

First Example of a 1,2,4-Triphosphabutadiene Complex: Synthesis and Crystal and Molecular Structure of $[\text{Co}(\eta^5\text{-C}_2\text{R}_2\text{P}_3)(\eta^4\text{-C}_2\text{R}_2\text{HP}_3)]$ ($\text{R} = \text{Bu}^t$)

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Treatment of CoCl_2 with $\text{Li}(\text{C}_2\text{R}_2\text{P}_3)$ ($\text{R} = \text{Bu}^t$) in 1,2-dimethoxyethane yields the complex $[\text{Co}(\eta^5\text{-C}_2\text{R}_2\text{P}_3)(\eta^4\text{-C}_2\text{R}_2\text{HP}_3)]$ containing a ligated 1,2,4-triphosphacyclopenta-1,3-diene ring system.

There is considerable current interest in the ligating behaviour of unsaturated organophosphorus ligands.¹ It has become increasingly apparent that formal replacement of a CH fragment by P or a CH_2 fragment by PR in organic compounds can lead to a variety of novel unsaturated organophosphorus compounds, many of which form complexes with transition metals. During the past few years the structural types (a)–(h) have been established (ML_n = transition metal and attendant ligands).^{2–14}

Very recently Mathey *et al.*¹⁵ described the first example of an η^4 -ligated 1-phosphabutadiene system (i) $\{\text{ML}_n = [\text{W}(\text{CO})_4]\}$ as its $[\text{W}(\text{CO})_5]$ adduct, formed *via* a coupling reaction of a vinyl carbene metal complex with the transient phosphinidene compound $[\text{W}(\text{CO})_5\text{PPh}]$. We now describe the first example of a 1,2,4-triphosphabutadiene metal complex of type (j), which further extends the range of ligated unsaturated phosphorus ligands.

Treatment of $\text{Li}(\text{C}_2\text{R}_2\text{P}_3)$ ($\text{R} = \text{Bu}^t$) in 1,2-dimethoxyethane with CoCl_2 gave a low yield (*ca.* 5%) of the red diamagnetic complex $[\text{Co}(\eta^5\text{-C}_2\text{R}_2\text{P}_3)(\eta^4\text{-C}_2\text{R}_2\text{HP}_3)]$ (1).[†] The $^{31}\text{P}\{^1\text{H}\}$ n.m.r. spectrum (Figure 1) exhibits two widely spaced ABX patterns for the two sets of three nonequivalent phosphorus nuclei in the two rings which have been fully analysed.[‡] A single crystal X-ray structural determination reveals the molecular structure shown in Figure 2, indicating that the complex should be formulated as an (η^5 -1,3,5-triphosphacyclopentadienyl)(η^4 -1,2,4-triphosphacyclopentadiene)cobalt derivative.[§]

The complex (1) may arise from the intermediate paramagnetic $[\text{Co}(\eta^5\text{-C}_2\text{Bu}_2\text{P}_3)_2]$ precursor by H abstraction from the solvent, since the yield is increased when the reaction is carried out in the presence of cyclopentadiene.

The η^5 -1,3,5-triphosphacyclopentadienyl ring is planar as expected and the bond distances and bond angles are similar to those previously described for $[\text{Fe}(\eta^5\text{-C}_2\text{Bu}_2\text{P}_3)_2]$ ¹⁴ and $[\text{Fe}(\eta^5\text{-C}_2\text{Bu}_2\text{P}_3)(\eta^5\text{-C}_5\text{H}_5)\text{W}(\text{CO})_5]$.¹⁶ The PPCP framework

of the η^4 -ligated 1,2,4-triphosphabutadiene fragment of the ($\text{C}_2\text{Bu}_2\text{HP}_3$) ring is also planar, and the C–P and P–P bond lengths (summarised in Figure 3) lie in the expected range for this unsaturated system.

Interestingly the H atom attached to the unligated carbon atom of the η^4 -($\text{C}_2\text{Bu}_2\text{HP}_3$) ring occupies an *endo*-position, but in contrast to analogous cyclopentadienylcyclopentadienecobalt complexes $[\text{Co}(\eta^5\text{-C}_5\text{H}_5)(\eta^4\text{-C}_5\text{H}_5\text{R})]$ ^{17,18} it has

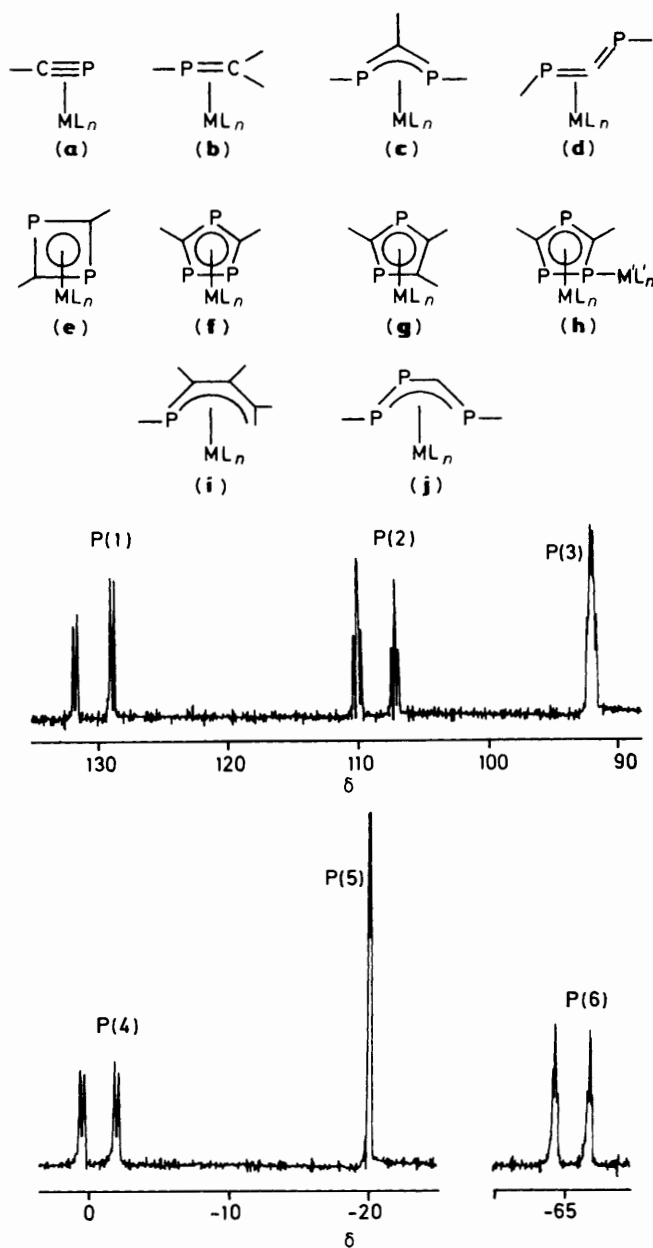


Figure 1. $^{31}\text{P}\{^1\text{H}\}$ N.m.r. spectrum of the complex (1).

[†] m/z 522 (M^+ , 40%) and 465 $\{[\text{Co}(\text{C}_2\text{Bu}_2\text{P}_3)(\text{C}_2\text{Bu}^t\text{HP}_3)]^+, 100\%$.

[‡] Selected spectroscopic data: ^{31}P δ (rel. H_3PO_4) P(1) +131.2, P(2) +109.5, P(3) +93.1, P(4) +0.2, P(5) –20.1, P(6) –64.7; $^1J_{\text{P}(1)\text{P}(2)}$ 419.7, $^2J_{\text{P}(1)\text{P}(3)}$ 41.6, $^2J_{\text{P}(2)\text{P}(3)}$ 40.2, $^1J_{\text{P}(4)\text{P}(6)}$ 360.6, $^2J_{\text{P}(5)\text{P}(6)}$ 24.5, $^2J_{\text{P}(4)\text{P}(6)}$ 24.5, $^2J_{\text{P}(2)\text{P}(4)}$ 39, $^2J_{\text{P}(3)\text{P}(6)}$ 24.2 Hz; ^1H δ 1.50 (Bu^t), 1.36 (Bu^t), 1.28 (Bu^t), 1.21 (Bu^t), and 0.8 (br., CH).

[§] Crystal data: $\text{C}_{20}\text{H}_{37}\text{CoP}_6$, $M = 522.3$, triclinic, space group $P\bar{1}$, $a = 10.489(10)$, $b = 10.631(10)$, $c = 12.573(8)$ Å $\alpha = 72.25(9)$, $\beta = 89.94(8)$, $\gamma = 79.65(9)^\circ$, $U = 1311.4$ Å³, $Z = 2$, $D_c = 1.32$ g cm^{–3}, monochromated Mo- K_α radiation, $\lambda = 0.71069$ Å, $\mu = 10.2$ cm^{–1}. The structure was solved by routine heavy-atom methods and refined by full-matrix least-squares with non-hydrogen atoms anisotropic using 2953 reflections with $I > \sigma(I)$ measured with an Enraf-Nonius CAD4 diffractometer. The hydrogen atom on C(3) was located on a difference map and refined isotropically. The final residuals were $R = 0.088$, $R' = 0.107$, with 248 parameters refined. The rather high R factor is a consequence of the weak diffraction from the crystal. Atomic co-ordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1.

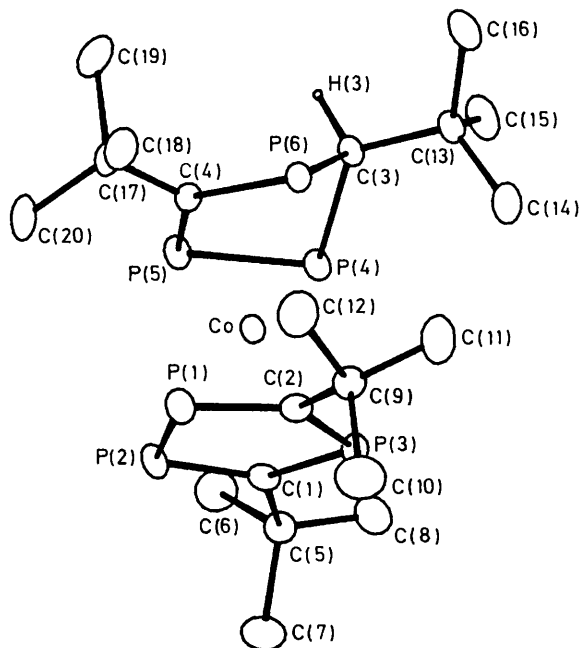
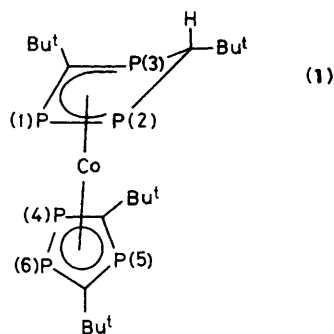


Figure 2. Molecular structure of complex (1).

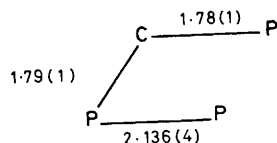


Figure 3. Bond lengths for the PPCP system in the complex (1).

not yet been possible to generate the $[\text{Co}(\eta^5\text{-C}_2\text{Bu}_2\text{P}_3)_2]^+$ cation {which would be isoelectronic with $[\text{Fe}(\eta^5\text{-C}_2\text{Bu}_2\text{P}_3)_2]$ }

by treatment of (1) with either CCl_4 or $\text{Ph}_3\text{C}^+\text{BF}_4^-$, presumably because of steric factors.

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