## Palladium-catalysed cross-coupling reactions of aryl-, alkenyl- and alkynyl-iodonium salts and iodanes with terminal alkynes in aqueous medium

## Suk-Ku Kang,\* Hong-Woo Lee, Su-Bum Jang and Pil-Su Ho

Department of Chemistry, Sung Kyun Kwan University, Natural Science Campus, Suwon 440-746, Korea

## The ligand-free $Pd(OAc)_2$ -catalysed coupling reaction of aryl-, alkenyl- and alkynyl-iodonium tetrafluoroborate and iodanes with terminal alkynes proceeds readily in the presence of NaHCO<sub>3</sub> in aqueous medium to afford the substituted alkynes in high yields under mild conditions.

The palladium-catalysed cross-coupling of aryl or vinyl halides with terminal alkynes gives substituted alkynes at elevated temperature.<sup>1</sup> However, in the presence of cuprous iodide as a co-catalyst, the palladium-catalysed coupling can proceed under mild conditions in excellent yields, a procedure known as the Sonogashira reaction,<sup>2</sup> and which is utilized in the synthesis of a variety of natural products. Recently, Linstrumelle<sup>3</sup> reported that vinyl and aryl halides or triflates reacted with terminal acetylenes without addition of a copper salt to provide aryl- and alkenyl-substituted acetylenes using piperidine or pyrrolidine as base. In connection with our programs to utilize iodonium salts<sup>4</sup> in palladium-catalysed cross-coupling,<sup>5</sup> we have investigated the coupling of iodonium salts and iodanes with terminal alkynes. Here we report Pd-catalysed crosscoupling of aryl- and alkynyl-iodonium salts and iodanes with terminal alkynes under aqueous conditions without using Cu<sup>I</sup> compounds as co-catalysts and with high catalytic turnovers (Scheme 1).

The results of palladium-catalysed cross-coupling<sup>6</sup> of hypervalent iodonium salts<sup>7</sup> with terminal alkynes are summarized in Scheme 1 and Table 1. The phenylacetylene **1a** (1 equiv.) was treated with diphenyliodonium tetrafluoroborate **2a** (1 equiv.) in the presence of Pd(OAc)<sub>2</sub> (0.2 mol%) and NaHCO<sub>3</sub> (1 equiv.)

R <sup>1</sup> ————————————————————————————————————	R²l+Ph X⁻
1a R <sup>1</sup> = Ph b R <sup>1</sup> = Bu c R <sup>1</sup> = CH <sub>2</sub> OH d R <sup>1</sup> = $O$ O O O O O O O	2a $R^2 = Ph$ $X^- = BF_4^-$ b $R^2 = 2$ -Thienyl $X^- = -OTs$ c $R^2 = (E)$ -PHCH=CH $X^- = BF_4^-$ d $R^2 = PhC = C$
e R <sup>1</sup> = SiMe <sub>3</sub> R <sup>1</sup>	X <sup>−</sup> = BF <sub>4</sub> <sup>−</sup> ↓ i B <sup>2</sup>
n.	
<b>3a</b> $R^1 = Ph$ , $R^2 = Ph$ <b>b</b> $R^1 = Ph$ , $R^2 = 2$ -Thienyl	$j R^{1} = \underbrace{OBn}_{OBn}, R^{2} = Ph$
c R <sup>1</sup> = Ph, R <sup>2</sup> = ( <i>E</i> )-PhCH=CH d R <sup>1</sup> =Ph, R <sup>2</sup> = PhC=C e R <sup>1</sup> = Bu, R <sup>2</sup> = Ph ( D1 - Bu, D <sup>2</sup> = 0. This is the	<b>k</b> $R^1 = \bigcirc_{O = O \\ O \\$
f R <sup>1</sup> = Bu, R <sup>2</sup> = 2-Thienyl g R <sup>1</sup> = Bu, R <sup>2</sup> = PhC=C h R <sup>1</sup> = CH <sub>2</sub> OH, R <sup>2</sup> = Ph i R <sup>1</sup> = CH <sub>2</sub> OH, R <sup>2</sup> = 2-Thienyl	I $R^1 = SiMe_3, R^2 = Ph$ m $R^1 = SiMe_3, R^2 = 2$ -Thienyl n $R^1 = SiMe_3, R^2 = PhC \equiv C$

Scheme 1 Reagents and conditions: i,  $Pd(OAc)_2$  (0.2 mol%),  $NaHCO_3$  (1 equiv.),  $MeCN-H_2O$  (4:1), room temp.

in MeCN-H<sub>2</sub>O (4:1) at room temperature for 10 minutes to afford diphenylacetylene 3a in 96% yield. Of the four bases (NaHCO<sub>3</sub>, triethylamine, Pri<sub>2</sub>NEt, and pyrrolidine) tested, NaHCO<sub>3</sub> proved most effective. It is notable that even in the absence of base, comparable yields were afforded (entry 1 in Table 1). As solvent system, MeCN– $H_2O(4:1)$  was better than DMF in terms of yield and reactivity. Under the same conditions, when 2-thienyl(phenyl)iodonium tosylate 2b<sup>8</sup> was employed as iodonium salt, 2-thienyl-substituted acetylene 3b was obtained as the sole product in 89% yield (entry 2). Alkenyl(phenyl)iodonium salt  $2c^7$  was also coupled with the terminal alkyne 1a to affored alkenyl-substituted alkyne 3c in 83% yield (entry 3). This cross-coupling was applied to alkynyl(phenyl)iodonium salt 2d7 to provide the alkynylsubstituted acetylene 3d in 93% yield (entry 4).9 Treatment of hex-1-yne 1b with diphenyliodonium salt 2a, 2-thienyl-(phenyl)iodonium salt 2b and alkynyliodonium salt 2d provided the coupled products 3e-g, respectively (entries 5-7). It is noteworthy that the coupling of prop-2-yn-1-ol 1c with iodonium salts 2a and 2b afforded the substituted prop-2-yn-1-ols 3h and 3i without the formation of unsaturated aldehyde resulting from base-catalysed iosmerization<sup>10</sup> (entries 8 and 9). For the prop-2-ynylic cyclic carbonates 1d, diphenyliodonium and alkynyliodonium tetrafluoroborates 2a and 2d were smoothly coupled to afford the substituted prop-2-ynylic carbonates 3j and 3k without any deprotection (entries 10 and 11). Finally, treatment of trimethylsilylacetylene le with iodonium salts 2a-b and 2d afforded the substituted acetylenes **3l-n** (entries 12–14).

In considering a plausible mechanism for the formation of the diyne 3d, it is presumed that facile oxidative addition of the highly electrophilic alkynyliodonium salt 2d with Pd<sup>0</sup> gives polar and reactive organopalladium tetrafluoroborate

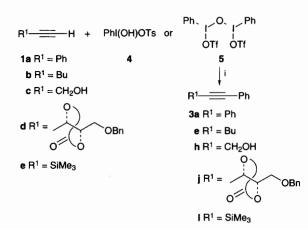
Table 1 Palladium-catalysed cross-coupling of iodonium salts with terminal alkynes a

Entry	Substrate	Iodonium salt	t/min	Product	Yield(%) <sup>b</sup>
1	1a	2a	10	3a	96(93)
2	1a	2b	20	3b	87(88)
3	1a	2c	15	3c	83(81)
4	1a	2d	10	3d	93(92)
5	1b	2a	20	3e	92(91)
6	1b	2b	25	3f	56(40)
7	1b	2d	20	3g	90(88)
8	1c	2a	5	3ĥ	98(95)
9	1c	2b	10	3i	93(91)
10	1d	2a	5	3j	97(92)
11	1d	2d	5	3k	93(91)
12	1e	2a	10	31	95(93)
13	1e	2b	10	3m	93(90)
14	1e	2d	10	3n	92(90)

<sup>a</sup> All the reactions were run with terminal alkynes (1 equiv.) and iodonium salt (1 equiv.) in the presence of Pd(OAc)<sub>2</sub> (0.2 mol%) and NaHCO<sub>3</sub> (1 equiv.) in MeCN-H<sub>2</sub>O (4:1) at room temperature. <sup>b</sup> The yields are isolated yields. The yields in parentheses are for reactions carried out in the absence of base.

(PhC=CPd+ BF<sub>4</sub><sup>-</sup>). This intermediate is subjected to alkynylation with phenylacetylene **1a** to form dialkynylpalladium(II) [(PhC=C)<sub>2</sub>Pd] followed by reductive elimination to afford the coupled diyne **3d**. Alternatively, we have investigated palladium-catalysed cross-coupling of terminal alkynes with iodanes. The results are summarized in Scheme 2 and Table 2.

We coupled the terminal alkyne **1a** with readily available [hydroxy(tosyloxy)iodo]benzene (HTIB), PhI(OH)OTs **4**<sup>11</sup> (also known as Koser's reagent), to afford the phenyl-substituted alkyne **3a** in 78% yield (entry 1). The reaction can also be carried out in the absence of base (entry 1 in Table 2).



Scheme 2 Reagents and conditions: i, Pd(OAc)<sub>2</sub> (0.2 mol%), NaHCO<sub>3</sub> (1 equiv.), MeCN-H<sub>2</sub>O (4:1), room temp., 5 or 10 min

**Table 2** Palladium-catalysed cross-coupling of iodanes with terminal alkynes<sup>a</sup>

Entry	Substrate	Iodonium salt	t/min	Product	Yield(%) <sup>b</sup>
1	1 <b>a</b>	4	25	3a	78(75) <sup>c</sup>
2	1a	5	20	3a	80(76)
3	1b	4	30	3e	76(70)
4	1b	5	20	3e	80(74)
5	1c	4	20	3h	91(89)
6	1c	5	20	3h	95(91)
7	1d	4	30	3j	91(86)
8	1d	5	20	3j	90(88)
9	1e	4	5	31	92(90)
10	1e	5	5	31	96(93)

<sup>*a*</sup> All the reactions were run with terminal alkynes (1 equiv.) and iodanes (1 equiv.) in the presence of Pd(OAc)<sub>2</sub> (0.2 mol%) and NaHCO<sub>3</sub> (1 equiv.) in MeCN-H<sub>2</sub>O (4:1) at room temperature. <sup>*b*</sup> The yields are isolated yields. The yields in parentheses are for reactions carried out in the absence of base. <sup>*c*</sup> The yield can be improved to 93% by addition of CuI.

When the terminal alkyne 1a was subjected to react with  $\mu$ -oxobis[(trifluoromethanesulfonato)(phenyl)iodine] 5 (also known as Zefirov's reagent<sup>12</sup>) under the same conditions, phenyl-substituted acetylene 3a was afforded in 80% yield (entry 2). Even without using base, a comparable yield was obtained. Treatment of hex-1-yne 1b with 4 or 5 furnished 3e, respectively (entries 4 and 5). It is notable that the coupling of prop-2-yn-1-ol 1c with Koser's or Zefirov's reagent afforded 3h without formation of any aldehyde (entries 6 and 7). For the prop-2-ynylic cyclic carbonate 1d, coupling with 4 or 5 afforded 3j without any deprotection (entries 8 and 9). Finally, reaction of trimethylsilylacetylene 1e with 4 or 5 afforded the substituted alkyne 3l (entries 10 and 11).

We thank generous financial support by KOSEF-OCRC and Ministry of Education (BSRI-95-3420).

## References

- 1 H. A. Dick and F. R. Heck, J. Organomet. Chem., 1975, 93, 259; L. Casar, J. Organomet. Chem., 1975, 93, 253.
- K. Sonogashira, Y. Tohda and N. Hagihara, *Tetrahedron Lett.*, 1975, 16, 4467;
  K. Sonogashira, in *Comprehensive Organic Synthesis*, ed. B. M. Trost and I. Fleming, Pergamon, Oxford, 1991, vol. 3, p. 521;
  R. Rossi, A. Carpita and F. Bellina, *Org. Prep. Proced. Int.*, 1995, 27, 129.
- 3 M. Alami, F. Ferri and G. Linstrumella, Tetrahedron Lett., 1993, 34, 6403.
- 4 For the recent reviews for iodonium salts: P. J. Stang, Angew. Chem., Int. Ed. Engl., 1992, 31, 274; R. M. Moriarty and R. K. Vaid, Synthesis, 1990, 431; M. Ochiai, Rev. Heteroat. Chem., 1989, 2, 92.
- 5 S.-K. Kang, K.-Y. Jung, C.-H. Park and S.-B. Jang, *Tetrahedron Lett.*, 1995, 36, 8047.
- 6 Pd-catalysed C-C bond formation: R. M. Moriarty, W. R. Epa and A. K. Awasthi, J. Am. Chem. Soc., 1991, 113, 6315; R. M. Moriarty and W. R. Epa, Tetrahedron Lett., 1992, 33, 4095; R. J. Hinkle, G. T. Poulter and P. J. Stang, J. Am. Chem. Soc., 1993, 115, 11626; K. Sugioka, M. Uchiyama, T. Suzuki and Y. Yamazaki, Nippon Kagaku Kaishi, 1985, 558; Chem. Abstr., 1986, 104, 109 137.
- 7 M. Ochiai, K. Sumi, Y. Takaoka, Y. Nagao, M. Shiro and E. Fujita, *Tetrahedron*, 1988, 44, 4095.
- 8 A. J. Morgida and G. F. Koser, J. Org. Chem., 1984, 49, 3643.
- 9 Diacetylene derivatives have been found to display liquid-crystalline and non-linear optical properties. See: A. E. Stiegman, E. Graham, K. J. Perry, L. R. Khundkar, L.-T. Cheng and J. W. Perry, J. Am. Chem. Soc., 1991, 113, 7658; C. Fouquey, J.-M. Lehn and J. Malthete, J. Chem. Soc., Chem. Commun., 1987, 1424.
- 10 K. Minn, Synlett, 1991, 115.
- 11 First synthesis: O. Neiland and B. Karek, J. Org. Chem. USSR (Engl. Transl.), 1970, 6, 889; G. F. Koser and R. H. Wettach, J. Org. Chem., 1977, 42, 1476.
- 12 N. S. Zefirov, V. V. Zhdankin and A. S. Koz'min, *Izv. Akad. Nauk SSR*, *Ser. Khim.*, 1983, 1682; V. V. Zefirov, Y. V. Dan'kov and A. S. Koz'min, *Zh. Org. Khim.*, 1984, **20**, 446; J. Gallos, A. Varvoglis and N. W. Alcock, *J. Chem. Soc.*, *Perkin Trans. 1*, 1985, 757; R. T. Hembre, C. P. Scott and J. R. Norton, *J. Org. Chem.*, 1987, **52**, 3650.

Received, 22nd November 1995; Com. 5/07639K