

Novel Backbone Rearrangement of Steroids: Formation of (20*R* and 20*S*) 1 β ,14 β -Dimethyl-18,19-dinor-5 β ,8 α ,9 β ,10 α -cholest-13(17)-enes

Odette Sieskind,^a Jean-Pierre Kintzinger,^b and Pierre Albrecht^{a*}

^a Laboratoire de Géochimie Organique, UA 31 du CNRS

^b Laboratoire de RMN et de Modélisation Moléculaire, UA 422 du CNRS, Département de Chimie, Université Louis Pasteur, 1, rue Blaise Pascal, 67008 Strasbourg, France

Two new 1-methyl rearranged sterenes were formed from cholestan-3 β -ol and related compounds by a novel backbone reaction, catalysed by K10-montmorillonite; their structures were determined by ¹H and ¹³C n.m.r. spectroscopy.

Under acidic conditions cholest-5-ene undergoes extensive isomerization reactions¹ leading, in particular, to $\Delta^{13(17)}$ -compounds by the well known backbone rearrangement.^{2–5} $\Delta^{13(17)}$ Rearranged sterenes ('diasterenes') have also been reported to occur in immature sediments where they are presumably formed by rearrangement of cholestenes catalysed by clay minerals.^{5–7} Furthermore laboratory experiments have shown that montmorillonite clays are the most efficient catalysts for converting cholest-5-ene into backbone rearranged products.⁸ We report here the occurrence of a novel backbone rearrangement of steroids leading to 1 β ,14 β -dimethyl-18,19-dinor-5 β ,8 α ,9 β ,10 α -cholest-13(17)-enes, the formation of which implies a carbocation migration pathway *via* C(1).

Typically, heating of cholestan-3 β -ol (1 g) in refluxing cyclohexane (50 ml) for one hour in the presence of K10-montmorillonite (FLUKA; 5 g) produced cholest-2-ene which was in turn rapidly converted into a mixture of isomeric sterenes. After work-up⁸ this mixture (85% of starting material) was shown by g.c. to contain four major components. Two of them (40% of the reaction product) were the previously described 5 β ,14 β -dimethyl-18,19-dinor-8 α ,9 β ,10 α -cholest-13(17)enes (20*R* and 20*S*), (**1**) ('diasterenes'). These compounds are formed by migration of a C(2) [or C(3)] carbenium ion towards C(5), followed by a classical backbone rearrangement and epimerization at C(20) (Figure 1).

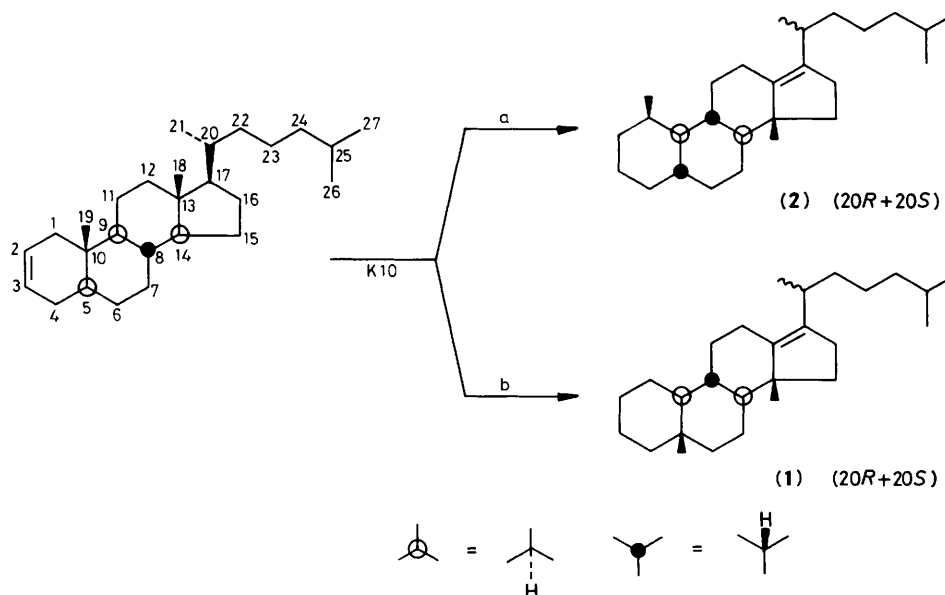
The two other major products (30% of reaction mixture) were sterenes of unknown structures displaying the same mass spectra as diasterenes (**1**) [electron impact, 70 eV, *m/z* 370

(7%, *M*⁺), 257 (100%, *M*⁺ – C₈H₁₇)] from which it was difficult to separate them by usual chromatographic methods.

The same reaction mixture was obtained when cholestan-3 β -ol was replaced by cholest-2-ene. However, when cholest-1-ene was used as the starting material the proportion of the new compounds was highly increased; they indeed became three times more important than diasterenes (**1**). With cholestan-1 α -ol⁹ they were the major products (30–40%) accompanied by only traces of (**1**).

These observations suggested, along with the similarity of chromatographic properties and mass spectra with those of (**1**), that the unknown sterenes were also backbone rearranged products, but formed by a different route implying a carbenium ion migration *via* C(1) instead of C(5). Indeed in such a rearrangement the angular C(19) methyl group would be shifted from C(10) to C(1) and the carbenium ion would migrate along the backbone as for classical diasterenes by a series of reversible carbocation–alkene interconversions^{5,10,11} leading finally to the thermodynamically more stable products (Scheme 1).

The experiment with cholestan-1 α -ol allowed the preparation of the new sterenes as a mixture of 20*R* and 20*S* diastereoisomers which were separated by t.l.c. (AgNO₃–SiO₂, elution with hexane–methylene chloride, 2:1) and further purified by reverse phase h.p.l.c. (RP-18, Merck, elution with acetone–water, 9:1). As in the case of the known diasterenes (**1**), the new products were obtained as oils. Their structure was established by ¹H and ¹³C n.m.r. on the 20*S* isomer (**2**). Spectra and chemical shifts of (**2**) (20*S*) were



Scheme 1. Backbone rearrangement pathways of cholest-2-ene on K10 montmorillonite: (a) via C(1); (b) via C(5).

Table 1. ^{13}C and ^1H chemical shifts for classical (1) and new (2*S*)-diasterenes (2) in C_6D_6 solutions. Chemical shifts were measured with respect to C_6D_6 or to C_6HD_5 and converted to the Me_4Si scale by $\delta_{\text{C}} = \delta_{\text{meas.}} + 128$ p.p.m.; $\delta_{\text{H}} = \delta_{\text{meas.}} + 7.15$ p.p.m.

Carbon number	(2 <i>S</i>)-Diasterene (1)		(2 <i>S</i>)-Diasterene (2)	
	$\delta^{13}\text{C}$	$\delta^1\text{H}$	$\delta^{13}\text{C}$	$\delta^1\text{H}$
1	24.81	0.92—1.66	28.30	2.03
2	27.72	1.13—1.73	34.45	1.36—1.53
3	22.03	1.42—1.48	20.73	1.40
4	42.73	1.05—1.33	35.49	0.86—1.67
5	34.15		38.61	1.13
6	42.42	1.08—1.37	35.24	0.93—1.63
7	22.57	1.35—1.44	27.42	1.18—1.56
8	56.16	0.89	55.24	0.96
9	36.84	1.22	35.25	1.24
10	51.05	0.71	51.32	0.74
11	31.86	0.69—1.90	30.86	0.76—1.97
12	23.25	1.80—2.41	23.18	1.88—2.47
13	141.78		141.77	
14	50.82		50.66	
15	38.41	1.48—1.69	38.43	1.55—1.73
16	28.47	2.15—2.28	28.42	2.18—2.33
17	134.12		134.15	
18	18.43	0.96	18.66	0.97
19	17.16	0.85	12.68	0.85
20	32.14	2.49	32.12	2.53
21	20.06	0.98	20.04	1.02
22	36.05	1.30—1.34	35.97	1.33
23	26.14	1.24—1.30	26.11	1.23—1.38
24	39.53	1.17	39.43	1.19
25	28.35	1.53	28.36	1.55
26	23.04	0.91	22.93	0.91
27	22.90	0.91	22.79	0.91

compared with those of the known diasterene (1) (2*S*). For the latter the ^{13}C signal assignments were essentially in agreement with those published previously,¹² except for a notable difference observed for C(5), also pointed out

recently by Peakman *et al.*⁵ They resulted from 2D-INADEQUATE experiments, whereas the corresponding proton signal assignments were obtained from COSY $\delta^{13}\text{C}/\delta^1\text{H}$ experiments (Table 1). The well resolved signals of H(12e), H(16), H(16'), H(11e), H(12a), H(15) and H(11a) (in CDCl_3 solution) allowed the measurements of the 3J coupling constants, which were in agreement with a ring C in chair conformation and a *B/C trans* ring junction (J_{11a-9a} 12.5 Hz; J_{11e-9a} 4.1 Hz; $J_{11a-12a}$ 12.5 Hz; $J_{11a-12e}$ 4.0 Hz; $J_{11e-12a}$ 4.2 Hz; $J_{11e-12e}$ 2.4 Hz), and of the highly stereospecific 5J coupling constants between H(12a) and H(16) and H(16') (4.4 and 2.0 Hz, respectively). The identity of the coupling patterns of diasterene (1) and of the new compound (2) clearly indicated that the *B/C* ring junction and the stereochemistry at C(14) were indeed preserved in the latter.

COSY $^1\text{H}/^1\text{H}$ and $^{13}\text{C}/^1\text{H}$ experiments enabled us to assign the signals of the remaining nuclei in rings A and B (Bruker AM 400, DISR 871 software) (Table 1). Significant differences were observed in the ^{13}C spectra of compounds (1) and (2) for several carbons of rings A and B only. From the ^1H and ^{13}C spectra, as well as from the above correlations it appeared that the C(19) methyl ($\delta^1\text{H} = 0.85$, d) was indeed shifted from C(10) to C(1).

J-spectroscopy allowed the determination of the fine structure of the signals corresponding to H(9) and H(10), and hence the nature of the *A/B* ring junction. Indeed H(9) was a quartet (J 10.6 Hz) further dedoubled (J 3.5 Hz), implying three *trans* and one *gauche* couplings, while H(10) appeared as a multiplet (J 10.8, 10.0, 4.0 Hz) corresponding to two *trans* and one *gauche* couplings. These values were consistent with H(9) being β , *trans* coupled with H(8) (α), H(11) (α), and H(10) (α), and *gauche* coupled with H(11) (β). They further showed that H(10) (α) was *trans* coupled to H(5) (β) and *gauche* coupled to H(1) (α), revealing an *A/B trans* ring junction and a C(19) methyl in an axial position.

These results therefore defined the new rearranged sterenes as being (2*0R* and 2*0S*) $1\beta,14\beta$ -dimethyl-18,19-dinor-5 $\beta,8\alpha,9\beta,10\alpha$ -cholest-13(17)-enes (2) (Scheme 1).

The C(19) methyl at C(1) remains axial most probably because the equatorial position is thermodynamically unfav-

oured due to periplanar interactions with ring C, a feature previously recognized in corresponding perhydrophenanthrenes.^{13,14} Data obtained on graphics computer system by modelling based on molecular mechanics (program SYBYL, Tripos Assoc.) were in agreement with this statement. They indeed showed an energy difference of about 5 kcal/mol (cal = 4.184 J) in favour of the more stable axial epimer.

We thank Dr. A. Milon for computer modelling calculations; Mrs. E. Krempf for technical assistance in n.m.r. measurements.

Received, 26th June 1988; Com. 8/02574F

References

- 1 R. B. Turner, W. R. Meodor, and R. E. Winkler, *J. Am. Chem. Soc.*, 1957, **79**, 4122.
 - 2 J. W. Blunt, M. P. Hartshorn, and D. N. Kirk, *Tetrahedron*, 1969, **25**, 149.
 - 3 D. N. Kirk and P. M. Shaw, *J. Chem. Soc., Perkin Trans. I*, 1975, 2284.
 - 4 D. M. Tal, E. Keinan, and Y. Mazur, *Tetrahedron*, 1981, **37**, 4327.
 - 5 T. M. Peakman, K. Ellis, and J. R. Maxwell, *J. Chem. Soc., Perkin Trans. I*, 1988, 1071.
 - 6 I. Rubinstein, O. Sieskind, and P. Albrecht, *J. Chem. Soc., Perkin Trans. I*, 1975, 1833.
 - 7 T. M. Peakman and J. R. Maxwell, in 'Advances in Organic Geochemistry 1987,' Pergamon, Oxford, in the press.
 - 8 O. Sieskind and P. Albrecht, *Tetrahedron Lett.*, 1985, **26**, 2135.
 - 9 See: G. H. Douglas, P. S. Ellington, G. D. Meakins, and R. Swindlees, *J. Chem. Soc.*, 1959, 1720; T. Nakano, M. Hasegawa, and C. Djerassi, *Chem. Pharm. Bull. Jpn.*, 1963, **11**, 465; C. Djefassi, D. H. Williams, and B. Berkoz, *J. Org. Chem.*, 1962, **27**, 2205, for preparation of cholest-2-ene, cholest-1-ene, and cholestan-1 α -ol respectively.
 - 10 D. E. Akporiaye, R. D. Farrant, and D. N. Kirk, *J. Chem. Res., (S)* 1981, 210.
 - 11 J. C. Jacquesy, R. Jacquesy, and G. Joly, *Tetrahedron Lett.*, 1972, 4739.
 - 12 M. Tal, H. E. Gottlieb, C. Ben-Ari, and Y. Mazur, *Tetrahedron*, 1981, **37**, 4331.
 - 13 T. R. Ames, J. L. Beton, A. Bowers, T. G. Halsall, and E. R. H. Jones, *J. Chem. Soc.*, 1954, 1905.
 - 14 W. Klyne, in 'Progress in Stereochemistry I,' ed. W. Klyne, Butterworths, London, 1954, p. 53.
-