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## Enantioselective Acylation of Enolates; the reaction of (4*R*)-*trans*-Diethyl 2-Alkyl-2-methoxy-1,3-dioxolane-4,5-dicarboxylates with *E*- and *Z*-Silyl Enol Ethers

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The stereoselectivity of the Lewis acid induced acylation of open-chain silyl enol ethers by chiral orthoesters is strongly affected by the C=C bond configuration: both Z and E silyl enol ethers are acylated in good isolated yields, but the Z isomers give rise to a 1:1 ratio of diastereoisomeric monoprotected 1,3-diketones, while excellent stereoselectivities are obtained with E enols.

The acylation of enolates represents an alternative to the more common aldol condensation for the synthesis of 1,3-difunctional compounds. While the number of methods for the enantioselective aldol condensation has increased recently, very few methods have been concerned with controlling the absolute stereochemistry of the acylation reaction.<sup>1</sup>

Recently we have introduced optically active orthoesters as precursors of chiral acyl cation equivalents, and reported their diastereoselective reaction with trimethylsilyl enol ethers derived from cyclic ketones.<sup>2</sup> Here we show the results obtained with open chain silyl enol ethers, in particular stressing the effect of the configuration of the enolic double bond.

The chiral orthoester 1 is reacted with a trimethysilyl enol ether 2 (2 equiv.) and BF<sub>3</sub>·OEt<sub>2</sub> (2 equiv.) at -60 °C for a few hours, affording a diastereoisomerically enriched monoprotected 1,3-diketone 3, (Scheme 1).

Table 1 shows some representative data obtained from pure  $Z^3$  and  $E^4$  silyl enol ethers.<sup>†</sup> The level of diastereoselectivity

that can be obtained using mixtures of silyl enol ethers prepared via metallation-silylation (using lithium diisopropylamide, in tetrahydrofuran at -78 °C followed by addition of Me<sub>3</sub>SiCl) of the parent ketone is also reported. The yields are only marginally dependent on the configuration of the enolic double bond, but the stereoselectivity is dramatically affected: E silyl enol ethers **2d**, **2e** and **2f** afford diastereoisomeric

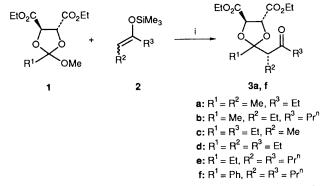
Table 1 Acylation of configurationally defined trimethyl<br/>silylenol ethers  $^a$ 

Product	Config. of 2	Yield $(\%)^b$	D.e. (%) <sup>c</sup>
3a	Z	65	0
3b	Ζ	61	0
3b 3c 3d 3e 3f	E: Z = 7:3	75	72
3d	E	70	90
3e	Ε	60	84
3f	Ε	90	88

<sup>*a*</sup> The reactions were carried out by mixing 1 (1 mmol), 2 (2 mmol) and BF<sub>3</sub>·OEt<sub>2</sub> (2 mmol) in anhydrous CH<sub>2</sub>Cl<sub>2</sub> (10 ml) at -60 °C for 3h. After quenching with sat. NaHCO<sub>3</sub> solution (10 ml), compounds **3a–e** were purified by flash chromatography. <sup>*b*</sup> Yields refer to purified isolated products. <sup>*c*</sup> Determined by NMR.

<sup>&</sup>lt;sup>+</sup> Z-Silyl enol ethers (Z: E > 98:2) were prepared by treatment of the parent ketones with Me<sub>3</sub>SiCH<sub>2</sub>CO<sub>2</sub>Et and cat. Bu<sub>4</sub>N<sup>+</sup>F<sup>-</sup> at -78 °C; pure *E* compounds (E: Z > 98:2) were obtained *via* oxidation of configurationally pure vinyllithium derivatives with Me<sub>3</sub>SiOOSiMe<sub>3</sub>.

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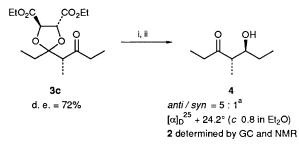


Scheme 1 Reagents and conditions: i, BF<sub>3</sub>·OEt<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, -60 °C

excess (d.e.) values slightly superior to those obtained with cyclic compounds,<sup>2</sup> while Z silyl enol ethers **2a** and **2b** are completely non-stereoselective, giving a 1:1 mixture of diastereoisomers. Of course the behaviour of the E:Z mixture **2c** is intermediate. Investigations aimed at finding a rationale for such a different trend are in progress.

To establish the absolute configuration of the stereocentre generated by the acylation reaction we transformed compound **3c** into 4-methyl-5-hydroxyheptan-3-one by reduction with excess diisobutylaluminium hydride and deprotection with BF<sub>3</sub>·OEt<sub>2</sub> in MeOH–CH<sub>2</sub>Cl<sub>2</sub> as shown in Scheme 2. According to Cram's and Felkin's rules the *anti* product is predominantly formed (*anti*:syn = 5:1).‡ After carefully checking that no epimerization could have occurred in the critical deprotection step§ we concluded that the configura-

§ To eliminate this possibility, *syn*-4-(1-hydroxyhexyl)hexan-3-one, independently prepared in a configurationally pure form according to G. P. Boldrini, F. Mancini, E. Tagliavini, C. Trombini and A. Umani-Ronchi, *J. Chem. Soc., Chem. Commun.*, 1990, 1680, was stirred with a five-fold excess of BF<sub>3</sub>·OEt<sub>2</sub> in MeOH-CH<sub>2</sub>Cl<sub>2</sub> for 24 h. After quenching with aq. NaHCO<sub>3</sub> and extraction with diethyl ether the <sup>1</sup>H NMR spectrum of the product (although some decomposition occurred) allowed any conversion to the *anti* isomer to be excluded.



Scheme 2 Reagents and conditions: i, Bu<sup>i</sup><sub>2</sub>AlH (5 equiv.), Et<sub>2</sub>O, -80 °C; ii, BF<sub>3</sub>·OEt<sub>2</sub>, MeOH, CH<sub>2</sub>Cl<sub>2</sub>, room temp.

tion at C-4 in 4 is the same as in 3c. Since the mixture obtained from the reduction-deprotection sequence showed a positive optical rotation, we attribute the (4S,5S) configuration to the major isomer. In fact both the *anti* and the *syn* isomers of (4S)-4-methyl-5-hydroxyheptan-3-one<sup>5</sup> have a positive specific rotation value.¶

As a consequence we can say that the configuration of 3c is (S) and on this basis we can assume that the Si face of the silylenol ether is preferentially acylated by (R, R) orthoesters.

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¶ All four stereoisomers of 4-methyl-5-hydroxyheptan-3-one have been prepared; they show the following specific rotations:  $(4R,5R)[\alpha]_D^{23} - 37.8^\circ$  (c 1.20 in diethyl ether),  $(4S,5S)[\alpha]_D^{22} + 36.8^\circ$ (c 1.25 in diethyl ether),  $(4R,5S)[\alpha]_D^{20} - 26.7^\circ$  (c 1.52 in diethyl ether),  $(4S,5R)[\alpha]_D^{20} + 27.0^\circ$  (c 1.24 in diethyl ether).<sup>5</sup>

<sup>&</sup>lt;sup>‡</sup> The <sup>1</sup>H NMR spectrum (Varian Gemini-300, C<sub>6</sub>D<sub>6</sub> solution) of the mixture showed two signals at  $\delta$  3.48 (dt, *J* 8.1 and 4.1 Hz) and at  $\delta$  3.64 (m) in a 5:1 integral ratio that were respectively attributed to the *anti* and *syn* isomer by comparison with the values reported in ref. 5. An analogous agreement with the reported values was found in the <sup>13</sup>C NMR spectrum. For 4 [ $\alpha$ ]<sub>D</sub><sup>25</sup> + 24.2° (*c* 0.8 in Et<sub>2</sub>O).