Total Syntheses of (+)- and (-)-Nonactate Esters using Silicon Compounds to Control the Stereochemistry

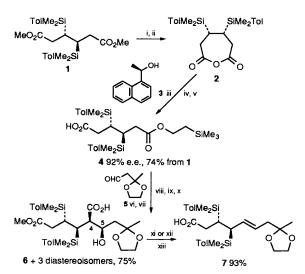
Ian Fleming* and Sunil K. Ghosh

University Chemical Laboratory, Lensfield Road, Cambridge, UK CB2 1EW

1,3-Transposition, using the epoxidation of the allylsilane **7**, derived from the meso 3,4-disilylated adipate **1**, gives the 1,4-related stereocentres of 8-oxo-6-hydroxy-3-dimethyl(tolyl)silylnonanoic acid ethylene acetal **10**; two highly stereocontrolled sequences from this common intermediate give methyl (+)-nonactate **13** and benzyl (-)-nonactate **17**.

Nonactic acid and its derivatives, such as the methyl ester **13**, with four stereogenic centres having 1,2-, 1,3- and 1,4-relationships, have been popular targets for demonstrating new methods of stereocontrol.¹⁻⁴ We have developed versatile methods, using organosilicon chemistry, to control 1,2- and 1,3-related centres,⁵ like those in nonactic acid, but hitherto we have had no method for the 1,4-related centres that it challenged us to. We now report a synthesis of the homochiral 6-hydroxy-3-silylnonanoic acid **10**, having 1,4-related stereocentres, and the conversion of this common intermediate into the nonactate esters **13** and **17**, one in each enantiomeric series, illustrating the versatility of our silicon-based methods of stereocontrol.⁶

We reported earlier⁷ an efficient and completely stereosclective synthesis of the 3,4-disilylated adipic ester 1 from the corresponding methyl β-silylacrylate, using Inanaga's samarium(11) coupling.8 We converted this meso diester into the meso anhydride 2, and opened it with Heathcock's homochiral alcohol 3,9 enriched to a high level (99.7% e.e.) of enantiomeric purity by Horeau's method, 10 as already reported.11 The inseparable diastereoisomeric esters were present in a ratio of 96:4, and gave therefore the homochiral mono ester 4 in 92% e.e. Using one of our allylsilane syntheses,¹² we converted this ester into the *E*-allylsilane 7. In the first step, aldol condensation with Kelly's aldehyde 513 gave largely the diastereoisomer 6, but with all four possible isomers detectable in a ratio of 76:9:9:6. We separately treated the two diastereoisomers with a syn relationship between the substituents on C-4 and C-5 (the 76% and 9%) with benzenesulfonyl chloride, and heated the derived β -lactone to induce a syn stereospecific decarboxylative elimination, and, without separating them from each other, we treated the two with an anti

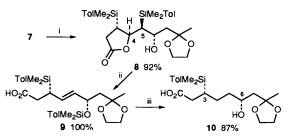


Scheme 1 Reagents and conditions: i, LiOH, THF, MeOH, H₂O, room temp., 48 h; ii, DCC, CH₂Cl₂, room temp., 12 h; iii, DMAP, CH₂Cl₂, -30° C, 4 d: iv, HOCH₂CH₂SiMe₃, DCC, DMAP, room temp., 12 h; v, H₂, Pd/C, EtOAc, room temp., 7 d; vi, 2LDA, THF, DMPU, $-78^{\circ} \rightarrow -50^{\circ}$ C: vii, -78° C, 2.5 h; viii, CH₂N₂; ix, separate diasterosisomers: x, TBAF, THF, 0° C \rightarrow room temp., 0.5 h; xi, (a) PhSO₂Cl, Py, 0° C, 20 h (b) collidine, reflux, 14 h; xii, Me₂NCH(OCH₂Bu¹)₂, CHCl₃, reflux, 7 h; xiii, KOH, MeOH, THF, H₂O, 45°C, 10 h

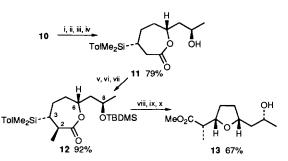
relationship with dimethylformamide dineopentyl acetal to induce an *anti* stereospecific decarboxylative elimination. This is the first time that we have used the full convergent possibilities of this synthetic method, and the overall yield of the *E*-allylsilane 7 from the ester **4** was a gratifying 68% (Scheme 1).

Epoxidation of the allylsilane (Scheme 2), a reaction known to be stereospecifically *anti*,¹⁴ gave the lactone **8** as a consequence of precedented¹⁵ silyl migration from C-4 to C-5. The selective elimination of the silyl group on C-5 and the lactone oxygen on C-4 took place on treatment with potassium hydride, with both the chemoselectivity and the stereoselectivity for the *trans* product **9** having good precedent in the work of Yamamoto.¹⁶ Hydrogenation of this silyl ether to give the key intermediate **10** took place only after the *O*-silyl group had been removed, in line with earlier difficulties we have had in trying to hydrogenate allylsilanes. Thus we had controlled the 1,4-related centres between C-3 and C-6, in 12 steps, in an overall yield of 40% from the ester **1**.

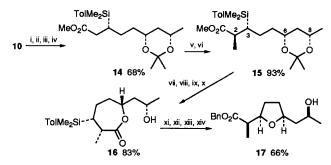
From this branch point, we carried out two sequences to give each of the nonactate esters. In the first (Scheme 3), we used Evans' method¹⁷ for reducing a 3-hydroxyketone to an *anti* 1,3-diol (*ca.* 95:5), which we separated and treated with Mukaiyama's reagent¹⁸ to give the lactone **11**. At this stage we were able to filter off a small amount of the crystalline racemic



Scheme 2 Reagents and conditions: i, MCPBA, Na₂HPO₄, CH₂Cl₂, room temp., 12 h; ii, KH, THF, 78 °C \rightarrow 0 °C, 5 h, iii, H₂, PtO₂, MeOH, room temp., 50 h



Scheme 3 Reagents and conditions: i, PyH, OTs, Mc₂CO, H₂O, reflux, 7 h; ii. Mc₄N, BH(OAc)₃. AcOH, MeCN, -25 °C, 48 h; iii, 2-chloro-*N*-methylpyridinium iodide, Et₃N, CH₂Cl₂, reflux, 10 h; iv, separate from racemate (92% e.e. \rightarrow 99% e.e.); v, TBDMSCl, DMF, imidazole, room temp. 15 h; vi, LDA, THF, DMPU, -78 °C, 1 h; vii, MeI, -78 °C, 36 h, -10 °C, 8 h; viii, KBr, AcO₂H, NaOAc, AcOH, room temp., 15 h; ix, TsCl, Py, DMAP, room temp., 5 d; x, TsOH, MeOH, room temp., 4 d



Scheme 4 Reagents and conditions: i, CH_2N_2 ; ii, PyH, OTs, Me_2CO , H_2O , reflux, 2.5 h; iii, NaBH₄, Bu₂BOMe, THF, MeOH, $-78 \,^{\circ}C$, 15 h; iv, PyH, OTs, $Me_2C(OMe)_2$, room temp., 15 h; v, LDA, THF, DMPU, $-78 \,^{\circ}C$, 16 h; vii, McI, $-78 \,^{\circ}C$, 36 h, $10 \,^{\circ}C$, 8 h; vii, PyH, OTs, MeOH, reflux, 2.5 h; viii, KOH, THF, MeOH, H_2O , $40 \,^{\circ}C$, 15 h; ix, 2-chloro-*N*-methylpyridinium iodide, Et₃N, CH_2Cl_2 , reflux, 10 h; x, separate from racemate (92% e.e. $\rightarrow 96\%$ e.e.); xi, TBDMSCl, DMF, imidazole, room temp., 15 h; xii, NaOCH₂Ph, PhCH₂OH, THF, room temp., 6 h; xiii, TSCl, Py, DMAP, room temp., 2 d; xiv, KBr, AcO₂H, AcOH, room temp., 15 h

lactone, and thereby raised the e.e. from 92 to 99%. Protection of the free hydroxy group and methylation gave only the lactone 12 with the anti relationship between the silyl and methyl groups. Although most of our work on siliconcontrolled enolate alkylations has been with open chain esters,¹⁹ we have seen a highly diastereoselective methylation of a caprolactone before.²⁰ The four stereocentres C-2, C-3, C-6 and C-8 were now in place, all that remained was to convert the silvl to a hydroxy group,²¹ to tosylate the new hydroxy group, and open the lactone, whereupon cyclisation took place with inversion of configuration at C-3 to give methyl (+)-nonactate 13, identifiably the correct diastereoisomer by its definitive² ¹H NMR spectrum, and with specific rotation, $[\alpha]_D{}^{22} + 22.55$ (*c* 1.14, CHCl₃), close to that reported, $[\alpha]_D{}^{25} + 22.1$ (*c* 0.7, CHCl₃).⁴ GC on a homochiral column confirmed the high enantiomeric purity of this material (99.2% e.e.). The overall yield for the 10 steps from the common intermediate 10 to methyl (+)-nonactate 13 was 49%.

In the second sequence (Scheme 4) starting from the 3-hydroxyacetal 10, we did everything in the opposite sense. We used Prasad's version²² of Narasaka's method²³ to give the syn 1,3-diol (90:10), which we separated from the minor diastereoisomer and protected it as its acetonide 14. We methylated the ester stereoselectively $14 \rightarrow 15$, following our well-established rule for the alkylation of open-chain esters having a β-silyl group.¹⁹ Methylation in an open-chain ester sets up the C-2 and C-3 stereocentres in the opposite sense along the carbon backbone to the methylation of the lactone $11 \rightarrow 12$, although the cause, electrophilic attack taking place anti to the silyl group, is of course the same. Acetal hydrolysis, ester hydrolysis and lactonisation gave the lactone 16, and again some of the crystalline racemate could be filtered off, raising the enantiomeric purity from 92 to 96% e.e. or better. Protection of the free hydroxy group and base-catalysed opening of the lactone with benzyl alcohol revealed the C-6 hydroxy group, which we tosylated to give the benzyl ester. Silyl-to-hydroxy conversion²¹ allowed the formation, with

inversion of configuration at C-6, of benzyl (-)-nonactate 17. Hydrogenolysis and treatment with diazomethane gave methyl (-)-nonactate, which appeared to be diastereomerically pure in its ¹H NMR spectrum, and enantiomerically pure (although we had calculated that it ought to be 96% e.e.) on the homochiral GC column. The overall yield for the 12 steps from the common intermediate 10 to benzyl (-)-nonactate 17 was 35%.

We thank the Commission of the European Communities for a grant, B/CII*-913098, to S. K. G and Professor S. V. Ley for further support.

Received, 28th July 1994; Com. 4/04627G

References

- 1 A review of syntheses up to 1980: P. A. Bartlett, Tetrahedron, 1980, 36, 2.
- 2 P. A. Bartlett, J. D. Meadows and E. Ottow, J. Am. Chem. Soc., 1984, 106, 5304.
- 3 For a complete list, see ref. 1 in the following paper.
- 4 R. E. Ireland and J.-P. Vevret, J. Org. Chem., 1980, 45, 4259; Can. J. Chem., 1981, 59, 572.
- 5 I. Fleming, J. Chem. Soc., Perkin Trans. 1, 1992, 3363.
- 6 For another solution to this problem, see M. Ahmar, C. Duyck and I. Fleming, *Pure Appl. Chem.*, 1994, in the press.
- 7 I. Fleming and S. K. Ghosh, J. Chem. Soc., Chem. Commun., 1992, 1775.
- 8 J. Inanaga, Y. Handa, T. Tabuchi, K. Otsubo, M. Yamaguchi and T. Hanamoto *Tetrahedron Lett.*, 1991, **32**, 6557.
- 9 P. D. Theisen and C. H. Heathcock, J. Org. Chem., 1988, 53, 2374 and 1993, 58, 142.
- 10 J. P. Vigneron, M. Dhaenens and A. Horeau, *Tetrahedron*, 1973, 29, 1055.
- 11 I. Fleming and S. K. Ghosh, J. Chem. Soc., Chem. Commun., 1994, 99.
- 12 I. Fleming, S. Gil, A. K. Sarkar and T. Schmidlin, J. Chem. Soc., Perkin Trans. 1, 1992, 3351.
- 13 T. R. Kelly, L. Ananthasubramanian, K. Borah, J. W. Gillard, R. N. Goerner, Jr, P. F. King, J. M. Lyding, W.-G. Tsang and J. Vaya, *Tetrahedron*, 1984, 40, 4569.
- 14 T. Hayashi, Y. Okamoto, K. Kabeta, T. Hagihara and M. Kumada, J. Org. Chem., 1984, 49, 4224; I. Fleming and N. K. Terrett, J. Organomet. Chem., 1984, 264, 99; E. Vedejs and C. K. McClure, J. Am. Chem. Soc., 1986, 108, 1094; G. Procter, A. T. Russell, P. J. Murphy, T. S. Tsan and A. N. Mather, Tetrahedron, 1988, 44, 3953; I. Fleming, N. J. Lawrence, A. K. Sarkar and A. P. Thomas, J. Chem. Soc., Perkin Trans. 1, 1992, 330.
- 15 I. Fleming and S. K. Ghosh, J. Chem. Soc., Chem. Commun., 1992, 1777.
- 16 K. Yamamoto, T. Kimura and Y. Tomo, *Tetrahedron Lett.*, 1984, 25, 2155.
- 17 D. A. Evans, K. T. Chapman and E. M. Carreira, J. Am. Chem. Soc., 1988, 110, 3560.
- 18 T. Mukaiyama, M. Usui and K. Saigo, Chem. Lett., 1976, 49.
- 19 R. A. N. C. Crump, I. Fleming, J. H. M. Hill, D. Parker, N. L. Reddy and D. Waterson, J. Chem. Soc., Perkin Trans. 1, 1992, 3277.
- 20 H.-F. Chow and I. Fleming, J. Chem. Soc., Perkin Trans. 1, 1984, 1815.
- 21 I. Fleming and P. E. J. Sanderson, *Tetrahedron Lett.*, 1987, 28, 4229.
- 22 K.-M. Chen, G. E. Hardtmann, K. Prasad, O. Repic and M. J. Shapiro, *Tetrahedron Lett.*, 1987, 28, 155.
- 23 K. Narasaka and F.-C. Pai Tetrahedron, 1984, 40, 2233.