

## Synthesis of Some Naturally-occurring Styrylamides

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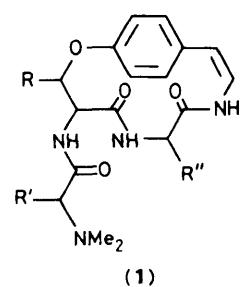
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Thermal elimination from *N*-( $\beta$ -phenyl- $\beta$ -phenylsulphinyloethyl)amides yielded styrylamides (**2**)—(**4**)

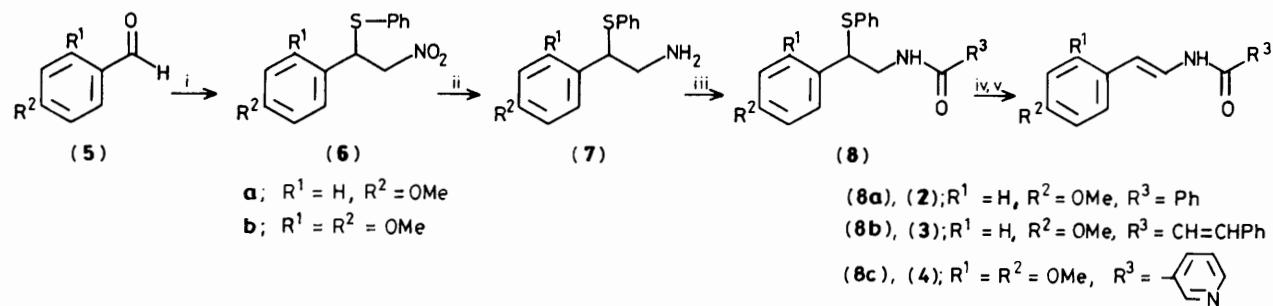
In connection with our programme of synthesis of cyclic peptides based on the route developed by Rapoport,<sup>1</sup> we needed a simple, efficient modification to permit introduction of the styrylamide group in such peptides, (**1**).<sup>2</sup> To explore such a synthetic goal, we selected as targets three naturally-occurring styrylamides (**2**)—(**4**), isolated from *Pleiosperium alatum*, (Wight and Arn) Swingle,<sup>3</sup> *Aegle marmelos* Corr.,<sup>4</sup> and *Amyris plumieri* D.C.,<sup>5a</sup> respectively. Several new styrylamides, including Tunichrome B-1, a V-complexing agent,<sup>6</sup> and Amathamide A and B, from a marine Bryozoa,<sup>7</sup> prompt us to report one successful route to the styrylamide group (Scheme 1).

Beginning with the appropriate aldehyde (**5**), the thiophenol present in the alkaline solution of (**5**) and nitromethane added to the intermediate  $\beta$ -nitrostyrene to give (**6**) in a one-pot sequence. Following reduction of (**6**) to (**7**), acylation with the appropriate acid derivative gave the three

amides (**8a**—**c**). Oxidation of (**8a**—**c**) with *m*-chloroperbenzoic acid (MCPBA) at  $-50\text{ }^\circ\text{C}$  gave sulfoxides that under reflux



Nummularine-M; R = Ph, R' = R'' = Bu<sup>s</sup> (ref. 2a)  
 Frangulanin; R = Pri, R' = Bu<sup>s</sup>, R'' = Bu<sup>i</sup> (ref. 2c)  
 Sativanine-A; R = Ph, R' = Bu<sup>s</sup>, R'' = Pri (ref. 2d)



**Scheme 1. Reagents and conditions:** i,  $\text{MeNO}_2$ ,  $\text{MeNH}_2\text{Cl}$ - $\text{Na}_2\text{CO}_3$ ,  $\text{PhSH}$ , 25 °C, 3 days [(6a) 86%, (6b) pure oil, t.l.c. quantitative]; ii,  $\text{Zn}, \text{HCl-HOAc}$ , heat, 2 h (pure oils, t.l.c., 84–90%); iii,  $\text{CH}_2\text{Cl}_2$ , a: (7a),  $\text{PhCOCl}$ - $\text{NaHCO}_3$ , 25 °C (83% MeOH), b: (7a), ( $\text{PhCH=CHCO}_2\text{O}$ ),  $\text{NaHCO}_3$ , 25 °C (66% MeOH), c: (7b), 3-pyCOOCOEt (from 3-pyCO<sub>2</sub>H,  $\text{CICO}_2\text{Et}$ ,  $\text{Et}_3\text{N}$ ), 0 °C (82% crude, glass solid); iv,  $\text{MCPBA}$ - $\text{CH}_2\text{Cl}_2$ - $\text{Et}_2\text{O}$ , –50 to 0 °C, 6 h (quantitative, pure by t.l.c., solids); v, toluene,  $\text{CaCO}_3$ , reflux, 5 h [(2) and (3) 65% from  $\text{CHCl}_3$ ; (4) 29%, recrystallised poorly from  $\text{C}_6\text{H}_6$ ].

yielded (2)–(4) in 20 [for (4)]–40% overall yields.<sup>†‡</sup> Presumably then, ring closure of appropriate substituents at

<sup>†</sup> The alkene of the styrylamide in Zizyphin-A, a cyclic peptide, was introduced via oxidative elimination on a selenide formed by substitution; U. Schmidt, A. Lieferknecht, H. Bokens, and H. Griesser, *Angew. Chem.*, 1981, **93**, 1121.

<sup>‡</sup> Satisfactory C, H analysis were obtained for (2)–(4).

(2) M.p. 192–193 °C (lit.<sup>3</sup> m.p. 178–180 °C); i.r. (KBr) 3325, 1655  $\text{cm}^{-1}$ ; u.v. (EtOH) 221 (log  $\epsilon$  4.21), 311 (4.39), 317 (4.39) nm;  $m/z$  253 ( $M^+$ ); <sup>1</sup>H n.m.r. (200 MHz) [ $(\text{CD}_3)_2\text{CO}$ ]  $\delta$  3.79 (3H, s, OMe), 6.42 (1H, d,  $J$  15 Hz, =CH), 6.89 (2H, d,  $J$  8.8 Hz, *o*-HArOMe), 7.35 (2H, d,  $J$  8.8 Hz, *m*-HArOMe), 7.52 (3H, m, ArH), 7.58 (1H, d,  $J$  15 Hz, =CHN), 8.00 (2H, dd,  $J$  7.7, 1.8 Hz, *o*-HArCO).

(3) M.p. 206.5–207.5 °C (lit.<sup>4</sup> m.p. 191 °C); i.r. 3462, 1640  $\text{cm}^{-1}$ ; u.v. 284 (log  $\epsilon$  4.47), 300 (4.49), 338 (4.52) nm;  $m/z$  279 ( $M^+$ ); <sup>1</sup>H n.m.r. [ $(\text{CD}_3)_2\text{CO}$ ]  $\delta$  3.78 (3H, s, OMe), 6.25 (1H, d,  $J$  14.7 Hz, =CHArOMe), 6.74 (1H, d,  $J$  15.4 Hz, =CHCO), 6.88 (2H, d,  $J$  8.8 Hz, *o*-HArOMe), 7.32 (2H, d,  $J$  8.8 Hz, *m*-HArOMe), 7.45 (3H, m, ArH), 7.55 (1H, d,  $J$  14.7 Hz, =CHN), 7.61 (2H, m, ArH), 7.65 (1H, d,  $J$  15.4 Hz, =CHAr).

(4) M.p. 160–161.5 °C (lit.<sup>5</sup> m.p. 159–160 °C); i.r. 3250, 1655  $\text{cm}^{-1}$ ; u.v. 220 (log  $\epsilon$  4.40), 262 (4.20) 331 (4.38) nm;  $m/z$  284 ( $M^+$ ); <sup>1</sup>H n.m.r. ( $\text{CDCl}_3$ )  $\delta$  3.82 (3H, s, OMe), 3.85 (3H, s, OMe), 6.46 [1H, m, 3-HAr(OMe)<sub>2</sub>], 6.49 [1H, dd,  $J$  8, 2 Hz, 5-HAr(OMe)<sub>2</sub>], 6.51 [1H, d,  $J$  14.7 Hz, =CHAr(OMe)<sub>2</sub>], 7.33 [1H, d,  $J$  8 Hz, 6-HAr(OMe)<sub>2</sub>], 7.43 [1H, dd,  $J$  8, 7.8 Hz, 5H-3 pyridyl(py)], 7.67 (1H, dd,  $J$  14.7, 14.3 Hz, =CHN), 8.06 (1H, br. d, NH), 8.19 (1H, dt,  $J$  7.8 Hz, 4H-3 py), 8.76 (1H, d,  $J$  4.3 Hz, 6H-3 py), 9.06 (1H, br. s, 2H-3 py); shaking with  $\text{D}_2\text{O}$  removes the signal at  $\delta$  8.06 and reduces  $\delta$  7.67 to a doublet.

$\text{R}^2$  and  $\text{R}^3$  of (8) followed by oxidation–thermal elimination as above would lead to the desired cyclic peptide having the styrylamide function.

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## References

- 1 J. C. Lagarias, R. A. Houghten, and H. Rapoport, *J. Am. Chem. Soc.*, 1978, **100**, 8202.
- 2 (a) V. B. Pandey, J. P. Singh, K. K. Seth, A. H. Shah, and G. Eckhardt, *Phytochemistry*, 1984, **23**, 2118; (b) A. H. Shah, V. B. Pandey, J. P. Singh, K. N. Singh, and G. Eckhardt, *ibid.*, p. 2120; (c) E. Haslinger and W. Robien, *Monatsh. Chem.*, 1982, **113**, 95; (d) R. Tschesche, A. H. Shah, and G. Eckhardt, *Phytochemistry*, 1979, **18**, 702.
- 3 A. Chatterjee, M. Chakrabarty, and A. B. Kundu, *Aust. J. Chem.*, 1975, **28**, 457.
- 4 T. R. Govindachari and M. S. Premila, *Phytochemistry*, 1983, **22**, 755.
- 5 (a) B. A. Burke and H. Parkins, *Tetrahedron Lett.*, 1978, 2723; (b) the *cis* isomer of (4) has been isolated: B. A. Burke and S. Philip, *Heterocycles*, 1985, **23**, 257.
- 6 R. C. Bruening, E. M. Oltz, J. Furukawa, K. Nakanishi, and K. Kustin, *J. Am. Chem. Soc.*, 1985, **107**, 5298.
- 7 A. J. Blackman and D. J. Matthews, *Heterocycles*, 1985, **23**, 2829.