Facile Synthesis of 2,5-Disubstituted Furans *via* Palladium Complex and Perfluorinated Resinsulfonic Acid Catalysed Isomerization–Dehydration of Alkynediols

Jianguo Ji and Xiyan Lu*

Shanghai Institute of Organic Chemistry, Academia Sinica, 345 Lingling Lu, Shanghai 200032, China

2,5-Disubstituted furans have been synthesized *via* the isomerization–dehydration of alkynediols catalysed by $Pd_2(dba)_3 \cdot CHCl_3$, Bun_3P and perfluorinated resinsulfonic acid at 130 °C in good yields; ($H_4dba = dibenzylideneacetone$).

Substituted furans, which not only occur widely in nature,¹ but also are versatile synthetic intermediates for the preparation of a wide range of cyclic and acyclic organic compounds,^{1d,2} enjoy wide application in a variety of commercially important products, such as pharmaceuticals, flavouring and aroma compounds, insect and fish antifeedants.^{1a,3} The synthesis of substituted furans, which is usually either by intramolecular reactions of 1,4-diketones or by the introduction of substituents to furan rings,^{2b,4} has received wide interest. Recently, we have reported that alkynediols could efficiently isomerize to 1,4-diketones under the catalysis of Pd₂(dba)₃·CHCl₃ 1 and Buⁿ₃P at 80 °C in good yields.⁵ In this paper, we report the synthesis of 2,5-disubstituted furans *via* the isomerization–



Table 1 Isomerization-dehydration of alkynediols catalysed by 1 and 3^a



^{*a*} A mixture of 1 (207 mg, 0.2 mmol), 2 (10 mmol), 3 (232 mg 0.2 mequiv.) and $Bu^{n}_{3}P$ (81 mg, 0.4 mmol) was heated at 130 °C under Ar. ^{*b*} Same as ^{*a*} except that the reaction was carried out under reduced pressure (20 mmHg) and the product was distilled during the reaction. ^{*c*} A mixture of 1 (21 mg, 0.02 mmol), 2 (1 mmol), 3 (23 mg, 0.02 mequiv.), $Bu^{n}_{3}P$ (8 mg, 0.04 mmol) and xylene (10 ml) was stirred at 130 °C under Ar. ^{*a*} The products were confirmed by ¹H NMR, IR and mass spectral data.

dehydration of alkynediols 2 catalysed by 1, $Bu_{3}^{n}P$ and perfluorinated resinsulfonic acid 3 (Nafion-H type, 0.86 mol equiv. Hg^{-1}).

Table 1 shows the results of the isomerization-dehydration of alkynediols catalysed by 1, 3 and $Bu^{n}_{3}P$. For those furans with a boiling point lower than 120 °C (4a-c), the reaction was carried out in the absence of solvent, and the product was distilled during the reaction (entries 1-3). For those furans with a boiling point higher than 170 °C (4d-g), xylene was chosen as the best solvent for the reaction, and the crude product was purified by column chromatography on silica gel (eluent : petroleum ether : ethyl acetate = 20 : 1) (entries 5-8).

This reaction can be applied to the bis(alkynediol) 5 and bis(2,5-disubstituted furan) 6[†] was obtained in 84% yield.

No reaction occurred when 2a was heated either in the absence of 1 or in the absence of 3. Pittman⁶ reported that hexane-2,5-dione can be transformed to 4a by catalysis with Nafion-H under severe conditions for 12 h but only in low yield. So the initial formation of 1,4-diketone in this reaction can be ruled out. A possible isomerization-dehydration mechanism is shown in Scheme 1. Being a superacid, 3 may not only play a role in the formation of the active species [PdH] which catalyse the isomerization of the alkynediol 2 to γ -hydroxy- α , β -unsaturated ketone 7^{5,7} followed by immediate formation of the lactol 8, but also catalyses the dehydration of 8 to yield 4.^{6,8}



This reaction combines the isomerization and dehydration in one pot and offers a convenient route for preparing 2,5-disubstituted furans from easily available alkyne-1,4-diols under mild conditions in good yields.

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⁺ Spectral data for 6: ¹H NMR (60 MHz, CCl₄) δ 5.70 (s, 2H), 2.90 (m, 1H), 2.50 (t, 2H, J 7 Hz), 1.60 (m, 2H), 1.15 (d, 6H, J 6 Hz); IR v/cm⁻¹ (neat) 3100, 1610, 1560, 1460, 1380, 1360, 960, 780; MS *m*/z 274 (M⁺), 232, 190, 164, 150, 136, 123, 105, 79, 66, 57.