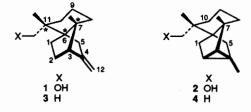
First Total Synthesis of (\pm) -Myltayl-4(12)-ene and Single-crystal X-ray Structure of *exo*-12-Normyltaylan-4-yl 4-Nitrobenzoate

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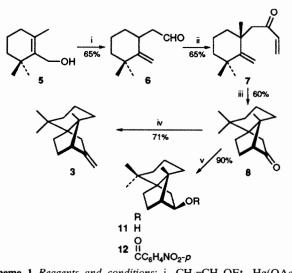
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The total synthesis of the unusual sesquiterpene (\pm) -myltayl-4(12)-ene **3** starting from the readily available cyclogeraniol **5** and the single-crystal X-ray structure of the 4-nitrobenzoate **12** of the noralcohol **11** are described.

Recently, Matsuo and coworkers have reported the isolation of novel, irregular sesquiterpene alcohols, myltaylenol 1^1 and cyclomyltaylenol 2^2 from the liverwort *Mylia taylorii* (Hook, S. Gray) and identified the new carbon frameworks present in them as myltaylane and cyclomyltaylanes. Later Wu and Chang isolated³ cyclomyltaylene **4** from the Taiwanese liverwort *Bazzania tridens*. A characteristic of the structure of myltaylane is the presence of a unique 2,4,4,7a-tetramethylperhydro-1,3a-ethanoindene carbon framework (cyclohexane fused to C-1, C-7 bond of norbornane) incorporating three contiguous quaternary carbon atoms (*), making it a challenging synthetic target. In continuation of our interest in the synthesis of sesquiterpenes containing multiple contiguous quaternary carbons,⁴ herein we describe the first total synthesis of myltayl-4(12)-ene **3** employing biogenetically patterned cation mediated cyclisation and rearrangements as key operations.

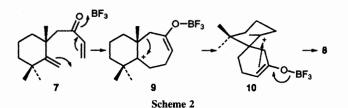


The synthetic sequence starting from cyclogeraniol is depicted in Scheme 1. The starting material cyclogeraniol 5 was obtained^{4b} by selective ozonation of the commercially available β -ionone followed by direct reduction of ozonide with sodium borohydride. One-pot Claisen rearrangement of



Scheme 1 Reagents and conditions: i, $CH_2=CH-OEt$, $Hg(OAc)_2$, toluene, sealed tube, 170 °C, 36 h; ii, $CH_2=CH-MgBr$, THF, room temp., 1 h, then PCC, NaOAc, CH_2Cl_2 , room temp., 0.5 h; iii, $BF_3 \cdot OEt_2$, CH_2Cl_2 , 0 °C, 20 min; iv, $Ph_3P+Me -Br$, $K^+ -O-Amyt$, Amyt-OH, C_6H_6 , room temp., 10 h; v, NaBH₄, MeOH, room temp., 45 min, then 4-NO₂C₆H₄COCl, Py, CH₂Cl₂, DMAP, room temp., 5 h

cyclogeraniol 5 using ethyl vinyl ether and mercuric acetate at 170 °C for 36 hours (sealed tube) furnished the enal 6^5 in 65% yield. Addition of vinylmagnesium bromide to the aldehyde followed by oxidation of the resultant allyl alcohol with buffered (NaOAc) pyridinium chlorochromate (PCC) in methylene chloride transformed the enal 6 into the key intermediate of the sequence, dienone 7, in 65% overall yield. Treatment of the dienone 7 with a catalytic amount of boron trifluoride etherate in methylene chloride furnished the normyltayl-4-one 8 (2,4-DNP derivative mp 166–168 °C), in



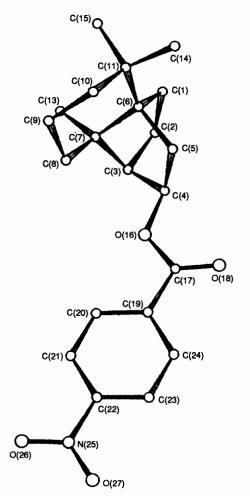


Fig. 1 X-Ray crystal structure of the p-nitrobenzoate 12

60% yield, whose structure was derived from the spectral data[†] in comparison with that of myltaylenol,¹ and mechanistic rationale. Finally Wittig methylenation (methyltriphenylphosphonium bromide and potassium *tert*-amylate (KOAmy) in benzene-*tert*-amyl alcohol) of norketone 8 furnished the myltayl-4(12)-ene 3[†] in 71% yield. The formation of the norketone 8 from the dienone 7 can be rationalised as depicted in Scheme 2, which also explains the stereochemical outcome at carbon C-7. First acid catalysed cyclisation of the dienone 7 generates the bicyclic tertiary carbonium ion 9, which rearranges to the spiro system 10. Reketonisation *via* cyclisation from the α -face of the carbonium ion centre furnishes the norketone 8 with β -methyl at C-7.

Since alternative pathways exist for the carbonium ions 9 and 10 for further rearrangements and cyclisation leading to different tricyclic systems, a single-crystal X-ray structure determination was undertaken to unambiguously establish the structure of the norketone 8. Thus stereospecific reduction of the norketone 8 with sodium borohydride furnished the *exo* alcohol 11, mp 123–25 °C, which was converted into the corresponding 4-nitrophenyl ester 12, mp 156–57 °C. Repeated crystallisation of the *p*-nitrobenzoate 12 from methylene chloride–hexane (1:10) afforded suitable crystals.‡

In conclusion we have achieved the first total synthesis of (\pm) -myltayl-4(12)-ene **3** starting from the readily available cyclogeraniol **5**. The X-ray crystal structure of the *p*-nitrobenzoate **12** confirms the structure of the norketone **8** and hence the myltayl-4(12)-ene **3**.

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Footnotes

[†] All the compounds exhibited spectral data consistent with their structures. Selected spectral data for 8: v_{max}/cm^{-1} (neat) 1745: δ_{H} (400 MHz, CDCl₃) 2.49 (1 H, dd, J 18.6, 3.3 Hz, H-5 *exo*), 1.77 (1 H, d, J 18.6 Hz. H-5 *endo*), 1.3–2.2 (11 H, m), 1.08 (3 H, s), 1.03 (3 H, s),

0.84 (3 H, s); 8c (22.5 MHz, CDCl3) 216.3 (s, C=O), 61.8 (d, C-3), 52.7 (s), 46.1 (2 C, s and t, ?), 35.8 (t), 33.5 (s), 30.2 (t), 28.6 (q), 26.7 (t), 23.0 (q), 22.3 (t), 18.7 (2 C, t and q). For 3: v_{max}/cm^{-1} (neat) 3070, 1660, 870, δ_H (90 MHz, CDCl₃) 4.71 (1 H, q, J 2 Hz) and 4.53 (1 H, q, J 2 Hz) (C=CH₂), 2.54 (1 H, d, J 16.5 Hz), 1.1-2.1 (12 H, m), 1.01 (3 H, s), 0.96 (3 H, s), 0.81 (3 H, s); δ_c (100 MHz, CDCl₃) 154.7 $(C=CH_2)$, 101.3 $(C=CH_2)$, 52.8, 46.9 and 33.6 (3 × quaternary C), 57.7, 40.4, 36.5, 30.2, 28.7, 27.9, 27.6, 23.2, 19.2, 19.1. \ddagger Crystal data: C₂₁H₂₇NO₄; M = 357.449, triclinic, space group $P\overline{1}$, a = 7.058(5), b = 7.655(4), c = 18.435(9) Å, $\alpha = 90.15(4)$, $\beta = 95.39(5)$, $\gamma = 107.62(5)^{\circ}$. V = 945(1) Å³, Z = 2, $D_c = 1.2567$ g cm⁻³, F(000) = 384, T = 291 K, Mo-K α ($\lambda = 0.7107$ Å). Intensity data were collected on an Enraf-Nonius CAD-4 diffractometer using a transparent colourless crystal of dimensions 0.4 \times 0.35 \times 0.3 mm in the $\omega\text{-}2\theta$ mode. A total of 3620 measured, 3177 unique and 2417 observed reflections with $|F_{\rm o}| > 5.0\sigma(F_{\rm o})$ were collected. The structure was solved by direct methods⁶ and full-matrix least-squares refinement⁷ with the non-hydrogen atoms anisotropic and hydrogen atoms isotropic, and converged at R = 0.071. The structure of the p-nitrobenzoate 12 thus determined is shown in Fig. 1. Atomic

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