## Stereoconvergent 'One-Pot' Tandem [2,3]-Wittig-Anionic Oxy-Cope Rearrangement of Acyclic Bis-Allylic Ethers in the Diastereoselective Synthesis of Substituted Tetrahydropyrans

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The unsaturated alcohols derived from a 'one-pot' tandem [2,3]-Wittig-anionic oxy-Cope (AOC) rearrangement undergo a halocyclisation reaction with iodine in acetonitrile to give substituted tetrahydropyrans with a high degree of stereocontrol.

Tandem reactions provide an opportunity for linking the synthetic power of two or more transformations in a single synthetic operation. Sigmatropic rearrangements such as the [2,3]-Wittig,<sup>2</sup> and the AOC<sup>3,4</sup> have been widely used to set up stereocentres by rearrangement through predictable five- and six-membered transition states. Nakai and coworkers have noted the synthetic potential of sequential [2,3]-Wittig-AOC rearrangements for acyclic stereocontrol and asymmetric transmission but, as with their earlier work, there are no true tandem reactions.5 We report here the first example of a 'one-pot' tandem [2,3]-Wittig-AOC rearrangement, which can be considered a homologated version of the Ireland ester enolate Claisen rearrangement without allylic transposition.<sup>6</sup> We have shown it to be a stereoconvergent reaction and to proceed with a high degree of acyclic stereocontrol. The methodology was applied to the diastereoselective synthesis of substituted tetrahydropyrans which are important structural sub-units in many natural products.7

The acyclic bis-allylic ether substrates 6 and 7 for the tandem reaction were synthesised from hex-1-yne as shown in Scheme 1. Deprotonation of hex-1-yne with Bu<sup>n</sup>Li followed by reaction with isobutyraldehyde gave the propynyl alcohol 1 in excellent yield. Stereoselective semi-hydrogenation of the propynyl alcohol 1 by aluminium hydride reduction or catalytic hydrogenation delivered the (E)- and (Z)-allylic alcohols respectively. Red-Al reduction<sup>8</sup> gave the (E)-allylic alcohol 3 with 100% geometric purity whilst the (Z)-allylic alcohol 5 could be obtained with 98% geometric purity by hydrogenation of the acetate 2 over palladium on barium sulfate.9 [Catalytic hydrogenation of the propynyl alcohol 1 with the same catalyst resulted in a lower (Z)-geometric purity of 93%; this could be increased slightly to 96% by use of Lindlar's catalyst in hexane-hexene. [10] Tetrabutylammonium iodide catalysed alkylation of the sodium alkoxide of the

H——Bu<sup>n</sup>

Bu<sup>n</sup>

Bu<sup>n</sup>

Bu<sup>n</sup>

RO

1; R = H
2; R = OAc

5; R = H

Vi

Ph

Ph

Scheme 1 Reagents and conditions: i, Bu<sup>n</sup>Li, isobutyraldehyde, THF, -78 °C, 3 h, 95%; ii, Ac<sub>2</sub>O, Et<sub>3</sub>N, DMAP, CH<sub>2</sub>Cl<sub>2</sub>, room temp., 22 h, 93%; iii, Red-Al, Et<sub>2</sub>O, reflux, 19 h; potassium sodium tartrate, 86%, [100% (*E*)-isomer]; iv, H<sub>2</sub>, 1 atm, Pd-BaSO<sub>4</sub>, MeOH, quinoline, 90%; v, K<sub>2</sub>CO<sub>3</sub>, H<sub>2</sub>O-MeOH, room temp., 16 h, 64%, [98% (*Z*)-isomer]; vi, NaH, cinnamyl bromide, Bu<sub>4</sub>NI (cat), 40 °C, 20 h, 76% 6 and 83% 7

alcohols 3 and 5 with cinnamyl bromide in THF delivered the bis-allylic ethers 6 and 7 in high yield

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The tandem [2,3]-Wittig-AOC reaction was accomplished by treatment of the bis-allylic ethers 6 and 7 with potassium hydride and 18-crown-6 in Me<sub>2</sub>SO (Scheme 2).<sup>11</sup> Reaction was typically complete after 1 h at room temp. even in the absence of 18-crown-6 (although the selectivity was slightly lower without the sequestering agent).† Analysis of the δ,ε-unsaturated aldehyde products 8 by <sup>1</sup>H NMR showed the tandem reaction to be stereoconvergent, with both geometric isomers **6** and **7** giving the same major (E)-product. The reactions also gave a minor (E)-product and a minor (Z)-product. All products resulted from initial regioselective deprotonation at the unsubstituted carbon atom, as reported previously by Nakai et al.<sup>12</sup> The stereochemistry of the [2,3]-Wittig-AOC aldehyde products 8 was analysed using Nakai's transition states for rearrangement of acyclic AOC substrate.<sup>11</sup> Both the major (E)-aldehyde and the (Z)-aldehyde were expected to have syn relative configuration from rearrangement via chair transition states. The minor (E)-product, with opposite (anti) relative configuration, is likely to be formed by rearrangement through a boat transition state. Hydrogenation of the double bond confirmed that the major (E)-isomer and the (Z)-isomer had the same relative configuration.‡ About 15% yield from the reaction was made up of isomers formed by another pathway, possibly a [1,2]-Wittig-AOC tandem reaction. The isomeric mixture of aldehydes from the tandem reaction was converted to the corresponding alcohols 9 by treatment with sodium borohydride in methanol.

The syn relative configuration of the major isomer from the tandem reaction was proved by conversion of a 75:17:10 mixture of isomeric unsaturated alcohols Esyn-9, Zsyn-9 and Eanti-9, obtained from a tandem reaction of the (E,E)-bisallylic ether 6, to iodotetrahydropyrans 10 (Scheme 3). The base-catalysed iodocyclisation reaction was accomplished

Scheme 2 Reagents and conditions: i, KH, 18-crown-6, Me<sub>2</sub>SO, room temp., 1 h, 44% (yield of 8 from 7); ii, NaBH<sub>4</sub>, MeOH, 0°C, 30 min, 57% (yield of 9 from 6)

Scheme 3 Reagents and conditions: i, I<sub>2</sub>, NaHCO<sub>3</sub>, MeCN, -23 °C-room temp., 24 h, 45%; ii, Bu<sub>3</sub>SnH, AlBN (cat), THF, room temp., 16 h, 97%

Fig. 1 Transition states for formation of minor tetrahydropyran isomer 12

using iodine and sodium hydrogen carbonate in acetonitrile, conditions for kinetic control. <sup>13</sup> Recovery of the starting material from the cyclisation reaction showed preferential reaction of the (E)-isomers. The <sup>1</sup>H NMR of the iodotetra-hydropyran product showed one major product (>80% of the mixture) and two stereoisomers; the major isomer was isolated by flash chromatography. Assignment of the <sup>1</sup>H NMR spectrum by decoupling and COSY experiments showed this isomer to be the all-equatorial iodotetrahydropyran from cyclisation through a transition state in which the iodonium cation is equatorial. The methine proton on the phenylbearing carbon,  $H_a$ , exhibited two large couplings (J 11.5 Hz) to the *trans*-diaxial protons  $H_b$  and  $H_c$ , and a smaller coupling to the equatorial proton  $H_d$  (J 4 Hz). Thus, the isopropyl and phenyl groups must be syn in the acyclic molecule.

De-iodination of the original mixture of iodotetrahydropyran stereoisomers with tributyltin hydride gave just two tetrahydropyrans 11 and 12 in a 4:1 ratio. The major isomer 11 was shown to be derived from the major iodotetrahydropyran isomer by comparison of the  $^{1}$ H NMR spectra. Examination of the  $^{1}$ H NMR spectrum of the minor isomer 12 showed it to have arisen from the tandem reaction product with *anti* relative configuration; the signal for the axial methine proton  $H_a$  (see assignment for 10) showed one large coupling to the diaxial proton  $H_c$  (J 12 Hz) and two small couplings to equatorial protons  $H_b$  and  $H_d$  (J 5 Hz). The axial orientation of the n-pentyl group was confirmed by an NOE experiment; irradiation of the axial proton  $H_a$  resulted in no enhancement of the signal due to the methine proton on the n-pentyl-bearing carbon,  $H_{\rm e}$ , (this contrasted with irradiation of the same proton in the major iodotetrahydropyran isomer which showed a strong enhancement of the axial methine proton resonance). We suggest that the minor isomer is formed via a cyclisation transition state,  $T_1$ , in which the iodonium ion is actually equatorial (Fig. 1) as the alternate transition state,  $T_2$ , would be destabilised by the presence of two axial substituents. The observed minor tetrahydropyran conformation must therefore be the thermodynamic product.

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## **Footnotes**

† Typical experimental procedure: KH (2-3 equiv.) was washed with THF to remove mineral oils, dried, and transferred under argon to a Schlenk flask holding an argon atmosphere before being treated with dry Me<sub>2</sub>SO (80 equiv.). After effervescence had subsided, 18-crown-6 (1.5 equiv.) was added to the clear, homogeneous solution. After 5 min a solution of the bis-allylic ether in a small volume of Me<sub>2</sub>SO was added and the deep-purple solution was stirred for ca. 1 h at room temp, before being poured onto an ice-brine mixture. After acidification to pH 7 with 1 mol dm<sup>-3</sup> HCl, the aqueous layer was extracted several times with ethyl acetate and the combined organic extracts were washed with H<sub>2</sub>O to remove Me<sub>2</sub>SO. Drying (Na<sub>2</sub>SO<sub>4</sub>) and concentration in vacuo gave the crude tandem reaction product. Analysis of the stereoselectivity of the reaction by GC was carried out on the crude reaction mixture. The aldehyde products 8 could be obtained pure by flash chromatography, eluting with ethyl acetatelight petroleum (bp 40-60 °C) containing triethylamine, or reduced directly to the alcohols 9.

‡ Hydrogenation of a 58:13:13:16 mixture of Esyn-9, Zsyn-9, Eanti-9 and other isomers with Adam's catalyst gave a 72:18:10 ratio of saturated alcohols; calculated ratio, 71:16:13.

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