Synthesis of 2,4-Diaza-4-deoxypodophyllotoxin, a New Analogue of Podophyllotoxin possessing Antitumour Activity

Yukio Hitotsuyanagi,*a Yoichi Naka,a Keiji Yamagami,b Akihiro Fujiib and Tetsuya Taharaa

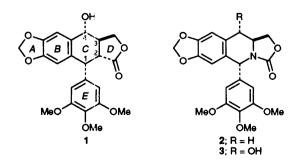
^a Research Laboratories, Yoshitomi Pharmaceutical Industries Ltd., 955 Yoshitomi-cho, Chikujo-gun, Fukuoka 871, Japan

^b Research Laboratories, Yoshitomi Pharmaceutical Industries Ltd., Koyata, Iruma-shi, Saitama 358, Japan

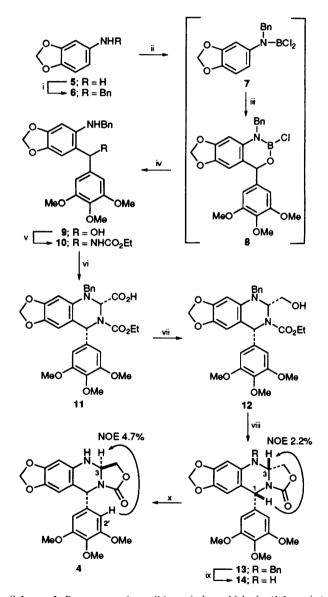
2,4-Diaza-4-deoxypodophyllotoxin **4** and its *cis* analogue **14** are synthesised stereoselectively from 3,4-methylenedioxyaniline **5**, and show significant activity against vincristine-resistant P-388 leukaemia *in vivo*.

Podophyllotoxin 1 has been known to possess antitumour activity and is still considered an important lead for antineoplastic agents.¹ Previous efforts to find a less toxic analogue of 1 have been conducted mainly through derivatization from 1. Recently 2-azapodophyllotoxin analogues, such as 2 and 3, have been synthesised by several groups² and attract much interest since some of them showed pronounced in vivo antitumour activity comparable to that of 1.2^{2b} However no 2-aza analogue incorporating an additional heteroatom at position 4 or other positions of ring C has been reported thus far due to the probable chemical instability and accordingly due to less accessibility. Thus, we designed and synthesised 2.4-diaza-4-deoxypodophyllotoxin 4 since it was expected that introduction of a nitrogen atom at this position should little alter the whole stereostructure, but might pose significant influence upon its biological properties through increasing the electron density on ring B.

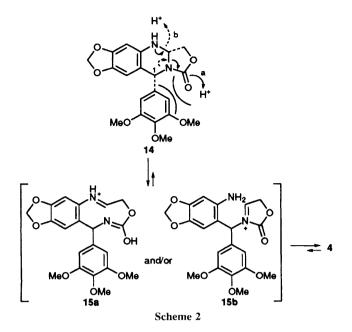
3,4-Methylenedioxyaniline 5 was reacted with benzaldehyde, and successive reduction of the resultant imine with sodium borohydride in a mixture of solvents, EtOH-THF-acetic acid (AcOH), gave N-benzyl-3,4-methylenedioxyaniline 6 (bp 186-189 °C/4 mmHg) in 82% yield (Scheme 1). Compound 6 was converted by boron trichloride (1.0 equiv.) and triethylamine (1.15 equiv.) in 1,2-dichloroethane at 0 °C to room temperature for 20 h into the N-benzyl-3,4-methylenedioxyanilinodichloroborane 7,3 which on treatment with 3,4,5-trimethoxybenzaldehyde (1.0 equiv.) and triethylamine (1.15 equiv.) at room temperature for 3 h gave the benzhydrol 9 in 38% yield from 6 after the hydrolysis of the boracyclic intermediate 8. The alcohol 9 was reacted with ethyl carbamate (5 equiv.) in polyphosphate ester $(PPE)^4$ -THF (1:4) to give the N-benzhydrylurethane 10 (mp 204–205 °C) in 66% yield. The urethane 10 was condensed with glyoxylic acid monohydrate (1.2 equiv.) in refluxing THF[†] for 1 h to give the tetrahydroquinazolinecarboxylic acid 11, which, without purification, was reduced with borane prepared in situ from sodium borohydride and boron trifluoride etherate in THF to afford the 3-hydroxymethyl derivative 12 (mp 147-149 °C) in 88% yield from 10 as a single stereoisomer. We could not determine its relative stereochemistry at this stage. Sodium methoxide treatment of 12 in refluxing MeOH effected the cyclization to form the oxazolidone 13 (mp 217-218 °C) in 97% vield, and successive debenzylation using palladium on carbon (Pd/C) under H₂ in CH₂Cl₂-MeOH-AcOH (6:3:1) afforded 14 (mp 249–250 °C) \ddagger in 95% yield. The relative stereochemistry of 14



between C-1 and C-3 was established to be a *cis* relation by the observation of 2.2% enhancement of NOE between 1-H (δ 5.47) and 3-H (δ 5.09).



Scheme 1 Reagents and conditions: i, benzaldehyde (1.1 equiv.), benzene, Dean–Stark, reflux, 1 h; NaBH₄ (0.5 equiv.), EtOH–THF–AcOH (60:10:1), room temp., 1 h (82%); ii, BCl₃ (1.0 equiv.), Et₃N (1.15 equiv.), 1,2-dichloroethane, 0 °C to room temp., 20 h; iii, 3,4,5-trimethoxybenzaldehyde (1.0 equiv.), Et₃N (1.15 equiv.), room temp., 3 h; iv, NH₄OH (38% from 6); v, ethyl carbamate (5 equiv.), PPE–THF (1:4), room temp., 1 h (66%); vi, OHC–CO₂H·H₂O (1.2 equiv.), THF, reflux, 1 h; vii, NaBH₄, BF₃·OEt₂, THF, room temp., 72 h (88% from 10); viii, MeONa, MeOH, reflux, 6 h (97%); ix, H₂, Pd/C, CH₂Cl₂–MeOH–AcOH (6:3:1), room temp., 9 h (95%); x, CHCl₃–CF₃CO₂H (3:1), 50 °C, 12 h (99%)



Surprisingly, treatment of **14** with a 3:1 mixture of CHCl₃ and trifluoroacetic acid at 50 °C for 12 h afforded exclusively the desired *trans* analogue **4** (mp 223–224 °C)§ in 99% yield, the stereochemistry of which was determined by the observation of NOE (4.7%) between 2'-H (δ 6.54) and 3-H (δ 4.96), and further supported by the upfield shift [δ 66.9 (**14**) $\rightarrow \delta$ 61.4 (**4**)] of the C-3 resonance in ¹³C NMR.⁵

This exclusive conversion from the *cis* analogue 14 to the *trans* analogue 4 can be rationalized by considering the intermediates 15a and/or 15b generated through path a or b shown in Scheme 2 and the relative thermodynamic stability between 4 and 14. The *cis* analogue 14 should suffer $A^{(1,2)}$ strain⁶ between the carbonyl oxygen atom of the oxazolidone and the pseudo-equatorially oriented trimethoxyphenyl group at C-1, but in the *trans* analogue 4 this strain is relieved by the C-1 substituent adopting a pseudo-axial position.¶

A goal of development of new antineoplastic drugs is to obtain molecules lacking cross resistance with classical agents.⁷ Analogues **4** and **14** showed significant activity against vincristine-resistant P-388 leukaemia (P-388/VCR) *in vivo.*

Received, 15th September 1994; Com. 4/05632I

Footnotes

[†] Upon this condensation, the use of THF as a solvent proved to be important. Other solvents such as MeOH or propan-2-ol resulted in the decrease of diastereoselectivity.

 $\label{eq:constraint} \begin{array}{l} \pm 14: \delta_{H} (CDCl_{3}; 400 \text{ MHz}; J/Hz) 3.80 (3H, s), 3.82 (6H, s), 3.90 (1H, br s, D_{2}O exchange), 4.12 (1H, dd, J 8.4, 8.4), 4.57 (1H, dd, J 8.4, 7.1), 5.09 (1H, br t, J.7.7), 5.47 (1H, s), 5.86 (1H, d, J 1.3), 5.88 (1H, d, J 1.3), 6.34 (1H, s), 6.42 (1H, s) and 6.57 (2H, s). \\ \delta_{C} (CDCl_{3}; 100 \text{ MHz}) 56.1 (q), 60.5 (d), 60.8 (q), 66.9 (d), 67.4 (t), 99.9 (d), 101.3 (t), 104.6 (d), 107.6 (d), 119.3 (s), 135.6 (s), 137.5 (s), 137.6 (s), 143.4 (s), 147.3 (s), 153.3 (s) and 154.9 (s). \end{array}$

§ 4: $\delta_{\rm H}$ (CDCl₃; 400 MHz) 2.48 (1H, br s, D₂O exchange), 3.81 (6H, s), 3.84 (3H, s), 4.16 (1H, dd, J 9.4, 1.5), 4.37 (1H, dd, J 9.4, 5.9), 4.96 (1H, dd, J 5.9, 1.5), 5.76 (1H, s), 5.87 (1H, d, J 1.3), 5.90 (1H, d, J 1.3), 6.29 (1H, s) 6.39 (1H, s) and 6.54 (2H, s). $\delta_{\rm C}$ (CDCl₃; 100 MHz) 55.5 (d), 56.3 (q), 60.8 (q), 61.4 (d), 68.6 (t), 98.7 (d), 101.1 (t), 105.9 (d), 108.1 (d), 112.5 (s), 136.3 (s), 137.8 (s), 138.0 (s), 142.2 (s), 147.8 (s), 153.4 (s), 156.8 (s).

¶ Koga *et al.* reported the *cis/trans* isomerization of **2** *via* a benzhydryl cation under strongly acidic conditions.^{2b} We cannot completely deny the possibility that this type of reaction might occur in **14** instead of, or concomitantly with, the reaction shown in Scheme 2.

|| The maximum T/C values for 4 and 14 against P-388/VCR *in vivo* (CDF₁ mice, ip-ip, days 1–5) are 152% (at 10 mg kg⁻¹ d⁻¹ level) and 137% (at 5 mg kg⁻¹ d⁻¹ level), respectively. Podophyllotoxin 1 showed marginal activity (T/C = 125%) upon this experiment.

References

- 1 D. C. Ayres and J.D. Loike, *Lignans. Chemical, biological and clinical properties*, CUP Cambridge, 1990.
- 2 (a) H. L. Pearce, N. J. Bach and T. L. Cramer, Tetrahedron Lett., 1989, 30, 907; (b) K. Tomioka, Y. Kubota and K. Koga, Tetrahedron Lett., 1989, 30, 2953; J. Chem. Soc., Chem. Commun., 1989, 1622; Tetrahedron, 1993, 49, 1891; (c) J. Van der Eycken, J.-P. Bosmans, D. Van Haver, M. Vandewalle, A. Hulkenberg, W. Veerman and R. Nieuwenhuizen, Tetrahedron Lett., 1989, 30, 3873; J.-P. Bosmans, J. Van der Eycken, M. Vandewalle, A. Hulkenberg, R. Van Hes and W. Veerman, Tetrahedron Lett., 1989, 30, 3877; (d) H. Itokawa, Y. Hitotsuyanagi and K. Takeya, Heterocycles, 1992, 33, 537; (e) P. Lienard, J. Royer, J.-C. Quirion and H.-P. Husson, Tetrahedron Lett., 1991, 32, 2489; P. Lienard, J.-C. Quirion and H.-P. Husson, Tetrahedron, 1993, 49, 3995; (f) F. Reteurtre, J. Madalengoitia, A. Orr, T. J. Guzi, E. Lehnert, T. Macdonald and Y. Pommier, Cancer Res., 1992, 52, 4478; (g) J. S. Madalengoitia and T. L. Macdonald, Tetrahedron Lett., 1993, 34, 6237; (h) S. W. McCombie, J. R. Tagat, W. A. Metz, D. Nazareno and M. S. Puar, Tetrahedron, 1993, 49, 8073.
- 3 T. Sugasawa, T. Toyoda, M. Adachi and K. Sasakura, J. Am. Chem. Soc., 1978, 100, 4842.
- 4 M. P. Cava, M. V. Lakshmikantham and M. J. Mitchell, J. Org. Chem., 1969, 34, 2665.
- 5 H. Beierbeck, J. K. Saunders and J. W. ApSimon, *Can. J. Chem.*, 1977, **55**, 2813.
- 6 F. Jhonson, Chem. Rev., 1968, 68, 375.
- 7 L. M. Hollingshead and D. Faulds, Drugs, 1991, 42, 690.