Advantageous Applications of Azabenzotriazole (Triazolopyridine)-based Coupling Reagents to Solid-phase Peptide Synthesis

Louis A. Carpino,* Ayman El-Faham, Charles A. Minor^b and Fernando Albericio* b

Department of Chemistry, University of Massachusetts, Amherst, Massachusetts 01003, USA
 Millipore Corporation, 75A Wiggins Avenue, Bedford, Masschusetts 01730, USA

1-Hydroxy-7-azabenzotriazole (HOAt) and its corresponding uronium and phosphonium salts are shown to be superior to their benzotriazole analogs in solid-phase peptide synthesis, thereby making possible the automated synthesis of peptides containing hindered amino acids.

Since the development by König and Geiger¹ in 1970 of 1-hydroxybenzotriazole (HOBt) as a peptide coupling additive, most common methods for peptide bond formation are carried out in its presence.² HOBt is used either in conjunction with carbodiimides or active esters, or built into a stand-alone reagent in the form of phosphonium³ or uronium⁴ salts. Recently, 1-hydroxy-7-azabenzotriazole (HOAt)⁵ has been described as a superior peptide coupling additive, which enhances coupling yields in solution by about 6–32 times, reduces the loss of chiral integrity by up to 50%, and provides visual indication (yellow-to-colourless) of the reaction endpoint. In this communication, the application of HOAt, as well as its uronium and phosphonium salt derivatives⁺ (Fig. 1) to solid-phase peptide synthesis is described.

In order to demonstrate the suitability of these derivatives and compare their performance relative to that of the corresponding benzotriazole analogs by a solid-phase strategy, several syntheses of fragment 65–74 (H-Val-Gln-Ala-Ala-Ile-Asp-Tyr-Ile-Asn-Gly-NH₂)⁸ of the acyl carrier protein were carried out by a fluorenylmethyloxycarbonyl



(Fmoc)/tert-butyl protection scheme [But for Tyr and Asp, and Trityl (Trt) for Asn and Gln]. Polyethylene glycol-polystyrene (PEG-PS)-resin⁹ bearing a peptide amide linker derived from 5-[4-(Fmoc)-aminomethyl-3,5-dimethoxyphenoxy]valeric acid (PAL)10 was used as a solid support along with a continuous-flow Millipore 9050Plus synthesizer.[‡] Coupling times were shortened and excesses of reagents were reduced in order to emphasize the differences between HOAt- and HOBt-mediated couplings. Under these conditions, incomplete incorporations were detected for Asn onto Gly, Ile72 onto Asn, Ile⁶⁹ onto Asp, and Val onto Gln. The purity of the peptides was judged, after cleavage of the peptides from the resin with trifluoroacetic acid (TFA)-H₂O (9:1) for 2 h at 25 °C, by reversed-phase high performance liquid chromatography (HPLC). The results, collected in Table 1, indicated clearly that (i) HOAt, HATU, and PyAOP are superior to HOBt, HBTU, PyBOP, and PyBroP, respectively (entries 1, 4 vs. 2, 5; 9 vs. 10; 11 vs. 12, 13); (ii) uronium and phosphonium salts are preferred to carbodiimide/active ester methods (entries 7, 8 vs. 1-6); (iii) addition of HOXt to HXTU (X = A, B) couplings did not significantly improve the coupling yields, with the exception of Asn coupling (entries 14, 15 vs. 9, 10); (iv) uronium salts are slightly better than phosphonium analogues (entries 9, 10 vs. 11, 12); and (v) uronium salts differ little among themselves (entries 9, 17-20).§

Equally satisfactory results were obtained when HATU was applied to the solid-phase synthesis of the pentapeptide

Table 1 Distribution of products, including various deletion peptides, according to HPLC analysis^a of the assembly of ACP(65-74) via HOAt- and HOBt-based coupling reagents

Entry	Coupling method	Equiv. ^b	Time/min	ACP	-2Ile	-Ile ⁷²	-Ile ⁶⁹	-Val	-Asn	
1	DIPCDI ^c -HOAt	4	3	65	2	7	9	1	2	_
2	DIPCDI-HOBt	4	3	31	13	15	18	3	1	
3	DIPCDI	4	3	14	4	7	22	2	32	
4	Pfp ^d -HOAt	4	3	82			15	1	1	
5	Pfp-HOBt	4	3	53	1	1	33	5	1	
6	Pfp	4	3	0e				_		
7	HÂTU	4	3	84		_		3	8	
8	HBTUf	4	3	73		1	2	3	5	
9	HATU	1.5	1.5	53	3	6	12	3	16	
10	HBTU	1.5	1.5	18	16	11	19	2	7	
11	PyAOP	1.5	1.5	45	5	7	16	1	12	
12	PyBOP ^g	1.5	1.5	12	19	12	16	1	2	
13	PyBroP ^h	1.5	1.5	0e						
14	HATU-HOAt	1.5	1.5	50	5	9	12	2	8	
15	HBTU-HOBt	1.5	1.5	17	18	12	18	3	3	
16	AOP	1.5	1.5	49	3	6	13	1	18	
17	HAPyU	1.5	1.5	50	4	7	15	2	13	
18	HAPipU	1.5	1.5	50	1	7	14	3	13	
19	HAMDU	1.5	1.5	53	1	6	13	3	13	
20	HAMTU	1.5	1.5	41	1	7	12	7	16	

^{*a*} Reversed-phase C-18 columns were used with elution by a linear gradient over 20 min of 0.1% TFA in MeCN and 0.1% aqueous TFA from 1:19 to 1:4, flow rate 1.0 ml min⁻¹. ^{*b*} Couplings were carried out in DMF in the presence of 2 equiv. of DIEA per equiv. of Fmoc-amino acid/coupling reagent. ^{*c*} N,N'-diisopropylcarbodiimide. ^{*d*} Pentafluorophenyl ester. ^{*e*} None of the desired product was obtained. ^{*f*} O-(benzotriazol-1-yl)-1,1,3,3-tetramethyluronium hexafluorophosphate. ^{*s*} Benzotriazol-1-yloxytris(pyrrolidino)phosphonium hexafluorophosphate.



Fig. 1 Structures of HOAt-based coupling reagents. [Abbreviations for these compounds have generally kept to the style previously used for the HOBt analogues; ^bO-(7-azabenzotriazol-1-yl)-1,1,3,3-tetra-methyluronium hexafluorophosphate; ^cO-(7-azabenzotriazol-1-yl)-1,1,3,3-bis(tetramethylene)uronium hexafluorophosphate; ^dO-(7-azabenzotriazol-1-yl)-1,1,3,3-bis(pentamethylene)uronium hexafluorophosphate; ^dO-(7-azabenzotriazol-1-yl)-1,3-dimethyl-1,3-dimethyl-1,3-dimethyl-1,3-dimethyl-1,3-trimethyleneuronium hexafluorophosphate; ^gO-(7-azabenzotriazol-1-yl)-1,3-dimethyl-1,3-trimethyleneuronium hexafluorophosphate; ^gO-(7-azabenzotriazol-1-yl)-1,3-dimethyl-1,3-trimethyleneuronium hexafluorophosphate; ^gO-azabenzotriazol-1-ylosytris(pyrrolidino)phosphonium hexafluorophosphate; ^hO-(hexafluorophosphate) hexafluorophosphate; ^gO-(hexafluorophosphate) hexafluorophosphate; ^hO-(hexafluorophosphate) hexafluorophosphate]



Fig. 2 HPLC chromatograms of crude peptide mixtures containing H-Tyr-Aib-Aib-Phe-Leu-NH₂ directly after extraction with chloroform and H₂O-acetic acid (7:3) of the cleavage reagent [TFA-H₂O (9:1)]. (A) HATU-HOAt; (B] HBTU-HOBt. Synthetic strategies were as specified in the text and footnote (‡). Reversed-phase C-18 columns were used for the analysis with elution by a linear gradient, over 20 min of 0.1% TFA in MeCN and 0.1% aqueous TFA from 1:19 to 3:7, flow rate, 1.0 ml min⁻¹.

H-Tyr-Aib-Aib-Phe-Leu-NH₂ under the same conditions. This peptide contains the sequence Aib-Aib, which is described as being particularly difficult for both solution¹¹ and solid-phase strategies.^{11,12} After 2 h of coupling using 4 equiv. of either Fmoc-Aib-OH or Fmoc-Tyr(Bu¹)-OH, 4 equiv. of HXTU, and 8 equiv. of DIEA (all remaining couplings were carried out for 30 min), the pentapeptide was obtained with a purity of 94% for the HATU synthesis, and only 43% for HBTU (Fig. 2).

Finally, synthesis of hexapeptide H-D-Ala-MeLeu-MeLeu-MeVal-Phe-Val-OH on a hypersensitive acid-labile (HAL)¹³-

Absorbance (220 nm)

Ó

t/ min Fig. 3 HPLC chromatograms of crude peptide mixtures containing H-D-Ala-MeLeu-MeLeu-MeVal-Phe-Val-OH directly after evaporation of cleavage reagent. (A) HATU-HOAt; (B) HBTU-HOBt. Synthetic strategies were as specified in the text and footnote (¶). Reversed-phase C-18 columns were used for the analysis with elution by a linear gradient over 20 min of 0.1% TFA in MeCN and 0.1% aqueous TFA from 0:1 to 1:0, flow rate 1.0 ml min⁻¹. The peak at 16.1 min corresponds to the title peptide. Peaks at 12.9, 13.7, 14.6 and 15.3 min correspond to H-MeVal-Phe-Val-OH, H-D-Ala-MeVal-Phe-Val-OH, H-MeLeu-MeVal-Phe-Val-OH, and H-D-Ala-MeLeu-

10

5

MeVal-Phe-Val-OH, respectively.

15

20

25

PEG-PS-resin was attempted.¶ The *N*-terminal tetrapeptide section of this peptide, containing three consecutive *N*-methyl amino acids, represents a fragment of cyclosporin.¹⁴ Synthesis of a peptide of this type constitutes an exceptionally difficult challenge even for conventional solution strategies.¹⁵ Incorporation of the last three amino acids required a double coupling protocol for 2 h. Using these conditions and after cleavage of the peptide with TFA–CH₂Cl₂ (5:95) for 1 h, the hexapeptide was obtained in a purity of 85% in the case of HATU, and only 8% for HBTU. In the HBTU synthesis, the major peak (48%) corresponded to the tripeptide H-MeVal-Phe-Val-OH, and the rest of the peaks to various deletion peptides (Fig. 3).

In conclusion, the marked efficiency of 7-azabenzotriazolebased coupling reagents in solid-phase peptide synthesis has been demonstrated. These derivatives enhance reactivity and are especially suited to the preparation of peptides containing hindered amino acids. Previously only Fmoc-amino acid fluorides have been used successfully for the solid phase assembly of peptides bearing contiguous Aib units.¹⁶

This work was supported in part by the National Institutes of Health (GM-09706) and the National Science Foundation (CHE-9003192). Drs G. William Griffin, Steven A. Kates, Hitesh Shroff, Salvatore A. Triolo, and Mr George A. Truran are thanked for fruitful discussions during the present work.

Received, 10th August 1993; Com. 3/04845D

Footnotes

[†] These derivatives were prepared from 1-hydroxy-7-azabenzotriazole according to published methods.^{6,7}

J. CHEM. SOC., CHEM. COMMUN., 1994

‡ All syntheses performed in this work were carried out automatically. The flow rate of the unit pump was set at 5.0 ml min⁻¹ and the following synthetic protocol was used: Fmoc group deprotection with 1,8-diazabicyclo[5,4,0]-undec-7-ene (DBU)-piperidine-N,N-dimethylformamide (DMF) (2:2:96) (6 min), DMF washing (7 min), Fmoc-amino acid coupling (see text and Table 1), and DMF washings (8 min).

§ A standard coupling protocol for routine peptide synthesis comprises 2 equiv. each of Fmoc-amino acid and HATU, and 4 equiv. of DIEA, for 10 min. For Asn and Gln couplings, 2 equiv. of HOAt are added. Using this protocol ACP (65-74) has been obtained with a purity of 96%. For more complex peptides, extended coupling times and extra equivalents of reagents may be advisable depending on the case.

¶ Incorporation of the first Fmoc-amino acid (4 equiv.) onto the hydroxymethyl resin was performed with DIPCDI (4 equiv.), in the presence of 4-(dimethylamino)pyridine (0.4 equiv.) in DMF for 1 h. Application of HOAt derivatives for this purpose is under study.

References

- 1 W. König and R. Geiger, Chem. Ber., 1970, 103, 788.
- 2 For reviews see: in The Peptides: Major methods of peptide bond formation, ed. E. Gross and J. Meienhofer, vol. 1, Academic, New York, 1979; G. B. Fields, Z. Tian and G. Barany, in Synthetic Peptides, A User's Guide, ed. G. A. Grant, W. H. Freeman, New York, 1992, pp. 77-183.
- 3 B. Castro, J. R. Dormoy, G. Evin and C. Selve, Tetrahedron Lett., 1975, 1219; J. Coste, D. Le-Nguyen and B. Castro, Tetrahedron Lett., 1990, 31, 205; E. Frérot, J. Coste, A. Pantaloni, M. N. Dufour and P. Jouin, Tetrahedron, 1991, 47, 259.

- 4 V. Dourtoglou, J. C. Ziegler and B. Gross, Tetrahedron Lett., 1978, 1269; R. Knorr, A. Trzeciak, W. Bannwarth and D. Gillessen, Tetrahedron Lett., 1989, 30, 1927; S. Chen and J. Xu, Tetrahedron Lett., 1992, 32, 647; Y. Kiso, Y. Fujiwara, T. Kimura, A. Nishitani and K. Akaji, Int. J. Peptide Protein Res., 1992, 40, 308; P. Henklein, M. Beyermann and R. Sohr, in Proc. 22nd Eur. Peptide Symp., ed. C. H. Schneider and A. N. Eberle, ESCOM Publishers B. V., Leiden, 1993, pp. 224-225.
- 5 L. A. Carpino, J. Am. Chem. Soc., 1993, 115, 4397.
- 6 J. R. Dormoy and B. Castro, *Tetrahedron Lett.*, 1979, 3321.
 7 V. Dourtoglou, B. Gross, V. Lambropoulou and C. Zioudrou, Synthesis, 1984, 572.
- 8 W. S. Hancock, D. J. Prescott, P. R. Vagelos and G. R. Marshall, J. Org. Chem., 1973, 38, 774; E. Atherton, D. L. J. Clive and R. C. Sheppard, J. Am. Chem. Soc., 1975, 97, 6584
- 9 G. Barany, F. Albericio, N. A. Solé, G. W. Griffin, S. A. Kates and D. Hudson, in Proc. 22nd Eur. Peptide Symp., ed. C. H. Schneider and A. N. Eberle, ESCOM Publishers B.V., Leiden, 1993, pp. 267-268.
- 10 F. Albericio, N. Kneib-Cordonier, S. Biancalana, L. Gera, R. I. Masada, D. Hudson and G. Barany, J. Org. Chem., 1990, 55, 3730.
- 11 C. Auvin-Guette, E. Frérot, J. Coste, S. Rebuffat, P. Jouin and B. Bodo, Tetrahedron Lett., 1993, 34, 2481.
- 12 P. Belton, R. Cotton, M. B. Giles, E. Atherton, J. Horton and J. D. Richards, in Proc. 20th Eur. Peptide Symp., ed. G. Jung and E. Bayer, Walter de Gruyter, Berlin, 1989, pp. 619-621.
- 13 F. Albericio and G. Barany, Tetrahedron Lett., 1991, 32, 1015.
- 14 M. Dreyfuss, E. Härri, H. Hofmann, H. Kobel, W. Pache and H. Tscherter, Eur. J. Appl. Microbiol., 1976, 3, 125
- 15 R. M. Wenger, Helv. Chim. Acta, 1984, 67, 502; U. Schmidt and B. Potzolli, Liebigs Ann. Chem., 1987, 935.
- 16 H. Wenschuh, M. Beyermann, E. Krause, L. A. Carpino and M. Bienert, Tetrahedron Lett., 1993, 34, 3733.