

A Concise Synthesis of 4-Unsubstituted Azetidion-2-ones†

Fernando P. Cossío, Begoña Lecea, and Claudio Palomo*

Departamento de Química Orgánica, Facultad de Ciencias Químicas, Universidad del País Vasco, Ap. 1072, 20080 San Sebastián, Spain

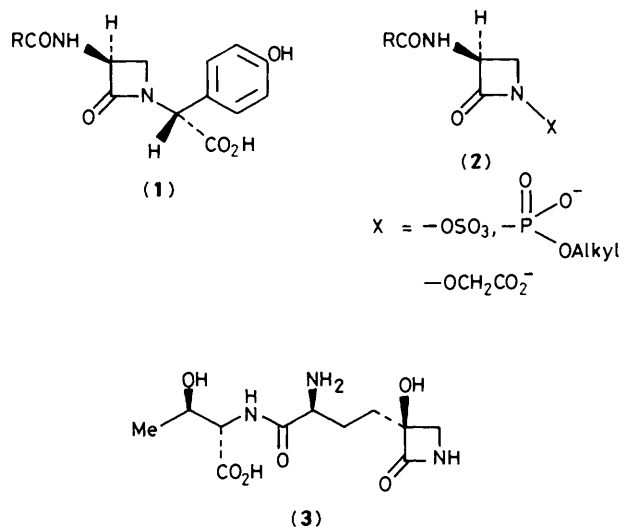
On catalysis by trimethylsilyl trifluoromethanesulphonate, 4-acetoxy- β -lactams react with hydrosilanes to give 4-unsubstituted- β -lactams in good to excellent yields.

β -Lactam antibiotics are of current widespread interest because of their applications.¹ Monolactams such as nocardicins (1), monobactams (2), and tabtoxin (3) are characterized by the absence of substituents at the 4-position of the β -lactam ring. Despite numerous suitable methods for the synthesis of β -lactams, *e.g.* the annelation of an imino compound with an activated acetic acid is a versatile procedure for the construction of the azetidione ring,² the inaccessibility of monomeric formaldehyde imines³ has necessitated the development of new strategies for the preparation of monocyclic 4-unsubstituted β -lactams.⁴

We first prepared 4-acetoxy- β -lactams by the acetic acid-imine approach⁵ and since the acetoxy group is readily removed by several reagents⁶ we investigated the conversion of 4-acetoxy- β -lactams into 4-unsubstituted β -lactams (see Scheme 1).

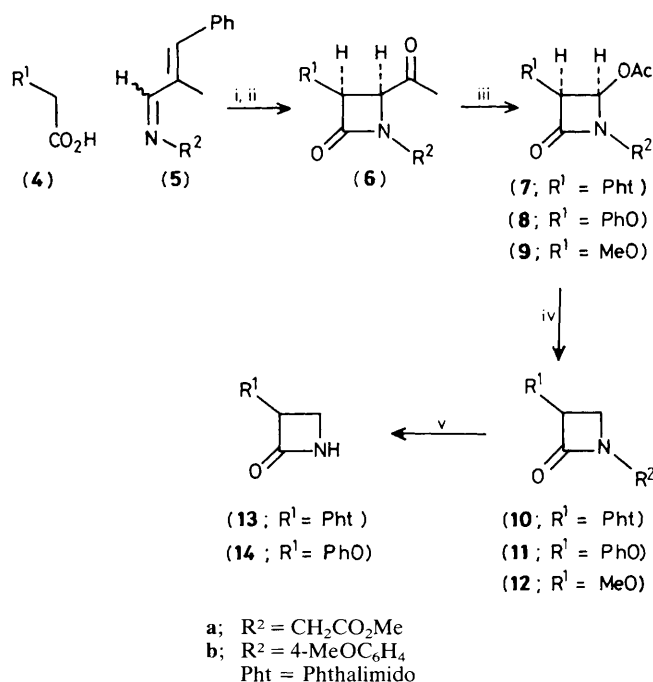
We found that the reaction between a 4-acetoxyazetidion-2-one and a sixfold excess of a hydrosilane, in the presence of a catalytic amount of trimethylsilyl trifluoromethanesulphonate (triflate), cleanly afforded a 4-unsubstituted β -lactam in good yield. Of the hydrosilanes examined, *i.e.* triethylsilane, polymethylhydrosiloxane, and tetramethyldisiloxane, only

tetramethyldisiloxane gave satisfactory results.‡ Of the catalysts zinc iodide, boron trifluoride, and trimethylsilyl triflate, only trimethyl triflate proved to be effective.



† For Part 62 of the series Reagents and Synthetic Methods see F. P. Cossío, I. Ganboa, J. M. García, B. Lecea, and C. Palomo, *Tetrahedron Lett.*, 1987, **28**, 1945.

‡ See R. D. G. Cooper in 'Topics in Antibiotic Chemistry', Vol. 3, ed. P. G. Sammes, Ellis Horwood Limited, Chichester, 1980, p. 156.



Scheme 1. Reagents and conditions: i, PhOPOCl_2 , NEt_3 , CH_2Cl_2 , room temp., 20–24 h; ii, O_3 , -78°C , CH_2Cl_2 , then Me_2S ; iii, *m*-chloroperbenzoic acid, C_6H_6 , reflux; iv, $\text{HMe}_2\text{SiOSiMe}_2\text{H}$, C_6H_6 , $\text{F}_3\text{CSO}_3\text{SiMe}_3$, reflux; v, cerium(IV) ammonium nitrate, $\text{MeCN}-\text{H}_2\text{O}$, ref. 7.

A representative procedure is as follows. A mixture of the 4-acetoxyazetidin-2-one (**7a**)⁵ (1 mmol), 1,1,3,3-tetramethyldisiloxane (6 mmol), and trimethylsilyl triflate (two drops) in benzene (5 ml) was refluxed for 57 h under nitrogen. The reaction mixture was evaporated and the waxy residue was treated with methanol. The solid product was filtered off affording (**10a**) in 70% yield: 60 MHz ^1H n.m.r. δ (CDCl_3) 8.02–7.33 (m, 4H, ArH), 5.47 (d,d, J 4.5 Hz, J' 3 Hz, 1H, CH), 4.33 (d, J 18 Hz, 1H, CH), 3.90 (d, J 18 Hz, 1H, CH), 3.95–3.58 (m, 2H, CH_2), 3.75 (s, 1H, OMe). Further examples are listed in Table 1. The β -lactams (**10b**)–(**12b**) were easily transformed into their corresponding *N*-unsubstituted β -lactams; thus removal of the *p*-methoxyphenyl group⁷ in (**10b**) led to (**13**) in 94% yield, m.p. 199–200 $^\circ\text{C}$ (decomp.); similarly (**14**), m.p. 109–110 $^\circ\text{C}$ (decomp.), was obtained in 86% yield. Using this synthesis diverse 4-unsubstituted β -lactams should be readily available.

Table 1. 4-Unsubstituted β -lactams prepared.

Compound ^a	<i>t</i> /h	Yield (%)	m.p./ $^\circ\text{C}$ ^b
(10a)	57	70	164–165
(10b)	15	95	226–228
(11a)	25	53	— ^c
(11b)	15	96	112–114
(12b)	15	70	74–75

^a All compounds were racemic mixtures and gave satisfactory spectral and analytical data. ^b From EtOH unless noted otherwise. ^c Colourless oil isolated by column chromatography. ^d From AcOEt–hexane.

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References

- W. Dürckheimer, J. Blumbach, R. Latrell, and K. H. Scheunemann, *Angew. Chem., Int. Ed. Engl.*, 1985, **24**, 180.
- For reviews see: A. K. Mukerjee and R. C. Srivastava, *Synthesis*, 1973, 373; A. K. Bose and M. S. Manhas, *Lect. Heterocycl. Chem.*, 1976, **3**, 43; N. S. Isaacs, *Chem. Soc. Rev.*, 1976, 181; A. K. Mukerjee and A. K. Singh, *Tetrahedron*, 1978, **34**, 1731.
- For recent methods for generating these intermediates *in situ* see: L. E. Overman and R. M. Burk, *Tetrahedron Lett.*, 1984, **25**, 1635; J. Barluenga, A. M. Bayon, and G. J. Asensio, *J. Chem. Soc., Chem. Commun.*, 1983, 1109; J. Barluenga, A. M. Bayan, and G. J. Asensio, *ibid.*, 1984, 427.
- T. Kamiya, H. Hashimoto, O. Nakaguchi, and T. Oku, *Tetrahedron*, 1979, **35**, 323; K. Ikeda, Y. Terao, and M. Sekiya, *Chem. Pharm. Bull.*, 1981, **29**, 1747; G. H. Hakimelahi, *Helv. Chim. Acta*, 1982, **65**, 1378; C. A. Townsend, G. M. Salituro, L. T. Nguyen, and M. J. DiNovi, *Tetrahedron Lett.*, 1986, **27**, 3819 and references cited therein; S. T. Hudgson, D. M. Hollinshead, S. V. Ley, C. M. R. Low, and D. J. Williams, *J. Chem. Soc., Perkin Trans I*, 1985, 2375 and references cited therein; L. E. Overman and T. Osawa, *J. Am. Chem. Soc.*, 1985, **107**, 1698 and references cited therein.
- J. M. Aizpurua, F. P. Cossío, B. Lecea, and C. Palomo, *Tetrahedron Lett.*, 1986, **27**, 4359.
- For review see: T. Kametani, *Heterocycles*, 1982, **17**, 463.
- D. R. Kronenthal, C. Y. Han, and M. K. Taylor, *J. Org. Chem.*, 1982, **47**, 2765.