1505

## Highly *exo*-Diastereoselective Diels–Alder Reactions of (2*S*)-*N*-Benzoyl-2-*tert*-butyl-4-methylene-1,3-oxazolidin-5-one

## Stephen G. Pyne,\* <sup>a</sup> Branko Dikic, <sup>a</sup> Peter A. Gordon, <sup>a</sup> Brian W. Skelton<sup>b</sup> and A. H. White<sup>b</sup>

<sup>a</sup> Department of Chemistry, University of Wollongong, PO Box 1144, Wollongong, NSW 2500, Australia <sup>b</sup> Department of Physical and Inorganic Chemistry, University of Western Australia, Nedlands, WA 6009, Australia

(2*S*)-*N*-Benzoyl-2-*tert*-butyl-4-methylene-1,3-oxazolidin-5-one undergoes highly *exo*-diastereoselective Diels–Alder reactions with cyclopentadiene and cyclohexa-1,3-diene.

 $\alpha,\beta$ -Dehydroamino acids are extremely useful substrates for preparing both natural and non-proteinogenic amino acids.<sup>1</sup> These substrates have been employed as Diels–Alder dienophiles<sup>2,3</sup> for the synthesis of racemic cycloalkane amino acids, which have a varety of biological activities.<sup>4</sup> Racemic  $\alpha$ -amino acids have also been prepared from the conjugate addition of nucleophilic reagents to  $\alpha,\beta$ -dehydroamino acids.<sup>1</sup> Enantioselective catalytic hydrogenation of  $\alpha,\beta$ -dehydroamino acids gives chiral  $\alpha$ -amino acids in high enantiomeric purity,<sup>1</sup> little attention however, has been devoted to the asymmetric synthesis of chiral  $\alpha$ -amino acids employing chiral  $\alpha,\beta$ -dehydroamino acids.

In 1987, Seebach *et al.*<sup>5</sup> reported the synthesis of the  $\alpha$ , $\beta$ -dehydroamino acid, (2*S*)-*N*-benzoyl-2-*tert*-butyl-4-methylene-1,3-oxazolidin-5-one **2** from the bromination and then dehydrobromination of the oxazolidinone **1**,<sup>6</sup> derived from (S)-alanine. More recently, a modified synthesis of 2 has been reported by Beckwith *et al.*<sup>7</sup> The latter author has employed 2 as a useful Michael acceptor for alkyl radicals for the enantioselective synthesis of  $\alpha$ -amino acids. We report here, the Diels-Alder reaction of 2 with cyclopentadiene and



Scheme 1 i, N-Bromosuccinimide (2 equiv.), hv, (PhCO<sub>2</sub>)O; ii, NaI, acetone, heat

 Table 1 Diels-Alder reactions of 2

	Diene	Lewis acid	t/days	<i>T</i> /°C	Diastereoselectivity	Yield (%)
	Cyclopentadiene	 LiClO.	14	25	>97(4) : <3 (others)	70
	Cyclopentaulene	$(5.0 \text{ mol dm}^{-3})$	3	25	4:5;35:65	67
	Cyclohexadiene		3	140	>97(6) : <3 (others)	62



Fig. 1 Molecular projection 4, 20% thermal ellipsoids are shown for the non-hydrogen atoms. Hydrogen atoms have an arbitrary radii of 0.1 Å.



cyclohexa-1,3-diene, which proceed in a highly diastereoselective and efficient manner.

Compound (2*S*)-2 was prepared from 1 by a modified version to that reported by Beckwith.<sup>7.8</sup> A side product from this reaction was the bromoalkene 3 (m.p. 123 °C) which was formed in 15–20% yield and as a single geometric isomer (Scheme 1). This material was extremely difficult to separate from 2 by column chromatography, however, most of 3 could be removed by crystallization from hexane. In this way 95% analytically pure  $2^{\dagger}$  could be obtained ( $[\alpha]_D^{24} - 92.6^{\circ}, c \ 0.49, CHCl_3; lit.^5 [\alpha]_D^{25} - 148.6^{\circ}, c \ 1.5, CHCl_3; lit.^7 [\alpha]_D^{25} - 186.3^{\circ}, c \ 1.5, CHCl_3).^7$  The enantiomeric purity of 2 was determined to be 50% using an ionically bound 3,5-dinitrobenzoyl leucine Pirkle HPLC column using a mixture of hexane, ethanol and acetonitrile (99:0.66:0.33) as eluent.

When a solution of 2 dissolved in cyclopentadiene (10 equiv.) was stirred at room temperature for 14 days the Diels-Alder adduct 4 (m.p. 119.5 °C,  $[\alpha]_D^{22} + 68^\circ$ , *c* 2.25, CHCl<sub>3</sub>) was isolated in 70% yield after purification by column chromatography. <sup>1</sup>H NMR analysis (400 MHz) of the crude reaction mixture indicated a product diasteroselection of >97 (4): <3 (other diastereoisomers) (Table 1). The stereochemistry of 4 was unequivocally determined by single crystal X-ray

analysis‡ (Fig. 1), which showed that the carbonyl group of 4 had the exo orientation and addition of the diene had occurred to the face of the alkene of 2 that was opposite to that of the bulky tert-butyl group of the 1,3-oxazolidin-5-one ring. A slight preference for exo-selectivity has been noted in the reaction of achiral  $\alpha,\beta$ -dehydroamino acids with cyclic dienes.<sup>2,3</sup> The high *exo*-diastereoselectivity shown by 2 is identical to that of the 1,3-dioxolan-4-one analogue of 2 in its reaction with cyclopentadiene.9 The Lewis acid catalysed reaction of 2 and cyclopentadiene with 5 mol  $dm^{-3}$  lithium perchlorate in diethyl ether (25 °C, 2 days) gave a mixture (65:35, 67% yield) of the endo- and exo-diastereoisomers 5 and 4 respectively (Table 1). The stereochemistry of 5 was based on the similarity of its <sup>1</sup>H NMR spectrum to that reported for the endo adduct from the 1,3-dioxolan-4-one analogue of 2 and cyclopentadiene.<sup>10</sup> Lithium perchlorate (5.0 mol  $dm^{-3}$ ) in diethyl ether is known to enhance the rate and endo-selectivity in most Diels-Alder reactions.10

The thermally induced reaction of 2 and cyclohexa-1,3diene at 140 °C for 2 days gave a single adduct (>97% diastereoselectivity) in 62% yield after purification (Table 1). This was assigned the *exo*-diastereoisomer 6 (m.p. 126 °C) on the basis of NOE difference <sup>1</sup>H NMR experiments which showed a 7% enhancement of the signal due to the allylic proton at H(1) when the methine of the 1,3-oxazolidinone ring was selectively irradiated.

Financial support by Johnson and Johnson Research Pty., Australia is gratefully acknowledged. We thank Professor A. L. J. Beckwith for the exchange of information on the synthesis of 2.

Received, 21st June 1991; Com. 1/03076K

## References

- 1 V. Schmidt, A. Lieberkneckt and J. Wild, Synthesis, 1988, 159.
- 2 H. Horikawa, T. Nishitani, T. Iwasaki, Y. Mushika, I. Inove and Miyoshi, *Tetrahedron Lett.*, 1980, **21**, 4101.
- 3 M. J. Crossley, T. W. Hambley and A. W. Stamford, Aust. J. Chem., 1990, 43, 1827.
- 4 H. S. Tager and H. N. Christensen, J. Am. Chem. Soc., 1972, 94, 968.
- 5 J. Zimmermann and D. Seebach, *Helv. Chim. Acta*, 1987, **70**, 1104.
- 6 D. Seebach and A. Fadel, Helv. Chim. Acta, 1985, 68, 1243.
- 7 A. L. J. Beckwith and C. L. Chai, J. Chem. Soc., Chem. Commun., 1990, 1087.
- 8 A. L. J. Beckwith, personal communication.
- 9 J. Mattay, J. Mertes and G. Maas, Chem. Ber., 1989, 122, 327.
- 10 P. A. Grieco, J. J.Nunes and M. D. Gaul, J. Am. Chem. Soc., 1990, 112, 4595.

‡ Single crystal X-ray structure determination of 4: C<sub>20</sub>H<sub>23</sub>NO<sub>3</sub>, M = 325.4, monoclinic,  $P_{21}/c$ , a = 6.194(4), b = 26.405(9), c = 10.847(4)Å,  $\beta = 98.31(4)^\circ$ , V = 1756Å<sup>3</sup>,  $D_c = 1.23$  g cm<sup>-3</sup>, Z = 4.3116 Unique diffractometer data to 20 max = 50° [monochromatic Mo-Kα radiation,  $\lambda = 0.71073$ Å,  $\mu = 0.5$  cm<sup>-1</sup> (no correction)]; 1381 'observed' [ $I > 3\sigma(I)$ ] used in full-matrix least-squares refinement [anisotropic thermal parameters for C, N, O: hydrogen atoms refined in (x, y, z, U<sub>iso</sub>)]. R = 0.051,  $R_w$  (statistical weights) = 0.049;  $T \sim 295$  K.

Atomic coordinates, bond lengths and angles, and thermal parameters have been deposited with the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1.

 $<sup>\</sup>dagger$  <sup>1</sup>H NMR analysis (400 MHz) indicated **2** was contaminated with 5% of **3**.