

The Late Stages of the Biosynthesis of the Diterpenoid Aphidicolin in *Cephalosporium aphidicola*

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Labelling studies (¹⁴C and ¹⁸O) have shown that the major biosynthetic pathway leading to the diterpenoid aphidicolin involves the aphidicolan-16 β -ols whilst the results of ²H-labelling studies are consistent with a minor pathway involving epoxidation of an aphidicol-16-ene and hydrolysis of the epoxide.

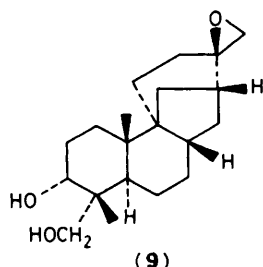
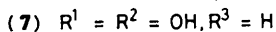
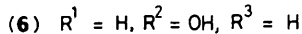
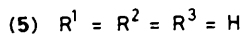
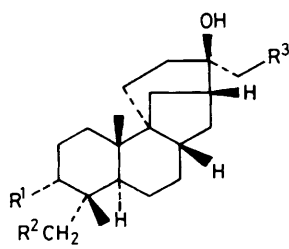
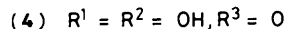
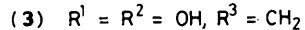
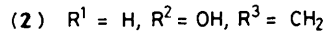
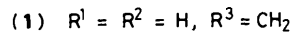
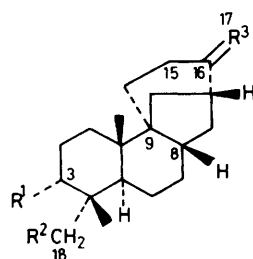
The tetracyclic diterpenoid fungal metabolite aphidicolin (**8**) has attracted interest¹ as a specific inhibitor of DNA polymerase α . In our previous biosynthetic work² we have defined the constituent isoprene units using the acetate and mevalonate labelling pattern and we have shown that a hydrogen shift from C-9 to C-8 occurs during the biosynthesis of this variant of the tetracyclic diterpenoid carbon skeleton. More recently we have shown³ that aphidicol-16-ene (**1**) and aphidicolan-16 β -ol (**5**) are incorporated to the extent of 0.09 and 7.9%, respectively. Since both aphidicol-16-ene and aphidicolan-16 β -ol are formed by the fungus,^{1,3} these results suggested that there might be a major pathway involving the 16 β -alcohols and a minor route through the 16-enes. Both intermediates could arise through a common carbocation generated during the cyclization.

In support of this we have now shown that 18-hydroxy-[17-¹⁴C]-aphidicol-16-ene (**2**) and 3 α ,18-dihydroxy[17-¹⁴C]-aphidicol-16-ene (**3**)[†] are specifically incorporated into aphidi-

colin (**8**) to the extent of 0.86 and 16.4%, respectively by *Cephalosporium aphidicola*. On the other hand 16 β ,18-dihydroxy[17-¹⁴C]aphidicolane (**6**) and 3 α ,16 β ,18-trihydroxy-[17-¹⁴C]aphidicolane (**7**) are incorporated to the extent of 20.5 and 52.6% respectively.

In accordance with the major pathway being that in which the 16 β -alcohols are formed by hydration of a carbocation, aphidicolin (**8**) biosynthesized in the presence of 20% H₂¹⁸O has been shown by mass spectrometry to incorporate an oxygen-18 label at C-16. The chemical ionization (ammonia) mass spectrum showed an ion at 356 a.m.u., (*M* + NH₄)⁺, and an additional ion at 358 a.m.u. in the spectrum of the labelled product. Deuterium labelling showed that the significant 307 a.m.u. ion in the electron-impact mass spectrum of aphidicolin represents an *M*-CH₂OH fragment derived from the loss of C-17. This 307 a.m.u. ion also retained the oxygen-18 label. The 17-nor-16-ketone (**4**), prepared by oxidation with periodic acid, had lost the label; thus the oxygen-18 must have been at C-16. Because the 16-ene pathway is a relatively minor route, this result does not necessarily provide any information on that pathway.

[†] The preparation of the labelled intermediates will be described in our full paper.



(9)

The conversion of the aphidicol-16-enes into the 16,17-glycols could proceed by hydration followed by hydroxylation. Alternatively the route may involve epoxidation and hydro-

lysis. Both pathways have precedent in terpenoid and steroid biosynthesis.⁴ If a 17-deuterio-16-ene is used as a substrate, the hydration/hydroxylation route would, in contrast to the epoxidation/hydrolysis pathway, involve some loss of label. 3 α ,18-Dihydroxy[15-²H,17,17-²H₂]aphidicol-16-ene (3) was fed to *C. aphidicola*. ²H N.m.r. spectroscopy of the resultant aphidicolin (8) showed that the ratio of the 15-²H and 17-²H signals remained constant, in support of an epoxidation and hydrolysis pathway. 3 α ,18-Dihydroxy-16 β ,17-epoxyaphidicolane (9) was incorporated into aphidicolin (8) by *C. aphidicola* to the extent of 42% although a control experiment showed that this was accompanied by some non-enzymic hydrolysis (ca. 30%). Nevertheless, these experiments between them establish the role of an epoxide in the minor pathway and an efficient hydration/hydroxylation route from the initial cyclization as the major pathway.

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- 4 For a review see H. L. Holland, *Chem. Soc. Rev.*, 1982, **11**, 371.