

Aziridination of Cyclohex-2-en-1-ol and Geraniol: Comparison with Epoxidation

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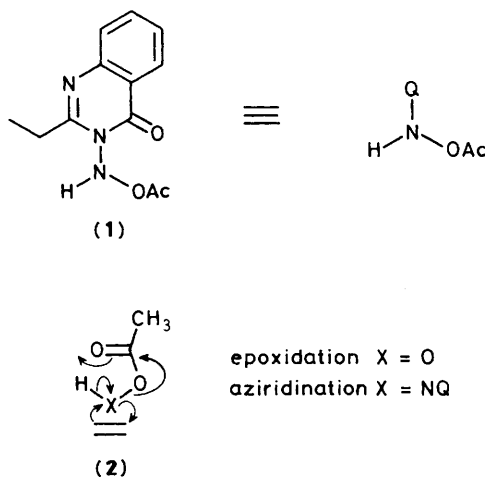
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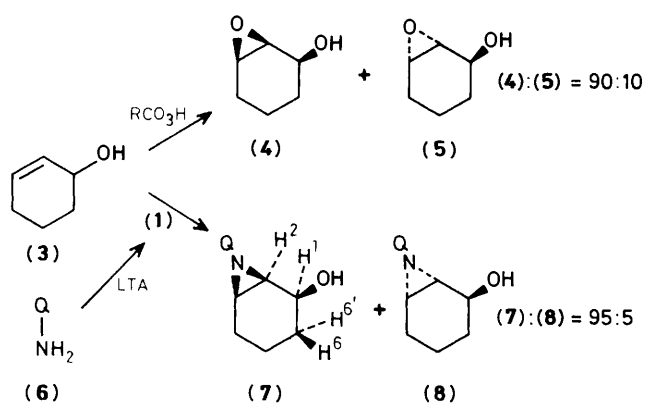
Aziridination of cyclohex-2-en-1-ol with 3-acetoxyamino-2-ethylquinazolone (1) is highly stereoselective, and reaction with geraniol is highly regioselective for the allylic alcohol double bond; comparisons are made with the corresponding reactions of peracids.

We have shown that the *N*-acetoxyaminoquinazolone (1) is an effective aziridinating agent for alkenes.¹ It was suggested that the mechanism for this aziridination (2) might be analogous to that accepted for peracetic acid oxidation of alkenes to give epoxides.

Seeking support for this analogy, we carried out the aziridinations of cyclohexenol (3) with (1). It is well known that epoxidation of (3) with peracids leads stereoselectively to the *syn*-epoxide (4); this has been attributed to hydrogen bonding in the transition state between the peracid and the OH group.²

Aziridination of (3) was carried out either by using solutions of (1) or by lead tetra-acetate (LTA) oxidation of the *N*-aminoquinazolone (6) in the presence of cyclohexenol (3) (3 mol. equiv.). The major product (7), m.p. 99–102 °C, isolated in 70% yield, was the *syn*-stereoisomer (7) with $J(\text{H1})(\text{H2}) = 4.5$ Hz, $J(\text{H1})(\text{H6}) = 9$ Hz, and $J(\text{H1})(\text{H6}') = 6$ Hz in its n.m.r. spectrum.





From a similar reaction of cyclohexenyl acetate (9), the only aziridine isolated was (10) (7%, m.p. 120–123 °C) having $J(\text{H}1)(\text{H}2) = \text{ca. } 1 \text{ Hz}$ in its n.m.r. spectrum.³

The amount of the minor stereoisomer (8) produced in the aziridination of cyclohexenol (3) was assayed by acetylation of the reaction mixture; n.m.r. examination of this product revealed only 5% of the acetate (10) to be present.

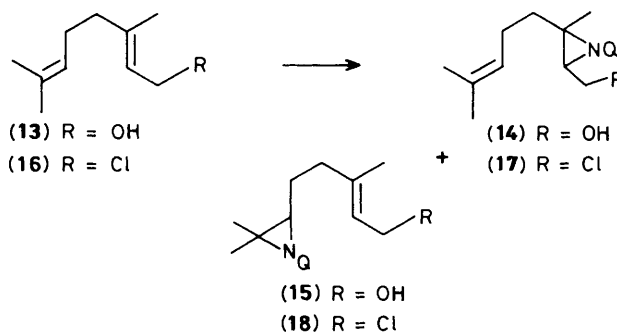
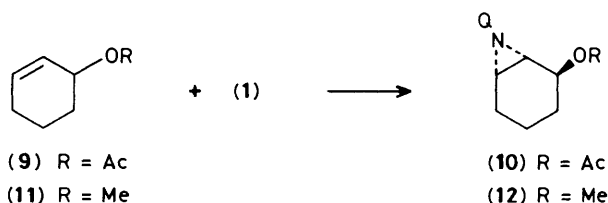
The importance of hydrogen bonding for efficient *syn*-aziridination of (3) was shown in the reaction of the derived methyl ether (11) with (1); only the *anti*-stereoisomer was isolated [19%, m.p. 107–109 °C, $J(\text{H}1)(\text{H}2) = \text{ca. } 1 \text{ Hz}$].

These aziridination results are analogous to those found for epoxidation of (3), (9), and (11),² except that aziridination of (9) and (11) proceeds far more stereoselectively (but in poorer yield) than the corresponding epoxidations.

Aziridination of geraniol (13) with the *N*-acetoxyaminoquinazolinone (1) or by LTA oxidation of (6) gave (14) (76%, m.p. 128–130 °C) and (15) (7%). Geranyl chloride (16), by contrast, is aziridinated preferentially on the 6,7-bond [ratio (17):(18) = 1:6].

Epoxidation of geraniol with peracids is not very regioselective: *m*-chloroperbenzoic acid gives a 2:1 ratio of epoxides from attack on the 6,7- and 2,3-double bonds respectively.⁴

Both the greater stereoselectivity in the aziridination of (3) and the higher regioselectivity in aziridination of geraniol with (1) can be rationalised by assuming stronger hydrogen bonding between the latter and the hydroxy group in the transition state for aziridination.



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- These values of $J(\text{H}1)(\text{H}2)$ of 4.5 Hz [for (7)] and 1 Hz [for (10) and (11)] are close to those found for 2,3-epoxides of cyclohexene-1-carboxylates of known configuration: S. G. Davies and G. H. Whitham, *J. Chem. Soc., Perkin Trans. 1*, 1977, 572.
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