## Diastereoselective Conjugate Additions Reactions of a Lithiated Allylic Sulfoximine to Acyclic Enones

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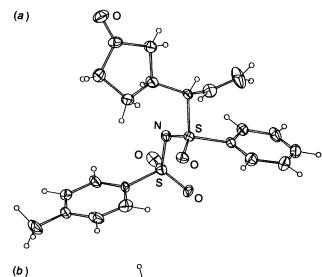
The conjugate addition reactions of lithiated N-p-tosyl S-phenyl S-prop-2-enyl sulfoximine  $\mathbf{4}$  with cyclic and acyclic enones gives exclusively 1,4- $\alpha$  adducts, the reactions with acyclic enones are highly diastereoselective.

In contrast to the chemistry of allylic sulfoxides<sup>1-4</sup> and sulfones,  $^{5.6}$  relatively little is known about the chemistry of allylic sulfoximines.  $^7$  In 1979, Johnson disclosed the synthesis of the first reported allylic sulfoximine. The synthesis of enantiomerically pure allylic sulfoximines has been recently reported,  $^{9.10}$  and Gais has demonstrated they undergo  $S_{\rm N}2$  or  $S_{\rm N}2'$  like displacement reactions with homocuprates. Harmata  $^{11}$  reported that the reaction of lithiated 1 with either 2-cyclopentenone or 2-cyclohexenone gave mixtures in which the 1,4- $\alpha$  adducts were slightly favoured over the 1,4- $\gamma$  adducts. More recently we have reported that the conjugate

Table 1 Conjugate addition reactions of lithiated 4 with enones

Enone		- Diastereoisomeric ratio	Yield (%)
R¹	R <sup>2</sup>		
-(CH <sub>2</sub> ) <sub>2</sub> -		49:33:10:8	87
-(CH <sub>2</sub> ) <sub>3</sub> -		47:25:14:14	92
Ph	Ph	93:7	90
Me	Ph	90:10	45
Ph	Me	94:6	61

addition reactions of lithiated 3 with cyclic and acyclic Michael acceptors give mainly 1,4- $\gamma$  and 1,4- $\alpha$  adducts respectively in THF and 1,4- $\alpha$  and 1,4- $\gamma$  adducts respectively in HMPA-THF.<sup>12</sup> Although the 1,4- $\gamma$  adducts from cyclic enones could be isolated in high diastereomeric purity, these reactions proceeded with modest regioselectivity with respect to  $\alpha$  verses  $\gamma$  attack on the lithiated sulfoximine. Furthermore acyclic enones gave products from 1,2 and 1,4 addition of lithiated 3. Here we report the conjugate addition reactions of lithiated N-p-tolyl S-phenyl S-prop-2-enyl sulfoximine 4 with cyclic and acyclic enones. In contrast to lithiated 1 and 3, the conjugate addition reactions of lithiated 4 are highly regioselective with respect to  $\alpha$  attack on the allylic anion and are highly diastereoselective for acyclic enones. Furthermore, we report the first stereochemical study of this type of reaction



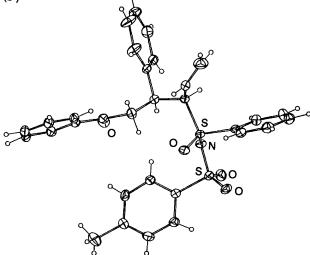


Fig. 1 Molecular projections of (a) 5 and (b) 6; 20% thermal ellipsoids are shown for the non-hydrogen atoms, hydrogen atoms having arbitrary radii of  $0.1~\text{\AA}$ 

from single crystal X-ray structural analyses of two of the reaction products.

Racemic allyl sulfoximine 4 (mp 82-83 °C) was prepared in 81% yield from sulfoximine  $2^{12}$  by treatment with p-toluenesulfonyl chloride (1.2 equiv.) and pyridine (1.2 equiv.) in dichloromethane at 22 °C for 1 h. Addition of n-butyllithium (Bu<sup>n</sup>Li, 1.1 equiv.) to a solution of 4 in THF at -78 °C gave an immediate yellow-orange solution of lithiated 4. After 15 min, the solution was treated with the enone (1.2 equiv.). After 3 min at -78 °C the almost colourless reaction mixture was quenched with acetic acid (1 equiv.) and then an aqueous solution of saturated ammonium chloride. The diastereoselectivities of these reactions were determined by <sup>1</sup>H NMR analysis of the crude reaction mixtures. The reaction products were purified by column chromatography on silica gel with ethyl acetate-hexanes as eluent and the chemical yields were determined on purified samples. The results of these investigations are summarized in Table 1.

In contrast to the chemistry reported for lithiated 1 and 3, lithiated 4 gave exclusively 1,4-α adducts with both cyclic and acyclic enones. Interestingly the regiochemistry of the reaction of lithiated 4 with enones is also different to that of lithiated allyl phenyl sulfone which gives exclusive 1,4-y adducts with cyclic enones and  $1,4-\alpha$  adducts with acyclic enones.5

When 2-cyclopentenone and 2-cyclohexenone were treated with lithiated 4, the  $1,4-\alpha$  adducts were obtained but as a mixture of the four possible diastereoisomers (Table 1). The relative  $(3S^*, 1'R^*, SS^*)$  stereochemistry of the major diastereomeric adduct 5 from the former reaction was secured by a single crystal X-ray structural analysis as shown in Fig. 1a. The relative stereochemistry of the major adduct from the reaction of lithiated 4 and 2-cyclohexanone is assumed to be the same as that in 5 on the basis of its similar <sup>1</sup>H NMR

In contrast, the reaction of lithiated 4 and the acyclic enones, benzylideneacetophenone, benzalacetone and (E)-1phenylbut-2-en-1-one were highly diastereoselective (Table 1). The relative  $(3R^*, 4R^*, SS^*)$  stereochemistry of the major diastereomeric adduct 6 from the reaction of lithiated 4 and benzylideneacetophenone was determined by a single crystal X-ray structural analysis (Fig. 1b). The relative stereochemistry of the major adducts from the reaction of lithiated 4 and benzalacetone and (E)-1-phenylbut-2-en-1-one is assumed to be the same as that in 6 on the basis of their similar <sup>1</sup>H NMR

The stereochemical outcome of these reactions with respect to the stereogenic centre  $\alpha$  to the sulfoximine group can be rationalised as arising from attack on the carbanion whose structure is as shown in 7 (only the monomeric species is considered) that may be similar to that of an α-lithiated benzyl sulfone or sulfoximine. 13,14 The α-substituent (CH.::CH<sub>2</sub>) of the sulfoximine would be expected to be anti to the bulky

N-p-tosyl group. Electrophilic attack on 7 should occur anti to the S-phenyl group and syn to lithium. The overall stereochemical outcome of these reactions can be rationalised as arising from the chelated transition states 8 and 9 in which the two bulky groups of each reaction partner, the sulfoximidoyl group and the  $\beta$ -enone substituent, are anti to minimise steric interactions.

Financial support by the Australian Research Council is gratefully acknowledged.

Received, 22nd October 1993; Com. 3/06322D

## **Footnote**

† Crystal data for 5:  $C_{21}H_{23}NO_4S_2$ , a = 21.534(5), b = 5.867(3), c =18.288(5) Å,  $\beta = 116.17(2)^{\circ}$ , V = 2074 Å<sup>3</sup>.  $2\theta_{\text{max}} = 47.5^{\circ}$ ; N = 2671,  $N_{\rm o} = 1177$ ; R = 0.055,  $R_{\rm w} = 0.050$ .  $(x, y, z, U_{\rm iso})_{\rm H}$  included constrained at estimated values.

6: C<sub>31</sub>H<sub>29</sub>NS<sub>2</sub>O<sub>4</sub>, a = 12.205(4), b = 12.566(6), c = 19.669(6) Å, β = 112.71(2)°, V = 2783 Å<sup>3</sup>. 2θ<sub>max</sub> = 60°; N = 4804,  $N_o = 2825$ ; R = 0.0000.052,  $R_{\rm w} = 0.053$ .  $(x, y, z, U_{\rm iso})_{\rm H}$  all refined.

Unique, room temp. diffractometer data sets (T ca. 295 K, 20/0 scan mode, monochromatic Mo-K $\alpha$  radiation,  $\lambda = 0.71073$  Å) yielding N independent, gaussian-absorption-corrected data,  $N_0$  with  $I > 3\sigma(I)$  used in the full-matrix least-squares refinement (anisotropic non-hydrogen thermal parameter form, statistical reflection weights). Both structures monoclinic,  $P2_1/c$ .

Atomic coordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Information for Authors, Issue No. 1, 1994.

## References

- 1 D. A. Evans and G. C. Andrews, Acc. Chem. Res., 1974, 7, 147.
- 2 D. J. Antonjuk, D. D. Ridley and M. A. Smal, Aust. J. Chem., 1980, 33, 2635
- 3 M. R. Binns, R. K. Haynes, A. G. Katsifis, P. A. Schober and S. C. Vonwiller, J. Am. Chem. Soc., 1988, 110, 5411.
- 4 D. H. Hua, S. Venkataraman, M. J. Coulter and G. Sinai-Zingde, J. Org. Chem., 1987, 52, 719.
- 5 G. A. Kraus and K. Frazier, Syn Commun., 1978, 8, 483.
- 6 M. Hirama, Tetrahedron Lett., 1981, 22, 1905.
- 7 For a recent review on diastereoselective reactions of sulfoximines see: S. G. Pyne, Sulfur Rep., 1992, 12, 57.
- 8 C. R. Johnson, E. U. Jonsson and A. Wambsgams, J. Org. Chem., 1979, 44, 2061.
- 9 J. Bund, H.-J. Gais and I. Erdelmeier, J. Am. Chem. Soc., 1991, 113, 1442
- 10 M. Reggelin and H. Weinberger, Tetrahedron Lett., 1992, 33,
- 11 M. Harmata and R. J. Claassen II, Tetrahedron Lett., 1991, 32, 6497.
- 12 S. G. Pyne and G. Boche, Tetrahedron, 1993, 49, 8449.
- 13 G. Boche, Angew. Chem., Int. Ed. Engl., 1989, 28, 277.
  14 S. G. Pyne, B. Dikic, B. W. Skelton and A. H. White, Aust. J. Chem., 1992, 45, 807.