Stereoselective Induction of an Axial Chirality by Suzuki Cross Coupling of Tricarbonyl(arene)chromium Complexes with Arylboronic Acids

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Tricarbonyl(2,6-disubstituted-1-bromobenzene)chromium was treated with *ortho*-substituted phenylboronic acids in the presence of a palladium(0) catalyst to give mono- $Cr(CO)_3$ complexes of biphenyl compounds with a complementary axial chirality depending upon the nature of *ortho*-substituent of the phenylboronic acids

Biphenyl or binaphthyl compounds with axial chirality are attractive targets not only as chiral ligands in asymmetric reactions but also for the synthesis of biologically active natural products.¹ Here we report the palladium(0) catalysed Suzuki cross-coupling reaction² giving mono-Cr(CO)₃ complexed biphenyl compounds, where the stereoselectivity is controlled by the nature of the *ortho*-substituent of the phenylboronic acid used.³

Oxidative addition of arylhalides to the palladium(0) is accelerated by coordination of an electron-withdrawing tricarbonylchromium group to the arene.⁴ Tricarbonyl(2-methoxy-6-methyl-1-bromobenzene)chromium **1a** was treated with *o*-methylphenylboronic acid **2a** catalysed by 10 mol% of Pd(PPh₃)₄ in the presence of sodium carbonate in aqueous MeOH at 75 °C to give the cross-coupling product **3a** in 96% yield without any formation of the corresponding atropisomer (Scheme 1). The stereochemistry of **3a** was assigned as $(S^*, S^*)^{\dagger}$ by comparison with an authentic sample derived from a stereodefined complex **3e** the configuration of which was determined by X-ray crystallography.⁵ The methyl group



on the B-ring of the complex 3a is directed toward the tricarbonylchromium group with respect to the complexed arene A-ring in spite of a severe steric interaction between the methyl and Cr(CO)₃ groups. Similarly, the cross-coupling reaction of o-methylphenylboronic acid 2a with the other chromium complexes 1b, c and d gave the products 3b, c and d with the same axial chirality (entries 2, 3, 4, Table 1). With o-methoxyphenylboronic acid 2d, the (S^*, S^*) -coupling products 3 were obtained as major products along with stereoisomeric complexes 4 (entries 10, 11, Table 1). Whereas the coupling of o-formylphenylboronic acid 2b with complexes 1a, **b** and **c** gave diastereoisomeric (S^*, R^*) -chromium complexes 4 ($R^2 = CHO$) as the only isolated products (entries 5, 6, 7, Table 1). Formation of these products 4 can be attributed to a thermodynamically controlled reaction via isomerization of the kinetically controlled coupling products (S^*, S^*) -3 to the corresponding axial isomeric (S^*, R^*) -complexes 4 under coupling reaction conditions, since oxidation of complex 3d to 3b at room temp. with Me₂SO-Ac₂O gave the corresponding diastereomerically axial complex 4 without isolation of 3b. Coupling of *o*-hydroxymethyl phenylboronic acid 2c with the complexes 1 catalysed by $Pd(PPh_3)_4$ produced the (S^*, S^*) chromium complex 3 and CO inserted product 5 in various ratios depending upon the ortho-substituent of the chromium complex (entries 8, 9, Table 1). Thus, reaction of 1a with o-hydroxymethylphenylboronic acid 2c produced (S^*, S^*)-3 as the major product, whereas coupling with 1c afforded only CO inserted compound 5. It is obvious from the results that the stereochemistry of axial chirality of the coupling products is unambiguously controlled by the nature of ortho-substituent of phenylboronic acids.

This diastereoselective cross-coupling reaction provides a promising approach to the formation of the enantiomerically pure biphenyl compounds with axial chirality in both enantiomeric forms from a single arene chromium complex (Scheme 2). Enantiomerically pure (+)-chromium complex 6^{\ddagger} was coupled with *o*-methylphenylboronic acid **2a** to give (+)-(*R*,*R*)-complex **7**, which was converted to (-)-(*R*)-2-methoxy-2'-methyl-6-(1,3-dioxolanyl)biphenyl **8**§ on exposure to sunlight, in good yield. Alternatively, diastereo-isomeric (+)-(*R*,*S*)-chromium complex **9** was obtained by the coupling of complex **6** with *o*-formylphenylboronic acid **2b**.

Table 1 Palladium(0) catalysed cross-coupling of (arene)chromium complexes 1 with arylboronic acids 2

	Complex	Phenylboronic acid		R ¹	R ²	Yiel	d (%)	
 Entry			Product			3	4	5
1	1a	2a	3a	Ме	Ме	96		
2	1b	2a	3b, 5	СНО	Me	82		7
3	lc	2a	3c	CHO(CH ₂) ₂ O	Ме	81	_	
4	1d	2a	3d	CH ₂ OH	Me	77	_	
5	1a	2b	4	Me	CHO		95	
6	1b	2b	4	СНО	СНО		43	
7	1c	2b	4	CHO(CH ₂) ₂ O	CHO		52	
8	1a	2c	3, 5	Me	CH_2OH	55		13
9	1c	2c	5	CHO(CH ₂) ₂ O	CH ₂ OH			40
10	1a	2d	3,4	Me	OMe	91	3	
11	1d	2d	3,4	CH ₂ OH	OMe	85	5	

2698



Scheme 2 Reagents and conditions: i, o-methylphenylboronic acid, $Pd(PPh_3)_4$, Na_2CO_3 , MeOH, H_2O , 70 °C, 30 min, 81%; ii, o-formylphenylboronic acid, $Pd(PPh_3)_4$, Na_2CO_3 , MeOH, H_2O , 70 °C, 30 min, 52%; iii, $hv-O_2$, 90%; iv, LAH, diethyl ether, 62%; v, MsCl, py, 38%; vi, LAH, diethyl ether, 51%; vii, $hv-O_2$, diethyl ether, 90%



The biphenyl complex 9 was converted to the antipode (+)-(S)-biphenyl compound 10 by reduction of the formyl group to a methyl group followed by demetallation.

Although the precise mechanism of the cross-coupling reaction is not clear enough to prove that the stereochemistry of the axial chirality depends on the ortho-substituent of the phenylboronic acids, the observed stereoselectivity can be rationalized by examining the transision states and palladium intermediates. Two cis diorgano palladium(II) intermediates 11 and 12 with a square configuration are the transient species prior to biaryl carbon-carbon bond formation. Complex 11 depicts a crowded system in which the R² substituent on the *B*-ring is close to the sterically larger \mathbb{R}^1 substituent. The alternative complex 12 appears to be free of any severe non-bonded interaction, in which the R² substituent is close to the smaller methoxy group, and the tricarbonylchromium group is further away from the bulky triphenylphosphine. In complex 12, both arene rings are coupled via overlapping7 of p-orbitals with evasion of severe non-bonding interactions between R² and the triphenylphosphine groups, giving the products 3.

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Footnotes

† (S^*, S^*) etc: the first symbol indicates a configuration of planar chirality of the chromium complexed arene ring (at C-1 position), and the second represents axial chirality. The * indicates a racemate but only one enantiomer is shown for clarity. ‡ Optically pure (+)-complex 6 ($[\alpha]_D^{17}$ +61.9 (c 0.5, EtOH) was

[‡] Optically pure (+)-complex **6** ($[\alpha]_D^{17}$ +61.9 (c 0.5, EtOH) was prepared from optically resolved (-)-tricarbonyl(2-bromo-3methoxybenzaldehyde)chromium { $[\alpha]_D^{24}$ -347.2 (c 1.415, CHCl₃)} by reaction with ethylene glycol and a catalytic amount of *p*-TsOH in MeCN at room temp. Resolution of the racemic tricarbonyl(2-bromo-3-methoxybenzaldehyde)chromium was achieved by the Davies method.⁶

§ The optical purity of the biphenyl compounds 8 and 10 was determined by HPLC with Chiralcel OJ (eluted with hexane–ethanol 75:25; column temperature 25 °C; flow rate 0.5 ml min⁻¹); retention times; 12.72 min for (-)-8; 15.44 min for (+)-10.

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