## Efficient Total Synthesis of Racemic and Optically Active Cyclobut-A and Simple Analogues<sup>†</sup>

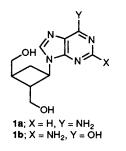
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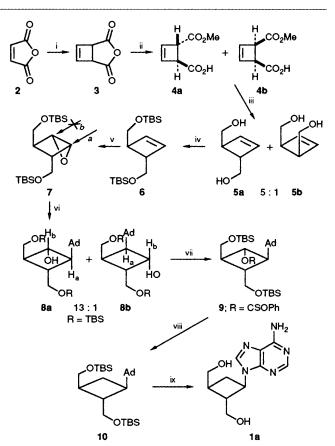
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Rapid synthesis of the potent antiviral agent, cyclobut-A, **1a**, from the inexpensive starting materials maleic anhydride and acetylene, in both racemic and optically active forms, is described; the route also allows for the preparation of several hydroxylated and keto analogues in good overall yield.

In 1989 Honjo and coworkers<sup>1</sup> prepared the unnatural products, cyclobut-A and cyclobut-G, **1a**, **b**, carbocyclic analogues of the known antiviral agent oxetanocin A,<sup>2</sup> and demonstrated that they were strongly antiviral.<sup>1</sup> Since then these compounds have been prepared by a number of different routes<sup>3</sup> and have been shown to exhibit very high activity against HIV infections.<sup>4</sup> We report here a new, very efficient synthesis which affords cyclobut-A **1a** in 20% overall yield from the inexpensive, readily available anhydride **3** and permits the preparation of novel analogues of cyclobut-A, namely the alcohols **11** and **13** and the ketone **14**.

Photocycloaddition of acetylene to maleic anhydride 2 has been carried out by several groups and proceeds in good yield (69–72%) to give the anhydride 3 (Scheme 1).<sup>5</sup> Treatment of 3 with sodium methoxide in methanol effected both methanolysis and epimerization to afford a mixture of the *trans* and *cis* monoesters 4 in which the *trans* isomer greatly predominated ( $\sim$ 5:1).‡ This mixture was not purified but rather directly

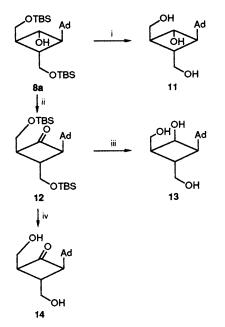




<sup>&</sup>lt;sup>†</sup> Presented at the CATG Meeting, NIH/NIAID, Washington, DC, April 1991 and at the 33rd GECO Conference, Corsica, France, September 1992.

Scheme 1 Reagents and conditions: i, HC=CH, acetone, hv, 69–72%; ii, NaOMe–MeOH; iii, LiAlH<sub>4</sub>, 62% **5a** from **3**; iv, *tert*-butyldimethylsilyl chloride (TBSCl), 1*H*-imidazole, dimethylformamide, 99%; v, *m*-chloroperbenzoic acid, CH<sub>2</sub>Cl<sub>2</sub>, 98%; vi, NaH, AdH, DMSO, 18-crown-6, 114°C, 28 h, 59%; vii, PhOCSCl, 4-dimethylaminopyridine, MeCN, 18 h, 83%; viii, Bu<sup>n</sup><sub>3</sub>SnH, azoisobutyronitrile, heat, dioxane, 30 min, 73%; ix, AcOH–H<sub>2</sub>O (4:1), 90°C, 1.5 h, quant.

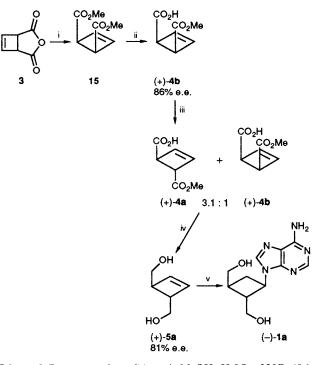
<sup>&</sup>lt;sup>‡</sup> The stereochemistry of the *cis* and *trans* dimethyl 3-cyclobutene-1,2dicarboxylates was easily assigned by heating them to  $\sim 100$  °C for 2 h and analysing the known *E,E*- and *E,Z*-isomers of dimethylhexa-2,4dienoates ( $\sim 5.5:1$ ) by <sup>1</sup>H NMR.



**Scheme 2** Reagents and conditions: i, AcOH- $H_2O(4:1)$ , 90 °C, 1.5 h, 96%; ii, PCC-CH<sub>2</sub>Cl<sub>2</sub>, 75%; iii, (*a*) L-Selectride, tetrahydrofuran (THF), 95% (*b*) AcOH- $H_2O$ , 91%; iv, AcOH- $H_2O$ , 94%

reduced to an easily separable 5:1 mixture of the trans and cis diols 5a,b, from which the desired trans isomer 5a could be isolated in 62% yield for the two steps from 3.6 Protection of this  $C_2$ -symmetric diol as the bis-(*tert*-butyldimethylsilyl)ether 6 and epoxidation furnished the epoxide 7 in nearly quantitative yield for the two steps (only one epoxide is possible due to the  $C_2$  symmetry of the bis-ether). Opening of the epoxide was accomplished with high regioselectivity by treating 7 with adenine (AdH) and sodium hydride in dimethyl sulfoxide (DMSO) in the presence of 18-crown-6 at 114 °C for 28 h. This produced a 13:1 mixture of the all-trans isomer 8a and a minor byproduct, tentatively assigned the trans-cis-trans structure  $\mathbf{8b}$ , in 59% yield (Ad = 7-adenyl). Compound  $\mathbf{8a}$  can be isolated free of its isomer by crystallization from dichloromethane-hexane (52% yield from 7). Not surprisingly, opening of the epoxide via path a is greatly favoured over that via path b due to the much larger steric hindrance associated with the latter mode of attack. The stereochemistry of 8a was tentatively assigned on the basis of the expected favoured approach of attack and the pattern of the coupling constants of the methine protons in the high field <sup>1</sup>H NMR spectrum of 8a  $(H_a: \delta 4.19, dd, J 7.7, 7.2 Hz; H_b: \delta 4.10, dd, J 8.1, 6.9 Hz).^7$ This tentative assignment was confirmed when 8a was converted into cyclobut-A la as follows. Deoxygenation of the secondary alcohol of 8a was carried out by a Barton radical process,<sup>8</sup> namely formation of the *O*-phenyl thiocarbonate 9 (83%) and reduction with tri(n-butyl)stannane (73%) to give the bis-silyl ether of cyclobut-A 10 in 61% overall yield for the two-step operation. Final acid-catalysed desilylation furnished the desired antiviral agent, cyclobut-A la, in quantitative yield, thus ending an efficient eight-step synthesis which proceeds in 20% overall yield from 3. The synthetic material was identical to an authentic sample by <sup>1</sup>H and <sup>13</sup>C NMR.

This synthetic approach allowed us to prepare several analogues of cyclobut-A unavailable by other routes. For example, hydrolysis of the silyl ethers of **8a** afforded in 96%



Scheme 3 Reagents and conditions: i, MeOH,  $H_2SO_4$ , 25 °C, 48 h, 94%; ii, immobilized PLE, acetone, pH 7 buffer, 92%, iii, LDA (2 equiv.), THF, -78 °C;  $H_3O^+$ , 95%; iv, LiAlH<sub>4</sub>; SiO<sub>2</sub>, 53% 5a from 4b; v, as before

yield the triol 11, a hydroxylated analogue of cyclobut-A (Scheme 2). The epimeric triol 13 could also be easily prepared. Oxidation of 8a with pyridinium chlorochromate (PCC) in dichloromethane produced in 75% yield the ketone 12 which was then reduced from the less hindered face with L-Selectride to give the *trans-cis-cis* isomer of 8a. Acidic hydrolysis furnished the desired triol 13, another hydroxylated analogue of 1a. Finally removal of the silyl protecting groups from the ketone 12 afforded compound 14, the keto analogue of cyclobut-A.

We have also completed a simple chiral synthesis of -)-cyclobut-A (-)-1a based on similar chemistry (Scheme 3). Opening of the anhydride 3 with acidic methanol afforded the known cis-diester 15 in high yield.9 This diester contains a plane of symmetry and therefore should be a good candidate for enzymatic enantioselective hydrolysis. Treatment of 15 with pig liver esterase (PLE) at pH 7 gave the desired monoester which was somewhat difficult to separate from the enzyme.10 However the use of PLE immobilized on a modified azlactone mixed polymer¶ in a pH 7 buffer containing 5% acetone afforded the desired optically active monoester (+)-4b in 92% chemical yield with an enantiomeric excess (e.e.) of 86% [determined by 1H NMR measurements of the diastereoisometric (S)- $\alpha$ -methylbenzylamine salts]. All that remained was to isomerize the correct carbonyl group, which was accomplished by treatment with 2 equiv. of lithium diisopropylamide (LDA) to generate the dianion followed by quenching at -78 °C to give a 3.1:1 mixture of the desired trans monoester (+)-4a and the starting material (+)-4b. Direct reduction of the mixture followed by chromatography produced the desired (S,S)-diol (+)-5a in an overall yield of 53% from (+)-4b. The e.e. of this diol was determined to be 81% by examination of the <sup>1</sup>H NMR spectra of the bis-Mosher's esters of both (+)-5a and the racemic compound 5a. The synthesis of optically active (-)-cyclobut-A, (-)-1a, was

<sup>§</sup> The small amount of this material, and the fact that it could not be obtained pure, did not permit a definite assignment of its structure. Although we favour structure  $\mathbf{8b}$ , we cannot rule out other reasonable alternatives, *e.g.* the corresponding 9-adenyl isomer of  $\mathbf{8a}$ .

<sup>¶</sup> Generously supplied by Dr Steve Heilman, 3M, St Paul, MN.

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carried out exactly as described for the racemic series in comparable vields.

We are currently extending this efficient synthesis to the preparation of other similar cyclobutyl nucleosides, e.g. cyclobut-G 1b, and further analogues in both the racemic and optically active series. The analogues 11, 13 and 14 are currently being tested for antiviral activity (both CMV and HIV), the results of which will be reported when they are available.

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|| The  $[\alpha]_D^{25}$  for our synthetic (-)-1a was -11.57 (c 1.0 in water), which compares to the value of -13.5 given by Slusarchyk *e al.* in ref. 3.

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