

Novel Behaviour of Dialkylamino-substituted Propargylic Alcohols in Reactions with Arene Ruthenium(II) Complexes

Didier Pilette,^{a,b} Samuel Moreau,^a Hubert Le Bozec,^a Pierre H. Dixneuf,^a John F. Corrigan^b and Arthur J. Carty^b

^a Laboratoire de Chimie de Coordination Organique, URA CNRS 415, Campus de Beaulieu, Université de Rennes, 35042 Rennes, France

^b Guelph-Waterloo Centre for Graduate Work in Chemistry, Department of Chemistry, University of Waterloo, Waterloo, Ontario, Canada N2L 3G1

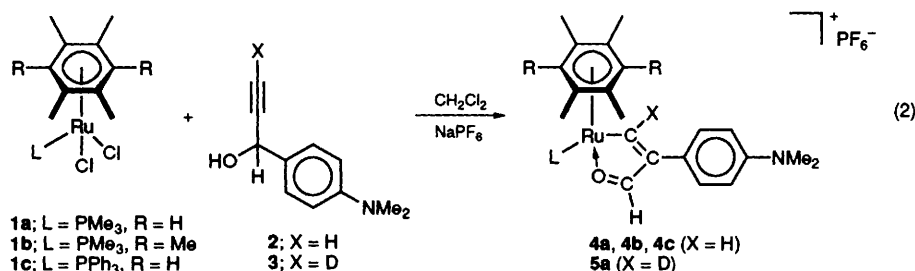
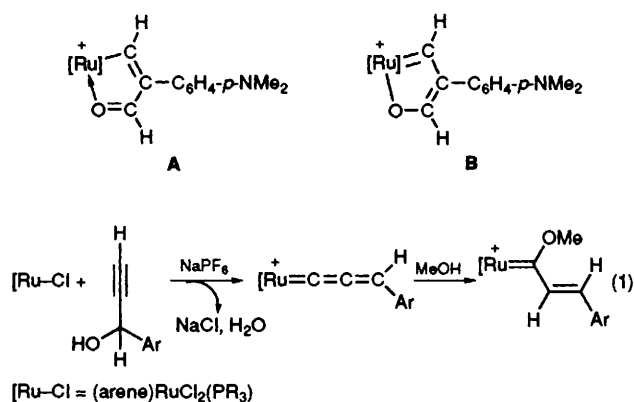
Reaction of arene ruthenium(II) derivatives $(C_6Me_4R_2)RuCl_2(L)$ ($R = H, Me; L = PMe_3, PPh_3$) with the prop-2-yn-1-ol derivative $HC\equiv C-C(H)(OH)(C_6H_4-p-NMe_2)$ in dichloromethane results in the formation of new oxametallacyclic ruthenium(II) complexes $\{(C_6Me_4R_2)Ru[CH=C(C_6H_4-p-NMe_2)CHO](L)\}PF_6^-$ via an unprecedented 1,2-migration of the dimethylaminophenyl substituent; the structure of one compound ($R = H; L = PPh_3$) is determined by an X-ray crystal structure analysis.

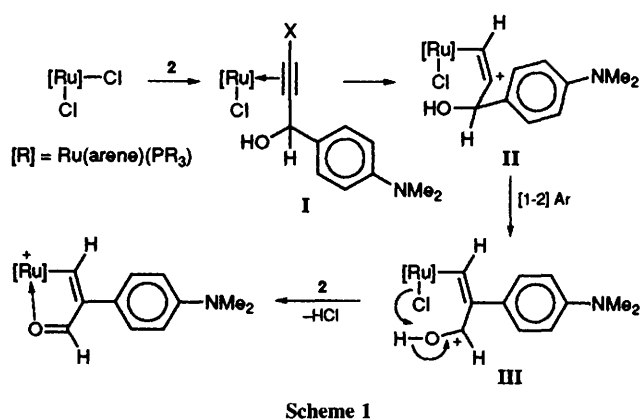
Prop-2-yn-1-ols have recently been shown to be convenient precursors for a variety of unsaturated carbene, allenylidene, alkenyl and polyenyl-carbene compounds, by activation with arene ruthenium(II) complexes.^{1,2} The key step of activation was shown to be the dehydration of the propargylic alcohol and formation of an allenylidene intermediate [eqn. (1)]. In attempting to manipulate the electronic properties of these polyunsaturated carbenes by incorporating electron-releasing substituents onto the hydrocarbon chain, we have discovered a new activation type which leads to ruthenium(II) oxametallacycles based on vinylaldehydes via an unprecedented 1,2-migration of a dialkylaminophenyl substituent of the prop-2-yn-1-ol derivative $HC\equiv C-C(H)(OH)(C_6H_4-p-NMe_2)$ **2**. This rearrangement, which is supported by a deuterium-labelling experiment, has a direct analogy in phenonium ion migrations in organic chemistry³ and can be rationalised in terms of polarisation and charge stabilisation in a slipped η^1 -alkyne complex leading to the preferred aryl over hydrogen migration. These results have relevance to the recent postulate that slipped η^1 -alkyne complexes may be intermediates along one reaction pathway for the rearrangement of η^2 -alkyne complexes to vinylidenes.⁴ Here, we report the generation of the arene ruthenium(II) oxametallacycles **4**, **5**, from the prop-2-yn-1-ol derivative **2** and the X-ray structure of the vinylaldehyde complex **4c**.

Treatment of a dichloromethane solution of $(C_6Me_4H_2)RuCl_2(PMe_3)$ **1a** with 1-(dimethylaminophenyl)prop-2-yn-1-ol **2** (3 equiv.) in the presence of $NaPF_6$ at room temp. resulted in the formation of the new oxametallacycle complex **4a**, isolated in 33% yield as red crystals after purification by column chromatography on alumina, [eqn. (2)]. The arene ruthenium complexes containing a more electron-releasing arene (C_6Me_6) **1b** or a bulkier phosphine (PPh_3) **1c** activate **2** and lead to the vinylaldehyde chelates **4b** and **4c** in 39 and 20% yield, respectively.† A labelling experiment with $DC\equiv C-C(H)(OH)(C_6H_4-p-NMe_2)$ **3** was carried out under similar conditions and compound **5a**, with the deuterium atom uniquely on C(1), was produced in 24% yield. This demonstrates that migration of the terminal hydrogen did not occur during the reaction. ¹H NMR spectra of **4a**, **4b** and **4c** exhibited two low field signals at δ 11–12.5 and δ 8–9, characteristic of the vinyl and aldehyde protons, respectively. On the other hand, only one low field doublet was observed for **5a** at δ 9.06. In the ¹³C{¹H} NMR spectra, the C(1) carbon appeared as a doublet at δ 240 ($^2J_{CP}$ 20 Hz) and as a sextuplet at δ 238 ($^2J_{CP}$ 18 Hz, $^1J_{CD}$ 21.5 Hz) for **5a**.

Details of the molecular structure of **4c** were revealed by a single crystal X-ray structure analysis (Fig. 1).† The principal structural feature of interest is the five membered Ru–O–C₃ metallacyclic ring formed by the rearrangement of the prop-2-yn-1-ol. The ring is planar and the C(1)–C(2) [1.377(6) Å] and O(1)–C(3) [1.260(5) Å] bond lengths suggest that the ligand is a vinylaldehyde coordinated to ruthenium via a Ru–C_{sp}² σ -bond [Ru(1)–C(1) 1.988(4) Å] and the oxygen atom O(1) of an aldehyde group [Ru(1)–O(1) 2.092(3) Å]. The former value is, however, ca. 0.1 Å shorter than the expected value for a Ru–C_{sp}² bond [Ru–C = 0.2061(10) Å in RuH(CHC(Me)CO₂C₄H₉)(PPh₃)₃]⁵ and approaches the bond lengths observed for related ruthenium(II) carbene complexes [Ru–C = 1.98(1) Å in {Ru(η^6 -C₆Me₆)[=C(OMe)–CH=CPh₂](Cl)(PMe₃)₃}PF₆].¹ There is thus a substantial contribution to the ground state of **4c** from the carbene structure B. This observation is supported by the spectroscopic data, with $\delta^{13}C$ for C(1) resonating in the region of a carbene carbon and the ¹H resonance of the attached proton exhibiting a downfield shift.

Although oxametallacycles of ruthenium(II) have been previously prepared, for example by reaction of acetylene





(2)] coupled with the non-migration of acetylenic hydrogen strongly suggests that vinylidene (allenylidene) intermediates are not involved in the formation of 4 and 5. A mechanism consistent with these observations (Scheme 1) involves the rearrangement of an η^2 -alkyne intermediate I to its slipped, polarised η^1 -isomer II bearing a positive charge at C(2). Slipped alkyne species have been postulated by Hoffmann⁴ as putative intermediates in the rearrangement of terminal alkynes to vinylidenes. In the present case however, 1, 2-aryl migration rather than hydrogen transfer is favoured by the electron-releasing dialkylamino substituent in the *para* position of the phenyl ring as is observed for such rearrangements in organic chemistry.⁸ The last step, loss of hydrogen chloride from III, is facilitated by the presence of an excess of the basic dimethylaminophenyl propargylic alcohol 2.

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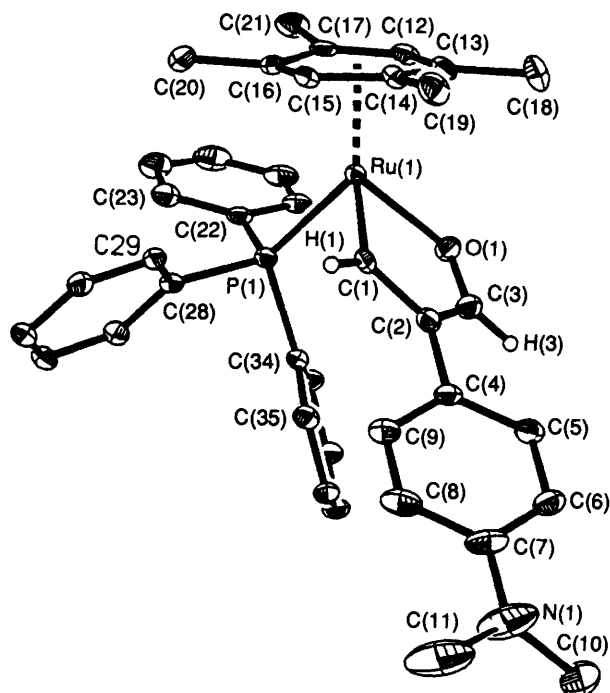


Fig. 1 Molecular structure of $[(C_6Me_4H_2)Ru(CH=C(C_6H_4-p-NMe_2)CHO)(PPh_3)]PF_6$ 4c showing the atomic numbering. Selected bond lengths (Å) and angles (°) are as follows: Ru(1)–P(1) 2.333 (1), Ru(1)–C(1) 1.9888 (4), Ru(1)–O(1) 2.092 (3), C(1)–C(2) 1.377 (6), O(1)–C(3) 1.260 (5), C(2)–C(3) 1.415 (6), C(7)–N(1) 1.389 (8); O(1)–Ru(1)–C(1) 76.9 (1), P(1)–Ru(1)–O(1) 89.1 (1).

dicarboxylates with Ru–H bonds,⁶ or by addition of water to a penta-2,3,4-trienylidene ruthenium arene complex,⁷ vinylaldehyde chelates are not accessible *via* insertion routes.

The distinctively different behaviour of dialkylaminophenyl substituted propargylic alcohols in these reactions [eqns. (1),

Footnote

† Crystal data for $[C_{39}H_{41}NOPRu]PF_6$, monoclinic, $P2_1/n$, $a = 11.122(2)$, $b = 14.078(3)$, $c = 23.061(5)$ Å, $\beta = 98.81(2)^\circ$, $V = 3568.2(12)$ Å³, $Z = 4$, $D_c = 1.520$ g cm⁻³, $F(000) = 1672$, $\mu(Mo-K\alpha) = 0.71073$ Å. Intensity data were collected on an LT-2 equipped Siemens R3m/V diffractometer (ω scans; 200 K). The structure solution (Patterson, Fourier methods) and refinement by full-matrix least-squares analysis were performed using the Siemens SHELXTL PLUS software and based on 5000 observed [$F > 6.0s(F)$] reflections from 6660 measured data. Final R and R_w values were 0.041 and 0.046, respectively. Hydrogen atoms (riding models) were included in the final cycles of refinement with isotropic values. Atomic coordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Information for Authors, Issue No. 1.

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