

A Novel Reaction of Coordinated Vinylidenes: Coupling with Hydrogen Sulfide to give a η^1 -Thioaldehyde

Claudio Bianchini,* Lionel Glendenning, Maurizio Peruzzini, Antonio Romerosa and Fabrizio Zanobini

Istituto per lo Studio della Stereochimica ed Energetica dei Composti di Coordinazione, CNR, Via J. Nardi 39, 50132 Firenze, Italy

The Ru^{II} fragment [(PNP)RuCl₂] assists the coupling of phenylacetylene with H₂S to give 2-phenylethanethial [PNP = CH₃CH₂CH₂N(CH₂CH₂PPh₂)₂] in a reaction that is initiated by alk-1-yne to vinylidene tautomerism at ruthenium, followed by electrophilic attack of H₂S on the vinylidene ligand; 2-phenylethanethial is recovered as either η^1 -S-ligand or *endo*-6-benzyl-1-thiabicyclo[2.2.1]hept-3-ene ligand, and the latter molecule is formed *via* a stereoselective Diels–Alder reaction between cyclopentadiene and the η^1 -2-phenylethanethial complex.

The chemistry of transition metal vinylidenes has been intensely investigated over the past two decades.¹ Interest in these compounds stems from both their wide potential in organic and organometallic syntheses² and their implication in several homo- and hetero-geneous catalytic processes.³ The reactivity of metal vinylidene complexes towards small inorganic molecules, *e.g.* O₂,⁴ H₂O⁵ and heteroallenes,⁶ remains an undeveloped although attractive area of organometallic chemistry. However, with reference to the large availability of these strategic inorganic molecules as well as the ubiquity of vinylidene derivatives along the series of transition metals, investigation of these reactions remains an exciting and prosperous area of vinylidene chemistry.

Here we report on the reaction between a Ru^{II} vinylidene complex and H₂S, which provides unprecedented access to η^1 -2-phenylethanethial.

Heating *mer,trans*-[(PNP)RuCl₂(PPh₃)][†] **1** in THF with an excess of phenylacetylene at reflux results in quantitative precipitation of the vinylidene *fac,cis*-[(PNP)RuCl₂{C=C(H)Ph}][†] **2** as pale orange needles crystals, while triphenylphosphine remains in solution. Bubbling H₂S into a refluxing CH₂Cl₂ solution of **2** for 2 h quantitatively converts the vinylidene complex to *fac,cis*-[(PNP)RuCl₂{ η^1 -S=CH(CH₂Ph)}][†] **3**.[†] The presence of a η^1 -2-phenylethanethial ligand in **3** is unambiguously shown by both ¹³C{¹H} and ¹H NMR spectroscopy[†] [δ C_α 230.34; δ C_β 54.14; δ H_α 12.10, to ³J(HH) 7.9 Hz] and the reaction with cyclopentadiene. Compound **3** in CH₂Cl₂ reacts at room temperature with an excess of 1,3-cyclopentadiene to give *fac,cis*-[(PNP)RuCl₂{ η^1 -*endo*-SC₆H₇(CH₂Ph)}][†] **4** as dark red microcrystals.[†] The Diels–Alder reaction with the diene is stereoselective and

occurs with a preference for the *endo* isomer, as shown by the chemical shift and coupling constants of H₆ in the η^1 -S-6-benzyl-1-thiabicyclo[2.2.1]hept-3-ene ligand [δ H₆ 4.91, td, ³J(H₆H₅) 4.0, ³J(H₆H_{Bz}) 8.0 Hz].^{8,9} The phenylethanethial/cyclopentadiene adduct can be removed from the metal by simply heating a CHCl₃ solution of **4** to reflux. As a result, the unsaturated metal fragment [(PNP)RuCl₂] dimerizes to the known face-sharing bioctahedral complex [Ru₂(μ -Cl)₃(PNP)₂]Cl **5**,⁷ while *endo*-6-benzyl-1-thiabicyclo[2.2.1]hept-3-ene[†] can be isolated by TLC (Al₂O₃, eluent Et₂O). The dimer **5** can also be obtained by refluxing **3** in CHCl₃. However, in the absence of a specific reagent for trapping the phenylethanethial, this spontaneously oligomerises.¹⁰

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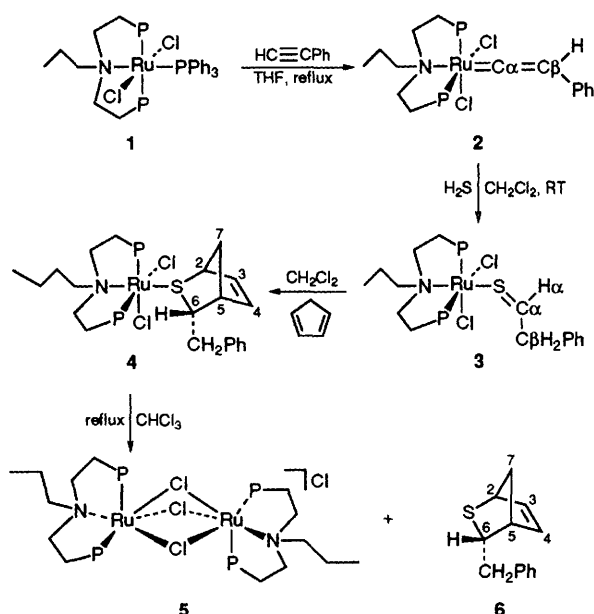
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Footnote

[†] Satisfactory spectroscopic data was obtained for all compounds.

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