Asymmetric Ring Transformation of *meso*-7-Substituted Bicyclo[3.3.0]octanones into Chiral Bicyclo[3.2.1]octene Derivatives

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Asymmetric ring transformation of *meso*-7-(2-oxoalkyl)bicyclo[3.3.0]octan-3-ones into chiral bicyclo[3.2.1]oct-2-ene derivatives is accomplished by treatment with chiral cyclic (six- or seven-membered ring) 1,2-diols–BF₃–Et₂O.

Generally, the conversion of *meso*-compounds into chiral compounds seems to be difficult, although this conversion can rarely be accomplished by an enzymatic¹ or chemical² procedure. In this communication, we describe the asymmetric ring transformation from *meso*-substrates into chiral compounds, by a method based on the drastic ring transformation reaction developed by our group.^{3–5}

meso-Substrates, 7-substituted bicyclo[3.3.0]octanones I⁵ can be prepared by the conventional method from bicyclo[3.3.0]octane-3,7-dione. Treatment of *meso*-substrates (1 equiv.) with BF₃-Et₂O (3–5 equiv.) and chiral (R,R or S,S)-cyclic-1,2-diols⁶ (2 equiv.) in CH₂Cl₂ at room temperature for 12–24 h afforded the ring transformation products,⁵ in fair to good yields, as an inseparable mixture[†] of diastereo-

isomers (Table 1). As shown in Table 1, in all cases, the ring transformation reaction proceeded in a diastereoselective manner, *i.e.* the substrate with R = Me (entries 1 and 2) was transformed by (S,S)-cyclohexane-1,2-diol or (R,R)cycloheptane-1,2-diol into the corresponding bicyclo-[3.2.1]octene derivatives with 26% diastereoisomeric excess (d.e.) or 27% d.e., respectively. Similarly the substrate with R = Bu^n (entries 3 and 4) was converted into the ring transformation product with 44% d.e. or 36% d.e. In the cases where R = cyclohexyl (entries 5 and 6) and R = Ph (entries 7 and 8), the diastereoselectivity was remarkably improved to afford products with 81-93% d.e. The best result (93% d.e.) (entry 7) was obtained in 63% yield by a combination of substrate (R = Ph) and (S,S)-cyclohexane-1,2-diol. In conclusion, this asymmetric ring transformation seems to be remarkably affected by the bulkiness of substituent (R). There is no precedent for the diastereoselective conversion from meso-compounds into chiral compounds, on the basis of drastic ring transformation. The absolute configuration of the ring transformation product II in entry 1 was determined from

[†] The d.e. of the products in entries 1–6 was estimated directly from the ¹³C NMR spectra, the products in entries 7 and 8 were determined by ¹³C NMR spectra after conversion into (+)-MTPA ester of the alcohol obtained by LiAlH₄ reduction. [MTPA = α -methoxy- α -(trifluoromethyl)phenylacetic acid].



Entry	R	Diol(n)	Yield (%)	D.e. (%)	Abs. config II	$[\alpha]^{25}_{D}$ (CHCl ₃) of III
1	Me	S,S(1)	66	26	1R.5R.7R	-29.3
2	Me	R, R(2)	70	27	15.55.75	+42.1
3	Bu ⁿ	S,S(1)	52	44	1R.5R.7R	-33.8
4	Bu ⁿ	R, R(2)	66	36	15,55,75	+33.3
5	Cyclohexyl	S,S(1)	71	81	1R, 5R, 7R	-54.7
6	Cyclohexyl	R, R(2)	65	81	15,55,75	+54.4
7	Ph	S,S(1)	61	93	1R, 5R, 7R	-54.4
8	Ph	R,R(2)	60	83	15,55,75	+47.7



Scheme 1 Reagents: i, 5% aq. KOH; ii, PhSeCl; iii *m*-chloroperbenzoic acid (*mCPBA*); iv, LiAlH₄; v, *tert*-butyldimethylsilyl chloride (TBDMSCl)-pyridine (Py); vi, PhCOCl-Py

the circular dichroism spectrum of the benzoate‡ derived from II by a sequence of reactions (Scheme 1). Its negative Cotton effect [$\Delta \varepsilon = -1.25$ (231 nm)] indicated that the absolute configuration of the benzoate is 1R,2S,5R,7R,7 i.e. the absolute configuration of II (entry 1) could be concluded to be 1R, 5R, 7R. The absolute configuration of the other products was determined on the basis of the preliminary finding that, in the ¹³C NMR spectrum of II (entry 1), the chemical shift of C-1 with the R configuration appeared at a lower field than that of the S configuration. The stereochemical course of this asymmetric induction is assumed to be as follows. This ring transformation seems to proceed *via* three steps; (i) intramolecular aldol condensation; (ii) acetalization; (iii) Grob fragmentation.5 Acetalization of intermediary enantiomeric aldols by chiral cyclic diol is considered to be a kinetic resolution process (Fig. 1), in which one enantiomer yields



Fig. 1

preferentially the acetal having the desirable orientation to cause the Grob fragmentation.

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[‡] The benzoate was prepared as follows. By conventional basecatalysed hydrolysis, selenolactonization using PhSeCl followed by oxidation with *m*CPBA⁸ and reduction with LiAlH₄, product in entry 1 was converted into the diol. The benzoate was obtained from the diol *via* the protection of primary alcohol with TBDMSCl and subsequent benzoylation of the secondary alcohol.