Diene-Dienophile-functionalized Carbene Ligands: Synthons for the Intramolecular Diels-Alder Reaction¹

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The diene–dienophile-substituted carbene complex pentacarbonyl(1-furfurylamino-2-methylprop-2-enylidene) tungsten undergoes a stereospecific intramolecular Diels–Alder reaction leading to a *trans*-fused hexahydroisoindolidene complex, the crystal structure of which has been determined.

The isolobal relation^{2,3} has become an efficient basis for the use of transition metal complexes in organic synthesis. This principle has also been applied to Fischer-type carbene complexes which are known to react as metal-tuned carbonyl compounds.⁴ For instance, the analogy of acrylates and alkoxy(vinyl)carbene complexes is demonstrated by facile

[4 + 2] cycloaddition reactions.⁵⁻⁷ Since Diels-Alder reactions, especially when carried out in an intramolecular fashion, provide a powerful tool in selective carbon-carbon bond formation, in, for example highly stereoselective syntheses of natural products,^{8.9} we have focused on intramolecular Diels-Alder reactions effected at a metal carbene



Scheme 1. Reagents and conditions: a, hexane, -20 °C, 3 h, 93%; b, toluene, 80 °C, 24 h, 68%; c, furfuryl alcohol/sodium, ether/pentane, -70 °C to room temp., 2 h, 39%; d, furfuryl thiol/potassium, ether, -5 °C, 5 min, 87%.



Figure 1. Molecular structure of (3) (ORTEP, thermal ellipsoids 50%, hydrogen atoms with arbitrary radii). Important distances (Å) and angles (°): W–C(1) 2.227(4), C(1)–C(2) 1.531(6), C(1)–N(1) 1.298(6), N(1)–C(8) 1.477(6); W–C(1)–C(2) 126.2(3), W–C(1)–N(1) 126.7(3), C(2)–C(1)–N(1) 107.1(4). Plane angles (°): C(1), C(2), N(1), C(8)/C(2), C(7), C(8) 30.1; C(7), C(2), C(3), C(4)/C(4), O(1), C(7) 118.6; C(4), C(5), C(6), C(7)/C(4), O(1), C(7) 128.5. Torsion angle (°): C(1)–C(2)–C(7)–C(8) 29.3.

template. This strategy is of general potential utility owing to the broad compatibility with group functionalization both in the carbene ligand and in the metal-coligand sphere.

We have used a two-step procedure which is based on the addition of nucleophiles to a carbene ligand and then to the carbonyl ligand formed therefrom. The propenylidene complex (1), obtained from a sequential addition of 2-lithiopropene and trimethyloxonium tetrafluoroborate to hexacarbonyltungsten reacts with furfurylamine to give a stereoisomeric mixture (which is not separable by chromatography) of aminocarbene complexes (2a) and (2b) (45:55). While the isomer (2b) remains unchanged upon warming of a toluene solution to 80 °C, the isomer (2a) undergoes cycloaddition to give (3)^{\dagger} (Scheme 1). In contrast, secondary amides analogous to (2), *e.g.* (A), fail to react. Tertiary amides such as (B) are required for Diels-Alder reactions;¹⁰ the resulting lactams, however, are prone to cycloreversion.¹¹

The X-ray analysis[‡] of the cycloadduct (3) confirms the *exo*-mode of cyclization (Figure 1). A similar stereochemistry has been suggested for analogous tertiary amides on the basis of n.m.r. spectroscopy.¹⁰ The five-membered ring containing the carbene carbon atom adopts an envelope conformation with C(7) being 0.47 Å above the best plane of the other four

† Selected spectroscopic data (numbering system according to I.U.P.A.C. nomenclature): (3): i.r.: v_{co} (ether) 2063m, 1968w, 1932vs, and 1916vs cm⁻¹; ¹H n.m.r. δ (C₆D₆) 7.19 (br., NH), 5.77 (dd, =CH), 5.64 (d, =CH), 4.35 ('dd, 'CH), 2.83 (dd, CH₂), 2.42 (dd, CH₂), 2.61 (d, NCH₂), 1.01 (d, NCH₂), and 0.58 (s, CH₃); ¹³C n.m.r. δ (C₆D₆) 263.05 (C_{carbene}), 202.52 (CO_{trans}), 198.72 (CO_{cis}), 138.44 and 131.42 (=C), 92.83 (7a-C), 79.01 (5-C), 73.09 (3a-C), 54.88 (1-C), 38.75 (4-C), and 21.85 (CH₃).

(4): v_{co} (hexane) 2069w, 1952vs, and 1942s cm⁻¹; ¹H n.m.r. δ (C₆D₆) 6.70 (d, =CH), 5.60 (m, =CH), 3.95 (s, OCH₃), 3.92 (s, OCH₂), 3.13 (d, OCH₂), 2.23 (t, CH), 1.28 (d, CH₃).

(5): v_{co} , (hexane) 2074w and 1943vs.(br.) cm⁻¹; ¹H n.m.r. δ (C₆D₆) 7.00 (d, =CH), 5.95 (m, =CH), 3.90 (s, OCH₃), 3.35 (s, SCH₂), 3.10 (d, SCH₂), 1.80 (t, 1-H), and 1.40 (d, CH₃).

(7): v_{co} (hexane) 2073w, 1954s, and 1944s cm⁻¹; ¹H n.m.r. δ (C₆D₆) 6.78 (d, =CH), 5.81 (dd, =CH), 5.00 (m, OCH₂), 3.30 (dt, 6-H), 2.13 (m, 7a-H), and 1.42 (m, 7-H).

‡ Crystal data for (3) PhMe: triclinic, space group P1, *a* = 8.886(1), *b* = 10.706(1), *c* = 12.807(1) Å, α = 109.50(1), β = 108.06(1), γ = 98.94(1)°, *U* = 1045.4 Å³, *D_c* = 1.796 g cm⁻³, *Z* = 2, μ (Mo-*K_α*) = 56.8 cm⁻¹, *T* = -35 °C. 3689 unique reflexions were recorded on a Syntex P2₁ diffractometer of which 3399 were 'observed' with *I* ≥ 2.0 σ(*I*) [ω-scan, Δω = 0.9°, (sinθ/λ)_{max} = 0.595, Mo-*K_α* radiation, $\lambda = 0.71069$ Å, graphite monochromator]. Lorentz polarization and empirical absorption corrections were applied (rel. transm.: 0.7– 1.0). Solution was by Patterson methods. *R* = 0.027, *R_w* = 0.032, *w* = 1/σ² (*F_o*) for 268 refined parameters (anisotropic, H atoms fixed, SHELX-76). Δρ_{fin}. (max./min.) = 1.0/-1.2 e/Å³. Atomic co-ordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1.



Scheme 2. Reagents and conditions: a, allyl alcohol/sodium, molecular sieve (4 Å), 0 °C, 24 h, 43%; b, toluene, 60 °C, 90 min, 25%.

ring atoms. The aminocarbene atom N(1) has a planar sp² configuration which results in a short N-C_{carbene} bond [1.298(6) Å] and a relatively long W-C_{carbene} bond [2.227(4) Å].

In (1) the nucleophilic addition to the carbon earbon atom has to compete with the conjugate addition to the C=C bond. In the presence of catalytic amounts of potassium or sodium, furfuryl thiol or furfuryl alcohol undergo Michael addition leading to (5) and (4), respectively.

In principle, the diene and the dienophile can be exchanged within the carbene ligand. In the allyloxy furfurylidene complex (6) which is accessible from the methoxy carbene complex by base-induced alcoholysis, however, the reactivity of the diene is reduced by the adjacent electrophilic carbene carbon atom. The cycloadduct (7) \dagger is obtained in an only moderate yield because of a facile cycloreversion (Scheme 2).

The diene-dienophile functionalization of carbene ligands offers a new means of stereocontrol for the Diels-Alder

reaction by a metal template. In combination with well known procedures for the cleavage of the metal–carbene bond,⁴ our further studies are aimed at the synthetic scope of this methodology which should further demonstrate the versatility of metal carbenes in cycloaddition reactions.¹²

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References

- For Part XXXI of the Series Reactions of Complex Ligands, see K. H. Dötz and J. Rott, J. Organomet. Chem., submitted for publication.
- 2 R. Hoffmann, Angew. Chem., Int. Ed. Engl., 1982, 21, 711.
- 3 F. G. A. Stone, Angew. Chem., Int. Ed. Engl., 1984, 23, 89.
- 4 Review: K. H. Dötz, H. Fischer, P. Hofmann, F.R. Kreissl, U. Schubert, and K. Weiss 'Transition Metal Carbene Complexes,' Verlag Chemie, Weinheim, 1983.
- 5 W. D. Wulff and D. C. Yang, J. Am. Chem. Soc., 1983, 105, 6726; 1984, 106, 7565.
- 6 K. H. Dötz and W. Kuhn, J. Organomet. Chem., 1985, 286, C23.
- 7 K. H. Dötz, W. Kuhn, G. Müller, B. Huber, and H. G. Alt, Angew. Chem., Int. Ed. Engl., 1986, 25, 812.
- 8 Reviews: (a) E. Ciganek, *Org. React.*, 1984, **32**, 1; (b) D. F. Taber, Intramolecular Diels-Alder and Alder Ene Reactions, Springer-Verlag, Berlin, 1984.
- 9 Reviews: W. Oppolzer, Angew. Chem., Int. Ed. Engl., 1977, 16, 10; G. Helmchen in 'Modern Synthetic Methods,' vol. 4, ed. R. Scheffold, Springer-Verlag, Berlin, 1986.
- 10 K. A. Parker and M. R. Adamchuk, Tetrahedron. Lett., 1978, 1689.
- 11 H. W. Gschwend, M. J. Hillman, B. Kisis, and R. K. Rodebaugh, J. Org. Chem., 1976, 41, 104.
- 12 Recent review: K. H. Dötz in 'Organometallics in Organic Synthesis—Aspects of a Modern Interdisciplinary Field,' ed. A. de Meijere and H. tom Dieck, Springer-Verlag, Berlin, 1988.