Approaches Towards the Synthesis of Papulacandin D: Preparation and Structural Elucidation of the Acyl Side Chain

Anthony G. M. Barrett,*^a Michael Peña^b and J. Adam Willardsen^a

^a Department of Chemistry, Imperial College of Science, Technology and Medicine, London, UK SW7 2AY
^b Department of Chemistry, Colorado State University, Fort Collins, CO 80523, USA

Both degradation and total synthesis from L-(+)-isoleucine are used to establish the absolute stereochemistry of the O-3'-acyl side chain of papulacandin D.

The papulacandins are a group of antifungal agents extracted from the fermentation broth of *Papularia spherosperma*.¹ These compounds are strongly active against *Candida albicans* and several other yeasts.^{1–3} Their mode of action involves the inhibition of β -1,3-glucan synthase, an enzyme vital in the construction of fungal cell walls.² Papulacandin D 1 consists of a *C*-arylglycosidic spiroketal esterified at the *O*-3 hydroxy by a tetra-unsaturated fatty acyl side chain. Although the gross structure of papulacandin D is well established, neither the relative nor absolute stereochemistries of the C-7 hydroxy group or the C-14 methyl residue of the acyl side chain are known.¹ We now report the determination of the absolute configuration and synthesis of the papulacandin D acyl side chain.⁴

Following a known method,⁵ L-(+)-isoleucine was transformed into (S)-2-methylpentanol 2. Toluene 4-sulfonation of 2 and subsequent displacement with potassium cyanide yielded the nitrile 3 (Scheme 1). † Reduction of 3 (DIBAL-H) and Wittig homologation gave the α,β -unsaturated ester 4 (E:Z, 98:2). Subsequent reduction of 4 (DIBAL-H), reoxidation to the aldehyde (PCC) and further Wittig homologation yielded the diene ester 5 (E:Z, 92:8). The two geometric isomers were easily separated by chromatography. The E,E-ester 5 was converted into the aldehyde 6 via a reduction (DIBAL-H)/ oxidation (MnO₂) sequence. Homologation of 6, with prop-2-ynyl zinc bromide,⁶ yielded alcohol 7 as a inseparable mixture of diastereoisomers (1:1). After protection as the triethylsilyl ether, the delicate tetraene ester 8 was elaborated via hydrozirconation followed by direct palladium(0) coupling of the derived vinyl zirconium intermediate with methyl 3-bromo-(E)acrylate.7 No traces of the fully conjugated pentaene 9, produced by the elimination of water, were detected.

With a route to the side chain secure, attention was directed to solving the stereochemical ambiguities. We considered that the C-7 stereochemistry could be defined *via* an asymmetric propynylation of aldehyde **6** (Scheme 1), but this approach was unsuccessful. Thus, a method of kinetic resolution was pursued. Sharpless asymmetric epoxidation⁸ of the allylic alcohol **7** resulted in recovery of alcohol **10** (45%) and isolation of the corresponding epoxide **11** (14%) (Scheme 2). Mosher ester analysis of the derived alcohol **19** (*vide infra*) revealed a diastereoisomeric excess of 96% (HPLC)‡. It is clear from this result that any of the four diastereoisomers of the acyl side chain should be accessible *via* the reactions in Schemes 1 and 2.



Papulacandin D

Fermentation of *Papularia spherosperma* followed by isolation of the crude mixture of papulacandins,¹ saponification and methylation^{1,9} (CH₂N₂) gave the natural fatty ester **12**. Reaction with (*S*)- α -methoxy- α -trifluoromethylphenylacetyl chloride gave the ester **13** (Scheme 3) which on analysis was shown to consist of one diastereoisomer (¹H and ¹³C NMR and HPLC). Additionally, this diastereoisomer was also shown to match (¹H



Scheme 1 Reagents and conditions: i, TsCl, py, ii, KCN, 18-crown-6, THF, 80% (both steps); iii, DIBAL-H, Et₂O, -78 °C; iv, Ph₃P=CHCO₂Et, CH₂Cl₂ 85% (both steps); v, DIBAL-H, Et₂O, -78 °C, 95%; vi, PCC, CH₂Cl₂, 80%; vii, Ph₃P=C(Me)CO₂Et, CH₂Cl₂, 85%; viii, DIBAL-H, Et₂O, -78 °C, 94%; ix, MnO₂, CH₂Cl₂, 95%; x, Zn dust, prop-2-ynyl bromide, THF, 95%; xi, Et₃SiCl, imidazole, DMAP, CH₂Cl₂, 88%; xii, (n⁵-C₃H₃)₂Zr(Cl)H, methyl 3-bromo-(*E*)-acrylate, [(Ph₃P)₂PdCl₂ + DIBAL-H], THF, 82%



Scheme 2 Reagents and conditions: i, $Ti(OPr^i)_4$. D-(-)-DIPT, Bu'OOH, CH_2Cl_2 , 4 Å mol. sieves, 10 45%, 96% de and 11 14%



Scheme 3 Reagents and conditions: i, S-(+)PhC(OMe)(CF₃)COCl, DMAP, CH₂Cl₂, 93%; ii, O₃, -78 °C, MeOH; Me₂S, -78 to 25 °C; iii, 2,4-dinitrophenylhydrazine, H₂SO₄, 63% (both steps)



Scheme 4 Reagents and conditions: i, TBAF, THF, 0 °C; ii, (+)-S-MTPA-Cl, DMAP, CH_2Cl_2 , 89%; iii, O_3 , -78 °C, MeOH; Me_2S , -78 to 25 °C; iv, 2,4-dinitrophenylhydrazine, H_2SO_4 , EtOH, 65% (both steps)

and ${}^{13}C$ NMR and HPLC) one of the synthetic diastereoisomers in the mixture of 15 and 16 (Scheme 4).

The allylic alcohol **10** is predicted to have the 7S configuration according to the model proposed by Sharpless and coworkers.¹⁰ Conversion of alcohol **10** into the corresponding Mosher ester and ¹H NMR analysis¹¹ gave additional proof of this stereochemical assignment. Protection of **10** as its triethylsilyl ether and chain extension gave tetraene ester **18** (Scheme 5). Deprotection with *tert*-butylammonium fluoride revealed the alcohol **19** and this was transformed into ester **20**. Again Mosher analysis of diester **20** was consistent with the stereochemistry of the C-7 hydroxy being *S*. Ester **20** also matched (¹H and ¹³C NMR and HPLC) the natural ester **13**. With all of this evidence the stereochemistry about the C-7 hydroxy can unequivocally be assigned as *S*.

Ozonolysis of the natural ester 13 (Scheme 3) gave the corresponding aldehyde which was isolated as its 2,4-dinitro-



Scheme 5 Reagents and conditions: i, TBAF, THF, 0 °C 98%; ii, (+)-S-MTPA-Cl, DMAP, CH₂Cl₂, 93%

phenyl hydrazone derivative 14. Comparison of the optical rotation of 14 { $[\alpha]_{D}^{25} = +13.2 \ (c = 0.29 \ in CHCl_3)$ } with the synthetic material derived directly from L-isoleucine 17 { $[\alpha]_{D}^{25} = +12.6 \ (c = 0.30 \ in CHCl_3)$ } (Scheme 4) was fully consistent with the configuration about the C-14 methyl group being S. We therefore conclude that papulacandin D has the structure 1 in which the acyl side chain has the 7S,14S configuration. Compounds 12 and 19 were also shown to be identical by their circular dichroic spectra.

We thank Myco Pharmaceutical Inc. for support of our programs on antifungal compounds; the United States Department of Agriculture–Agriculture Research Service for a culture of *Papularia spherosperma* (NRRL 8086); Dr D. Leak and Mr J. Mansfield at Imperial College Department of Biochemistry for their help in growing this culture; the EPSRC National Chiroptical Centre for performing CD measurements; Glaxo Group Research Ltd for the most generous endowment; the Wolfson Foundation for establishing the Wolfson Centre for Organic Chemistry in Medical Science at Imperial College and the NIH for support when this work started in the USA.

Received, 28th March 1995; Com. 5/01962A

Footnotes

[†] All new compounds were fully characterised by spectroscopic data and microanalysis and/or HRMS.

[‡] Apex silica 5U (25 cm \times 4.6 mm), eluted with 2.5% ethyl acetate in hexanes at 1.0 cm³ min⁻¹ on a ATI Unican crystal 200 HPLC system with diode array detection from 220 to 360 nm.

References

- 1 P. Traxler, J. Gruner and J. A. L. Auden, J. Antibiot., 1977, 30, 289.
- 2 G. Rommele, P. Traxler and W. Wehrli, J. Antibiot., 1983, 36, 1539.
- 3 F. Van Middlesworth, M. N. Olmstead, D. Schmatz, K. Bartizal, R. Fromtling, G. Bills, K. Nolstadt, S. Honeycutt, M. Zweerink, G. Garrity and K. Wilson, J. Antibiot., 1991, 44, 45.
- 4 A. G. M. Barrett, M. Peña and J. A. Willardsen, in *Recent Advances in the Chemistry of Anti-Infective Agents*, ed. P. H. Bentley, RSC, Cambridge and London, 1993, p. 234.
- 5 V. Schurig, U. Leyrer and D. Wistuba, J. Org. Chem., 1986, 51, 242.
- 6 L. E. Friedrich, N. de Vera and M. Hamilton, Synth. Commun., 1980, 10, 637.
- 7 E.-I. Negishi, T. Takahashi, S. Baba, D. E. Van Horn and N. Okukado, J. Am. Chem. Soc., 1987, **109**, 2393.
- 8 V. S. Martin, S. S. Woodward, T. Katsuki, Y. Yamada, M. Ikeda and K. B. Sharpless, *J. Am. Chem. Soc.*, 1981, **103**, 6237.
- 9 P. Traxler, H. Fritz, H. Fuhrer and W. Richter, J. Antibiot., 1980, 33, 967.
- 10 T. Katsuki and K. B. Sharpless, J. Am. Chem. Soc., 1980, 102, 5974.
- 11 J. A. Dale and H. S. Mosher, J. Am. Chem. Soc., 1973, 95, 512.