Homochiral synthesis of an aza analogue of S-adenosyl-L-methionine (AdoMet) and its binding to the *E. coli* methionine repressor protein (MetJ)

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A synthesis of 5'-{N-[(S)-3-amino-3-carboxypropyl]methylamino}-5'-deoxyadenosine from p-adenosine and (S)-glutamic acid is described; this product, AzaAdoMet 2, has a p K_a of 7.10 for the tertiary amino group and so acts as a charge-switchable analogue of AdoMet 1, a key component of polyamine biosynthesis; the binding of both 1 and 2 to the *E. coli* methionine repressor protein is investigated by X-ray crystallography.

(S)-Adenosyl-L-methionine (AdoMet) 1 is well known as an essential co-factor involved in many biological processes. In particular, it has a central role as a biological methylating agent and in the ethylene-polyamine cycle.¹ We describe here a homochiral synthesis of the aza-desthio analogue of 1, AzaAdo-Met 2. This analogue has previously been prepared as a mixture of epimers² that can inhibit *E.coli* tRNA uracil-5-methyl-transferase³ and *Eco*RI DNA methylase.⁴ However, these earlier syntheses² have failed to control chirality at the amino acid centre, so all existing biological work has necessarily used a mixture of epimers. We therefore planned a homochiral synthesis based on D-adenosine and (S)-glutamic acid 3.

The protected product 4 was assembled from the two key intermediates; 5, derived from D-adenosine, and 6, derived from (S)-glutamic acid. 2',3'-O,O-(1-Methylethylidene)-D-adenosine⁵ was converted into the 5'-O-tosylate by treatment with toluene-*p*-sulfonyl chloride in dry pyridine,⁶ which was stirred in liquid methylamine to give the amine intermediate 5.7 The synthesis of 6 is summarised in Scheme 1. (S)-Glutamic acid 3



Scheme 1 Reagents and conditions: i, paraformaldehyde, p-TsOH (cat.), benzene, Dean-Stark, reflux, 7 h, (61%); ii, (COCl)₂ (1.1 equiv.), CH₂Cl₂, DMF (cat.), room temp., 1 h, (94%); iii, N-hydroxypyridine-2-thione sodium salt, CF₃CH₂I (5 equiv.), DMAP (cat.), CH₂Cl₂, reflux, irradiate, 1 h, (39–63%); iv, Pri₂NEt, MeCN, 55 °C, 2 d, (51%); v, BF₃.Et₂O, EtSH, CH₂Cl₂, room temp., 24 h, (33–69%)

was protected⁸ as the oxazolidinone 7, which was treated with oxalyl chloride to afford the corresponding acid chloride 8. This was converted into the alkyl iodide 6 via a Barton-type⁹ decarboxylative iodination using 2-iodo-1,1,1-trifluoroethane as iodine donor.^{10†} The key intermediates 5 and 6 were then heated together in acetonitrile solution in the presence of *N*,*N*diisopropylethylamine to give 4. Attempts to remove the benzyloxycarbonyl group from 4 by catalytic hydrogenation with 10% Pd–C proved unsuccessful; reactions were incomplete after several days and a mixture of products was detected by TLC. Catalytic transfer hydrogenation¹¹ also afforded mixtures of products. However, treatment of 4 with BF₃–EtSH complex¹² effected complete removal of all protecting groups. The final product 2^{\ddagger} was purified on Dowex 50WX4-400 ion-exchange resin.^{2a}

Compound 2 was dissolved in 10% D₂O in H₂O (500 mm³)§, and acetone (2 mm³) was added as the NMR internal reference (δ 2.214). Spectra were recorded on a Bruker AMX 500 spectrometer at 298 K. The pK_a of the tertiary amino group was determined from a plot of the chemical shift of the *N*-methyl group *vs* pH. The data were fitted using ULTRAFIT¹³ to give a value of pK_a = 7.10 ± 0.07. As a result, AzaAdoMet may be employed in either the unprotonated, uncharged form or in its cationic, protonated form by controlling the pH.

The MetJ repressor protein, the product of the *metJ* gene, is a stable dimer of two identical subunits of molecular mass 11,996 Da. It is responsible for the repression of its own gene and of those coding for enzymes involved in methionine and AdoMet synthesis.14 The protein dimer binds two molecules of the co-repressor AdoMet and this elevates over 100-fold the affinity of the protein for the operator DNA sequence.15 AzaAdoMet is isosteric with AdoMet¹⁶ and shows a pHdependent affinity for the repressor protein in the presence of DNA. At pH 7.4, where the unprotonated form dominates, binding to the MetJ protein was found to be very weak. However, at pH 5.5, when AzaAdoMet exists predominantly as the protonated form, binding to the protein is 10 times stronger than for AdoMet itself at the same pH.¹⁶ This result supports the hypothesis that the net positive charge on the co-repressor is an essential part of the mechanism of action of the complex.^{15a}

Crystals of the binary complex between the *E. coli* MetJ repressor protein and AzaAdoMet were grown from a solution of PEG 600 by vapour diffusion methods.¹⁷ They were isomorphous with the crystals of the MetJ–AdoMet complex (space group *P*2₁; *Z* = 2; cell dimensions *a* = 35.91, *b* = 63.08, *c* = 44.20 Å, $\alpha = \gamma = 90.0$, $\beta = 102.7^{\circ}$).¹⁸ Data to 2.0 Å resolution were recorded from crystals containing the corepressor analogue AzaAdoMet and were processed with XDS,¹⁹ then merged and scaled using programs from the CCP4 program suite²⁰¶ A model of the MetJ–AzaAdoMet complex was built using the molecular graphics program FRODO²¹ and refined in TNT²² using all data on the resolution range 2.0–2.3 Å∥.

The structure of AzaAdoMet bound to the protein confirms the (S)- configuration of the amino acid. For both the natural



Fig. 1 (a) The X-ray crystal structure of the ribofuranose moiety of 2, showing its $O^{4'}$ -endo character. (b) The X-ray structure of AdoMet 1 as bound in a crystalline complex with MetJ showing the syn-conformation of the glycosylic bond. (c) The X-ray structure of AzaAdoMet 2 as bound in a crystalline complex with MetJ with the same orientation for the adenine base as in (b) above.

co-repressor, AdoMet, and the isosteric analogue, AzaAdoMet, the purine ring lies in a hydrophobic pocket and is secured by two hydrogen bonds to the protein. The glycosylic bond of the nucleoside adopts the unusual *syn*-conformation with torsion angle χ +76° for AdoMet (+77° for AzaAdoMet). The pucker of the ribose ring is ${}^{0}T_{4}$, with largely O⁴-*endo* character (Fig. 1). AdoMet and AzaAdoMet are almost superimposable in these respects.

The ammonium and carboxylate groups of the co-repressor lie at the surface of the protein where they are exposed to solvent. This part of the AdoMet is flexibly bound and the amino acid moieties in AdoMet and its analogue differ in conformation, making hydrogen bonds to different protein amino acid side-chains. This divergence between the structures begins with the exocyclic ribose torsion angle γ [C³'-C⁴'-C⁵'-S(N)], which is +17° in AdoMet and +62° in AzaAdoMet, and continues through to the amino acid carboxyl group. This work has established the value of AzaAdoMet as an isosteric, chargeswitchable analogue of AdoMet. Its binding to other proteins, *e.g.* M.*HhaI* and M.*MspI*, is under investigation.

Footnotes

 \dagger Benzene could be replaced by toluene in step 1 but with significant reduction in yield of 7. The yield of pure 6 was variable because of its instability on silica; isolated yields ranged from 39 to 63%. Step 2 was also

performed replacing dichloromethane by benzene as solvent, but lower yields were obtained.

[‡] Spectral data for **2** (free base form): $\delta_{\rm H}$ (D₂O, 400 MHz, J/Hz) 8.10 (s, 1 H), 7.98 (s, 1 H), 5.88 (d, 1 H, J 5), 4.61 (t, 1 H, J 5), 4.25–4.19 (m, 1 H), 4.11 (t, 1 H, J 5), 3.60 (dd, 1 H, J 5, 7), 2.88–2.76 (m, 2 H), 2.73–2.61 (m, 2 H), 2.27 (s, 3 H), 2.01–1.91 (m, 1 H) and 1.88–1.78 (m, 1 H). $\delta_{\rm C}$ (D₂O, 100 MHz) 174.0, 154.7, 152.0, 147.9, 139.4, 118.1, 87.5, 79.9, 72.6, 71.3, 58.5, 53.8, 53.5, 40.2 and 25.7.

§ The pK_a was corrected to compensate for the D₂O-H₂O mixture.²³

¶ X-Ray diffraction data collection and refinement statistics for MetJ-AzaAdoMet: $d_{\min} = 2.3$ Å, 7497 independant reflections, multiplicity = 2.1, completeness = 83%, $R_{\text{symm}} = 5.3\%$ [$R_{\text{symm}} = \Sigma_h |I(h) - \langle I(h) \rangle | / \Sigma_h |I(h)|$].

|| Summary of structure refinement of MetJ-AzaAdoMet: 1690 protein atoms, 54 water atoms, 52 AdoMet atoms, resolution range = 20–2.3 Å, 7497 reflections, R factor = 16.7% [*R*-factor = $\Sigma_h |F_o(h) - F_c(h)| / \Sigma_h F_o(h)$], Rmsd bonds = 0.019 Å, Rmsd angles = 2.63°.

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