Synthesis of 1*H*- and 5*H*-1,3-diazepines from azido- and tetrazolo-pyridines

Ales Reisinger and Curt Wentrup*

Chemistry Department, The University of Queensland, Brisbane, Queensland 4072, Australia

Stable 1*H*-1,3-diazepines 7–9, 10, 13, 14, 17 and 19 are obtained, often in high yields, by photolysis of trifluoromethyl-substituted azido- or tetrazolo-pyridines in the presence of alcohols or amines; in some cases, 5*H*-1,3-diazepines are also formed (11, 21 and 23).

While 1,4-diazepines have been investigated extensively,¹ the 1,3-diazepines are comparatively little known.^{1,2} Herein we report the first syntheses of stable, monocyclic *N*-unsubstituted 1*H*-1,3-diazepines, often in high yields. The compounds carrying CF₃ groups are of special interest in view of their physical and chemical properties.³

We have shown previously that 1,3-diazacyclohepta-1,2,4,6-tetraenes 4 are formed very efficiently on matrix photolysis of tetrazoles 1 and/or azides 2 via the 2-pyridylnitrenes 3.⁴ The trifluoromethyl groups appear to accelerate the tetrazole-azide tautomerism $(1 \rightleftharpoons 2)$ and increase the nitrene reactivity,⁴ thereby facilitating the formation of 4. We now report the synthesis of 1,3-diazepines by trapping of 4 with nucleophiles.

Photolysis[†] of azide **5** in dioxane-methanol (3:1) for 80 min at room temperature gave the 2-methoxy-4-trifluoromethyl-1,3-diazepine **7** in 92% yield after Kugelrohr distillation. The same compound was obtained in 47% yield by analogous photolysis of the isomeric tetrazolopyridine **6**. Evidence for the 1*H* structure of **7** is given by the ¹H and ¹³C NMR spectra.[‡] The broad NH signal at δ 4.80 vanished on addition of D₂O. The proton and ¹³C NMR assignments of this and the following compounds are based on the observed proton and fluorine coupling constants, homonuclear decouplings, DEPT and COSY experiments and Eu(fod)₃-induced chemical shifts.⁵

Proof of the diazepine ring structure was obtained by X-ray crystallography of the benzoylation product $8,\ddagger$ obtained in 87% yield (Scheme 1 and Fig. 1). The ring has a pronounced

boat structure. This compound was hydrogenated to 9 in 87% yield.[‡]

Similar photolysis of 5 in diisopropylamine (DIPA)-dioxane (1:3) (Scheme 2) afforded the 1*H*-1,3-diazepine 10 (76%) which, however, isomerised to the 5*H* tautomer 11 on distillation (100 °C) or preparative GC (130 °C). \ddagger

In compounds 7 and 10 the NH proton is specifically located at the 1-position shown as established by ¹H and ¹³C NMR spectroscopy, including homonuclear proton decoupling experiments as well as the H- and F-coupled ¹³C NMR spectra. In all 1*H*-1,3-diazepines where a choice exists, the NH proton is observed only in the sterically least hindered position; this includes also compounds **17a** and **19a** mentioned below. The isomeric 7- and 6-trifluoromethylpyridines **12** and **15** give rise to 1:2 mixtures of the 1*H*-1,3-diazepines **13a**,c and **14a**,c in overall yields of 70–72%. These compounds were identified by their ¹H and ¹³C NMR spectra. Benzoylation again gave 1-benzoyl derivatives **13b**,d and **14b**,d (Scheme 3).

The bis(trifluoromethyl)tetrazolopyridine 16 afforded the 1H-1,3-diazepines 17a,c in 93–95% yield after room temperature photolysis for 90 min.‡ The *N*-benzoyl derivative 17c was prepared as above.

Chlorotrifluoromethyl-substituted 1H-1,3-diazepines **19a–c** are obtained analogously, albeit in somewhat inferior yields (Scheme 3).

6,8-Dichlorotetrazolo[1,5-*a*]pyridine **20** was prepared from 2,3,5-trichloropyridine with hydrazine hydrate (95% yield of 2-hydrazino-3,5-dichloropyridine; mp 179 °C) followed by diazotization (87% yield of **20**; mp 78–79 °C; also obtainable by direct azidation of the trichloropyridine with NaN₃ in DMF). Photolysis of **20** in the appropriate amine gave **21a** and **b** in *ca*. 46% yield whereby one of the ring chlorines has been substituted for the dialkylamino group. The dichloropyridine **20**



Scheme 1 Reagents and conditions: i, hv, dioxane-methanol (3:1), 25 °C, 80 min, 92%; ii, as in (i), 47%, *ca*. 5% of 2-amino-3-trifluoromethylpyridine also formed (GC-MS); iii, BuLi, THF, -80 °C; iv, PhCOCl, -80 °C, 87%; v, H₂/Pd, MeOH, 25 °C, 14 h, 87%.



Fig. 1 ORTEP drawing of 8 (20% probability level)



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undergoes an exchange reaction with sodium alkoxide in alcohol in the dark at room temperature (3 days) to provide the alkoxychlorotetrazolopyridines 22 in quantitative yields. Photolysis of the latter in the presence of amines gives exclusively 5H-1,3-diazepines 23a-d‡ in good yields (Scheme 4).

Investigation of the factors influencing the stabilities and tautomerisation of 1H-, 5H-, *etc.* 1,3-diazepines is in progress and will be reported in due course.





This research was supported by the Australian Research Council. We are indebted to Dr C. H. L. Kennard and Mr K. Byriel for the X-ray structure determination.

Footnotes

[†] All photolyses were carried out using a 1000 W high pressure Hg lamp and quartz vessels. Solutions were purged with N₂. Products were purified by column chromatography on silica gel 100 and/or Kugelrohr distillation at 100 °C, 0.01 millibar (1 bar = 10^5 Pa).

‡ All new compounds gave satisfactory analytical (C, H, N) and/or HRMS data. ¹H and ¹³C NMR data are available for all compounds; J values are given in Hz. Selected data for 7 ¹H NMR (CDCl₃) & 3.73 (s, OMe), 4.80 (br, 1 H, NH), 4.94 (dd, 1 H, 6-H, $J_{5,6}$ 5.4, $J_{6,7}$ 7.8), 5.45 (dd, 1 H, 7-H, $J_{6,7}$ 7.8, $J_{1,7}$ 5.1) and 5.84 (d, 1 H, 5-H, $J_{5,6}$ 5.4); ¹³C NMR (CDCl₃) δ 56.1 (OMe), 110 (d, 6-C), 117 (d, 5-C), 121 (CF₃), 133 (d, 7-C), 138 (4-C) and 158 (2-C); v/cm⁻¹ 1688s. For 8 ¹H NMR (CDCl₃) & 3.52 (s, OMe) 5.98 (dd, 1 H, 6-H, J_{5,6} 5.7, J_{6,7} 7.1), 6.48 (dq, 1 H, 5-H, J_{5,6} 5.7, J_{5,CF3} 1.1), 6.54 (dd, 1 H, 7-H, $J_{6,7}$ 7.1, $J_{5,7}$ 0.4) and 7.40–7.62 (m, 5 H, Ar-H); ¹³C NMR (CDCl₃) δ 57 (OMe), 116 (5-C), 119 (6-C), 121 (CF₃), 129 (7-C), 138 (4-C), 149 (2-C), 167 (C=O), 127, 129, 132 and 134 (phenyl); v/cm⁻¹ 1691 (C=O) and 1648 (C=N); X-ray structure: see below. For 9 ¹H NMR (CDCl₃) δ 1.8–2.0 (m, 2 H, 6-H), 1.55-2.15 (m, 2 H, 5-H), 2.95 and 4.55 (m, 2 H, 7-H), 3.31 (OMe), 3.95 (m, 1 H, 4-H, ${}^{3}J_{CF3}$ 7) and 7.4 (m, 5 H, Ar-H); ${}^{13}C$ NMR δ 23.8 (5-C), 25.8 (6-C), 42.9 (7-C), 55 (OMe), 60.0 (4-C), 125.5 (CF₃), 156.3 (2-C), 169.2 (C=O), 127, 128, 131 and 136 (phenyl); v/cm⁻¹ 1692 (C=O) and 1659 (C=N). For 17a ¹H NMR (CDCl₃) δ 3.76 (OCH₃), 5.1 (br, 1 H, NH), 5.79 (s, 1 H, 5-H) and 6.04 (d, 1 H, 7-H, J_{1,7} 7.25); ¹³C NMR (CDCl₃) δ 57 (OCH₃), 112.6 (5-C), 112.8 (6-C), 120.6 (6-CF₃), 122.8 (4-CF₃), 136.5 (7-C), 139.5 (4-C) and 157.3 (2-C). For 23a ¹H NMR & 1.08 (t, 6 H, CH₃), 2.99 (s, 2 H, 5-H), 3.37 (q, 4 H, CH₂), 3.73 (OMe) and 6.69 (s, 1 H, 7-H); ¹³C NMR (CDCl₃) & 13 (2 CH₃), 39.5 (5-C), 42 (2 CH₂), 55 (OMe), 103 (6-C), 136 (7-C), 154 (4-C) and 159 (2-C).

Crystal data for 8: $C_{13}H_{11}F_{3}N_{2}O_{2}$, M = 296.25, monoclinic, space group $P2_{1}/n$, a = 10.742(2), b = 11.2020(10), c = 11.830(4) Å, $\alpha = 90^{\circ}$, $\beta = 104.434(9)^{\circ}$, $\gamma = 90^{\circ}$, U = 1378.6(5) Å³, F(000) = 608 electrons, Z = 4, $D_{c} = 1.427$ g cm⁻¹, μ (Mo-K α) = 0.124 mm⁻¹, Mo-K α radiation ($\lambda = 0.71073$ Å). Crystal size $0.10 \times 0.10 \times 0.10$ mm, T = 298(2) K. Data were collected on an Enraf-Nonius CAD4 four-circle diffractometer by theta scan type, within the limits $2.29 < \theta < 24.97^{\circ}$. 2414 unique reflections were collected ($R_{int} = 0.0180$); 1339 refl. with $I > 2\sigma(I)$, $R_1 = 0.0437$, $wR_2 = 0.1401$, s = 0.784. Structure solved by direct methods (SHELXS 86) and refined (F^2) (SHELXL 93). Hydrogens were located by difference methods and refined isotropically. All other atoms were refined anisotropically. Atomic coordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1. Any request to the CCDC for this material should quote the full literature citation and the reference number 182/1.

References

- J. T. Sharp, Comprehensive Heterocyclic Chemistry, ed. A. R. Katritzky and C. W. Rees, Pergamon, Oxford, 1984, vol. 7, pp. 593–651; B. A. Bunin and J. A. Ellman, J. Am. Chem. Soc., 1992, 114, 10997 and reference 3 therein.
- 2 J. Kurita, H. Kojima, M. Enkaku and T. Tsuchiya, Chem. Pharm. Bull., 1981, 29, 3696; H. Sawanishi, K. Tajima and T. Tsuchiya, Chem. Pharm. Bull., 1987, 35, 4101; H. Sawanishi and T. Tsuchiya, J. Chem. Soc., Chem. Commun., 1990, 723; P. Molina, A. Arques, A. Alias, M. Foces-Foces and A. L. Llamas-Saiz, J. Chem. Soc., Chem. Commun., 1992, 424.
- 3 Organofluorine Chemistry: Principles and Commercial Applications, ed. R. E. Banks, B. E. Smart and J. C. Tatlow, Plenum, New York, 1994.
- 4 R. A. Evans and C. Wentrup, J. Chem. Soc., Chem. Commun., 1992, 1062; R. A Evans, M. W. Wong and C. Wentrup, J. Am. Chem. Soc., 1996, 118, in the press.
- 5 For relevant NMR data on 1H-, 2H- and 3H-azepine see: E. Vogel, H.-J. Altenbach, J.-M. Drossard, H. Schmickler and H. Stegelmeier, Angew. Chem., Int. Ed. Engl., 1980, 19, 1016; D. Hamprecht, K. Polborn and W. Steglich, Angew. Chem., Int. Ed. Engl., 1995, 34, 1469; C. M. Daly, B. Iddon, H. Suschitzky, U. Jordis and F. Sauter, J. Chem. Soc., Perkin Trans. 1, 1988, 1933.

Received, 12th December 1995; Com. 5/08096G