

Highly Enantioselective Route to (*R*)-Proline Derivatives via Enzyme Catalysed Hydrolysis of *cis*-*N*-Benzyl-2,5-bismethoxycarbonylpyrrolidine in an Aqueous Dimethyl Sulphoxide Medium

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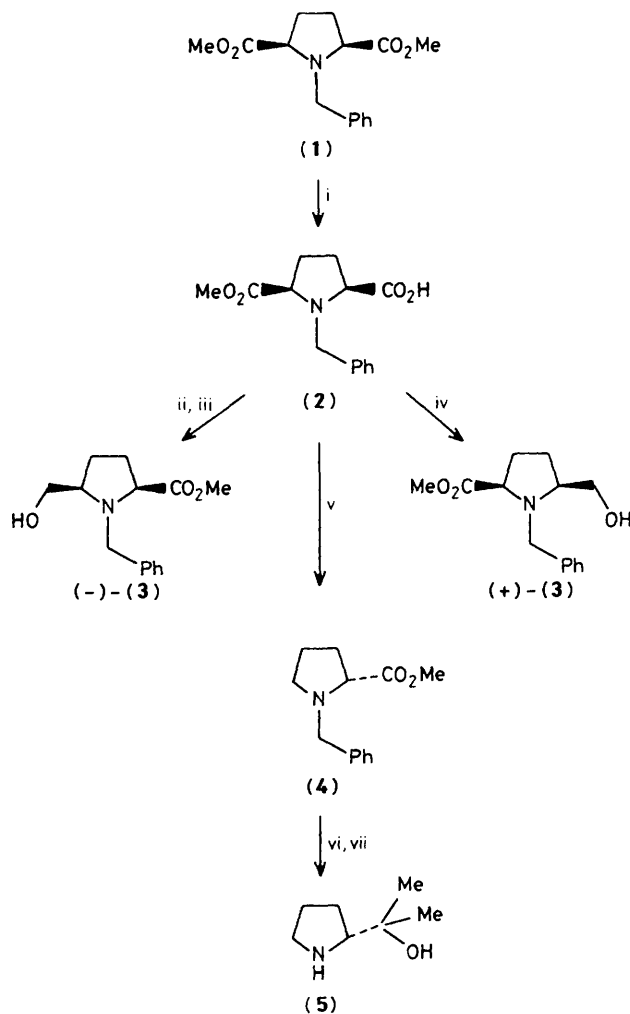
Pig liver esterase catalysed hydrolysis of diester (**1**) with dimethyl sulphoxide as cosolvent in buffered solutions gave optically pure monoester (**2**) which was further transformed to the (*R*)-proline related ester (**4**) by radical decarboxylation.

Enzyme catalysed hydrolysis of symmetrical diesters as an effective method of producing many interesting optically active compounds on a practical scale is currently under intensive investigation.¹ (*2S,5R*)-*N*-Benzyl-5-methoxycarbonylproline (**2**) should have significant potential as a starting material for the preparation of pyrrolidine alkaloids (*e.g.* ant signal substances²) and has been considered as a chiral synthon candidate for use in the synthesis of carbapenem antibiotics. In this context Ohno *et al.*³ recently reported the pig liver esterase (PLE) catalysed hydrolysis of the readily

available *meso* diester (**1**)⁴ giving (**2**) in 80% enantiomeric excess (*e.e.*).

In this communication we describe the synthesis of the optically pure monoester (**2**) employing dimethyl sulphoxide (DMSO) as cosolvent in the enzymatic hydrolysis of (**1**). † The

† Several related pyrrolidine and piperidine diester substrates have been subjected to this hydrolysis. Our results with respect to changes in reaction conditions *etc.* will be presented in a forthcoming full paper.



Scheme 1. Reagents and conditions: i, PLE, DMSO-H₂O, pH 7.5; ii, LiOAc, LiBH₄-tetrahydrofuran (THF); iii, MeOH/H⁺; iv, BH₃-THF; v, isobutyl chlorocarbonate, *N*-methylmorpholine, *N*-hydroxypyridine-2-thione, Et₃N, *t*-butyl thiol, irr. 300 W; vi, MeMgI; vii, H₂/Pd-C, MeOH.

dramatic influence of DMSO on enantioselectivity in the PLE-catalysed hydrolysis of dialkylated propanedioic acid dimethyl esters⁵ and other diesters⁶ has been shown in previous studies.

A DMSO concentration of 25% in the reaction medium[‡] proved to give an enantiospecific hydrolysis of diester (1) (Table 1, entry 3). The chiral half-ester (2) was obtained in 39% chemical yield and 100% e.e. as directly determined from its ¹H n.m.r. spectrum in the presence of enantiomerically pure (*R*)-(+)-1-phenylethylamine.⁷ Further experimental work to improve and explain the moderate chemical yield is in progress.

In order to demonstrate the usefulness of the monoester (2) as a versatile source of optically active *cis*-2,5-bifunctionalized pyrrolidines it was selectively transformed to either enantiomer of *N*-benzyl-*cis*-5-methoxycarbonylprolinol [*i.e.* (-)-

Table 1. Optical purity of monoester (2) obtained by pig liver esterase catalysed hydrolysis of diester (1).[‡]

Entry	% DMSO	% e.e. ^a
1	0	17
2	10	61
3	25	100
4	50	97

^a These values (±2%) were determined by ¹H n.m.r. spectroscopy (200 MHz) of the monoesters in the presence of enantiomerically pure (*R*)-(+)-1-phenylethylamine.

(3) and (+)-(3)][§] (Scheme 1). Furthermore we have found (2) to be a valuable and general entry to (*R*)-proline derivatives through radical decarboxylation⁸ to the ester (4) (46% yield). No racemisation occurs in this process as was shown by decarboxylation of monoester (2) to (4) of the same e.e. The optical purity of (4) was determined by ¹H n.m.r. spectroscopy in the presence of tris[3-(heptafluoropropylhydroxymethylene)-(+)-camphorato]europium(III) [Eu(hfc)₃]. Racemic (4) required for this study was produced by treatment of (*S*)-(4) [derived from (*S*)-proline] with NaOMe.

The (*R*)-enantiomer of the crystalline aminol (5) (m.p. 36–38°C) is thus now available *via* Grignard reaction of (4) with MeMgI followed by hydrogenation. The (*S*)-enantiomer of this interesting chiral auxiliary⁹ has been conveniently prepared in four steps from (*S*)-proline *via* (*S*)-(4) (55% overall yield).

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[‡] A typical batch contained 1 mmol substrate (1) dissolved in 75 ml of 0.25 M (CH₂OH)₃CNH₂-HCl (Tris-HCl), pH 7.5, 25% DMSO. Using 1.5 mg PLE (Sigma, batch no. 34F-8110) the hydrolysis was completed within 6 h at 30°C.

[§] Satisfactory spectroscopic data (¹H and ¹³C n.m.r., mass spectra) were obtained for all new compounds.