A Biogenetic Synthesis of (\pm) -Secotrinerviten-2 β ,3 α -diol

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Based on biogenetic considerations, (\pm)-secotrinerviten-2 β ,3 α -diol was synthesized from acetyl epoxy dehydromukulol (**4c**) by the action of BF₃-Et₂O.

Termite soldiers of nastitermes genera secrete several kinds of diterpenoids as part of their defence mechanisms, of which secotrinervitanes (1) and (2a) are the typical examples.¹ These natural products are biogenetically related to cembrene A

(3a), a trail pheromone of the termite workers:² bonding between C-4 and C-16[‡] of the epoxy derivatives (4a,b) of (3a,b) leads to the secotrinervitane skeleton,³ in which a macrocyclic ring is connected at the C-1 and C-4 positions of the cyclohexane ring with a *trans* diequatorial configuration. The biological activity and particularly the unique structure

‡ Numbering is based on that of secotrinervitane skeleton.

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aroused our interest in the synthesis of these terpenoids. After completion of our synthesis of (1),⁴ we have undertaken the biogenetic type synthesis of (\pm) -secotrinerviten-2 β ,3 α -diol (**2a**)⁵ starting from the epoxide (**4b**). The result is described in this communication.

In our previous paper, we reported⁶ the efficient preparation of epoxy alcohol (4b) from *trans* dehydromukulol (3b). Although treatment of (4b) with $BF_3 \cdot Et_2O$ in diethyl ether at -40 °C resulted in the formation of a complex mixture, the corresponding acetate (4c), prepared from (4b) by the action of Ac₂O in pyridine at room temperature for 24 h, afforded the expected bicyclic hydroxy acetate (2b) in 82% yield when treated with $BF_3 \cdot Et_2O$ in diethyl ether at -20 °C for 17 h.§ The cyclization of (4c) depends largely on the solvent employed and a complex mixture was formed when CH_2Cl_2 at -78 °C was used instead of diethyl ether. Reduction of (2b) with LiAlH₄ gave free diol (2a). Physical data of (2a), thus synthesized, were identical with those of natural secotriner-viten-2 β ,3 α -diol.

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§ (**2b**) M.p. 78—79 °C; ¹H n.m.r. (CDCl₃, 90 MHz) δ 5.3—5.1 (1H, m), 4.8—4.7 (1H, br.s), 4.72 (1H, d, J 1.5 Hz), 4.53 (1H, dd, J 11.3, 8.4 Hz), 3.64 (1H, d, J 8.4 Hz), 2.89 (1H, br.d. J 8.5 Hz), 2.13 (3H, s), 1.63 (3H, s), 1.57 (3H, t, J 1.4 Hz), and 0.87 (3H, s).