## **Total Synthesis of Naturally Occurring Granaticin**

## Kousuke Okazaki, Keiichi Nomura, and Eiichi Yoshii\*

Faculty of Pharmaceutical Sciences, Toyama Medical and Pharmaceutical University, Sugitani 2630, Toyama 930-01, Japan

Natural granaticin (1) has been synthesized by joining the chiral left and right fragments, (5) and (10), and subsequent lactonization of the resulting product (11) and oxidative *O*-demethylation.

Granaticin  $(1)^1$  is a structurally unique member of the pyranonaphthoquinone antibiotics, and its synthesis has been the subject of considerable interest. Recently we reported the first total synthesis of (1) in racemic form.<sup>2</sup> We now describe a

stereocontrolled synthesis of optically active natural granaticin, which, in a key step, employs benzannulation between the pentacyclic phthalide (5) and dihydropyranone (10), both having the correct absolute stereochemistry.



(1)



Scheme 1. Reagents and conditions: i,  $OsO_4$ ,  $Me_3N(O)$ ,  $Bu'OH-H_2O$ , 4 h; ii, *N*-bromosuccinimide, azoisobutyronitrile,  $CCl_4$ , 40 °C, 30 min, then AgClO<sub>4</sub>, tetrahydrofuran (THF), room temp., 10 min; iii, excess of 2-methoxypropene, THF, camphorsulphonic acid, room temp., 1 h; iv,  $Bu^nLi$  (1.1 equiv.), THF, -100 °C, 3 min, then  $Et_2NCOCl$  (2.8 equiv.), -70 °C to room temp., 1.5 h; v, Bu'Li (3 equiv.), THF, -78 °C, 1 h, then dimethylformamide (4.3 equiv.), 0 °C, 15 min; vi,  $Me_3SiCN$ ,  $CH_2Cl_2$ , catalytic amount of 18-crown-6-KCN complex, 0 °C, 15 min, then AcOH, room temp., 24 h.

The optically active starting material (*R*)-(2)<sup>†</sup> leading to (5) was first converted into compound (3), m.p. 153–55°C,  $[\alpha]_D^{25}$  +52.2° (*c* 1.53, CHCl<sub>3</sub>), according to the procedure reported for the racemate;<sup>3</sup> highly diastereoselective catalytic osmylation followed by *N*-bromosuccinimide-mediated cyclization (46% overall yield) (Scheme 1). The bromine substituent in (3) was then replaced by the *N*,*N*-diethylcarbamoyl group by sequential treatment of its acetonide with n-butyl-lithium and Et<sub>2</sub>NCOCl to give (4) in 74% yield. Transformation of benzamide (4) into cyanophthalide (5)<sup>‡</sup> was carried out by our improved procedure<sup>4</sup> (32% overall yield).

‡ For the synthesis of regioisomeric cyanophthalide (racemic), see ref. 2.



Scheme 2. Reagents and conditions: i,  $CH_2=C(OMe)OSiMe_2Bu^{t}$  (1.25 equiv.),  $BF_3$ -ether (1.5 equiv.), toluene,  $-78 \,^{\circ}C$ , 1.5 h, then 1% KOH-MeOH, room temp., 10 min; ii, MeSO<sub>2</sub>Cl (2.5 equiv.),  $Et_3N$  (8 equiv.),  $CH_2Cl_2$ , 0  $^{\circ}C$  to room temp.; iii, silica gel chromatography (hexane-AcOEt, 2:1); iv, Me<sub>2</sub>AlNMe<sub>2</sub> (1.3 equiv.),  $CH_2Cl_2$ , 0  $^{\circ}C$  to room temp., 1.5 h, then pyridinium chlorochromate, AcONa, molecular sieves 3Å,  $CH_2Cl_2$ ,  $-10 \,^{\circ}C$ , 40 min.



Scheme 3. Reagents and conditions: i, MeSOCH<sub>2</sub>Li (3.4 equiv.), Bu<sup>t</sup>OH (3.1 equiv.), tetrahydrofuran (THF), -78 °C, 15 min, then (10), -78 °C to room temp., 14 h; ii, Me<sub>2</sub>SO<sub>4</sub> (8 equiv.), K<sub>2</sub>CO<sub>3</sub> (17 equiv.), acetone, reflux, 40 min; iii, LiBBus<sub>3</sub>H (2 equiv.), THF, -78 °C, 1 h, then Me<sub>3</sub>SiCl (6 equiv.), THF, room temp., 23 h; iv, excess of 2-methoxypropene, camphorsulphonic acid, THF, room temp.; v, cerium(Iv) ammonium nitrate, MeCN, then AlCl<sub>3</sub> (6 equiv.), Et<sub>2</sub>S (24 equiv.), CH<sub>2</sub>Cl<sub>2</sub>, room temp.

Preparation of the Michael acceptor (10) starting from di-O-acetylrhamnal (6) (Scheme 2) was commenced by reaction of (6) with ketene methyl t-butyldimethylsilyl acetal in the presence of  $BF_3$ -Et<sub>2</sub>O in toluene at low temperature.<sup>5</sup>

<sup>&</sup>lt;sup>†</sup> This enantiomer,  $[\alpha]_D^{25}$  +12.4° (c 0.67, CHCl<sub>3</sub>), was obtained by resolution of (±)-(**2**) in high yield *via* the (-)-*N*-(1-phenylethyl)carbamate, and the absolute configuration was determined by the exciton chirality method as applied to the *O*-benzoate of the corresponding naphthyl carbinol (K. Okazaki, Dissertation for Doctor's degree, March 1988). For the preparation of (±)-(**2**), see ref. 3.

The *C*-glycosides, (7) and the C-1 epimer (3:2 ratio, 64% yield), were obtained after *O*-deacetylation and chromatographic separation. The *O*-methanesulphonate (8) of (7) readily underwent an intramolecular  $S_N 2'$  reaction on silica gel chromatography to afford the  $\gamma$ -lactone (9)§ [87% yield from (7)], m.p. 88—91 °C,  $[\alpha]_D^{25}$  +185° (*c* 0.805, CHCl<sub>3</sub>). The lactone (9) was then converted to the  $\gamma$ -keto carboxamide (10) by aminolysis with Me<sub>2</sub>AlNM<sub>2</sub><sup>6</sup> followed by oxidation with pyridinium chlorochromate<sup>7</sup> (82% yield).

Coupling of the chiral fragments (5) and (10) was carried out in the presence of LiCH<sub>2</sub>SOMe (excess)<sup>2</sup> affording the hexacyclic naphthalene-1,4-diol derivative (11) in *ca*. 50% yield, which on *O*-methylation gave (12) (Scheme 3). The ketone carbonyl group in (12) was then reduced with lithium tri-s-butylborohydride to produce the *cis*- $\gamma$ -hydroxycarboxamide predominantly, which could be easily lactonized by treatment with chlorotrimethylsilane in wet CH<sub>2</sub>Cl<sub>2</sub><sup>8</sup> to afford (13) (56% overall yield), [ $\alpha$ ]<sub>D</sub><sup>25</sup> -158° (*c* 0.203, CHCl<sub>3</sub>).

Finally, oxidative O-demethylation of (13) according to the same method as applied to racemate<sup>2</sup> afforded (1) in 62% yield from (13) after chromatographic purification. The

synthetic granaticin [deep red crystals from AcOEt, m.p. and mixed m.p. 211-213 °C (decomp.)] proved to be identical chromatographically and spectroscopically (<sup>1</sup>H n.m.r., i.r., u.v., and c.d.¶) with an authentic sample of natural granaticin provided by Dr. H. H. Peter, Ciba-Geigy Ltd., Switzerland.

Received, 19th September 1988; Com. 8/03571G

## References

- 1 'The Merck Index,' 10th edn., ed. M. Windholz, Merck & Co., Inc., New Jersey, 1983, p. 652.
- 2 K. Nomura, K. Okazaki, K. Hori, and E. Yoshii, J. Am. Chem. Soc., 1987, 109, 3402.
- 3 Y. Takeuchi, M. Sudani, and E. Yoshii, J. Org. Chem., 1983, 48, 4151; E. Yoshii, Y. Takeuchi, K. Nomura, K. Takeda, S. Odake, M. Sudani, and C. Mori, Chem. Pharm. Bull., 1984, 32, 4767.
- 4 K. Okazaki, K. Nomura, and E. Yoshii, Synth. Commun., 1987, 17, 1021.
- 5 Reaction of enol silane with (6): N. Greenspoon and E. Keinan, J. Org. Chem., 1988, 53, 3723.
- 6 A. Basha, M. Lipton, and S. M. Weinreb, *Tetrahedron Lett.*, 1977, 4171
- 7 E. J. Corey and J. W. Suggs, Tetrahedron Lett., 1975, 2647.
- 8 T. Wakamatsu, H. Hara, and Y. Ban, J. Org. Chem., 1985, 50, 108.

¶ C.d. data of (1) [ $\lambda$ /nm ( $\Delta\epsilon$ ) in EtOH]: 202 (+5.52), 221 (-5.70), 230 (+0.35), 240 (-1.55), 258 (+0.17), 276 (-0.52), 310 (+1.73), 386 (-1.73), 430 (-1.03), 526 (-2.42).

<sup>§</sup> The stereochemistry was determined by differential nuclear Overhauser enhancement performed with (9) and the diastereoisomeric *cis*-lactone derived from the C-1 epimer of (7). <sup>1</sup>H N.m.r. data for (9) (270 MHz, CDCl<sub>3</sub>):  $\delta$  1.30 (d, *J* 6.8 Hz, 3H), 2.60 (d, *J* 17.3 Hz, 1H), 2.82 (dd, *J* 17.8, 5.1 Hz, 1H), 4.40 (qdd, *J* 6.8, 3.7, 2.0 Hz, 1H), 4.55 (m, 2H), 6.00 (ddd, *J* 10.3, 4.2, 2.0 Hz, 1H), 6.21 (dd, 10.3, 3.7 Hz, 1H).