

## Biosynthesis of Estrogens by Microsomal Placental Aromatase; Isolation and Metabolism of 10 $\beta$ -Hydroxyestr-4-ene-3,17-dione

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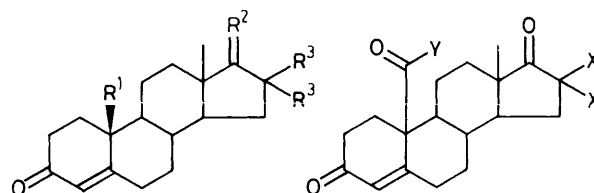
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Incubation of [16,16,19-<sup>2</sup>H<sub>3</sub>][19-<sup>3</sup>H]-19-oxoandrost-4-ene-3,17-dione (**2a**) with microsomal aromatase gave *inter alia* [16,16-<sup>2</sup>H<sub>2</sub>]-10 $\beta$ -hydroxyestr-4-ene-3,17-dione (**1**), whose analogue (**1b**) on incubation with the aromatase gave 10 $\beta$ ,17 $\beta$ -dihydroxyestr-4-en-3-one (**1d**) (*ca.* 10%) and no estrogens which would indicate that (**1b**) is not an obligatory estrogen precursor.

For the past several years we have systematically explored the scope of transformations of [16,16,19-<sup>2</sup>H<sub>3</sub>][19-<sup>3</sup>H]-19-oxoandrost-4-ene-3,17-dione (**2a**) by human microsomal placental aromatase.<sup>1-4</sup> In the course of our recent studies,<sup>1</sup> we isolated several 'new' products. One of these products showed an HPLC<sup>†</sup> retention time (*R<sub>t</sub>*) of 7.8 min, and the GC-MS of its bis-*O*-methyloxime trimethylsilyl (MO-TMS) derivative showed *m/z* 420 (*M*<sup>+</sup>), 389, 299, 226, *etc.* The mass spectrum was consistent with a C<sub>18</sub>H<sub>22</sub><sup>2</sup>H<sub>2</sub>O<sub>3</sub> compound, *e.g.*, [2H<sub>2</sub>]-( $\xi$ -hydroxy)-estr-4-ene-3,17,dione. These results were most interesting in view of the hypothesis that the 'third' mole of oxygen is utilized for the oxidative transformation of (**2b**) to 10 $\beta$ -hydroxyestr-4-ene-3,17-dione (**1b**) which, following dehydration, yields estrone.<sup>5</sup> This hypothesis seemed rather unusual since Covey *et al.*<sup>6</sup> demonstrated that the 10 $\beta$ -hydroxy compound (**1b**), and Akhtar *et al.*<sup>8</sup> showed that the 10 $\beta$ -hydroxy formate (**1e**), are not metabolized to estrogens by microsomal aromatase.

Considering the gross composition of the unknown, the mechanistic implications of the proposed scheme<sup>5</sup> and the contradicting evidence,<sup>6,7</sup> prompted us to determine the structure of the metabolite. As a working hypothesis, we assumed that the metabolite is [16,16-<sup>2</sup>H<sub>2</sub>]-10 $\beta$ -hydroxyestr-4-ene-3,17-dione (**1a**) and we undertook the synthesis of an authentic reference sample. To this end, a moderate stream of



(**1a**) R<sup>1</sup> = OH; R<sup>2</sup> = O; R<sup>3</sup> = D

(**1b**) R<sup>1</sup> = OH; R<sup>2</sup> = O; R<sup>3</sup> = H

(**1c**) R<sup>1</sup> = OOH; R<sup>2</sup> = O; R<sup>3</sup> = H

(**1d**) R<sup>1</sup> = OH; R<sup>2</sup> =  $\beta$ -OH; R<sup>3</sup> = H

(**1e**) R<sup>1</sup> = HCOO; R<sup>2</sup> = O; R<sup>3</sup> = H

(**2a**) X = D; Y = D(T)

(**2b**) X = H; Y = H

dry air was bubbled (72 h) through a toluene solution (5 ml) of a mixture of (**2b**) (150 mg) and azoisobutyronitrile (AIBN); (7.5 mg) maintained at 50 °C (reflux condenser).<sup>8</sup> The toluene was removed in a stream of N<sub>2</sub>, and the residue was purified by PLC (silica; EtOAc-cyclohexane, 3:1) to yield (**1b**) (30 mg) and the 10 $\beta$ -hydroperoxide (**1c**) (75 mg). The hydroperoxide (**1c**) showed m.p. 176–178 °C (from MeOH); NMR ( $\delta$ ) 9.23 (br., OOH), 6.0 (4-H), 0.9 (13-Me); mass spectrum of MO-TMS *m/z* 434, 417, 403, 386, 334, 320, 313, *etc.* Treatment of a solution<sup>9</sup> of (**1c**) (40 mg in 2.0 ml of EtOH) with PPh<sub>3</sub> (70 mg) for 3 h at room temperature gave, following PLC (as above), (**1b**) (25 mg), m.p. 198–200 °C (from acetone); NMR ( $\delta$ ) 5.76 (4-H), 0.94 (13-Me); mass spectrum of MO-TMS gave *m/z* 418, 387, 328, 297, *etc.* The HPLC *R<sub>t</sub>* of

<sup>†</sup> Altech Co. column; Nucleosil 50, 5 $\mu$ ; i.d. 4.6 mm  $\times$  25 cm; 20% propan-2-ol in iso-octane; flow rate 1 ml/min; the eluates were monitored at 240 and 280 nm.

authentic (**1b**) (7.8 min) was identical to that of the metabolite. Except for the shift by two mass units, the mass spectrum of the metabolite and of the synthetic (**1b**) were identical. This established the structure of the metabolite as the 10 $\beta$ -hydroxy compound (**1a**).

We then undertook the evaluation of intermediacy of (**1b**) in the biosynthesis of estrone. The experiments were carried out at pH 6 and 7.2 exactly as previously described for the incubations of (**2a**) with placental aromatase from which estrone and several other metabolites including (**1a**) were isolated.<sup>1</sup> The 10 $\beta$ -hydroxy compound (**1b**) (1 mg) was incubated (pH 7.2) with placental microsomal aromatase, from the same batch employed above, in the usual manner and the products were recovered with chloroform.<sup>1</sup> The residue obtained was analysed by GC-MS of the derived MO-TMS. In addition to the starting material, (**1b**) MO-TMS, only one product (ca. 10%) which showed  $m/z$  463 ( $M^+$ ), 431, 416, 342, 252, etc. was detected. The metabolite was identified as 10 $\beta$ ,17 $\beta$ -dihydroxyestr-4-en-3-one (**1d**).

The absence of detectable amounts of estrogens indicates that (**1b**) is not an obligatory intermediate in the biosynthesis of estrogens.

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