Palladium-catalysed cross-coupling reactions of aryl-, alkenyl- and alkynyl-iodonium salts and iodanes with terminal alkynes in aqueous medium

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The ligand-free Pd(OAc)₂-catalysed coupling reaction of **aryl-, alkenyl- and alkynyl-iodonium tetrafluoroborate and iodanes with terminal alkynes proceeds readily in the** presence of NaHCO₃ in aqueous medium to afford the **substituted alkynes in high yields under mild conditions.**

The palladium-catalysed cross-coupling of aryl or vinyl halides with terminal alkynes gives substituted alkynes at elevated temperature.' However, in the presence of cuprous iodide as a co-catalyst, the palladium-catalysed coupling can proceed under mild conditions in excellent yields, a procedure known as the Sonogashira reaction,² and which is utilized in the synthesis of a variety of natural products. Recently, Linstrumelle³ reported that vinyl and aryl halides or triflates reacted with terminal acetylenes without addition of a copper salt to provide aryl- and alkenyl-substituted acetylenes using piperidine or pyrrolidine as base. In connection with our programs to utilize iodonium salts⁴ in palladium-catalysed cross-coupling,⁵ we have investigated the coupling of iodonium salts and iodanes with terminal alkynes. Here we report Pd-catalysed crosscoupling of aryl- and alkynyl-iodonium salts and iodanes with terminal alkynes under aqueous conditions without using Cu^I compounds as co-catalysts and with high catalytic turnovers (Scheme 1).

The results of palladium-catalysed cross-coupling⁶ of hypervalent iodonium salts⁷ with terminal alkynes are summarized in Scheme 1 and Table 1. The phenylacetylene **la** (1 equiv.) was treated with diphenyliodonium tetrafluoroborate **2a (1** equiv.) in the presence of $Pd(OAc)_2 (0.2 \text{ mol\%)}$ and NaHCO₃ (1 equiv.)

R1- -н $\ddot{}$	R^2 l ⁺ Ph X^-
1a R ¹ = Ph b $R^1 = Bu$ $c R1 = CH2OH$ $dR^1 =$ OBn	2a R ² = Ph $X^- = BF_A^-$ b $R^2 = 2$ -Thienyl $X^- = -OTs$ c $R^2 = (E)$ -PHCH=CH $X^- = BF_A^-$ $d R^2 = PhC \equiv C$
e R ¹ = SiMe ₃ R1.	X^- = BF_4^- ∙R.
3a R ¹ = Ph, R ² = Ph $bR^1 = Ph$, $R^2 = 2$ -Thienyl	Î. $R^1 =$ OBn, R^2 = Ph
c R ¹ = Ph, R ² = (<i>E</i>)-PhCH=CH $dR^1 = Ph$, $R^2 = PhC = C$ e R^1 = Bu, R^2 = Ph	$k R^1 =$, $R^2 = PhC \equiv C$ OBn
f R ¹ = Bu, R ² = 2-Thienyl $gR^1 = Bu$, $R^2 = PhC \equiv C$ $h R1 = CH2OH$, $R2 = Ph$ $i R1 = CH2OH, R2 = 2-Thienvl$	$I \ R^1 =$ SiMe ₃ , R ² = Ph m R ¹ = SiMe ₃ , R ² = 2-Thienyl n R^1 = SiMe ₃ , R^2 = PhC=C

Scheme 1 *Reagents and conditions: i, Pd(OAc)₂ (0.2 mol%), NaHCO₃* (1 **equiv.), MeCN-H20 (4** : l), **room temp.**

in MeCN-H₂O $(4:1)$ at room temperature for 10 minutes to afford diphenylacetylene **3a** in 96% yield. Of the four bases (NaHCO₃, triethylamine, Prⁱ₂NEt, and pyrrolidine) tested, $NaHCO₃$ proved most effective. It is notable that even in the absence of base, comparable yields were afforded (entry 1 in Table 1). As solvent system, MeCN $-H_2O(4:1)$ was better than DMF in terms of yield and reactivity. Under the same conditions, when **2-thienyl(phenyl)iodonium** tosylate **2b8** was employed as iodonium salt, 2-thienyl-substituted acetylene **3b** was obtained as the sole product in 89% yield (entry 2). Alkenyl(pheny1)iodonium salt **2c7** was also coupled with the terminal alkyne **la** to affored alkenyl-substituted alkyne **3c** in 83% yield (entry 3). This cross-coupling was applied to alkynyl(pheny1)iodonium salt **2d7** to provide the alkynylsubstituted acetylene **3d** in 93% yield (entry 4).9 Treatment of hex- 1 -yne **lb** with diphenyliodonium salt **2a,** 2-thienyl- (pheny1)iodonium salt **2b** and alkynyliodonium salt **2d** provided the coupled products **3e-g,** respectively (entries 5-7). It is noteworthy that the coupling of prop-2-yn-1-01 **lc** with iodonium salts **2a** and **2b** afforded the substituted prop-2-yn-1-01s **3h** and **3i** without the formation of unsaturated aldehyde resulting from base-catalysed iosmerization¹⁰ (entries 8 and 9). For the prop-2-ynylic cyclic carbonates **Id,** diphenyliodonium and alkynyliodonium tetrafluoroborates **2a** and **2d** were smoothly coupled to afford the substituted prop-2-ynylic carbonates **3j** and **3k** without any deprotection (entries 10 and 1 1). Finally, treatment of trimethylsilylacetylene **le** with iodonium salts **2a-b** and **2d** afforded the substituted acetylenes **31-n** (entries 12–14).

In considering a plausible mechanism for the formation of the diyne **3d,** it is presumed that facile oxidative addition of the highly electrophilic alkynyliodonium salt 2d with Pd^o gives polar and reactive organopalladium tetrafluoroborate

Table 1 Palladium-catalysed cross-coupling of **iodonium salts with terminal alk ynesa**

Entry	Substrate	Iodonium salt	t/min	Product	Yield $(\%)^b$
1	1a	2а	10	3a	96(93)
2	1a	2 _b	20	3b	87(88)
3	1a	2c	15	3c	83(81)
4	1a	2d	10	3d	93(92)
5	1b	2a	20	3e	92(91)
6	1 _b	2 _b	25	3f	56(40)
7	1 _b	2d	20	3g	90(88)
8	1c	2a	5	3h	98(95)
9	1c	2 _b	10	3i	93(91)
10	1d	2a	5	3j	97(92)
11	1d	2d	5	3k	93(91)
12	1e	2a	10	31	95(93)
13	1e	2 _b	10	3m	93(90)
14	1e	2d	10	3n	92(90)

a **All the reactions were run with terminal alkynes (1 equiv.) and iodonium** salt (1 equiv.) in the presence of Pd(OAc)₂ (0.2 mol%) and NaHCO₃ (1 **equiv.)** in **MeCN-H20 (4** : **1) at room temperature.** *b* **The yields are isolated yields. The yields in parentheses are for reactions carried out in the absence of base.**

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(PhC \equiv CPd⁺ BF₄⁻). This intermediate is subjected to alkynylation with phenylacetylene **la** to form dialkynylpalladium(11) $[(PhC\equiv C)₂Pd]$ followed by reductive elimination to afford the coupled diyne **3d.** Alternatively, we have investigated palladium-catalysed cross-coupling of terminal alkynes with iodanes. The results are summarized in Scheme 2 and Table 2.

We coupled the terminal alkyne **la** with readily available **[hydroxy(tosyloxy)iodo]benzene** (HTIB), PhI(0H)OTs **4l** (also known as Koser's reagent), to afford the phenylsubstituted alkyne **3a** in 78% yield (entry **1).** The reaction can also be carried out in the absence of base (entry **1** in Table 2).

Scheme 2 Reagents and conditions: i, $Pd(OAc)_2$ (0.2 mol%), $NaHCO_3$ (1 equiv.), MeCN-H₂O (4:1), room temp., 5 or 10 min

Table **2** Palladium-cataly sed cross-coupling **of** iodanes with terminal alk ynes⁴

Entry	Substrate	Iodonium salt	t /min	Product	Yield $(\%)^b$	
	1a	4	25	3a	78(75)c	
2	1a	5	20	3a	80(76)	
3	1b	4	30	3e	76(70)	
4	1b	5	20	3e	80(74)	
5	1c	4	20	3h	91(89)	
6	1c	5	20	3h	95(91)	
	1d	4	30	3j	91(86)	
8	1d	5	20	3j	90(88)	
9	1e	4	5	31	92(90)	
10	1e	5	5	31	96(93)	

*^a*All the reactions were **run** with terminal alkynes (1 equiv.) and iodanes (1 equiv.) in the presence of Pd(OAc)₂ (0.2 mol%) and NaHCO₃ (1 equiv.) in MeCN-H₂O $(4:1)$ at room temperature. ^b The yields are isolated yields. The yields in parentheses are for reactions carried out in the absence of base. *^c*The yield can be improved to 93% by addition of CuI.

When the terminal alkyne **la** was subjected to react with ~-oxobis[(trifluoromethanesulfonato)(phenyl)iodine] *5* (also known as Zefirov's reagent¹²) under the same conditions, phenyl-substituted acetylene **3a** was afforded in 80% yield (entry 2). Even without using base, a comparable yield was obtained. Treatment of hex-1-yne **lb** with **4** or **5** furnished **3e,** respectively (entries **4** and 5). It is notable that the coupling of prop-2-yn-1-01 **lc** with Koser's or Zefirov's reagent afforded **3h** without formation of any aldehyde (entries *6* and 7). For the prop-2-ynylic cyclic carbonate **Id,** coupling with **4** or **5** afforded **3j** without any deprotection (entries 8 and **9).** Finally, reaction of trimethylsilylacetylene **le** with **4** or **5** afforded the substituted alkyne **31** (entries 10 and 11).

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References

- 1 H. A. Dick and F. R. Heck, *J. Organornet. Chem.,* 1975, 93, 259; L. Casar, J. *Organornet. Chem.,* 1975, 93, 253.
- 2 K. Sonogashira, Y. Tohda and N. Hagihara, *Tetrahedron Lett.,* 1975,16, 4467; K. Sonogashira, in *Comprehensive Organic Synthesis,* ed. B. M. Trost and **I.** Fleming, Pergamon, Oxford, 1991, vol. 3, p. 521; R. Rossi, A. Carpita and F. Bellina, *Org. Prep. Proced. Znt.,* 1995, **27,** 129.
- 3 M. Alami, F. Fem and *G.* Linstrumella, *Tetrahedron Lett.,* 1993, 34, 6403.
- 4 For the recent reviews for iodonium salts: P. J. Stang, *Angew. Chem., Int. Ed. Engl.,* 1992,31,274; R. M. Moriarty and R. K. Vaid, *Synthesis,* 1990,431; M. Ochiai, *Rev. Heteroat. Chem.,* 1989, **2,** 92.
- 5 S.-K. Kang, K.-Y. Jung, C.-H. Park and **S.-B.** Jang, *Tetrahedron Lett.,* 1995,36, 8047.
- 6 Pd-catalysed C-C bond formation: R. M. Moriarty, W. R. Epa and A. K. Awasthi, *J. Am. Chem. SOC.,* 1991,113,6315; R. M. Moriarty and W. R. Epa, *Tetrahedron Lett.,* 1992, 33,4095; R. J. Hinkle, G. T. Poulter and P. J. Stang, *J. Am. Chem. SOC.,* 1993, 115, 11626; K. Sugioka, M. Uchiyama, T. Suzuki and Y. Yamazaki, *Nippon Kagaku Kaishi,* 1985, 558; *Chem. Abstr.,* 1986, 104, 109 137.
- 7 M. Ochiai, K. Sumi, Y. Takaoka, Y. Nagao, M. Shiro and E. Fujita, *Tetrahedron,* 1988,44,4095.
- 8 A. J. Morgida and *G.* F. Koser, *J. Org. Chem.,* 1984, 49, 3643.
- 9 Diacetylene derivatives have been found to display liquid-crystalline and non-linear optical properties. See: A. **E.** Stiegman, E. Graham, K. J. Perry, L. R. Khundkar, **L.-T.** Cheng and J. W. Perry, *J. Am. Chem. SOC.,* 1991,113,7658; C. Fouquey, J.-M. Lehn and J. Malthete, J. *Chem. SOC., Chem. Commun.,* 1987, 1424.
- 10 K. Minn, *Synlett,* 1991, 115.
- 11 First synthesis: 0. Neiland and B. Karek, *J. Org. Chem. USSR (Engl. Transl.),* 1970, 6, 889; G. F. Koser and R. **H.** Wettach, *J. Org. Chem.,* 1977,42, 1476.
- 12 N. S. Zefirov, V. V. Zhdankin and A. **S.** Koz'min, *Zzv. Akad. Nauk SSR, Ser. Khim.,* 1983, 1682; V. V. Zefirov, Y. V. Dan'kov and A. **S.** Koz'min, *Zh. Org. Khim.,* 1984, **20,** 446; J. Gallos, A. Varvoglis and N. W. Alcock, *J. Chem. SOC., Perkin Trans. I,* 1985,757; R. T. Hembre, C. P. Scott and **J.** R. Norton, *J. Org. Chem.,* 1987, 52, 3650.

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