

## Doubly Helical, Chiral Crown Thioether fully preorganized for Planar Coordination

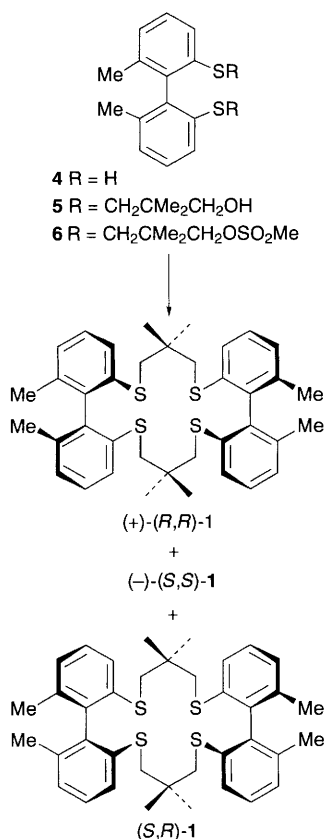
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Optically active 18-membered quadridentate crown thioethers with double helicity, (+)-(*R,R*)- and (-)-(*S,S*)-bitol<sub>2</sub>Me<sub>4</sub>[18]S<sub>4</sub> and the non-helical *meso*-compound are prepared and their reactions with ruthenium(II) give *trans*-[RuCl<sub>2</sub>{(+)-(*R,R*)-bitol<sub>2</sub>Me<sub>4</sub>[18]S<sub>4</sub>}] **2** and the *meso*-analogue of *D*<sub>2</sub> and *C*<sub>2h</sub> symmetries, respectively; free (+)-(*R,R*)-bitol<sub>2</sub>Me<sub>4</sub>[18]S<sub>4</sub> and the racemic ruthenium(II) complex have been structurally characterized.

There is increasing interest in the transition metal complexes of crown thioethers;<sup>1</sup> however, those of chiral crown thioethers are extremely rare.<sup>2</sup> Here we report the preparation, characterization and optical resolution of an 18-membered quadridentate crown thioether bitol<sub>2</sub>Me<sub>4</sub>[18]S<sub>4</sub> **1** and its coordination to ruthenium(II). The ligand **1** consists of two 2,2'-bitolyl-3,3'-dithiolato groups bonded alternately to two 2,2-dimethylpropane subunits. Incorporation of two structurally rigid and axially chiral bitolyl groups determines unequivocally doubly helical and non-helical arrays for the racemate and *meso*-isomer of **1**, respectively. This is also the case for the ruthenium complexes *trans*-[RuCl<sub>2</sub>{(+)-(*R,R*)-**1**}] **2** and *trans*-[RuCl<sub>2</sub>{(*R,S*)-**1**}] **3**. The former represents the first example of an optically active and doubly helical transition metal complex containing sulfur ligands. Almost all helicates so far reported have been produced with open-chain and macrocyclic polydentate ligands containing nitrogen donors such as pyridine and bipyridine.<sup>3</sup>

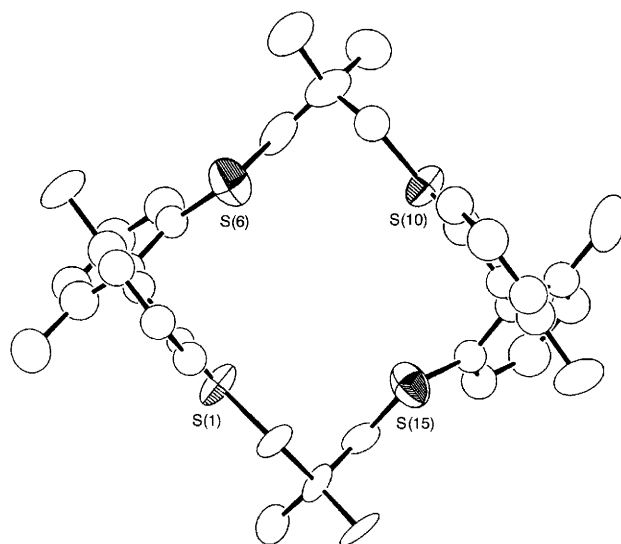
The macrocycle **1** was prepared through S-alkylation of the disodium salt of (±)-2,2'-dimethyl-6,6'-disulfanylbiphenyl **4**<sup>4</sup> with 3-bromo-2,2-dimethylpropan-1-ol in EtOH under reflux, mesylation of the resulting diol **5** with mesyl chloride in pyridine, and cyclization of the dimesylate **6** with **4** in the presence of Cs<sub>2</sub>CO<sub>3</sub> in DMF (Scheme 1).<sup>5</sup> The total yield is 13.5%. Complete optical resolution of the three diastereo-

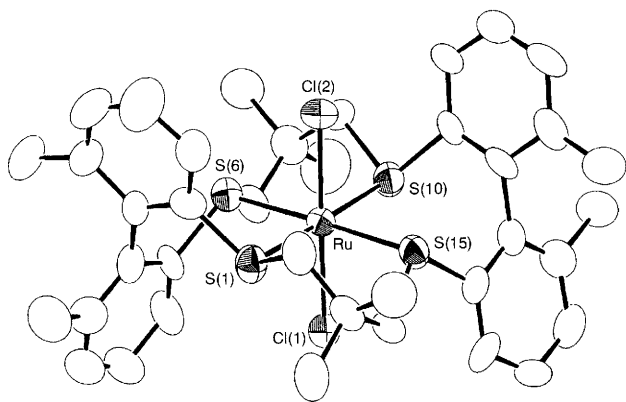


Scheme 1

isomers was achieved by HPLC with a column packed with cellulose tris(3,5-dimethylphenylcarbamate) on silica gel;<sup>6</sup> elution with 0.1% Pr<sup>i</sup>OH-hexane gave optically pure (-)-(*S,S*)-**1**, (+)-(*R,R*)-**1** {[α]<sub>D</sub><sup>20</sup> -157 and +152 (c 0.4, hexane), respectively} and the *meso*-compound (*R,S*)-**1** in a 1 : 1 : 2 ratio. The absolute configuration of the two chiral diastereoisomers was deduced by comparing their CD spectra with those of (6,6'-dimethylbiphenylene-2,2'-diyl)bis(diphenylphosphine)<sup>7</sup> and was determined unequivocally by an X-ray structural study for the (+)-(*R,R*) enantiomer.† (+)-(*R,R*)-**1** assumes a quadrangular structure of approximate *D*<sub>2</sub> symmetry with an *endo*-orientation of all S atoms directed towards the centre of the macrocycle (Fig. 1). Owing to the inherent twist about the pivotal bond of 2,2'-bitolyl groups, the four S atoms adopt an up-down-up-down conformation where up and down indicate the direction of the lone electron pair orbitals on the S atoms with respect to the approximately square plane (4S) defined by the four S atoms. Two aromatic rings of the 2,2'-bitolyl groups are orthogonal with a mean dihedral angle of 87.8°. The corresponding angle between the least-squares 4S plane and each aromatic ring is 79.3°. The non-bonded S...S distances of 3.873(6)–4.031(8) Å are longer than the sum of the van der Waals radius (3.7 Å). The *D*<sub>2</sub> structure is retained in solution as deduced from <sup>1</sup>H NMR spectra of the two optically resolved enantiomers, which exhibit one Me singlet and two CH<sub>2</sub> doublets due to the two CH<sub>2</sub>CMe<sub>2</sub>CH<sub>2</sub> groups.‡ The observation of two signals for each proton of the *meso*-compound is consistent with *C*<sub>2h</sub> symmetry with an up-up-down-down (*syn*) conformation.

The X-ray structure suggests that the chiral crown thioether with double helicity is fully preorganized for planar coordination to metal ions. Indeed, treatment of (+)-(*R,R*)-**1** with K<sub>2</sub>[RuCl<sub>5</sub>(H<sub>2</sub>O)] in MeOCH<sub>2</sub>CH<sub>2</sub>OH under reflux for 24 h followed by chromatography (silica gel, CHCl<sub>3</sub>) and recrystallization from CH<sub>2</sub>Cl<sub>2</sub>-hexane gave *trans*-[RuCl<sub>2</sub>{(+)-(*R,R*)-**1**}] (**2**) {[α]<sub>D</sub><sup>20</sup> +35 (c 0.1, CHCl<sub>3</sub>)} as reddish brown crystals in

Fig. 1 Molecular structure of (+)-(*R,R*)-bitol<sub>2</sub>Me<sub>4</sub>[14]S<sub>4</sub>-1



**Fig. 2** Molecular structure of *trans*-[RuCl<sub>2</sub>{(+)-(R,R)-1}] **2** in the racemate. Selected bond distances (Å) and angles (°): Ru–S(1) 2.373(7), Ru–S(6) 2.368(5), Ru–S(10) 2.361(7), Ru–S(15) 2.370(5), Ru–Cl(1) 2.417(5), Ru–Cl(2) 2.395(5); S(1)–Ru–S(6) 89.0(2), S(1)–Ru–S(10) 179.1(2), S(1)–Ru–S(15) 90.2(2), S(6)–Ru–S(10) 90.2(2), S(6)–Ru–S(15) 178.7(2), S(10)–Ru–S(15) 90.7(2), S(1)–Ru–Cl(1) 90.7(2), S(1)–Ru–Cl(2) 89.7(2), S(6)–Ru–Cl(1) 89.8(2), S(6)–Ru–Cl(2) 91.1(2), S(10)–Ru–Cl(1) 89.6(2), S(10)–Ru–Cl(2) 90.1(2), S(15)–Ru–Cl(1) 89.2(2), S(15)–Ru–Cl(2) 89.9(2).

82% yield. A similar reaction with (*R,S*)-**1** afforded *trans*-[RuCl<sub>2</sub>{(*R,S*)-**1**}] (**3**) (43% yield). The <sup>1</sup>H NMR spectra of **2** and **3** indicate *D*<sub>2</sub> and *C*<sub>2h</sub> structures as observed for the corresponding free ligands, respectively.† Although attempts to obtain a single crystal of **2** failed, recrystallization of the racemic mixture from toluene–EtOH gave crystals suitable for an X-ray analysis (Fig. 2).‡ The geometry at Ru is octahedral, with four S atoms in the equatorial plane and two Cl atoms in the axial sites. An up-down-up-down conformation of a coordinated quadridentate thiamacrocycle is rare, having a sole precedent in the seven-coordinate [Hg([16]aneS<sub>4</sub>)(ClO<sub>4</sub>)<sub>2</sub>].<sup>9</sup> The Ru–S distance (average 2.368 Å) is significantly longer than those [2.291(3)–2.308(2) Å] found for the saturated quadridentate thiamacrocycle with a *syn*-conformation in *trans*-[RuH(Cl)(L)] (L = Me<sub>4</sub>[14]aneS<sub>4</sub>, Me<sub>6</sub>[15]aneS<sub>4</sub>) and [Ru<sub>2</sub>H(μ-H)Cl(Me<sub>4</sub>[14]aneS<sub>4</sub>)<sub>2</sub>]<sup>2+</sup>.<sup>10</sup> The Cl–Ru–Cl angle [179.1(2)°] is essentially linear. To reduce the steric strain induced by the planar coordination, the dihedral angle (average 76.5°) between two tolyl planes becomes more acute than that of the free ligand, while the decrease in the corresponding angle (74.5°) between the 4S and tolyl planes is not so extensive. As expected from the observed conformation of the thiamacrocycle, the Ru atom lies exactly in the least-squares 4S plane, the deviation being only 0.009(5) Å.

The above results suggest that the M[(+)-(R,R)-**1**] moiety possesses the helical cavity at the axial sites capable of discriminating the chirality of prochiral substrates through coordination and may serve as a novel type of chiral catalyst. Indeed, the latter possibility is shown by the asymmetric hydrosilylation of PhCOMe with Ph<sub>2</sub>SiH<sub>2</sub> with the catalyst prepared from [RhCl(C<sub>8</sub>H<sub>12</sub>)<sub>2</sub>] and (+)-(R,R)-**1** *in situ* which gave (*S*)-PhCH(OH)Me after hydrolysis in 35 and 57% chemical and optical yields, respectively.

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## Footnotes

† The structures were solved by direct methods for (+)-(R,R)-**1** and by the Patterson method for the racemate of *trans*-[RuCl<sub>2</sub>(**1**)] and refined by full-matrix least-squares for intensity data collected on a Rigaku AFC-5R diffractometer using graphite-monochromated Mo-Kα radiation (λ = 0.7107 Å). Atomic coordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Information for Authors, Issue No. 1.

Crystal data for (+)-(R,R)-**1**: C<sub>38</sub>H<sub>44</sub>S<sub>4</sub>, M = 629.0, monoclinic, P2<sub>1</sub>, a = 11.671(7), b = 25.839(8), c = 12.687(5) Å, β = 110.10(3)°, U = 3592(5) Å<sup>3</sup>, Z = 4, D<sub>c</sub> = 1.163 g cm<sup>-3</sup>, μ(Mo-Kα) = 2.76 cm<sup>-1</sup>, F(000) = 1344, final R(R<sub>w</sub>) = 0.0677 (0.0534) for 3984 reflections [I > 3.0σ(I)] and 415 variables. There are two independent molecules in the asymmetric unit, the structures of which are essentially the same. The possibility of the alternative enantiomer, (–)-(S,S)-**1** is ruled out by Roger's test,<sup>8</sup> the final R(R<sub>w</sub>) factors then being 0.0681(0.0537).

For *trans*-[RuCl<sub>2</sub>(**1**)] (racemate): C<sub>38</sub>H<sub>44</sub>S<sub>4</sub>Cl<sub>2</sub>Ru·2C<sub>6</sub>H<sub>5</sub>Me·EtOH, M = 1031.3, triclinic, P1̄, a = 15.174(4), b = 15.977(5), c = 13.304(4) Å, α = 114.13(2), β = 109.07(2), γ = 66.51(2)°, U = 2647(1) Å<sup>3</sup>, Z = 2, D<sub>c</sub> = 1.294 g cm<sup>-3</sup>, μ(Mo-Kα) = 5.78 cm<sup>-1</sup>, F(000) = 1080, final R(R<sub>w</sub>) = 0.066 (0.057) for 2839 reflections [I > 3.0σ(I)] and 444 variables.

‡ (+)-(R,R)-**1**: mp 168–169 °C; EI-MS (70 eV) m/z 628 ([M]<sup>+</sup>). (–)-(S,S)-**1**: mp 167–168 °C; EI-MS (70 eV) m/z 628 ([M]<sup>+</sup>). <sup>1</sup>H NMR (δ, CDCl<sub>3</sub>) for both enantiomers 0.97 (s, 12H, MeC), 1.92 (s, 12H, MeAr), 2.65 (d, J 11.6 Hz, 4H, CH<sub>2</sub>), 2.85 (d, J 11.6 Hz, 4H, CH<sub>2</sub>), 7.06 (d, J 7.6 Hz, 4H, Ar), 7.11 (d, J 7.6 Hz, ArH), 7.18 (dd, J 7.6, 7.6 Hz, 4H). (*R,S*)-**1**: mp 282–283 °C; EI-MS (70 eV) m/z 628 ([M]<sup>+</sup>); <sup>1</sup>H NMR (δ, CDCl<sub>3</sub>) 0.77 (s, 6H, MeC), 1.05 (s, 6H, MeC), 1.91 (s, 12H, MeAr), 2.52 (d, J 11.8 Hz, 4H, CH<sub>2</sub>), 3.01 (d, J 11.8 Hz, 4H, CH<sub>2</sub>), 7.03 (d, J 7.8 Hz, 4H, Ar), 7.09 (d, J 7.8 Hz, ArH), 7.18 (dd, J 7.8, 7.8 Hz, 4H). *trans*-[RuCl<sub>2</sub>{(+)-(R,R)-**1**}] : mp > 300 °C; FAB-MS (3-nitrobenzyl alcohol) m/z 802 ([M]<sup>+</sup>); <sup>1</sup>H NMR (δ, CDCl<sub>3</sub>) 1.25 (s, 12H, MeC), 1.91 (s, 12H, MeAr), 2.74 (d, J 11.2 Hz, 4H, CH<sub>2</sub>), 3.45 (d, J 11.2 Hz, 4H, CH<sub>2</sub>), 7.19–7.26 (m, 12H, Ar). *trans*-[RuCl<sub>2</sub>{(*R,S*)-**1**}] : mp > 300 °C; FAB-MS (3-NOBA) m/z 802 ([M]<sup>+</sup>); <sup>1</sup>H NMR (δ, CDCl<sub>3</sub>) 1.04 (s, 6H, MeC), 1.17 (s, 6H, MeC), 1.98 (s, 12H, MeAr), 2.62 (brd, J 11.3 Hz, 4H, CH<sub>2</sub>), 3.29 (d, J 11.3 Hz, 4H, CH<sub>2</sub>), 7.28 (d, J 7.8 Hz, 4H ArH), 7.32 (dd, J 7.8, 7.8 Hz, 4H, Ar), 7.47 (d, J 7.8 Hz, 4H, Ar).

## References

- S. R. Cooper and S. C. Rawle, *Struc. Bonding (Berlin)*, 1990, **72**, 1; A. J. Blake and M. Schröder, *Adv. Inorg. Chem.*, 1990, **35**, 1; T. Yoshida, T. Adachi and T. Ueda, *Pure Appl. Chem.*, 1990, **62**, 1127.
- M. Lemaire, J. Buter, B. K. Vriesema and R. M. Kellogg, *J. Chem. Soc., Chem. Commun.*, 1984, 309; M. Lemaire, B. K. Vriema and R. M. Kellogg, *Tetrahedron Lett.*, 1985, **26**, 3499.
- E. C. Constable, *Prog. Inorg. Chem.*, 1994, **42**, 67; E. C. Constable, *Tetrahedron*, 1992, **48**, 10013; R. Krämer, J.-M. Lehn, A. D. Lian and J. Fischer, *Angew. Chem., Int. Ed. Engl.*, 1993, **32**, 703; J. K. Judice, S. J. Keipert and D. J. Cram, *J. Chem. Soc., Chem. Commun.*, 1993, 1323; S. W. A. Bligh, N. Choi, E. G. Evagorou, W.-S. Li and M. McPartlin, *J. Chem. Soc., Chem. Commun.*, 1994, 2399; D. E. Fenton, R. W. Matthews, M. McPartlin, B. P. Murphy, I. J. Scowen and P. A. Tasker, *J. Chem. Soc., Chem. Commun.*, 1994, 1391 and references cited therein.
- T. Yoshida and T. Ueda, to be published.
- J. Buter and R. M. Kellogg, *J. Org. Chem.*, 1981, **46**, 4481.
- Y. Okamoto, M. Kawashima and K. Hatada, *J. Chromatogr.*, 1986, **363**, 173.
- R. Schmid, M. Cereghetti, B. Heiser, P. Schönholzer and H. J. Hansen, *Helv. Chim. Acta*, 1988, **71**, 897.
- D. Roger, *Acta Crystallogr., Sect. A*, 1981, **37**, 734.
- T. E. Jones, L. S. W. L. Solol, D. B. Rorabacher and M. D. Glick, *J. Chem. Soc., Chem. Commun.*, 1979, 140.
- T. Yoshida, T. Adachi, T. Ueda, F. Goto, K. Baba and T. Tanaka, *J. Organomet. Chem.*, 1994, **473**, 225.