Single Step Synthesis of 2,4-Dicyanonaphthylamines from Synthetic Equivalents of α -Cyanoalkynes

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Nitrile thioethers R–CBr=C(SEt)CN 4 react like α -alkynenitriles with ambiphilic derivatives, as shown by a new synthesis of naphthylamines.

The simplest way of access to bicyclic compounds **3** seems to be the condensation of an α -alkynenitrile with a 1,4-dipole. To our knowledge this reaction has never been reported (Scheme 1). We now report the first carbocyclisation reaction in accordance with this principle, starting from bromoacrylonitriles **4**.¹ These compounds behave in fact like synthetic equivalents of α -alkynenitriles R-CBr=C(SEt)-CN = R-C=C-CN.

Naphthylamine 6, for example, was readily obtained when dipole 5 was reacted with 4 (Scheme 2), using the following general procedure: the carbanion 5 generated from NaH (14

mmol) and 2-cyanophenylacetonitrile (or 2-cyano-5-methoxy-phenylacetonitrile²) (15 mmol) in MeCN (15 cm^3) was



Table 1 Phys	ical data fo	r naphthy	lamines 6ª
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					IR (Nujol): v/c	m ⁻¹
	R1	R ²	M.p/°C	Yield (%)	NH ₂ (free)	C≡N
6a	Ме	Н	275-276	78	3452, 3362	2220
6b	Me	OMe	300-301	56	3447, 3374	2223, 2215
6c	Bu ⁿ	Н	161-163	73	3444, 3365	2221
6d ^b	Ph	Н	357-360	86	3478, 3376	2228, 2218
6e	Ph	OMe	317-320	64	3470, 3363	2233, 2225
6f	$p-ClC_6H_4$	Н	296-297	83	3478, 3376	2228, 2218
6g	p-MeOC ₆ H ₄	Н	273–275	79	3480, 3380	2230, 2225

^a Satisfactory microanalytical and/or accurate mass measurements were obtained for all new compounds. ^b The reaction of freshly prepared phenylpropynenitrile, m.p. 38–39 °C (7 mmol), 5 ($R^2 = H$) (7.7 mmol) and NaH (7 mmol) under the same conditions gives 6d in 80% yield.





Scheme 3

References

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warmed to 45 °C, and acrylonitrile 4[†] (7 mmol, in the same solvent) was added all at once. Stirring was pursued for 45 min to 45 °C, then the solution was refluxed for 45 min and the solvent was evaporated. The residue was stirred for 30 min with H_2O-Et_2O (Et₂O extracts completely by-product 7) and the naphthylamine 6 filtered off and recrystallised (Table 1).

Isolation of the by-product 7 (characterised by unequivocal synthesis from $2\text{-}CN\text{-}C_6H_4\text{-}CH\text{=}O^3$) suggests to us the following mechanism with a sulphenium leaving group in the aromatisation step (Scheme 3).

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[†] The bromo-compounds 4 were prepared by the method in ref. 1.

$$R-CHO + \langle SMe \xrightarrow{EIO^{-}} R-CH = C \langle SMe \xrightarrow{Br_{2}(1.3 \text{ equiv.})} R-CBr = C \langle SMe \xrightarrow{CN} (CCI_{4} \text{ or } MeCN)^{i} \rangle$$