

Synthesis of a New Liquid Crystal with a 4,4'-Biphenanthryl Core

Kazuo Yamamura,* Yukihiro Okada, Satoru Ono, Masakazu Watanabe, and the late Iwao Tabushi

Department of Synthetic Chemistry, Kyoto University, Sakyo-ku, Kyoto 606, Japan

4,4'-Biphenanthryl carrying six long tails forms a stable thermotropic liquid crystalline mesophase.

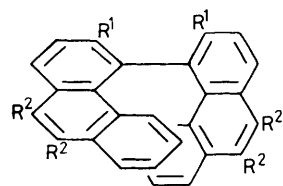
The control of liquid crystalline molecular alignment including that in the discotic mesophase is being actively investigated for elaboration of molecular materials.¹ Of particular interest are chiral mesogens such as those giving phases of the cholesteric or chiral smectic type,² from the viewpoint of electric stimulus-response. In the course of investigation of new functional mesogens,³ we found that the 4,4'-biphenanthryl derivative (**1**), which carries six long tails, forms a stable liquid

crystalline mesophase. This is the first example of a 'pseudo'-discotic mesogen, the core of which may potentially possess molecular asymmetry, opening a route to a new class of chiral liquid crystal other than the cholesteric and chiral smectic types.

The mesogen (**1**) was prepared as follows (Scheme 1): (\pm)-3,3'-dihydroxy-4,4'-biphenanthryl (**2**) was obtained through the oxidative coupling of 3-phenanthrol⁴ as previ-

ously reported.⁵ Treatment of (2) with dodecyl bromide in the presence of Bu^tOK in tetrahydrofuran (THF) gave 3,3'-bis(dodecyloxy)-(±)-4,4'-biphenanthryl (3) in 85% yield. Oxidation of (3) to the corresponding bis-*o*-quinone, 3,3'-bis(dodecyloxy)-(±)-4,4'-biphenanthryl-9,9',10,10'-tetraone (4), was successfully carried out using CrO₃ in acetic acid [37% yield; *m/z* 783 (*M*⁺); ¹H n.m.r. (CDCl₃) δ 0.7–1.4(46H), 3.6–3.9(4H), 6.8–8.4(12H); ν_{C=O} (KBr) 1660 cm⁻¹. The bis-*o*-quinone (4) was readily reduced with Na₂S₂O₄ to afford 3,3'-bis(dodecyloxy)-9,9',10,10'-tetrahydroxy-(±)-4,4'-biphenanthryl (5). Compound (5) was, however, re-oxidized by air very readily to (4), therefore the reduced (5) (without isolation) was condensed with dodecanoic acid using dicyclohexylcarbodiimide (DCC) in THF. Silica gel chromatographic purification provided the desired compound, (1), the spectral (i.r. and n.m.r.) data for which were in agreement with the assigned structure [see Figure 1 for the 400 MHz ¹H n.m.r. spectrum of (1)].

The liquid crystalline properties of (1) were studied by

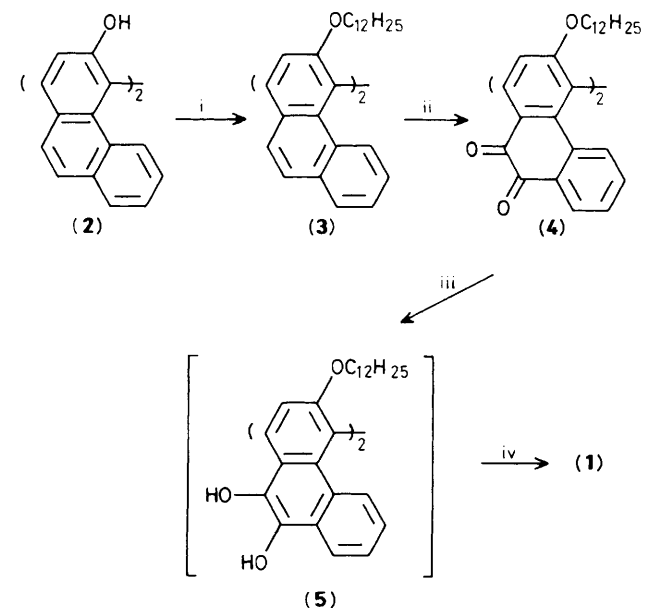


(±)-(1)

R¹ = OC₁₂H₂₅

R² = OCOC₁₁H₂₃

[Only the (+) isomer structure is shown in this figure for simplicity.]



Scheme 1. Reagents: i, C₁₂H₂₅Br, Bu^tOK; ii, CrO₃, AcOH; iii, Na₂S₂O₄, aq. THF; iv, C₁₁H₂₃CO₂H, DCC-THF.

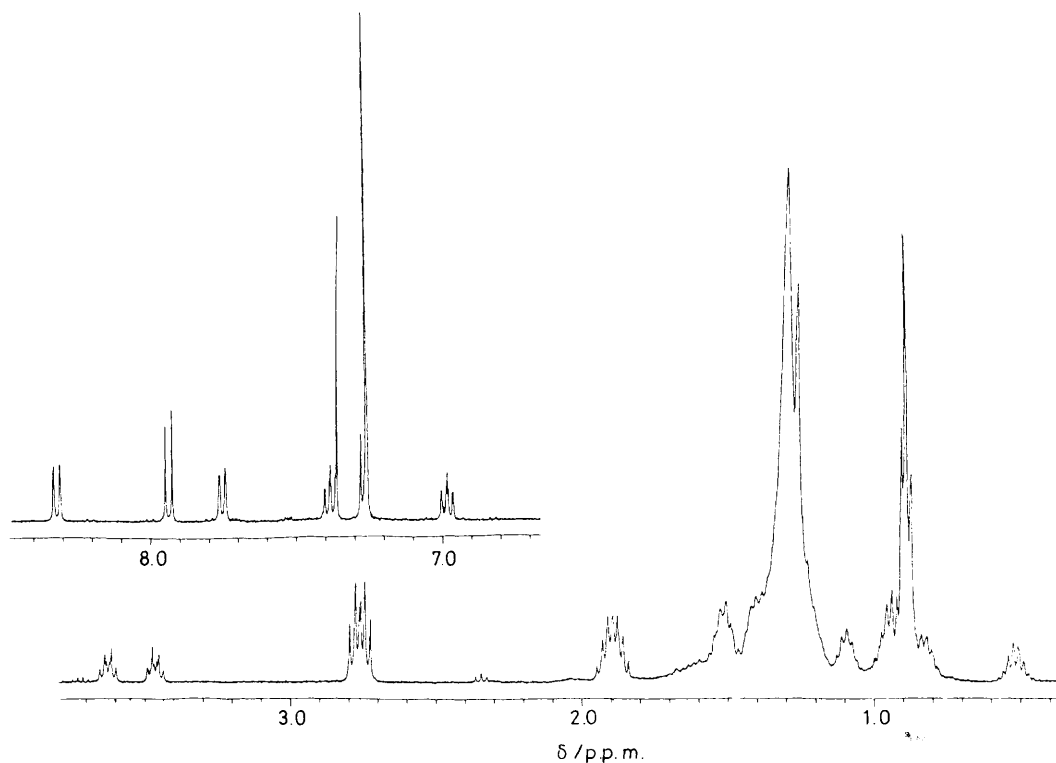


Figure 1. The 400 MHz ¹H n.m.r. spectrum of mesogen (1) in CDCl₃.

colouration of the mesogen began, suggesting a thermal decomposition, which continued up to *ca.* 300°C. We are currently attempting to determine the molecular organization and alignment of the mesophase including *X*-ray studies.

There has been no previous report of a mesogen having a 'pseudo'-discotic core with a skewed structure like that of (**1**). The synthetic route described herein promises to provide a method for preparing new, chiral biphenanthryl-based mesogens, since the chiral resolution of biphenanthryl has been achieved by Yamamoto.⁵

We thank the Japanese Ministry of Education, Science, and Culture for a grant-in-aid for specially promoted research (No. 61065003).

Received, 2nd October 1987; Com. 1434

References

- 1 J.-M. Lehn, J. Malthete, and A.-M. Levelut, *J. Chem. Soc., Chem. Commun.*, 1985, 1794; C. Piechoki and J. Simon, *ibid.*, p. 259; J. Malthete and A. Collet, *Nouv. J. Chim.*, 1985, **9**, 151; B. A. Gregg, M. A. Fox, and A. J. Bard, *J. Chem. Soc., Chem. Commun.*, 1987, 1134; C. Sirlin, L. Bosio, and J. Simon, *ibid.*, p. 379.
 - 2 G. W. Gray, 'Molecular Structure and the Properties of Liquid Crystals,' Academic Press, London, 1962; For a review see, R. B. Meyer, *Mol. Cryst. Liq. Cryst.*, 1977, **40**, 33.
 - 3 I. Tabushi, K. Yamamura, and K. Kominami, *J. Am. Chem. Soc.*, 1986, **108**, 6409; I. Tabushi, K. Yamamura, and Y. Okada, *J. Org. Chem.*, 1987, **52**, 2502; I. Tabushi, K. Yamamura, and Y. Okada, *Tetrahedron Lett.*, 1987, **28**, 2269.
 - 4 L. F. Fieser, *J. Am. Chem. Soc.*, 1929, **51**, 2420.
 - 5 K. Yamamoto, K. Noda, and Y. Okamoto, *J. Chem. Soc., Chem. Commun.*, 1985, 1065.
-