

Enantioconvergent Preparation of the A-Ring Precursors of Calcitriol from Either (*R*)- or (*S*)-Epichlorohydrin

Kazuki Tazumi and Kunio Ogasawara*

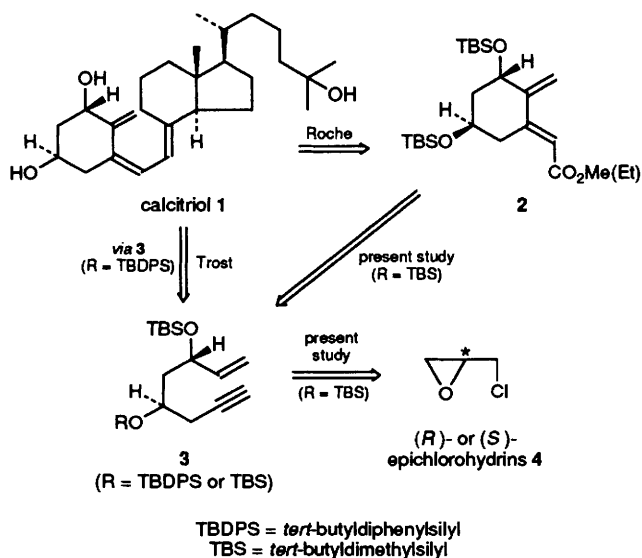
Pharmaceutical Institute, Tohoku University, Aobayama, Sendai 980-77, Japan

Linear and cyclic A-ring precursors of calcitriol have been prepared in both synthetically and enantiomerically convergent ways starting from either (*R*)- or (*S*)-epichlorohydrin.

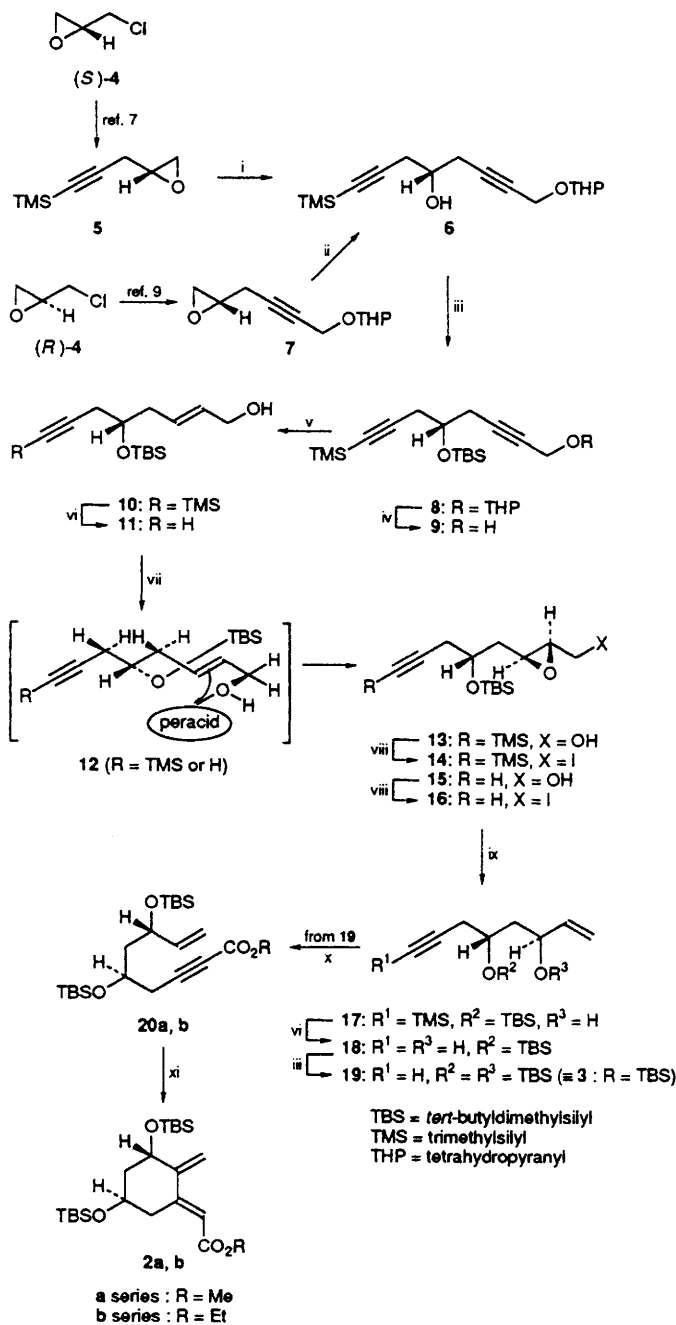
Medicinal importance of calcitriol **1** and the related vitamin D derivatives has stimulated a great deal of recent interest in development of synthetic routes to these molecules.^{1,2} In particular, most of the synthetic efforts have been devoted to the construction of the A-ring moiety because of the difficulty of the conversion from the naturally occurring steroidal precursors.³ Generally, the A-ring moiety has been elaborated from either cyclic precursors such as **2**⁴ or linear precursors such as **3**^{1a,5} though both have never been prepared from a common starting material.⁶ Herein, we describe the first synthetically and enantiomerically convergent approach to the both types of the A-ring precursors of calcitriol **1** utilizing either (*R*)- or (*S*)-enantiomer of epichlorohydrin **4** (Scheme 1).

Exposure of the five-carbon chiral epoxyacetylene⁷ **5**, obtained from (*S*)-epichlorohydrin [(*S*)-**4**], to the prop-1-ynyl ether in the presence of *n*-butyllithium and boron trifluoride etherate⁸ gave the diynol† **6** in 67% yield. The same compound could also be prepared from (*R*)-epichlorohydrin [(*R*)-**4**] in four steps (>70% overall) via the six-carbon epoxyacetylene **7** on sequential reaction with prop-1-ynyl tetrahydropyranyl ether⁹ and trimethylsilylacetylene. The diynol **6** was then transformed into the primary alcohol **9**, [α]_D³⁰ + 9.0 (*c* 1.01, MeOH), in 88% yield via the silyl ether **8** on sequential protection and deprotection.¹⁰ Treatment of **9** with an excess (\approx 5 equiv.) of sodium bis(2-methoxyethoxy)-aluminum hydride¹¹ in ether afforded a readily separable mixture of the (*E*)-allylic alcohols, **10**, [α]_D²⁸ + 3.9 (*c* 0.92, MeOH), and **11**, [α]_D²⁵ -6.0 (*c* 0.86, CHCl₃), in yields of 65 and 32%, the former of which was transformed into the latter (97%) on stirring with methanolic potassium carbonate.¹²

Very fortunately, the oxidation of **10** with *m*-chloroperoxybenzoic acid (MCPBA) in dichloromethane proceeded in a stereospecific way to give the desired β -epoxide **13**, [α]_D³² -43.3 (*c* 1.28, CHCl₃), in 73% yield accompanied by the separable α -epoxide in 7% yield. The observed stereochem-



Scheme 1



Scheme 2 Reagents and conditions: i, HC \equiv CCH₂O-THP, LiBuⁿ, BF₃·OEt₂, THF, -78°C, 2.5 h; ii, TMSC \equiv CH, LiBuⁿ, BF₃·OEt₂, THF, -78°C, 2.5 h; iii, TBS-Cl, imidazole, DMF, room temp., 6 h; iv, cat. PPTS, MeOH, room temp., 17 h; v, Na(MeOCH₂CH₂O)₂AlH₂, Et₂O, 0°C → room temp., 1 h; vi, K₂CO₃, MeOH, 40°C, 30 min. vii, MCPBA, NaHCO₃, CH₂Cl₂, 0°C, 2 h; viii, I₂, PPh₃, imidazole, THF-MeCN (4:1), room temp.; ix, activated Zn, cat. AcOH, MeOH, sonication, 40°C, 1 h; x, a, LiBuⁿ then ClCO₂Me, THF, -78°C, 2 h; b, Li⁺ then ClCO₂Et, THF, -78°C, 2.5 h; xi, [Pd₂(dba)₃]-CHCl₃ (20 mol%), pivalic acid (50 mol%), Ph₂PCH₂CH₂PPh₂ (10 mol%), benzene, 50°C, \approx 22 h

ical outcome may be due to the bulky TBS ether at the homoallylic centre which forced the allylic hydroxy group to be the opposite face in the transition state **12**, directing the oxidant in a favourable way for the present purpose. Both diastereomeric epoxides could also be generated selectively from **10** by employing the Katsuki–Sharpless asymmetric epoxidation conditions:¹³ thus, the β -epoxide **13** was formed selectively in quantitative yield in the presence of diisopropyl L-tartrate, while a 11:1 mixture of the α - and the β -epoxides was formed in an excellent total yield in the presence of diisopropyl D-tartrate.

Having introduced the requisite stereogenic centres, **13** was first transformed into the iodide **14**, $[\alpha]_D^{31} -26.3$ (*c* 1.19, CHCl₃), in 85% yield on exposure to iodine in the presence of triphenylphosphine and imidazole.¹⁴ The iodide **14** was then treated with activated zinc powder in methanol containing acetic acid ($\approx 3\%$) under sonication to give rise to the enynol **17**, $[\alpha]_D^{28} -26.0$ (*c* 1.06, CHCl₃), in 87% yield, which on detrimethylsilylation with methanolic potassium carbonate¹² afforded the terminal enynol **18**, $[\alpha]_D^{27} -32.3$ (*c* 1.15, CHCl₃), in 97% yield. Finally, **18** was transformed into the di-TBS ether **19**, $[\alpha]_D^{30} -10.3$ (*c* 1.50, CHCl₃), in 94% yield which may be taken as an equivalent of the Trost intermediate^{1a,5} [3: R = *tert*-butyldiphenylsilyl (TBDPS)].

Quite similarly, the detrimethylsilyl product **11** could also be transformed into the terminal enynol **18** in a comparable overall yield. Thus, the oxidation of **11** with *m*-chloroperbenzoic acid afforded a separable mixture of the desired β -epoxide **15**, $[\alpha]_D^{27} -51.4$ (*c* 1.04, CHCl₃), and the diastereomeric α -epoxide in yields of 74 and 5%. The major product **15** was transformed into the iodide **16**, $[\alpha]_D^{29} -26.9$ (*c* 1.17, CHCl₃), in 84% yield as above, which on reductive treatment left **18** in 87% yield.

In order to transform the linear precursor **19** into the cyclic intermediate **2**, **19** was treated with methyl chlorocarbonate in the presence of *n*-butyllithium to give the methyl propiolate **20a**, $[\alpha]_D^{28} -4.7$ (*c* 1.15, CHCl₃), in 78% yield. Similarly, the ethyl propiolate **20b**, $[\alpha]_D^{25} -3.5$ (*c* 1.23, CHCl₃), could be obtained from **19** in 81% yield. After considerable examination employing palladium-based conditions,^{1a,15} we have found that cycloisomerization of **20** could best be carried out in the presence of tris(dibenzylideneacetone)dipalladium–chloroform (1/1) [Pd₂(dba)₃·CHCl₃]¹⁶ (0.2 equiv.), ethylenebis(diphenylphosphine) (0.1 equiv.) and pivalic acid (0.5 equiv.) to give the desired dialkylidenecyclohexanes, **2a**, $[\alpha]_D^{25} -13.8$ (*c* 0.86, CHCl₃) and **2b**, $[\alpha]_D^{26} -5.3$ (*c* 0.56, EtOH) [lit.:^{4b} $[\alpha]_D^{25} -4.7$ (*c* 0.5, EtOH); $[\alpha]_D^{23} -4.9$ (*c* 0.5, EtOH);^{1b} $[\alpha]_D -4.2$ (*c* 0.48, EtOH)^{1c}], in yields of 64 and 69%, respectively (Scheme 2).

In conclusion, we have shown an efficient route to the linear and the cyclic A-ring precursors of calcitriol in both synthetically and enantiomerically convergent ways starting from either (*S*)- or (*R*)-enantiomer of epichlorohydrin.

We would like to express our gratitude to DAISO Co. Ltd., Osaka, Japan for donation of a substantial amount of (*S*)- and

(*R*)-epichlorohydrins (>99% ee) and to Professor Seiichi Takano for kind encouragement.

Received, 19th April 1994; Com. 4/02331E

Footnote

† Satisfactory analytical (combustion and high resolution mass) and spectral (IR, ¹H NMR, Mass) data were obtained for all new isolable compounds.

References

- For recent full accounts describing references of most of the previous work, see: (a) B. M. Trost, J. Dumas and M. Villa, *J. Am. Chem. Soc.*, 1992, **114**, 9836; (b) K. Nagasawa, H. Ishihara, Y. Zako and I. Shimizu, *J. Org. Chem.*, 1993, **58**, 2523; (c) C. Chen and D. Crich, *Tetrahedron*, 1993, **49**, 7943.
- For a recent review, see: S. R. Wilson and A. Yasmin, *Stud. Nat. Prod. Chem.*, 1992, **10**, 43.
- B. Lythgoe, *Chem. Soc. Rev.*, 1980, 449.
- (a) E. G. Baggolini, J. A. Iacobelli, B. M. Hennessy and M. R. Uskokovic, *J. Am. Chem. Soc.*, 1982, **104**, 2945; (b) E. G. Baggolini, J. A. Iacobelli, B. M. Hennessy, A. D. Batcho, J. F. Sereno and M. R. Uskokovic, *J. Org. Chem.*, 1986, **51**, 3098.
- B. M. Trost and J. Dumas, *J. Am. Chem. Soc.*, 1992, **114**, 1924.
- Recently an interesting approach based on the same strategy has appeared by the synthesis of the A-ring precursors lacking 1-hydroxy group, see: J. M. Nuss, M. M. Murphy, R. A. Rennels, M. H. Heravi and B. J. Mohr, *Tetrahedron Lett.*, 1993, **34**, 3079.
- S. Takano, T. Kamikubo, T. Sugihara and K. Ogasawara, *Tetrahedron Asymmetry*, 1992, **3**, 853; S. Takano, T. Kamikubo, T. Sugihara, M. Suzuki and K. Ogasawara, *Tetrahedron Asymmetry*, 1993, **4**, 201.
- M. Yamaguchi and I. Hirao, *Tetrahedron Lett.*, 1983, **24**, 391.
- K. C. Nicolaou, G. Skokotas, P. Maligres, Z. Zuccarello, E. J. Sweiger, K. Toshima and S. Wendeborn, *Angew. Chem., Int. Ed. Engl.*, 1989, **28**, 1272.
- M. Miyashita, A. Yoshikoshi and P. A. Grieco, *J. Org. Chem.*, 1977, **42**, 3772.
- Y. Gao and K. B. Sharpless, *J. Org. Chem.*, 1988, **53**, 4081. Both lithium aluminium hydride and diisobutylaluminium hydride gave a mixture containing non-acetylenic compounds.
- P. A. Wender, J. A. McKinney and C. Mukai, *J. Am. Chem. Soc.*, 1990, **112**, 5369; J. Suffert, *Tetrahedron Lett.*, 1990, **31**, 7437.
- cf.* R. A. Johnson and K. B. Sharpless, in *Comprehensive Organic Synthesis*, ed. B. M. Trost and I. Fleming, Pergamon, Oxford, 1990, vol. 7, p. 389.
- P. J. Garegg and B. Samuelsson, *J. Chem. Soc., Chem. Commun.*, 1979, 978.
- B. M. Trost, D. C. Lee and F. Rise, *Tetrahedron Lett.*, 1989, **30**, 651; A. Knierzinger, A. Grieder and P. Schonholzer, *Helv. Chim. Acta*, 1991, **74**, 517; B. M. Trost and Y. Shi, *J. Am. Chem. Soc.*, 1993, **115**, 9421, 12491; B. M. Trost and O. J. Gelling, *Tetrahedron Lett.*, 1993, **34**, 8233.
- T. Ukai, H. Kawazura, Y. Ishii, J. J. Bonnet and J. A. Ibers, *J. Organomet. Chem.*, 1974, **65**, 253.
- G. Stork, D. Hutchinson, M. Okabe, D. Parker, C. S. Ra, F. Ribereau, T. Suzuki and T. Zebovitz, *Pure Appl. Chem.*, 1992, **64**, 1809.