

## A Novel Route to Usefully Functionalised Spiro[*n*.4] Systems; Application to a Formal Synthesis of ( $\pm$ )- $\alpha$ -Cedrene

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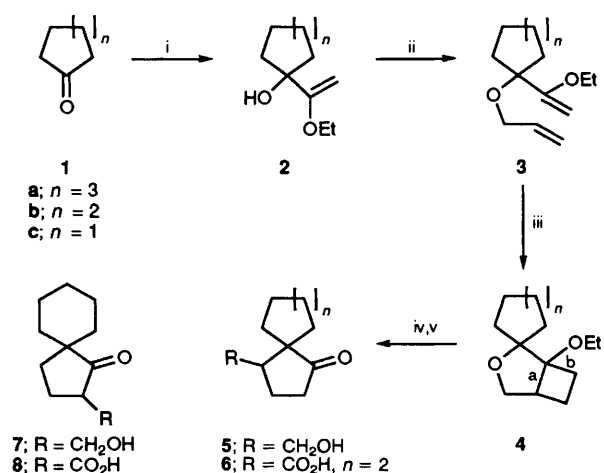
A facile route for the construction of functionalised spiro[*n*.4] systems through pinacol type rearrangement of appropriately designed cyclobutane derivatives with an application to a synthesis of ( $\pm$ )- $\alpha$ -cedrene is described.

Development of new methods for the construction of usefully functionalised spiro systems continues<sup>1</sup> to be the focus of intense interest. 1,4-Bifunctionalised spiro cyclopentanes are particularly useful in providing an entry into a wide variety<sup>2</sup> of natural products in addition to the spiranes. Despite the large number of syntheses<sup>1,3</sup> of spiro natural products, a direct method for constructing spiro subunits at the carbonyl carbon of cyclic ketones remains elusive. Herein we report a facile route for the construction of functionalised spiro[*n*.4] systems via a pinacol type rearrangement of alkoxy-cyclobutane derivatives prepared by intramolecular alkene-enol ether photocycloaddition.

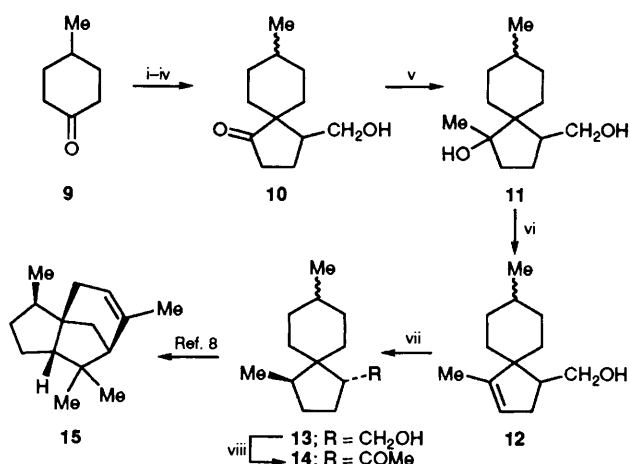
In a representative sequence (Scheme 1), reaction<sup>4</sup> of cycloheptanone **1a** with ethoxyvinyl lithium in tetrahydrofuran (THF) at  $-70^{\circ}\text{C}$  afforded in 81% yield the alcohol **2a**, b.p.  $130\text{--}135^{\circ}\text{C}$  (5 mm Hg). The alcohol **2a** was transformed to the allyl ether derivative **3a**, b.p.  $160\text{--}165^{\circ}\text{C}$  (0.5 mmHg) in 83% yield on treatment with NaH and allyl bromide in refluxing THF-hexamethylphosphoramide (HMPA) (5:1). The diallyl ether derivative **3a** in diethyl ether solution, on irradiation through a quartz immersion well in the presence of  $\text{CF}_3\text{SO}_2\text{Cu}$  as catalyst, underwent smooth cycloaddition<sup>5</sup> to afford after chromatographic purification the oxabicyclo[3.2.0] heptane **4a**<sup>†</sup> in 54% yield. The rearrangement of the four-membered ring in **4a** could be accomplished efficiently by use of trifluoromethanesulfonic acid (TFSA) in dichloromethane at room temperature to afford exclusively the spiro cyclopentanone **5a** in 76% yield. That the rearrangement of **4a** had taken place to produce a hydroxy-cyclopentanone derivative was indicated by its IR absorptions at  $1725$  ( $\text{C}=\text{O}$ ) and  $3420\text{ cm}^{-1}$  ( $\text{OH}$ ) and the disappearance of  $\text{OCH}_2\text{CH}_3$  signals in the  $^1\text{H}$  NMR spectrum of the product. Similarly cyclohexanone **1b** and cyclopentanone **1c** afforded the spiro ketones **5b** and **5c** respectively in good yields.

That the structure of the rearrangement product is **5** as predicted to arise by exclusive migration of the internal bond 'a' having maximum continuous orbital overlap<sup>6</sup> with the p-orbital at the cationic centre was established as follows. The

product obtained by rearrangement of **4b** was oxidised to afford in 86% yield the keto-acid **6**, m.p.  $103^{\circ}\text{C}$ . The failure of the keto-acid thus obtained to undergo decarboxylation excluded the  $\beta$ -keto acid structure **8** which would arise if rearrangement of **4b** had involved migration of the peripheral bond 'b' to produce **7**. Thus, through a four-step sequence, cyclic ketones can be converted very efficiently to spiro[*n*.4] systems with the carbonyl and the hydroxymethylene groups on the five-membered ring disposed in a manner so as to lead entry into a variety of natural products.



**Scheme 1 Reagents and conditions:** i,  $\text{Bu}^t\text{Li}$ , ethyl vinyl ether, THF,  $-70^{\circ}\text{C}$  to room temp., 81–90%; ii, NaH–THF, allyl bromide, HMPA, reflux, 81–86%; iii,  $h\nu$ ,  $\text{Et}_2\text{O}$ ,  $\text{CF}_3\text{SO}_2\text{Cu}$ , 54–60%; iv, TFSA,  $\text{CH}_2\text{Cl}_2$ ,  $-78^{\circ}\text{C}$  to room temp., 2 h, 50–78%; v,  $\text{CrO}_3$ ,  $\text{H}_2\text{SO}_4$ ,  $\text{H}_2\text{O}$ , acetone,  $0^{\circ}\text{C}$  to room temp., 1 h



**Scheme 2 Reagents and conditions:** i,  $\text{Bu}^t\text{Li}$ , ethyl vinyl ether, THF,  $-70^{\circ}\text{C}$ , 81%; ii, NaH–THF, allyl bromide, HMPA, reflux, 89%; iii,  $h\nu$ ,  $\text{Et}_2\text{O}$ ,  $\text{CF}_3\text{SO}_2\text{Cu}$ , 58%; iv, TFSA,  $\text{CH}_2\text{Cl}_2$ , room temp., 76%; v, dihydrofuran, pyridinium toluene-*p*-sulfonate (PPTS),  $\text{CH}_2\text{Cl}_2$ , then MeLi, THF, reflux, then PPTS, MeOH, 67% overall; vi,  $\text{Me}_2\text{SO}$ ,  $165^{\circ}\text{C}$ , 73%; vii,  $\text{H}_2$ ,  $\text{PtO}_2$ , EtOH, 99%; viii,  $\text{Me}_2\text{SO}$ ,  $(\text{COCl})_2$ ,  $\text{NEt}_3$ ,  $\text{CH}_2\text{Cl}_2$ ,  $-65^{\circ}\text{C}$ , then MeLi,  $\text{Et}_2\text{O}$ , then Jones reagent, acetone, 55% overall

<sup>†</sup> Compounds **4**, **5**, **10**, **12** and **13** gave satisfactory microanalytical data. Satisfactory spectral data were obtained for all new compounds.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ) data for selected compounds: **2a**,  $\delta$  (200 MHz) 1.29 (3H, t,  $J$  7 Hz), 1.43–2.06 (13H, m), 3.72 (2H, q,  $J$  6.9 Hz), 3.87 (1H, d,  $J$  2.4 Hz) and 4.18 (1H, d,  $J$  2.6 Hz). **3a**,  $\delta$  (200 MHz), 1.28 (3H, t,  $J$  7 Hz), 1.40–1.80 (8H, m), 1.88 (4H, m), 3.74 (2H, q,  $J$  7 Hz), 3.82 (2H, m), 4.03 (1H, d,  $J$  2.4 Hz), 4.19 (1H, d,  $J$  2.4 Hz), 5.06–5.38 (2H, m) and 5.84–6.10 (1H, m). **4a**,  $\delta$  (200 MHz) 1.18 (3H, t,  $J$  7 Hz), 1.36–1.90 (12H, m), 2.02–2.38 (4H, m), 2.70–2.98 (1H, m), 3.46–3.64 (3H, m) and 3.84 (1H, ABq,  $J$  10 Hz). **5a**,  $\delta$  (100 MHz), 1.32–2.44 (18H, m), 3.56 (1H, q, A of ABX,  $J_{\text{AB}}$  10 Hz,  $J_{\text{AX}}$  3 Hz) and 3.89 (1H, q, B of ABX,  $J_{\text{AB}}$  10 Hz,  $J_{\text{BX}}$  3 Hz). **6** (as methyl ester),  $\delta$  (200 MHz), 1.20–1.82 (11H, m), 2.08–2.70 (3H, m), 3.10 (1H, m) and 3.68 (3H, s). **10**,  $\delta$  (100 MHz) 0.92 (3H, br s), 1.12–2.84 (15H, m) and 3.36–3.96 (2H, m). **12**,  $\delta$  (100 MHz) 0.96 (3H, d,  $J$  6.4 Hz), 1.12–2.72 (16H, m, with d,  $J$  1.7 Hz at  $\delta$  1.73 for the vinylic Me), 3.32–3.96 (2H, m) and 5.3 (1H, br s). **14**,  $\delta$  (100 MHz) 0.68–2.48 (20H, m), 2.14, 2.20 and 2.24 (3H, all s COMe) and 2.52–3.16 (1H, m). The component with a COMe peak at  $\delta$  2.24 comprising ca. 15–20% in this mixture was completely equilibrated with NaOMe–MeOH to produce a 3:1 mixture of the components having COMe peaks at  $\delta$  2.14 and 2.20 indicating the less stable *syn*-orientation of the Me and the COMe groups in this component (cf. ref. 8).

The synthetic potential of this novel strategy is demonstrated by a formal synthesis of ( $\pm$ )- $\alpha$ -cedrene as delineated in Scheme 2. Following the above sequence spiroannulation on 4-methylcyclohexanone **9** afforded the spiro cyclopentanone **10**. Protection of the hydroxy group in **10** as the tetrahydropyranyl ether followed by MeLi addition and deprotection afforded the diol **11** in overall 67% yield. Selective dehydration<sup>7</sup> of the tertiary hydroxy group in **11** was achieved by heating in dimethyl sulfoxide to afford the spiro cyclopentene **12**. Hydrogenation of **12** over Adam's catalyst afforded mainly **13**. Transformation of **13** to the known<sup>8</sup> methyl ketones **14**† through three consecutive steps involving Swern oxidation, MeLi addition and Jones oxidation, dictated that hydrogenation proceeded stereoselectively. The ketones **14** have already been transformed<sup>8</sup> in three steps to  $\alpha$ -cedrene **15**.<sup>9</sup> Thus with the synthesis of **14**, a formal synthesis of  $\alpha$ -cedrene **15** is achieved.

We are grateful to the Department of Science and Technology, Government of India for financial support. G. S. thanks the CSIR for a fellowship.

Received, 19th January 1993; Com. 3/00318C

† The ketones **14** displayed IR, <sup>1</sup>H NMR and mass spectra comparable with those reported (ref. 8).

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