New potential access to ethylenic pseudodipeptides through catalytic alkene metathesis

Florence Garro-Helion and Franqois Guibe*

Institut de Chimie Moléculaire d'Orsay, Laboratoire des Réactions Organiques Sélectives, associé au CNRS, Bât. 420, Université *Paris-Sud, 91405 Orsay, France*

Dialkenenic amide 1 (\mathbb{R}^1 **=** CH_2Ph **,** \mathbb{R}^2 **= H) is readily converted, through ruthenium-catalysed metathetic ring closure, to ethylenic lactam 2, a close precursor of Z-ethylenic pseudodipeptide 3.**

Backbone modifications of biologically active peptides constitute an increasingly popular strategy which may be aimed at various goals, such as the improvement of metabolic stability, the obtention of analogues with agonistic or antagonistic properties or the elucidation of structure-activity relationships.¹ Isosteric replacement of the amide bond by an ethylenic linkage is of particular interest since the three-dimensional structures of the *E* and Z HC=CH linkages closely resemble those of the *truns* and cis amide bonds respectively.2 Although disfavoured compared to the *trans* one, the *cis* peptide bond geometry is nevertheless not uncommonly encountered in naturally occurring peptides and it has been suggested that it could play an important role in their biological activity.3 In that respect, replacement of the amide bonds by surrogates in **a** locked cis or *trans* configuration such as the *Z-* and E-ethylenic bonds is of obvious interest.

No stereoselective syntheses of dipeptide Z-ethylenic isosteres have been reported to date.4 Recently, Rich and coworkers5 have prepared the Z-ethylenic isostere of *(N*methylleucy1)leucine as a mixture of (easily separable) diastereoisomers by alkylation with iso-butyl iodide of the *(N*methylleucy1)glycine analogue. Syntheses of dipeptide *E*ethylenic isosteres which incorporate stereocontrol at the C-2 position of the 5-aminopent-3-enoic unit are relatively few. If we except Hopkins' convergent approach,⁶ which is based on Julia's alkene synthesis and in which coupling partners are first synthesized in optically pure form, all other protocols involve simultaneous positioning of the double bond and stereoselective building of the C-2, or more rarely the C-5, chiral centres by way of sigmatropic,⁷ anti S_{N^2} or $S_{E'}^{8,9}$ processes. Such strategies require, as a preliminary step, the preparation of optically pure allylic precursors which are not always easily accessible.

We propose a different, convergent approach whose key step is an intramolecular alkene metathesis reaction on diethylenic amides of general structure **1** leading to ethylenic lactams **2,** which are direct precursors of the dipeptide Z-ethylenic isosteres **3.** For our purpose, use of the ruthenium metathesis catalyst 4,^{10,11} recently introduced by Grubbs and coworkers, seemed to be particularly well suited, owing to its exceptional resistance to poisoning by polar molecules. **As** for the precursors of **1,** allylic mines **of** general structure *5* are readily available from the chiral pool of amino acids, while optically active α -substituted β , y-ethylenic carboxylic acids 6 may conceivably be obtained by α -regioselective^{12,13} and diastereoselective13 alkylation (or aldolisation) of crotonic or vinylacetic acid derivatives containing chiral auxiliary groups.

Setting aside temporarily the problem of stereoselective introduction of the $C²$ substituent, and as a test of the general feasibility of our project, we report herein the results of our preliminary investigations into the synthesis of the Phe-Gly Zethylenic isostere, **1** with $R^1 = CH_2Ph$, $R^2 = H$.

For nitrogen protection of the amide bond in **1,** we selected the ferrocenylmethyl (Fcm) group **7** because: (i) unlike many other N-substituted aminoacids, the N-Fcm derivatives may be used in conventional peptidic coupling reactions (DCC/DMAP) without apparent difficulties;¹⁴ (ii) the Fcm group is readily removed by trifluoroacetolysis;¹⁴ (iii) the Fcm group could be expected to be bulky enough to induce, at least partially, a *syn* orientation of the two ethylenic appendages in **1** (as effectively represented here). Using well documented chemistry, the *N*ferrocenylmethylamine **lOf** was prepared, as shown in Scheme 1, in six steps and 58% overall yield from N-Boc-Lphenylalanine **8.**

Acylation of 10 with vinylacetyl chloride (Et₃N, Et₂O, -15 °C then 0 °C, 1 h) gave 11 in 95% yield which exists (NMR) as a *ca.* 1:1 mixture of its two rotameric forms. Treatment of **11** with 5-10 moI% of catalyst **4** (Scheme 2) produced lactam **12** in 85-90% yield. Removal of the Fcm group from **12** was achieved by trifluoroacetolysis in the presence of triethylsilane as the carbocation scavenger. Attempted hydrolysis of the deprotected lactam **13** under the

Scheme 1 *Reagents and conditions:* **i, Bu'OCOCl(1 equiv.),** *N-* methylmorpholine (1 equiv), thf, -20 °C, 5 min, then HN(OMe)Me+HCl (1 equiv.), **NEt₃** (1.1 equiv.), -20 °C to room temp., 95%; ii, LiAlH₄ (1.25 mol. equiv.), Et₂O, room temp., 20 min, 83%; iii, H₂C=PPh₃ (2 equiv.), Me₂SOthf, -78 °C, 15 min, then 40 °C, 12 h, 85%; iv, $CF_3CO_2H-CH_2Cl_2$ 1:1 *(v/v),* **room temp., 2 h, 100°C; v, ferrocenecarbaldehyde (I equiv.), 4 A** molecular sieves, CH_2Cl_2 , room temp., 6 h; vi, NaB H_4 (0.5 mol. equiv.), **room temp., 40 min, 86% (two steps)**

Chem. Commun., **1996 641**

rather harsh standard conditions of hydrolytic cleavage of amide bonds **(6** mol dm-3 HC1, 100°C, several h) did not give satisfactory results. Compound **13** was therefore converted (75% yield) by treatment with triethyloxonium tetrafluoroborate to lactim ether **14** which was then quantitatively hydrolysed under mild acidic conditions $(0.25 \text{ mol dm}^{-3}$ HCI, room temp., 6 days) similar to those used by Schollkopf for cleavage of cyclic bis-lactim ethers.15 The desired dipeptide Zethylenic isostere **15** was thus obtained in 35% overall yield from Boc-L-phenylalanine. Neither during the metathetic nor during the hydrolytic process could any migration of the ethylenic bond be detected. The fact that the hydrolytic cleavage of imidate **14** does not stop at the ethyl ester stage, notwithstanding the mild acidic conditions used, is worthy of note. Such pronounced lability of the ester group is likely to be the result of intramolecularl6 general acid catalysis by the neighbouring NH3+ group. Ring cleavage on **14** was also carried out in deuterium oxide, under which conditions no detectable (NMR, *GC/MS)* deuterium incorporation at the carbon next to the carbonyl group could be detected. Such absence of deuterium incorporation is a strong indication that the hydrolytic ring opening of lactim ethers 14 bearing an R² substituent should occur without epimerization.

To check the conservation of enantiomeric purity during our synthesis, the reaction sequence described above was repeated starting from optically pure N-Boc-D-phenylalanine and from racemic N-Boc-DL-phenylalanine as well. Optical purities of intermediates were determined by HPLC on a chiral column (Chiralcel OD, J. T. Baker Co). Up to the Boc-allylamine stage **9,** virtually complete retention of chiral integrity was observed, resulting in enantiomeric excesses > 98.5%. On the other hand, important racemization was found to occur, lowering the enantiomeric excesses to 62%, during the Boc to Fcm transprotection operations. No further racemization was observed during the metathetic process or the final steps leading to **15.**

Work is in progress to modify our protective group strategy in order to avoid the racemization problem mentioned and to generalize the present synthetic approach to homochiral 2,5-disubstituted dipeptide Z-ethylenic isosteres. The transformation of ethylenic Z-isosteres to the corresponding *E*isomers by one of the many methods available for alkene inversion¹⁷ is also under study.

Scheme 2 *Reagents and conditions:* i, **4** (5-10 mol%), benzene, Ar atmosphere, room temp., 6-7 h, 85-90%; ii, CF₃CO₂H-CH₂Cl₂ 1:1 (v/v), Et₃SiH (2 equiv.), room temp., 9 h, 92%; iii, Et₃O·BF₄ (1.25 equiv.), CH2C12, Ar atmosphere, room temp., 30 min, then phosphate buffer; iv, 0.4 mol dm⁻³ HCl (2 equiv.), Ar atmosphere, room temp., 7 days, then Dowex **50-X8,** 100%

We thank the Glaxo Company (Les Ulis, France) for a doctoral scholarship (to F. G.-H).

Footnotes

t All compounds were fully characterized by IR, 'H and 13C NMR spectroscopy, MS and by elemental analysis. When necessary, 'H NMR attributions were ascertained by irradiation or COSY experiments.

References

- 1 A. F. Spatola in: *Chemistry and Biochemistry of Amino Acids, Peptides and Proteins,* ed. B. Weinstein, Marcel Dekker, New York, 1983, vol. 7, pp. 267-357; J. **S.** Kaltenbronn, J. P. Hudspeth, E. A. Lunney, B. M. Mischiewicz, E. D. Nicolaides, J. T. Repine, W. H. Roark, M. A. Stier, F. J. Tinney, P. **K.** W. Woo and A. D. Essenberg, *J. Med. Chem.,* 1990, 33, 838; D. Tourwé, *Janssen Chim. Acta*, 1985, 3, 3.
- 2 M. **M.** Hann, P. G. Sammes, P. D. Kennewell and J. B. Taylor, *J. Chem. SOC., Perkin Trans. I,* 1982, 307.
- 3 J. Zabrocki, J. B. Dunbar, Jr., K. W. Marshall, M. V. Toth and G. R. Marshall, *J. Org. Chem.,* 1992,57, 202; J. Zabrocki, G. D. Smith, J. B. Dunbar, Jr., H. Iijima and G. **R.** Marshall, *J. Am. Chem. SOC.,* 1988, 110, 5875.
- 4 *cis*-Disubstituted benzene rings and heterocycles have been used as surrogates of the cis-amide bond: M. Elseviers, L. Van der Auwera, H. Pepermens, D. Tourwé and G.VanBinst, *Biochem. Biophys. Res. Commun.,* 1988,154,515; A. D. Abell, D. A. Hoult and E. J. Jamieson, *Tetrahedron Lett.,* 1992, 33, 5831; L. W. Boteju and V. J. Hruby, *Tetrahedron Lett.,* 1993,34, 1757; see also ref. 3.
- *5* A. C. Bohnsted, **J.** V. N. Vara Prasad and D. H. Rich, *Tetrahedron Lett.,* 1993,34, 5217.
- 6 A. Spaltenstein, P. A. Carpino, F. Miyake and **P.** B. Hopkins, *J. Org. Chem.,* 1987,52,3759; *Tetrahedron Lett.,* 1986,27,2095; L. **S.** Lehman de Gaeta, M. Czarniecki and A. Spaltenstein, J. *Org. Chem.,* 1989, 54, 4004.
- 7 Y. F. Yong and M. A. Lipton, *Bioorg. Med. Chem. Lett.,* 1993,3,2879; K. M. Bol and R. M. J. Liskamp, *Tetrahedron,* 1992,48,6425; J. **S.** Wai, T. E. Fisher and M. W. Embrey, *Tetrahedron Lett.,* 1995,36, 3461; T. Allmendinger, P. Furet and E. Hungerbuhler, *Tetrahedron Lett.,* 1990, 31, 7297; J. K. Whitesell and R. M. Lawrence, *Chirality,* 1989, 1, 89; J. A. McKinney, D. **F.** Eppley and R. M. Keenan, *Tetrahedron Lett.,* 1994,35, 5985.
- **8** T. Ibuka, **K.** Nakai, **H.** Habashita, Y. Hotta, N. Fujii, N. Mimura, Y. Miwa, T. Taga and Y. Yamamoto, *Angew. Chem., Int. Ed. Engl.,* 1994, 33, 652; T. Ibuka, H. Habashita, A. Otaka, N. Fujii, Y. Oguchi, T. Uyehara and Y. Yamamoto, *J. Org. Chem.,* 1991,56,4370; P. Wipf and P. C. Fritch, *J. Org. Chem.,* 1994, *59,* 4875.
- 9 R. Beresis and J. **S.** Panek, *Bioorg. Med. Chem. Lett.,* 1993, 3, 1609.
- 10 **S.** T. Nguyen, R. H. Grubbs and J. W. Ziller, *J. Am. Chem. SOC.,* 1993, 115,9858; G. C. Fu, **S.** T. Nguyen, R. H. Grubbs and J. W. Ziller,J. *Am. Chem. SOC.,* 1993,115,9856.
- 11 While this work was in progress, several synthetic applications of the metathetic ring closure reaction using Grubbs' ruthenium catalyst appeared in the literature: B. C. Borer, **S.** Derenberg, H. Bieraugel and U. K. Pandit, *Tetrahedron Lett.,* 1994, 35, 3191; **A.** Kinoshita and M. Mori, *Synlett.,* 1994, 1020; S.-H. Kim, N. Bowden and R. H. Grubbs, *J.Am. Chem. Soc.,* 1994,116, 10801; J. P. Morken, M. **T.** Didiuk, M. **S.** Visser and A. H. Hoveyda, *J. Am. Chem.* SOC., 1994, 116, 3123; **S.** J. Miller, S.-H. Kim, Z. R. Chen and R. H. Grubbs, *J. Am. Chem. Soc.,* 1995,117,2108; **S.** J. MillerandR. H. Grubbs,J.Am. *Chem.* SOC., 1995, 117, 5855.
- 12 N. Petragnani and M. Yonashiro, *Synthesis,* 1982, 521.
- 13 D. Caine, in *Comprehensive Organic Synthesis,* ed. B. M. Trost, Pergamon, Oxford, 1991, vol. 3, pp. 42-53.
- 14 H. Eckert and C. Seidel, *Angew. Chem., Int. Ed. Engl.,* 1986, 25, 159.
- 15 See for instance: U. Schollkopf, W. Hartwig and K.-H. Pospischil, *Synthesis,* 1981,966; U. Schollkopf, U. Groth, K. 0. Westphalen and C. Deng, *Synthesis,* 1981, 969; **R.** M. Williams, *Synthesis* of *Optically Active a-Amino Acids,* Pergamon, Oxford, 1989, ch. 1.
- 16 A. J. Kirby, *Adv. Phys. Org. Chem.,* 1980,17, 183; F. M. Menger, *Acc. Chem. Res.,* 1985,18, 128; **K.** N. Houk, J. A. Tucker and A. E. Dorigo, *Acc Chem. Res.,* 1990,23, 107.
- 17 P. E. Sonnet, *Tetrahedron,* 1980, 36, 557.

Received, I st December 1995; Com. 51078236