

Muonic Alkyl Radicals with α - or β -Phosphorus Substituents: $(\text{EtO})_2\text{P}(\text{O})\dot{\text{C}}\text{HCH}_2\text{Mu}$ and $\text{H}_2\dot{\text{C}}\text{CH}(\text{Mu})\text{P}(\text{O})(\text{OEt})_2$

Christopher J. Rhodes,^a Martyn C. R. Symons,^{*a} Christopher A. Scott,^b Emil Roduner,^c and Martin Heming^c

^a Department of Chemistry, The University, Leicester LE1 7RH, U.K.

^b Rutherford Appleton Laboratory, Chilton, Didcot, Oxon. OX11 0QX, U.K.

^c Physikalisches-Chemisches Institut, Zürich University, CH-8057, Zürich, Switzerland

The magnitudes of the muon–electron hyperfine coupling constants for the title radicals show that the muonium atom favours the out-of-plane site for the α -P substituted radical, but is constrained towards the in-plane site for the β -P substituted radical.

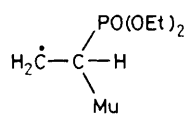
While many alkyl radicals containing muonium atoms β - to the trigonal carbon are known,¹ there has been little work on muonic radicals containing hetero-atoms, especially 'heavy-atom' substituents,² and we know of no previous work on muonic phosphorus-containing radicals.

We have succeeded in detecting both radicals (1) and (2) (Figure 1), the results making an interesting contrast. The hyperfine coupling for (1) is less than $a'(\text{Mu})$ for $\dot{\text{C}}\text{H}_2\text{CH}_2\text{Mu}$ at room temperature^{1,3} [*ca.* 37.2 G (1 G = 10^{-4} T): $a'(\text{Mu})$ = the true coupling divided by 3.184, the ratio of the muon and proton magnetic moments], whereas that for (2) is close to that for muonic ethyl.

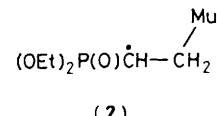
The ratio of the muon coupling in $\text{H}_2\dot{\text{C}}\text{CH}_2\text{Mu}$ to the methyl proton coupling in $\text{H}_2\dot{\text{C}}-\text{CH}_3$ ¹ is *ca.* 1.4. This large isotope effect is, at least in part, caused by the muonium atom exhibiting a preference for the 'out-of-plane' site. For radical (2), this remains the case, the α -P(O)(OEt)₂ group having no specific effect on the muon coupling. However, for radical (1), the muon coupling is clearly reduced relative to $\text{H}_2\dot{\text{C}}\text{CH}_2\text{Mu}$,

suggesting that the P(O)(OEt)₂ group dominates the conformation in this case.

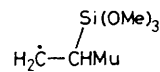
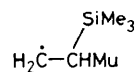
It is well established, both from the large hyperfine coupling to ³¹P nuclei in β -positions,⁴ and the small couplings to



$$a'(\text{Mu}) = 29.88 \text{ G}$$



$$a'(\text{Mu}) = 37.03 \text{ G}$$



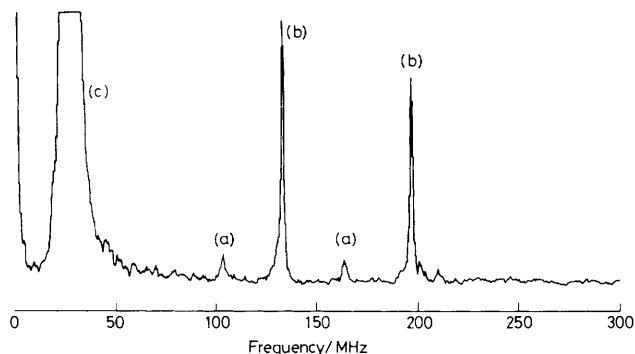


Figure 1. Muon spin rotation (μ .s.r.) spectrum recorded during muon implantation in $\text{CH}_2=\text{CHP}(\text{O})(\text{OEt})_2$, showing features assigned to: (a) $(\text{EtO})_2\text{P}(\text{O})\text{CHMu}\dot{\text{C}}\text{H}_2$ and (b) $\text{CH}_2\text{Mu}\dot{\text{C}}\text{HP}(\text{O})(\text{OEt})_2$; (c) is the signal from muons in diamagnetic environments.

β -protons⁴ in normal protiated radicals of type (1), that the preferred conformation is indeed that in which the σ - π overlap for the C-P bond is maximised.

These results are in line with those recently obtained² for radicals (3) and (4). In both these radicals, $a'(\text{Mu})$ is less than that in $\dot{\text{C}}\text{H}_2\text{CH}_2\text{Mu}$, and falls on cooling, in contrast with the result for $\text{H}_2\dot{\text{C}}\text{CH}_2\text{Mu}$ where the coupling increases on cooling.^{1,3} Thus, again, we conclude that the heavy atom substituents (SiR_3) dominate over the bound muonium atom for the conformational control of these radicals. However,

when compared with the corresponding protiated species,^{2,5} there is still a large muon isotope effect (*ca.* 1.4) for these radicals, suggesting that the mean conformations are still skewed in favour of the muon to some extent.

Finally, we note that the muon hyperfine coupling for the $\text{H}_2\dot{\text{C}}\text{CHMuSi}(\text{OMe})_3$ radical (29.09 G) is slightly less than that for the phosphorus-containing radical, $\text{H}_2\dot{\text{C}}\text{CHMuP}(\text{O})(\text{OEt})_2$ (29.88 G) at room temperature. This fall on going from Si to P accords with the concept of conformational control through hyperconjugative electron transfer from the C-X bond, since this is expected to be stronger for C-Si than for C-P σ -electrons.

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References

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