## Bis-organoimido Complexes of Tungsten(vi). The Crystal and Molecular Structure of Bipyridyldichlorobis(phenylimido)tungsten(vi), [WCl<sub>2</sub>(NPh)<sub>2</sub>(bipy)]

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Bis-organoimido complexes of tungsten(v<sub>1</sub>) are prepared by reaction of WCl<sub>6</sub> with Me<sub>3</sub>SiNHCMe<sub>3</sub>, or by reaction of mono-organoimido complexes with the silylamines Me<sub>3</sub>SiNHR; the structure of [WCl<sub>2</sub>(NPh)<sub>2</sub>(bipy)] (bipy =  $\alpha$ , $\alpha$ -bipyridyl) has been established by X-ray crystallography.

Preparative routes leading to bis-organoimido complexes,  $[M(NR)_2]$ , are less well developed than those for monoorganoimido complexes. We have previously shown that for  $MCl_5$  molecules, (M = Nb, Ta), substitution of two chlorides by alkylamido ligands, NHR, results in formation of an alkylimido ligand and co-ordinated primary amine [equation (1)] by an interligand proton-transfer process which is independent of steric size. A third alkylamido ligand could be substituted [equation (2)] but the complex could not be induced to eliminate HCl to form a second imido function. We have now extended these reactions to establish whether two alkylimido ligands can be generated from an MCl<sub>6</sub> molecule and, if so, whether steric size is important.

$$MCl_5 + 2Me_3SiNHR \rightarrow [MCl_3(NR)(NH_2R)] + 2Me_3SiCl (1)$$

$$[MCl3(NR)(NH2R)] + 2RNH2 \rightarrow [MCl2(NR)(NHR)(NH2R)] + RNH3Cl (2)$$

Reaction of WCl<sub>6</sub> with two equivalents of Me<sub>3</sub>SiNHCMe<sub>3</sub> benzene gives a yellow product analysing as [WCl<sub>4</sub>(NHCMe<sub>3</sub>)<sub>2</sub>] (1) but <sup>13</sup>C n.m.r. spectroscopy shows it to be a mixture of (1) and  $[WCl_4(NCMe_3)(NH_2CMe_3)]$  (2). Reaction of the mixture with a further two equivalents of Me<sub>3</sub>SiNHCMe<sub>3</sub>, or reaction of WCl<sub>6</sub> with 4 or more equivalents of Me<sub>3</sub>SiNHCMe<sub>3</sub> gives [WCl<sub>2</sub>(NCMe<sub>3</sub>)<sub>2</sub>(NH<sub>2</sub>CMe<sub>3</sub>)]<sub>x</sub> (3) indicated by analytical and <sup>13</sup>C n.m.r. spectral data. A single absorption at 210 cm<sup>-1</sup> in the i.r. spectrum indicates trans chloro ligands. Complex (3) reacts with y-picoline (pic) or  $\alpha, \alpha$ -bipyridyl (bipy) to form [WCl<sub>2</sub>(NCMe<sub>3</sub>)<sub>2</sub>(pic)<sub>2</sub>] (4) and [WCl<sub>2</sub>(NCMe<sub>3</sub>)<sub>2</sub>(bipy)] (5) for which i.r. data indicate trans chlorine atoms, while reaction of (3) with PMe<sub>3</sub> gives  $[WCl_2(NCMe_3)_2(PMe_3)]_x$  (6), for which absorptions at 218 and 180 cm<sup>-1</sup> indicate cis chlorine atoms. Complex (3) reacts with 4 equivalents of t-butylamine in light petroleum (b.p. 40-60°C) to give the known complex  $[W(NCMe_3)_2(NHCMe_3)_2]$  (7),<sup>3</sup> which we have as yet been unable to convert into a tris-imido complex.

$$[WCl_4(NHCMe_3)_2] \qquad [WCl_2(NR)_2(bipy)]$$
 (1) (5)  $NR = NCMe_3$  (9)  $NR = NPh$  
$$[WCl_4(NCMe_3)(NH_2CMe_3)] \qquad [WCl_2(NCMe_3)_2(PMe_3)]_x$$
 (6) 
$$[WCl_2(NCMe_3)_2(NH_2CMe_3)]_x \qquad [W(NCMe_3)_2(NHCMe_3)_2]$$
 (3) (7) 
$$[WCl_2(NCMe_3)_2(pic)_2] \qquad [WCl_2(NCMe_3)(NPh)(PMe_3)_2]$$
 (4) (8)

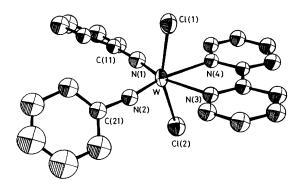
The silvlamines  $Me_3SiNHR$  (R = CHMe<sub>2</sub>, Et) do not react cleanly with WCl<sub>6</sub> or the mixture of (1) and (2). However, the mono-organimido tungsten(vi) complexes  $[WCl_4(NR)]_2$  (R = Ph, CHMe<sub>2</sub>, Me), prepared from  $[WCl_4(O)]_2$  and aryl or alkyl isocyanates,  $^4$  react with the silylamines Me<sub>3</sub>SiNHR' (R' = Ph, PhMe, CMe<sub>3</sub>, CHMe<sub>2</sub>, CH<sub>2</sub>Me) in benzene to give the bis-organoimido complexes [WCl<sub>2</sub>(NR)(NR')(NH<sub>2</sub>R')]<sub>2</sub>, for which i.r. absorptions in the vicinity of 310 and 270 cm<sup>-1</sup> indicate cis-metal dichloride ligands. Bridging and terminal phenylimido ligands are characterised in the <sup>13</sup>C n.m.r. spectra by ipso carbon resonances at δ 162 and 151 respectively but the  $\alpha$ -carbon resonance position does not distinguish these for the bis alkylimido dimers. However the least sterically hindering organoimido ligand is expected to form the bridge. An X-ray crystal structure determination of  $[WCl_2(NCMe_3)(\mu-NPh)(NH_2CMe_3)]_2^5$  has shown bridging NPh and terminal NCMe<sub>3</sub> ligands.

The dimers do not react further with primary amines to give complexes similar to (7), but they may be used to form  $[WCl_2(NR)(NR')L_2]$  complexes which cannot be prepared via the reaction of the dioxo species [WCl<sub>2</sub>(O)<sub>2</sub>L<sub>2</sub>] with aryl or alkyl isocyanates. Thus reaction of [WCl<sub>2</sub>(NCMe<sub>3</sub>)(µ-NPh)- $(NH_2CMe_3)_2$  with PMe<sub>3</sub> gives  $[WCl_2(NCMe_3)(NPh)(PMe_3)_2]$ (8) which contains trans orientated phosphines as indicated by a single, virtually coupled triplet for the PMe<sub>3</sub> ligands in both the <sup>13</sup>C{<sup>1</sup>H} and <sup>1</sup>H n.m.r. spectra and by a singlet (9.6 p.p.m., external H<sub>3</sub>PO<sub>4</sub>) in the <sup>31</sup>P{<sup>1</sup>H} spectrum. Reaction gives the complexes dimers with bipy  $[WCl_2(NR)(NR')(bipy)]$  for which a single band in the far i.r. spectrum indicates the metal dichlorides are now orientated trans. This isomerisation is confirmed by an X-ray crystal structure determination of [WCl<sub>2</sub>(NPh)<sub>2</sub>(bipy)] (9).†

The asymmetric unit contains two crystallographically distinct, but structurally similar, molecules. The structure of one of these molecules is depicted in Figure 1 which shows a distorted octahedral geometry about tungsten, with two cis-orientated phenylimido groups, trans chloro ligands, and the nitrogen atoms of the bipyridyl ring co-ordinated trans to the imido functions. The four separate determinations of the imido W-N bond lengths and W-N-C bond angles are not

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<sup>†</sup> Crystal data for (7):  $C_{22}H_{18}Cl_2N_4W$ , M=593.15, triclinic, space group  $P\overline{1}$ , a=14.616(6), b=17.778(6), c=8.532(2) Å,  $\alpha=94.71(2)$ ,  $\beta=100.41(3)$ ,  $\gamma=84.31(3)^\circ$ , U=2165.1 ų, F(000)=1144,  $D_c=1.819$  g cm<sup>-3</sup>, Z=4,  $\mu(\text{Mo-}K_{\alpha})=64.2$  cm<sup>-1</sup>. Intensity data were recorded on a Nonius CAD4 diffractometer with Mo- $K_{\alpha}$  radiation and corrected for Lorentz, polarisation, and absorption effects. The structure was solved from Patterson and heavy-atom electron density syntheses and refined by full-matrix least-squares, omitting hydrogens, with tungsten and chlorine atoms anisotropic, to an R value of 0.048 for 3357 independent reflections for which  $I>3\sigma(I)$ . Atomic co-ordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1.



**Figure 1.** Molecular structure of [WCl<sub>2</sub>(NPh)<sub>2</sub>(bipy)] (9). Important averaged bond lengths (Å) and bond angles (°): W-Cl(1) 2.390(4), W-Cl(2), 2.385(4), W-N(1) 1.782(8), W-N(2) 1.775(9), W-N(3) 2.319(8), W-N(4) 2.315(8); W-N(1)-C(11) 165.3(8), W-N(2)-C(21) 166.3(8), N(1)-W-N(2) 104.3(4), Cl(1)-W-Cl(2) 159.3(1).

significantly different [mean values 1.789(6) Å and 165.9(6)° respectively]. Electronically this represents delocalisation of 6 donor electrons over the two phenylimido ligands and thus an 18 electron count for the complex is maintained.

Bis-organoimido complexes of tungsten(vI) may thus be prepared by reactions involving proton transfer from one primary alkylamido group to another. The process is indepen-

dent of steric size, which contrasts with reactions of high-valent early transition metals where, with less acidic  $\alpha$ -hydrogens, metal-alkyl ligands require steric bulk before forming an alkylidene ligand. The new complexes are organoimido analogues of the well-known *cis*-dioxo tung-sten(v1) complexes, and the reaction sequences described herein allow their formation for a variety of alkyl and aryl groups.

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