

Tandem Intramolecular Michael Addition and 1,3-Dipolar Cycloaddition Reactions of Oximes; Versatile New Carbon–Carbon Bond Forming Methodology

Paul Armstrong, Ronald Grigg,* Sivagnanasundram Surendrakumar, and William J. Warnock

Chemistry Department, Queen's University, Belfast BT9 5AG, Northern Ireland

Oximes undergo intramolecular Michael addition to proximate electronegative alkenes, generating cyclic nitrones which can be trapped inter- and intra-molecularly in 1,3-dipolar cycloaddition reactions taking place either separately or in tandem.

Oximes react with Michael acceptors and dipolarophiles in a two-step process [(1) → (2) → (3)] to give isoxazolidines in good yields.¹ There are four broad variants of the consecutive Michael addition–cycloaddition sequence (Table 1), and examples of classes 1 and 2 have been described.¹

Classes 3 and 4 provide an easy new route to cyclic nitrones; this can be achieved conceptually *via* either an *exo-trig* (or *exo-dig*) cyclisation (4) → (5), or an *endo-trig* (or *endo-dig*) cyclisation (6) → (7). The *exo-trig* cyclisation (4) → (5) is related to, and complements, Gallagher's metal-catalysed allenic oxime cyclisations (8) → (9).²

Class 3 processes can be carried out stepwise with isolation of the intermediate nitron, or the tandem sequence can be carried out in a one-pot process. Thus (10) reacts with

hydroxylamine in water at room temperature over 3.5 h to give the nitron (11) (72%).[†] When (11) is treated in chloroform at room temperature with *N*-methylmaleimide it gives, *via* a diastereofacially specific cycloaddition, a 3:1 mixture (100%) of (12) and (13). In boiling toluene however, the product is predominantly (> 90%) (12). Proton H_A in (11) gives rise to an n.m.r. multiplet at δ 4.27 due to coupling to the two adjacent methylene groups and long-range coupling to the C(Me) group. The coupling of H_A to the ring methylene protons H_B and H_C ($J_{AB} + J_{AC}$) is 14 Hz, showing that H_A is pseudoaxial.

The ketone (14) similarly reacts with hydroxylamine to give a 3.3:1 mixture (58%) of nitrones (15) and (16). This mixture on treatment with *N*-methylmaleimide in boiling chloroform

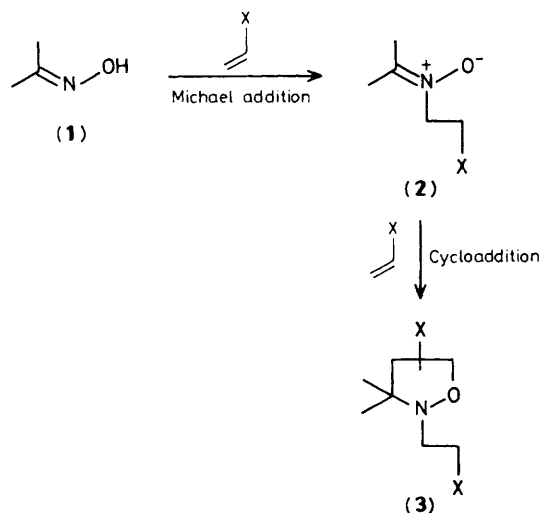
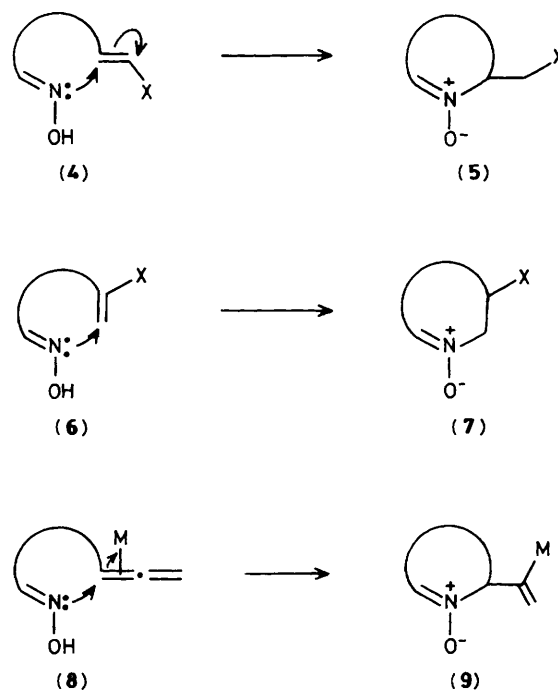
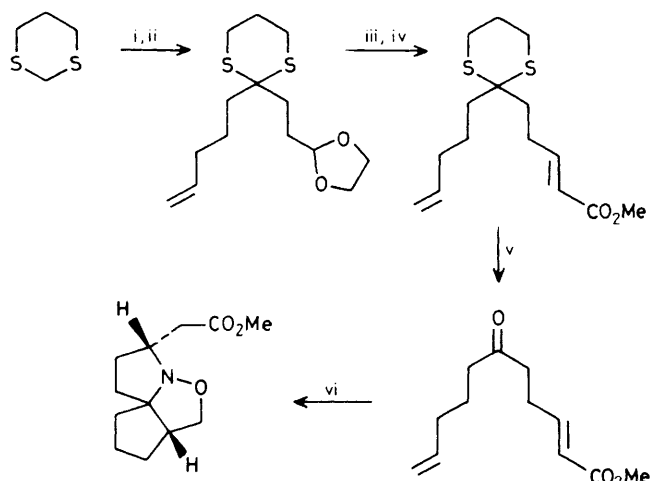
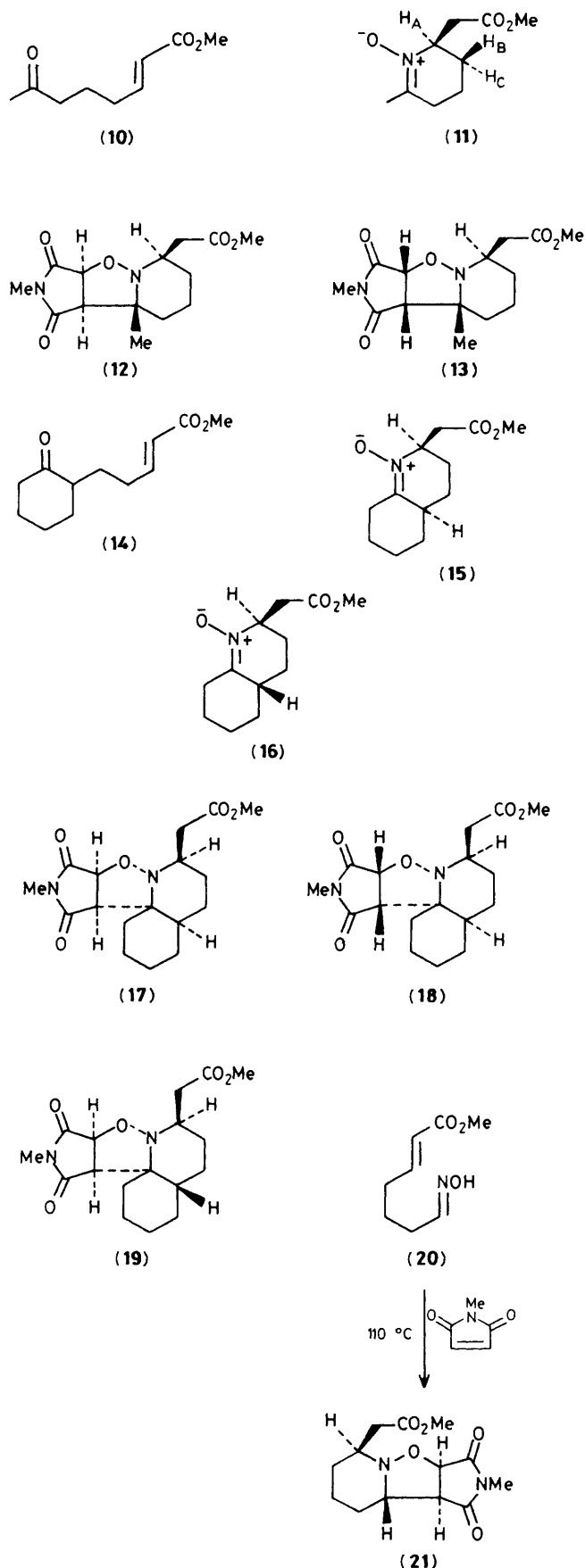


Table 1. Synthetic variants of the two-step tandem oxime cycloaddition process.

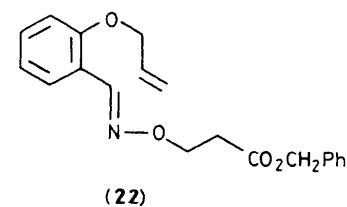
Class	Michael addition	Cycloaddition
1	Intermolecular	Intermolecular
2	Intermolecular	Intramolecular
3	Intramolecular	Intermolecular
4	Intramolecular	Intramolecular



[†] All new compounds gave satisfactory analytical and spectroscopic data.



Scheme 1. Reagents: i, Bu^nLi , THF, -78°C $\text{H}_2\text{C}=\text{CH}[\text{CH}_2]_3\text{Br}$; ii, Bu^nLi , $\text{PO}(\text{NMe}_2)_3$, -78°C , 1,3-dioxolan-2-ylethyl bromide; iii, 1M HCl/tetrahydrofuran, 60°C , 3 h; iv, $\text{Ph}_3\text{P}=\text{CHCO}_2\text{Me}$, CH_2Cl_2 , 25°C ; v, *N*-chlorosuccinimide, AgNO_3 , aq. MeCN, 15 min; vi, $\text{NH}_2\text{OH}\cdot\text{HCl}$, NaOAc, wet xylene, 140°C , 16 h.



gives a 4:3:2 mixture (100%) of (17)—(19) via a diastereofacially specific cycloaddition. The one-pot reaction is exemplified by the reaction of (20) with *N*-methylmaleimide in toluene (110°C ; 3 h) to give (21) in 90% yield.

The examples involving (10), (14), and (20) generate nitrones via *exo-trig* processes. Examples of nitron formation by the *endo-trig* Michael addition process (6) \rightarrow (7) are under study.

Class 4 reactions are exemplified by the sequence shown in Scheme 1. All steps occur in $> 80\%$ yield and in the final cyclisation step (82% yield) it is convenient to generate the oxime *in situ*.

We have briefly explored one further aspect of the mechanism of these tandem Michael addition–1,3-dipolar cycloaddition processes. The possible intervention of an initial *O*-Michael adduct, e.g. (22), was studied. Heating (22) at 140°C for 24 h in xylene failed to give any cycloadduct. In cases where the oxime is not isolated, the possibility of initial Michael addition of hydroxylamine to the α,β -unsaturated ester/sulphone exists.

The cycloadditions reported herein are illustrative of a large number of such reactions we have carried out. The ready reductive cleavage of the N–O bond of isoxazolidines and the potential for further synthetic manipulations involving the $\text{NCH}_2\text{CH}_2\text{X}$ moiety augur well for application of this methodology to natural product synthesis.

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References

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