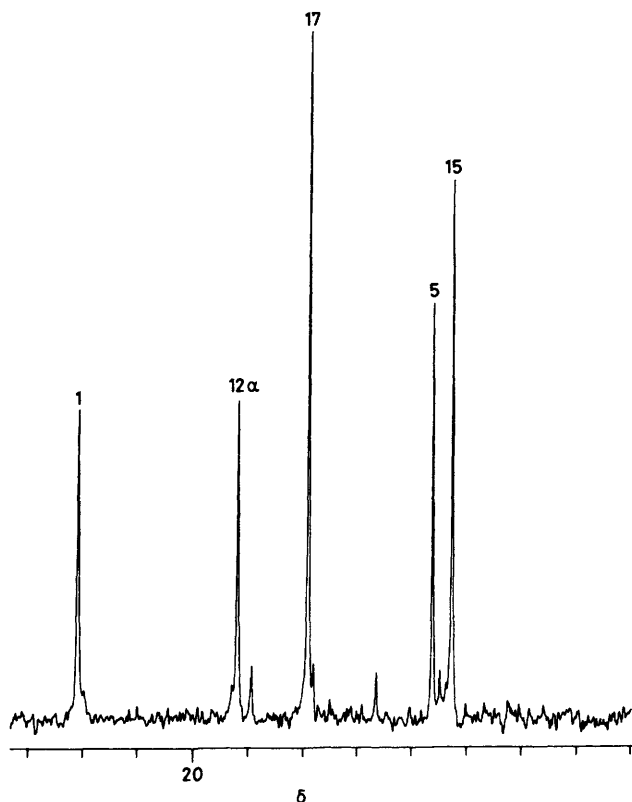


Figure 1. Proton-noise decoupled ^{13}C -spectrum showing enhanced signals from labelled cobester [as (5)] from the Inverse Pulse experiment with $^{13}\text{C}[\text{SAM}]$. Spectrum recorded in C_6D_6 at 100.6 MHz with 1.5 Hz line-broadening.

contraction. The illustrated mechanism for the contraction (13)→(14) is based on Eschenmoser's important biomimetic model studies⁹ together with the firm knowledge that during ring-contraction, C-20 and its attached methyl group are extruded as acetic acid.^{10,11} It is not yet possible to indicate the methylation state at C-5 and C-15 over the stages (12) and (13) but by having both methyls in place on the 19-acetylcobyrinic acid (14), then the deacetylation step directly yields cobyrinic acid (4) ready for conversion finally into vitamin

B_{12} .¹ Experiments to prepare and/or detect several of the key intermediates in Scheme 1‡ are in hand.

The pulse labelling approach has recently been adopted elsewhere¹² and used with our signal assignments for (5)¹³ to confirm our earlier finding³ that the fourth methylation site is at C-17 and to extend this to indicate the later order as 12, then 1 followed by 5 and 15 in unknown sequence.

We thank the Deutsche Forschungsgemeinschaft for a postdoctoral grant (to H. C. U.), the S.E.R.C. and Roche Products Ltd. for financial support, Dr. R. Hirschmann of Merck Sharp & Dohme Research Laboratories for a generous gift of vitamin B_{12} , and Drs. T. A. Carpenter and F. J. Leeper for running many of the n.m.r. spectra.

Received, 31st January 1985; Com. 145

References

- 1 Reviewed by A. R. Battersby and E. McDonald, in 'Vitamin B_{12} ,' ed. D. Dolphin, Wiley, New York, 1982, vol. 1, p. 107.
- 2 A. R. Battersby, K. Frobel, F. Hammerschmidt, and C. Jones, *J. Chem. Soc., Chem. Commun.*, 1982, 455.
- 3 H. C. Uzar and A. R. Battersby, *J. Chem. Soc., Chem. Commun.*, 1982, 1204.
- 4 D. M. Doddrell, H. C. Uzar, and A. R. Battersby, *J. Org. Chem.*, 1983, **48**, 3613.
- 5 T. E. Podschun and G. Muller, *Angew. Chem., Int. Ed. Engl.*, 1985, **24**, 46.
- 6 C. Nussbaumer and D. Arigoni, work described in E. T. H. Diss. No. 7623.
- 7 For metal-free corrins, see V. B. Copenhagen in 'Vitamin B_{12} ,' ed. D. Dolphin, Wiley, New York, 1982, vol. 2, p. 105.
- 8 For corphin chemistry see A. P. Johnson, P. Wehrli, R. Fletcher, and A. Eschenmoser, *Angew. Chem., Int. Ed. Engl.*, 1968, **7**, 623.
- 9 V. Rasetti, A. Pfaltz, C. Kratky, and A. Eschenmoser, *Proc. Natl. Acad. Sci. USA*, 1981, **78**, 16.
- 10 L. Mombelli, C. Nussbaumer, H. Weber, G. Muller, and D. Arigoni, *Proc. Natl. Acad. Sci. USA*, 1981, **78**, 9.
- 11 A. R. Battersby, M. J. Bushell, C. Jones, N. G. Lewis, and A. Pfenninger, *Proc. Natl. Acad. Sci. USA*, 1981, **78**, 13.
- 12 A. I. Scott, N. E. Mackenzie, P. J. Santander, P. E. Fagerness, G. Muller, E. Schneider, R. Sedlmeier, and G. Worner, *Bioorg. Chem.*, 1984, **12**, 356.
- 13 A. R. Battersby, C. Edington, C. J. R. Fookes, and J. M. Hook, *J. Chem. Soc., Perkin Trans. 1*, 1982, 2265; see also L. Ernst, *Liebigs Ann. Chem.*, 1981, 376.

‡ Scheme 1 revised (March 1985) to include results in ref. 5.