## 2-Bromoamides as Synthons for Pseudopeptides containing Aminodicarboxy Units

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The monoalkylating and enantioselective behaviour of a chiral 2-bromoamide allows the synthesis of pseudopeptides in which a dipeptide component is changed into an aminodicarboxy moiety, with overall retention of configuration.

The well known role of peptides in biology has prompted a wealth of synthetic and structure–activity studies of partially altered structures, e.g. pseudopeptides where the CONH link is replaced by NHCO, CH<sub>2</sub>NH, CH<sub>2</sub>S, CH<sub>2</sub>CO, CSNH, etc. We considered it interesting to provide a simple access to pseudopeptides where the NH<sub>2</sub> group of an amino acid or peptide provides the NH link of a built-in aminodicarboxy unit. Accordingly, we looked again at the early Fisher peptide synthesis where a 2-bromoacyl amino acid was allowed to react with ammonia, <sup>2a</sup> as well as the early synthesis of 'imino acids.' <sup>2b</sup> Aminodicarboxylic acids (imino acids, opines, alines, etc.), extensively studied in connection with some sea organisms, are obtained by nucleophilic substitution (with inversion

Br 
$$CONH$$
  $CO_2R^3$   
3;  $R^1 = Me$ ,  $R^2 = R^3 = CH_2Ph$  (Bzl)

$$R^4NH$$
 $CONH$ 
 $CO_2R^3$ 
 $C_6H_4OBu^4.p$ 
4;  $R^4 = MeO_2C$ 
5;  $R^4 = allyl$ 

of configuration); $^{2b,3a,b}$  related compounds have been obtained upon amination–reduction (as racemates). $^{4,5}$  Following our recent demonstration that a 2-bromopropanamide reacts slowly with inversion of configuration with representative amines, but faster, and with retention of configuration, with the same amines in the presence of silver oxide, $^6$  we have now found that the free amino group of an amino acid ester substitutes the bromine of a chiral 2-bromoacylamino acid ester in the presence of  $Ag_2O$ , with relevant retention of configuration.

We believe that, under the present conditions, Ag<sub>2</sub>O behaves as a 'coupling agent' and promotes a neighbouring group mechanism, with the observed reaction rates and stereochemistry. The mechanism operating in the homogeneous phase<sup>7a</sup> as well as the role of Ag<sub>2</sub>O are under active investigation.<sup>7b</sup> We report as examples, the synthesis of: (a) two pseudopeptides 1 and 2† where the original Tyr<sup>1</sup>-D-Ala<sup>2</sup> moiety of the opioid tetrapeptide dermorphin<sup>8</sup> and heptapeptide deltorphin-C<sup>9</sup> have been changed into a related aminodicarboxy unit; and (b) a modified peptide 5 carrying a substituted N-terminal allylamino group.

† All new compounds gave satisfactory analytical and spectral data. Selected data for 1: m.p. 179–181 °C;  $[\alpha]_D^{20}$  + 7.6 (c 1, DMF); K' 5.54 (capacity factor) was determined using Vydac  $C_{18}$  with a gradient consisting of two mobile phases: B = 60% acetonitrile in 0.1%  $CF_3CO_2H$ ; A = 10% acetonitrile in 0.1%  $CF_3CO_2H$ . A 25 min linear gradient was run from 0% B to 50% B in 25 min, flow rate 1.0 ml min<sup>-1</sup>,  $\lambda$  = 220 nm; FAB-MS (MH+) m/z 471; amino acid analysis: Phe 0.97, Gly 1.0 [PITC methodology; peptides (50–1000 pmol) were hydrolysed in 200  $\mu$ l 6 mol 1<sup>-1</sup> HCl containing 1% phenol for 1 h at 150 °C].

Selected data for **2**: m.p. 151–153 °C;  $[\alpha]_D^{20} + 10.8$  (c 1, DMF); K' 8.28; FAB-MS (MH+) m/z 784; amino acid analysis: Phe 1.02, Asp 0.97, Val 1.89, Gly 1.0.

Intermediate 4.  $^1$ H NMR (200 MHz, CDCl<sub>3</sub>):  $\delta$  1.12 (d, 3H), 1.32 (s, 9H), 2.69 (d, 2H), 2.81–3.16 (m, 3H), 3.34–3.49 (m, 1H), 3.6 (s, 3H), 4.69–4.8 (m, 1H), 5.17–5.22 (m, 2H) and 6.8–7.37 (m, 15H); yield 80%,  $R_{\rm f}$  0.43 (n-hexane–AcOEt, 4:1 v/v).

*N*-Allylamine **5**. <sup>1</sup>H NMR:  $\delta$  1.19 (d, 3H), 1.25 (br s, 1H), 2.98–3.22 (m, 5H), 4.86–5.2 (m, 4H), 5.6–5.85 (m, 1H), 7.02–7.36 (m, 10H) and 7.66 (d, 1H); yield 89%,  $R_f$  0.2 (n-hexane–AcOEt, 1:1 v/v).

$$C_6H_4OBu^t-p$$
 $MeO_2C$ 
 $NH_2$ 
 $+$  3
 $MeO_2C$ 
 $NH_2$ 
 $MeO_2C$ 
 $MeO_2C$ 
 $NH_2$ 
 $MeO_2C$ 
 $MeO_2C$ 
 $MeO_2C$ 
 $MeO_2C$ 
 $MeO_2C$ 
 $MeO_2C$ 
 $MeO_2C$ 

Scheme 1 Reagents and conditions: i, 6:3: Ag<sub>2</sub>O (molar ratio 2:1:1), room temp., 3 h, toluene, 75%; ii, H<sub>2</sub>, Pd/C, EtOH, room temp., 1 h, 95%; iii, N-(3-dimethylaminopropyl)-N'-ethylcarbodiimide hydrochloride (EDC), Et<sub>3</sub>N, hydroxybenzotriazole (HOBt), dimethylformamide (DMF), H-Gly-NH<sub>2</sub>-HBr or H-Asp(OBu<sup>1</sup>)-Val-Val-Gly-NH<sub>2</sub>, 1<sup>3</sup> –5 °C to room temp., 16 h, 71%; iv, CF<sub>3</sub>CO<sub>2</sub>H (95%), 0 °C, 1 h, 100%. Compounds indicated in the text as 1', 2' etc. bear the opposite configuration of the alanine unit.

(S,S)-2-Bromopropanoyl-Phe-OBzl 3 was prepared by acylating Phe-OBzl with (S)-2-bromopropanoyl chloride. The latter was obtained, in turn, from L-alanine, via the diazonium salt and the 2-bromoacid, with overall retention of configuration. The diastereoisomeric mixture of 3 and (R)-2-bromopropanoyl-Phe-OBzl 3' was obtained, in turn, starting from commercial (R,S)-2-bromopropanoyl bromide (Fluka).

The monoalkylating and enantioselective behaviour of a chiral non-racemic 2-bromoamide synthon becomes apparent when the mixture  $\bf 3$ ,  $\bf 3'$ , or pure  $\bf 3$  is independently allowed to react with Tyr(Bu<sup>t</sup>)-OMe  $\bf 6$ , in the presence of Ag<sub>2</sub>O (Scheme 1). Whereas two diastereoisomeric products, *i.e.* (S,S,S)-4 and (S,R,S)-4' were obtained starting from the diastereoisomeric mixture  $\bf 3$ ,  $\bf 3'$ , only one diastereoisomer was obtained starting from diastereoisomerically pure  $\bf 3$ , as confirmed by careful HPLC screening. We assign the (S,S,S)-configuration to the diastereoisomer  $\bf 4$  arising from  $\bf 3$ , by assuming that the Ag<sub>2</sub>O-promoted substitution of bromine occurs with retention of configuration. We believe that traces of the undesired diastereoisomer are due to impurities in the reagent rather than to a leak in the mechanism of the substitution at the  $C^{\alpha}$ -Br bond.  $^{6.7a.b}$ 

Intermediate 4, including an aminodicarboxylic unit, was debenzylated at the C-terminal benzyl ester function and condensed with the C-terminal part of dermorphin tetrapeptide or deltorphin-C, with no protection of the novel secondary amino group. The resulting 1 and 2 have the final (S,S)-configuration of the new moieties, as shown. The diastereoisomeric 1', 2' with the related (S,R)-configurations were also obtained, starting from the diastereoisomeric mixture 4, 4'; preparative HPLC allowed the separation of the final products.

The N-allylamine 5 was obtained upon reaction of allyl-

amine with 3 or 3, 3'; no HPLC separation could be achieved, in this case, for the diastereoisomeric mixture.

The four compounds 1, 2; 1', 2' were tested in quantitative opioid binding assays: 1' and 2' proved better<sup>11</sup> than compounds containing the D-Tyr<sup>1</sup>-D-Ala<sup>2</sup> unit.<sup>12</sup>

In conclusion, in pseudopeptides such as 1 and 2, the secondary amino function results from the amino acid reacting as a nucleophile with the bromoamide, whereas the peptide linkages result from routine peptide synthesis. The scope and limitations of our findings are under investigation.

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